# Fifty-Second Report

OF THE

# CONNECTICUT AGRICULTURAL EXPERIMENT STATION

NEW HAVEN

FOR THE YEAR

1928

PRINTED IN COMPLIANCE WITH STATUTE

 $\begin{array}{c} \text{NEW HAVEN} \\ \\ \text{PUBLISHED BY THE STATE} \end{array}$ 

1929

Publication
Approved By
The Board of Control

# Press of The Wilson H. Lee Co., New Haven, Conn.

# CONNECTICUT AGRICULTURAL EXPERIMENT STATION

### BOARD OF CONTROL

As of

### October 31, 1928

His Excellency,	Governor John H. Trumbull, ex-officio President			
Charles R. Treat, Vice-President. Orange George A. Hopson, Secretary. Mt. Carmel Wm. L. Slate, Director and Treasurer. New Haven Joseph W. Alsop				
Elijah Rogers	Southington Middletown			
Francis F. Lincoln.				
Е	. H. JENKINS, Ph.D., Director Emeritus.			
Administration.	WM. L. SLATE, B.Sc., Director and Treasurer. MISS L. M. BRAUTLECHT, Bookkeeper and Librarian. MRS. R. A. HUNTER, Secretary. G. E. Graham, In charge of Buildings and Grounds.			
Chemistry: Analytical Laboratory.	E. M. BAILEY, PH.D., Chemist in Charge. C. E. SHEPARD OWEN L. NOLAN HARRY J. FISHER, A.B. W. T. MATHS DAVID C. WALDEN, B.S. FRANK C. SHELDON, Laboratory Assistant. V. L. CHURCHILL, Sampling Agent. Mrs. A. B. VOSBURGH, Secretary.			
Biochemical Laboratory.	MRS. A. B. VOSBURGH, Secretary.  T. B. OSBORNE, PH.D., Consulting Biochemist. H. B. VICKERY, PH.D., Biochemist in Charge. GEORGE W. PUCHER, PH.D., Assistant Biochemist. MISS HELEN C. CANNON, B.S., Dietitian.			
Botany.	G. P. CLINTON, Sc.D., Botanist in Charge. E. M. STODDARD, B.S., Pomologist. MISS FLORENCE A. MCCORMICK, PH.D., Pathologist. H. B. BENDER, M.S., Graduate Assistant. A. D. McDonnell, General Assistant. MRS. W. W. Kelsey, Secretary.			
Entomology.	W. E. BRITTON, Ph.D., Entomologist in Charge: State Entomologist. B. H. WALDEN, B.AGR. M. P. ZAPPE, B.S. PHILIP GARMAN, Ph.D. ROGER B. FRIEND, Ph.D., JOHN T. ASHWORTH, Deputy in Charge of Gipsy Moth Work. R. C. BOTSFORD, Deputy in Charge of Mosquito Elimination. J. P. JOHNSON, B.S., Deputy in Charge of Asiatic and Japanese Beelle Quarantines. MRS. GLADYS BROOKE, B.A., Secretary.			
Forestry.	WALTER O. FILLEY, Forester in Charge. H. W. HICOCK, M.F., Assistant Forester. J. E. RILEY, JR., M.F., In Charge of Blister Rust Control. MISS PAULINE A. MERCHANT, Secretary.			
Plant Breeding.	DONALD F. JONES, S.D., Geneticist in Charge. W. R. SINGLETON, S.M., Assistant Geneticist. H. R. Murray, B.S., Graduate Assistant. MISS EVELYN GREY, Secretary.			
Soil Research.	M. F. Morgan, M.S., Agronomist in Charge. H. G. M. JACOBSON, M.S., Assistant Agronomist. HERBERT A. LUNT, M.S., Assistant in Forest Soils. DWIGHT B. DOWNS, General Assistant.			
Tobacco Sub-station at Windsor.	PAUL J. ANDERSON, Ph.D., Pathologist in Charge. T. R. SWANBACK, M.S., Agronomist. MISS DOROTHY LENARD, Secretary.			

#### TABLE OF CONTENTS

	PAGE
Officers and Staff of the Station	iii
Table of Contents	iv
Letter of Transmittal, Board of Control	v
Report of the Treasurer	vi
Report of Insect Pest Appropriation	viii
Report of Mosquito Elimination Appropriation	ix
Report of Tobacco Research Appropriation	x
Report of White Pine Blister Rust Appropriation	xi
Report on Commercial Fertilizers, Bulletin 296	3
The Effect of Topping and Suckering on Havana Seed Tobacco,	
Bulletin 297	99
Report of the Director, Bulletin 298	115
Report of the Tobacco Substation, Bulletin 299	145
The Composition of Some Commercial Insecticides, Fungicides,	
Bactericides, Rodenticides, and Weed Killers, a Compilation,	
Bulletin 300	207
Control Studies on the Plum Curculio in Connecticut Apple Or-	
chards, Bulletin 301	373
The Willow Scab Fungus, Bulletin 302	443
Report on Commercial Feeding Stuffs, Bulletin 303	477
The Asiatic Beetle in Connecticut, Bulletin 304	585
Report of the State and Station Entomologist, Bulletin 305	669
Soil Reaction and Liming as Factors in Tobacco Production in	
Connecticut, Bulletin 306	773
Report on Food and Drug Products, Bulletin 307	813
Index	851
Bulletin of Immediate Information No. 62	I
Bulletin of Immediate Information No. 63	IX
Bulletin of Immediate Information No. 64	XIII
Bulletin of Immediate Information No. 65	xvII
Bulletin of Immediate Information No. 66	XXI
Bulletin of Immediate Information No. 67	XXVI

# Letter of Transmittal

Large Room of the State of the

To His Excellency, John H. Trumbull, Governor of Connecticut:

The Board of Control of the Connecticut Agricultural Experiment Station, as required by law, herewith submits its fifty-second annual report for the year ending October 31st, 1928.

No extended statement is presented in this letter. The Report of the Director (Bulletin 298, pages 113 to 140), which constitutes a part of this report, is a summary of the work accomplished during the year, lists the changes in staff, the additions to equipment and the publications.

Immediately following this letter will be found the financial statements of the several appropriations for which the station is responsible. These are for the fiscal year ending June 30th, 1928.

All of which is respectfully submitted,

George A. Hopson,

Secretary of the Board.

# Report of the Treasurer

July 1, 1927—June 30, 1928

W. L. SLATE, in account with THE CONNECTICUT AGRICULTURAL EXPERIMENT STATION for the fiscal year ended June 30, 1928.

#### RECEIPTS.

Balance on hand, July 1, 1927		
State Comptroller (General or Current Expense)  " " (General—Additions)  " (Food and Drug).  United States Appropriation (Hatch).  " " (Adams).  " " (Purnell).  " " (Clark-McNary).  Fertilizer Analysis Fees. Feed Analysis Fees. Lockwood Trust Fund.	\$53,000.00 827.75 7,500.00 7,500.00 7,500.00 20,000.00 959.48 13,000.00 7,000.00 11,000.00 \$128,287.23	
Miscellaneous Receipts: Balance on hand, July 1, 1927. Sales of gasoline	\$201.46 826.12	
LESS MISCELLANEOUS RECEIPTS DEPOSITED WITH STATE TREASURER	1,027.58 908.35 119.23	4
	\$128,406.46	

#### DISBURSEMENTS.

Salaries	\$76,441.66
Labor	19,986.22
Stationery and Office Supplies	784.79
Scientific Supplies (chemicals)	1.074.25
" (other laboratory supplies)	835.55
" (photographic supplies)	192.71
Feeding Stuffs	77.22
Insecticides, etc	74.97
Lumber and Small Hardware	41.76
Miscellaneous Supplies	747.60
(vi)	
하는 '마'는 사이들은 보고 있는 것이 없는 것이다.	

Automobile Oil.  Food and Drug Samples. Fertilizers.  Telegraph and Telephone.  Postage.  Travel Expense (outlying-investigations).  " (meetings, conferences, etc.)  " (gasoline for automobiles).	135.34 4.38 774.65 549.26 457.79 2,886.85 1,336.34 1,041.06 129.93	
Freight, Express and Parcels Post  Publications (bulletins, etc.)  (miscellaneous)	$126.70 \\ 170.50$	
Coal	2,330.00 1,849.34 204.55	
Furniture and Fixtures (new)	1,129.41 111.39	
Library (books and periodicals)	1,063.61 381.70 2,093.28	
Lave Stock	$140.45 \\ 110.00$	
Automobiles (new)	1,653.75 533.71 1,205.91	
" (repairs)  New Buildings and Structures  Buildings (repairs and alterations)	300.09 2,951.78 3,142.13	
Grounds	6.80 1,079.48 57.50	
Miscellaneous Contingent Expenses		8,214.41
Balance on hand, June 30, 1928:		
State General Appropriation (in hands of State Comptroller)	72.82	
Treasurer)	119.23	192.05
	\$12	8,406.46

#### REPORT OF

# W. L. SLATE, Treasurer

IN ACCOUNT WITH

# Insect Pest Appropriation

(Section 2109 of General Statutes, Revision of 1918)

#### July 1, 1927—June 30, 1928

#### RECEIPTS.

Balance on hand, July 1, 1927  Insect Pest Appropriation		
" " (odditional)	\$30,000.00	
(additions)	27.36	#00 00 <del>7</del> 00
Miscellaneous Receipts:	LI SEMENTE DE	\$30,027.36
Mileage for use of Automobiles \$27.36 LESS MISCELLANEOUS RECEIPTS DE-		
POSITED WITH STATE TREASURER 27.36		
TOSTIED WITH DIATE TREASURER 27.50		\$30,027.36
		=======================================
DISBURSEMENTS.		
SalariesDISBURSEMENTS.	\$15,450.00	
Labor	16,709.25	
Labor	125.35	
Scientific Supplies (chemicals)	42.93	
" (other laboratory supplies)	25.61	
(photographic supplies)	57.38	
Insecticides, etc	130.67	
Lumber and Small Hardware	.73	
Miscellaneous Supplies	841.64	
Automobile Oil	62.31	
Telegraph and Telephone	167.74	
Postage	91.22	
Travel Expense (outlying investigations)	2,219.31	
" (meetings, conferences, etc.)	259.53	
" (gasoline for automobiles)	507.06	
Freight, Express and Parcels Post	15.33	
Electricity	45.30	
Furniture and Fixtures (new)	180.65	
" (repairs)	19.15	
Library (books and periodicals)	201.05	
" (binding)	82.35	
Scientific Equipment (new)	21.48	
Livestock	1.35	
Automobiles (new)	544.00	
_ " (repairs)	256.90	
Tools, Machinery and Appliances (new) " (repairs)	48.63	
	7.10	
New Buildings and Structures	234.24	
Buildings (repairs and alterations)	10.58	
Rent of Land and Buildings	314.66	
Insurance (automobile)	169.47	
Insurance (automobile)	72.44	
Total Disbursements		\$38,915.41
Charged to allotment for second half of fiscal period		
(June 30, 1928)		8,888.05
		\$30,027.36

# REPORT OF

# W. L. SLATE, Director

IN ACCOUNT WITH

# Mosquito Elimination Appropriation

(Sections 2409 and 2410 of General Statutes, Revision of 1918; amended by Chapter 68, Public Acts of 1923)

July 1, 1927-June 30, 1928

#### RECEIPTS

		RECEIPTS.
	\$7,500.00 88.46	State Appropriation
\$7,588.46		Total
		DISBURSEMENTS.
8,365.86	\$2,500.00 4,668.39 31.40 32.11 3.51 2.00 4.45 109.65 225.64 34.20 8.50 229.00 166.27 59.97 \$8,078.29	Salary (R. C. Botsford) Labor Stationery and Office Supplies Photographic Supplies Miscellaneous Supplies Automobile Oi! Oil for Mosquito Spraying Telephone and Telegraph Travel (outlying investigations) " (gasoline) Tools, Machinery and Appliances (new) " " (repairs) Automobile Repairs and Accessories Buildings and Land (Repairs and Alterations) Insurance (automobile)  Transferred to New Mosquito Elimination Fund
777.40		Charged to allotment for second half of fiscal period June 30, 1928
	_	

\$7,588.46

### REPORT OF

# W. L. SLATE, Director

IN ACCOUNT WITH

# **Tobacco Research Appropriation**

(Public Acts, 1921, Chapter 184.)

July 1, 1927—June 30, 1928

radioographa nedamental musikar

#### RECEIPTS

Contributions 1 Sales of Tobacco 1  DISBURSEMENTS.	5,000.0 <b>0</b> ,814.24 ,860.49	
		4.73
To all Old On the Hand on more share authority of her		
	7,400.00 5,863.93 113.71 111.82 93.06 20.91 25.01 36.93 31.81 358.84 28.08 512.36 75.07 317.77 143.91 89.07 51.23 531.35 158.69 54.96 272.77 15.40 18.92 562.50 2.61 135.41 676.73 19.84 477.37 75.00 62.78	37.84

Charged to allotment for second half of fiscal period June 30, 1928.....

663.11

\$18,674.73

### REPORT OF

# W. L. SLATE, Director

IN ACCOUNT WITH

# White Pine Blister Rust Appropriation

(Section 2117 of General Statutes, Revision of 1918)

July 1, 1927-June 30, 1928

#### RECEIPTS.

State Appropriation	\$7,500.00	
Refund for Experimental Work 204.83 Refund for Camp Meals 7.83	1 591 54	
	1,531.54	\$9,031.54
Expenditures.		
By the State Comptroller on vouchers submitted by W. L. Slate, Director: Salaries. Labor.	\$1,646.46 7,365.28	
Stationary	21 00	

Datatics	\$1,040.40
Labor	7,365.28
Stationery	31.88
Lumber and Small Hardware	15.32
Miscellaneous Supplies	12.72
relegraph and Telephone	31.05
Travel (outlying investigations)	1,271.08
(meetings, conferences, etc.)	64.55
Cartage	43.80
ublications	5.50
rurniture and Fixtures	45.00
Automobiles (new)	657.70
100Is, Machinery and Appliances (new)	41.16
Automobiles (repairs)	45.35
illsurance (automobile)	243.42
Miscellaneous Contingent Expenses	60.19
그 경우 그 그 그 사람들이 있었다면 하는 것이 없는 것이 없는 것 같아 없는 것이 없는 것이 없다.	

Total	Disbursements
Charged to allo June 30, 1	otment for second half of fiscal period

2,548.92

\$9,031.54

\$11,580.46

# Connecticut Agricultural Experiment Station

Nem Haven, Connecticut

# Report on Inspection

of

# Commercial Fertilizers for 1928

E. M. BAILEY, Chemist in Charge of the Analytical Laboratory.

CONTENTS.	
	Page 3
The Fertilizer Law	lane.
Inspection of 1928.	17
Raw Materials Containing Nitrogen	
Raw Materials Containing Phosphotic Acid	~ ~ ~
Raw Materials Containing Nitrogen and Potash	45
Raw Materials Containing Nitrogen and Phosphoric Acid	46
Mixed Fertilizers:	
Containing Nitrogen and Phosphoric Acid	
Containing Phosphoric Acid and Potash	
Special Mixtures and Home Mixtures	
Migaellanana Bartilizara Amandmanta ata	
Miscellaneous Fertilizers, Amendments, etc.: Sheep Manure, etc	84
Lime, etc	. 84
Other miscellaneous	
Collaborative Work	04
Index	. I

The Bulletins of this Station are mailed free to citizens of Connecticut who apply for them, and to other applicants as far as the editions permit.

# CONNECTICUT AGRICULTURAL EXPERIMENT STATION

#### OFFICERS AND STAFF

#### BOARD OF CONTROL

	DOARD OF CONTROL
George A. Hops	ncy, Governor John H. Trumbull, ex-officio President t, Vice-President
Joseph W. Alsop Elijah Rogers Edward C. Sch	Director and Treasurer Mt. Carmel New Haven Avon neider Southington oln Cheshire
	STAFF.
	E. H. JENKINS, Ph.D., Director Emeritus.
Administration.	WM. L. SLATE, B.SC., Director and Treasurer. MISS L. M. BRAUTLECHT, Bookreeper and Librarian. MRS. R. A. HUNTER, Secretary, G. E. GRAHAM, In charge of Buildings and Grounds.
Chemistry: Analytical Laboratory.	E. M. BAILEY, PH.D., Chemist in Charge. C. E. SHEPARD OWEN L. NOLAN HARRY J. FISHER, A.B. W. T. MATHIS DAVID C. WALDEN, B.S. FRANK C. SHELDON, Laboratory Assistant. V. L. CHURCHILL, Sampling Agent. MRS. A. B. VOSBURGH, Secretary.
Biochemical Laboratory.	T. B. OSBORNE, PH.D., Consulting Biochemist. H. B. VICKERY, PH.D., Biochemist in Charge. GEORGE W. PUCHER, PH.D., Assistant Biochemist. MISS HELEN C. CANNON, B.S., Dietitian.
Botany.	G. P. CLINTON, Sc.D., Botanist in Charge. E. M. STODDARD, B.S., Pomologist. MISS FLORENCE A. McCormick, Ph.D., Pathologist. H. B. BENDER, M.S., Graduate Assistant. A. D. McDonnell, General Assistant. MRS. W. W. Kelsey, Secretary.
Entomology.	W. E. BRITTON, PHD., Entomologist in Charge: State Entomologist. B. H. WALDEN, B.AGR. M. P. ZAPPE, B.S. PHILIP GARMAN, PH.D. ROGER B. FRIEND, PH.D. JOHN T. ASHWORTH, Deputy in Charge of Gipsy Moth Work. R. C. BOTSFORD, Deputy in Charge of Mosquito Elimination. J. P. JOHNSON, B.S., Deputy in Charge of Asiatic and Japanese Beetle Quarantines. MRS. GLADYS BROOKE, B.A., Secretary.
Forestry.	WALTER O. FILLEY, Forester in Charge. H. W. HICOCK, M.F., Assistant Forester. J. E. RILEY, JR., M.F., In Charge of Blister Rust Control. Miss Pauline A. Merchant, Secretary.

Donald F. Jones, S.D., Geneticist in Charge. W. R. SINGLETON, S.M., Assistant Geneticist. H. R. MURRAY, B.S., Graduet Assistant. Miss Evelyn Grey, Secretary.

M. F. Morgan, M.S., Agronomist in Charge. H. G. M. Jacobson, M.S., Assistant Agronomist. Herbert A. Lunt, M.S., Assistant in Forest Soils. Dwight B. Downs, General Assistant.

Tobacco Sub-station PAUL J. ANDERSON, PH.D., Pathologist in Charge. T. R. SWANBACK, B.S., Agronomist. at Windsor,

# Report on Inspection Commercial Fertilizers 1928

E. M. BAILEY,

Chemist in Charge, Analytical Laboratory.

#### THE FERTILIZER LAW.

The provisions of the Connecticut fertilizer law have been discussed in previous reports but for more ready reference its essential features may be noted here.

SIGNIFICANCE OF THE TERM "COMMERCIAL FERTILIZERS."

Explaining what is meant by the term "commercial fertilizers" the law says:

"The term 'commercial fertilizers' shall be construed to mean any and every substance imported, manufactured, prepared or sold for fertilizing or manuring or soil amendment purposes, except barnyard manure and stable manure which have not been artificially treated or manipulated, marl and lime. Cottonseed meal, rapeseed meal, castor pomace and all other vegetable products used as fertilizers, including the ashes of cotton hulls and wood ashes, shall be included as fertilizers within the meaning of this act and separate analysis fees shall be paid on each different grade which is sold or offered for sale in the state. The person responsible for paying the fees above prescribed may deduct from the total tonnage sold such sales of cottonseed meal or other vegetable products as are made to anyone who gives a written certificate on a form supplied by the Connecticut Agricultural Experiment Station stating that the material bought by him was to be used exclusively for feed and not for fertilizer."

#### CONCERNING COTTONSEED MEAL.

Cottonseed meal is a fertilizer within the meaning of the Statute but it is provided that when this product is sold for feeding purposes only, it shall be exempt from the tonnage tax. When sold as a feed, cottonseed meal is subject to registration under the terms of the feed law. By regulation, however, if it is sold exclusively as a fertilizer, or exclusively as a feed it may be registered but once, and under that law which applies.

The status of cottonseed meal under the fertilizer law has been clearly stated in bulletin1 from this Station from which the following may be quoted:

Registration and analysis fees. "Each brand of cottonseed meal must be registered on forms provided by this Station and an analysis fee of ten dollars paid on it before it is sold, offered or exposed for sale, and on the first day of January annually thereafter."

<sup>1</sup>Bull. of Information No. 9, 1919.

Plant Breeding.

Soil Research.

"A distinctive name constitutes a distinct brand. If shipments have different guaranties of composition they are held to be different brands."

Branding or tagging. "Since nitrogen is the only fertilizer ingredient considered in the trade in cottonseed meal no guaranty of phosphoric acid or potash is required. If either is guaranteed by the manufacturer, however, an additional fee of ten dollars must be paid on each element. The statement of composition now legal for feeds may be used hereafter if the percentage of nitrogen is stated.

"Note that the law regarding feeding stuffs forbids the use of metal in attaching tags and requires that each package shall be branded or tagged with the statement required by law."

Duties of shippers. "It is assumed from correspondence with shippers outside the state that they will register the brands which they sell in Connecticut, will pay analysis fees as has been done in the past by manufacturers of commercial fertilizers, and will semi-annually thereafter pay the tonnage fees.

"They will report to this Station their total sales and if they wish, may report what part has been sold for feed exclusively. From the reports of dealers within the state it will be possible to determine quite closely the

amounts of each brand actually used as feed.

"In the case the jobber outside the state neglects or refuses to register a brand, the dealer who sells it within the state is responsible under the law."

Duties of dealers. "Dealers are required to file with the director of the Station on July first of each year and semi-annually thereafter a sworn statement of their total sales of each brand of cottonseed meal and the amount of each sold exclusively for feed, during the preceding six months."

### REQUIREMENTS TO BE COMPLIED WITH BY SELLERS OF COMMER-CIAL FERTILIZERS.

The seller is responsible for the proper labeling of each package, for the registration at the Station of every brand sold by him an for the payment of the analysis fee, before offering for sale, and annually thereafter on January 1st.

The law specifies the information which shall be given on the

label as follows:

- 1. Weight of each package in pounds.
- 2. Brand name or trade mark.

3. Analysis:

- Available phosphoric acid, per cent.
- Total phosphoric acid, per cent.

Nitrogen, per cent.

- Equivalent ammonia, per cent. Potash soluble in water, per cent.
- 4. Name and address of the manufacturer or of the person who is responsible for the statement of the guaranty.

In the case of bone meal, tankage or other organic products, and in basic slag and mineral phosphates in which a large percentage of the phosphoric acid is not available by laboratory methods, the

phosphoric acid shall be claimed as total phosphoric acid unless it is desired to claim available phosphoric acid instead, in which case the guaranty shall take the form set forth above.

The label may be a tag attached to the package or a statement printed thereon. Percentages shall be minimum percentages only.

The presence of leather in its various forms, wool waste, hair, or any inert nitrogenous material shall be declared on the label imless, by processing, the activity of these materials has been rendered satisfactory as determined by official methods.

When potash is derived from sulphate or carbonate of potash

it may be so claimed.

No claim or guaranty for less than 0.82 per cent of nitrogen or for less than 1 per cent of phosphoric acid, or for less than 1 per cent of potash shall be regarded in the registration or analysis of any commercial fertilizer.

The seller must also, on the 1st of January and July, report the tonnage of fertilizer sold within the preceding six months and pay to the director of the Station a tonnage fee of 6 cents per ton. On request, copies of the law and blanks for registration and

for tonnage reports will be supplied by the Station.

If, however, proper labeling, registration and payments have been provided for by the manufacturer of the brands or by another responsible person all sellers of such brands are released from the above mentioned requirements. The retailer, therefore, should assure himself that the requirements of the law have been met by the manufacturers of the brands which he handles, or himself be prepared to meet all these requirements.

#### PRECAUTIONS TO BE OBSERVED IN DRAWING SAMPLES FOR ANALYSIS.

The analysis of a fertilizer is of no value unless the sample analyzed represents as nearly as possible the stock from which the sample was drawn. The law prescribes the procedure to be followed by authorized agents of this Station when taking official samples for analysis as follows:

"When samples are taken from fertilizers in bags, a tube shall be used, and it shall be inserted at one end of the bag and shall pass substantially the entire length of the bag, so as to take a core of the material being sampled from substantially the entire length of the bag. Samples thus taken from individual bags shall be thoroughly mixed, and the official samples shall be taken from the mixture so drawn by the method known as 'quartering.' Samples of fertilizers taken as herein provided shall be taken from at least five per centum of the separate original unopened packages in the lot, for the mixture from which the official samples shall be taken. If less than one hundred bags are in the lot, at least five bags shall be sampled; if less than five bags, all shall be sampled. Broken packages shall not be sampled."

#### GRATUITOUS ANALYSES.

CONNECTICUT EXPERIMENT STATION

Under the fertilizer law the Station is charged only with the analysis of samples drawn by its own agents. It does, however, each year analyze a considerable number of samples drawn by individuals, representing stock purchased by them for their own use. The object of the purchaser is to satisfy himself as to whether he has obtained goods of the grade represented and, perhaps, to obtain evidence upon which to base a claim for shortage should the materials not meet their guaranties. The Station assumes no responsibility for the sampling in case of such unofficial samples and can only vouch for the accuracy of the results obtained on the materials as submitted. Since a representative sample is as essential as an accurate analysis in judging the quality of a shipment of fertilizer, it is evident that a satisfactory adjustment will seldom be effected on the basis of an unofficial sample. Notwithstanding certain objections which may be raised to the practice of analyzing samples submitted by individuals, the Station is disposed to continue such work so long as there is evidence that it constitutes a useful service.

# REGISTRATIONS REGISTRATIONS FOR 1928.

For 1928, 61 firms and individuals registered at this Station for sale in this State 433 brands of fertilizers. As required by Statute, the brands so registered are listed as follows:

# American Agricultural Chemical Company, New Haven Sales Dept., New Haven, Conn.

A.A.C. Acme Fertilizer

A.A.C. Aroostook Potato Manure

A.A.C. Castor Pomace

A.A.C. Complete General Fertilizer A.A.C. Double Manure Salts

A.A.C. Double A Tobacco Fertilizer

A.A.C. Dry Ground Fish

A.A.C. Gladiator Fertilizer
A.A.C. Grass and Lawn Top Dressing
A.A.C. Ground Tankage
A.A.C. Hi-Grade Tobacco Manure

A.A.C. Monarch Fertilizer A.A.C. Muriate of Potash

A.A.C. Nitrate of Soda A.A.C. Old Hickory Fertilizer

A.A.C. Princess Fertilizer

A.A.C. Pulverized Sheep and Goat Manure

A.A.C. Special Ground Bone A.A.C. Sulphate of Ammonia

A.A.C. Sulphate of Potash

A.A.C. 16% Superphosphate (Acid Phosphate) A.A.C. Tobacco Ash Element Agrico Fertilizer for Corn

Agrico Fertilizer for Potatoes

Agrico Fertilizer for Truck Bowker's All Round Fertilizer

Bowker's Lawn and Garden Dressing

Bowker's Market Garden Fertilizer

Bowker's Potato and Vegetable Phosphate Bowker's Stockbridge Early Crop Manure Bowker's Stockbridge Hill and Drill Fertilizer Bowker's Stockbridge Tobacco Manure

Bowker's Sure Crop Fertilizer Bradley's Blood, Bone and Potash

Bradley's Complete Manure for Potatoes and Vegetables

Bradley's Complete Tobacco Manure

Bradley's Eclipse Fertilizer

Bradley's Northland Potato Grower Bradley's Potato Fertilizer Bradley's Potato Manure Bradley's XL Superphosphate of Lime

Cottonseed Meal

National Aroostook Special Fertilizer National Complete Tobacco Fertilizer

National Market Garden Fertilizer

National Pine Tree State Potato Fertilizer National Premier Potato Manure

Sanderson's Atlantic Coast Mixture

Sanderson's Complete Tobacco Grower

Sanderson's Corn Superphosphate Sanderson's Formula A

Sanderson's Formula B Sanderson's Potato Manure

#### American Cyanamid Company, 535 Fifth Ave., New York City. Ammo-Phos

#### Apothecaries Hall Company, Waterbury, Conn.

Acid Phosphate Basic Slag Phosphate Bone Meal 3-22 Bone Meal 4-20 Bone and Meat Tankage Carbonate of Potash Castor Pomace Cotton Seed Meal Dry Ground Fish Liberty Corn and All Crops, 2-8-2 Liberty Corn, Fruit and All Crops, 2-12-4 Liberty Fish, Bone and Potash, 3-8-3 Liberty High Grade Market Gardeners, 5-8-7 Liberty High Grade Tobacco Manure, 7-3-7 Liberty Home Vegetable Garden Fertilizer Liberty Lawn Fertilizer Liberty Onion Special (Potash as Sulphate), 4-8-7 Liberty Potato and General Crops, 4-8-10 Liberty Potato and Market Gardener's Special, 4-8-4 Liberty Potato and Vegetable, 2-8-10 Liberty Special Fertilizer for Fruit, 7-8-6 Liberty Tobacco Special, 5-3-5 Liberty Tobacco Starter Liberty Top Dresser for Grass and Grain, 10-31/2-8 Muriate of Potash Nitrate Soda Nitrate Soda and Potash Precipitated Bone Sheep Manure Sulphate Ammonia Sulphate Potash Sulphate Potash-Magnesia Tankage

#### Armour Fertilizer Works, 50 Broad St., New York City.

Armour's Big Crop Bone Meal 3-48 Armour's Big Crop Fertilizer 2-12-4 Armour's Big Crop Fertilizer 3-8-4 Armour's Big Crop Fertilizer 4-8-4 Armour's Big Crop Fertilizer 4-8-7 Armour's Big Crop Fertilizer 4-8-7
Armour's Big Crop Fertilizer 4-16-10
Armour's Big Crop Fertilizer 4-16-4
Armour's Big Crop Fertilizer 5-8-7
Armour's Big Crop Fertilizer 7-11-10
Armour's Big Crop Fertilizer 7-12-7
Armour's Big Crop Fertilizer 8-6-6 Armour's Big Crop Fertilizer 8-6-6 Armour's Big Crop Super Phosphate (Acid Phosphate) 16% Armour's Big Crop Super Phosphate 20% Armour's Big Crop Tobacco Fertilizer 7-3-7 Armour's Big Crop Tobacco Special 5-3-5 Armour's Castor Pomace 5½% Armour's Cotton Seed Meal 8% Armour's Lawn and Garden Grower 6-8-6 Armour's Ground Tankage 9% Armour's Muriate of Potash 48%

Armour's Nitrate of Soda 18% Armour's Sheep and Goat Manure 11/2-1-2 Armour's Sulphate of Ammonia 25% Armour's Sulphate of Potash 48%

# Ashcraft-Wilkinson Co., Trust Co. of Georgia Bldg., Atlanta, Ga.

"Helmet" "Monarch"

BULLETIN 296

#### Atlantic Packing Company, New Haven, Conn.

Atlantic 5-4-16 Atlantic 5-8-7 Atlantic Grain Fertilizer 2-10-2 Atlantic Special Vegetable 4-8-4

## The Baker Castor Oil Company, 120 Broadway, New York City.

Castor Pomace

#### Barrett Company, 40 Rector St., New York City.

Arcadian Nitrate of Soda Arcadian Sulphate of Ammonia Sulphate of Ammonia

#### F. A. Bartlett Tree Expert Co., Stamford, Conn.

Bartlett's Green Tree Food

#### The Berkshire Chemical Company, Bridgeport, Conn.

Berkshire Castor Pomaçe Berkshire Complete Tobacco Berkshire Dry Ground Fish Berkshire Economical Grass Berkshire Fine Ground Bone Berkshire Grass Special Berkshire Ground Tankage Berkshire Long Island Special Berkshire Market Garden Fertilizer Berkshire Potato and Vegetable Fertilizer Berkshire Sheep Manure Berkshire Super Phosphate (Acid Phosphate) Berkshire Tobacco Special Berkshire Tobacco Starter Berkshire Truck Fertilizer Carbonate of Potash High Grade Sulphate Potash Muriate Potash Nitrate of Soda Precipitated Bone Phosphate

### Amos D. Bridge's Sons, Incorporated, Hazardville, Conn.

Corn. Onion. Potato and General Purpose Special Tobacco Fertilizer

# F. W. Brode Corp., 119 Madison Ave., Memphis, Tenn.

"Owl Brand," 36 % Cotton Seed Meal

# A. H. Case & Company, Inc., 965 William St., Buffalo, N. Y.

Par Plus Brand Pulverized Sheep Manure

11

CONNECTICUT EXPERIMENT STATION

Chittenden's Basic Slag Chittenden's Castor Pomace Chittenden's Complete Grain Chittenden's Dry Ground Fish Chittenden's Fine Ground Bone Chittenden's High Grade Potato Chittenden's High Grade Potato Chittenden's Potato Special 4% Potash Chittenden's Special Top Dresser Chittenden's Tobacco Special Chittenden's Valley Wrapper Brand Chittenden's Vegetable and Onion Grower Special Mixture

#### Everett B. Clark Seed Company, Milford, Conn., Succeeded by Associated Seed Growers, Inc., New Haven, Conn.

Nitrate of Soda 16% Acid Phosphate Special Mixture for General Use 4-8-4 Super Phosphate Special Mixture with 6% Potash 4-8-6 Tip Top Brand

# The Conn. Fat Rendering & Fertilizer Corp.. West Haven. Conn.

Tankage

### Consolidated Rendering Company, 40 North Market St., Boston, Mass.

Castor Pomace Corenco Sheep Manure Dry Ground Fish Ground Bone Muriate of Potash Nitrate of Soda Superphosphate (Acid Phosphate) Sulphate of Ammonia Sulphate of Potash Tankage 6-30 Tankage 9-20

#### C. & R. Sales Co., Worcester, Mass.

C. & R. Lawn Shrub Fertilizer

### The Davey Tree Expert Company, Kent, Ohio.

Davey Shredded Cattle Manure Davey Tree Food

Eastern States Nitrate of Soda

Eastern States 0-16-8 Open Formula

# Eastern States Farmers' Exchange, 38 Lyman St., Springfield, Mass.

Buckland's Formula A 6-2-7 Special Tobacco Mixture Eastern States Acid Phosphate Eastern States Calcium Nitrate Eastern States Dried Ground Fish Eastern States Fine Bone Meal Eastern States Ground Animal Tankage Eastern States Muriate of Potash Eastern States Nitrate of Potash

Eastern States 3-12-3 Open Formula Eastern States 4-8-10 Open Formula Eastern States 4-8-10 Open Formula
Eastern States 4-16-4 Open Formula
Eastern States 5-8-7 Open Formula
Eastern States 5-10-5 Open Formula
Eastern States 8-16-6 Open Formula
Eastern States 8-16-8 Open Formula
Eastern States 8-16-20 Open Formula
Eastern States 8-16-14 Open Formula
Eastern States 10-16-14 (Potash from Sulphate) Open Formula
Eastern States Open Formula 9-3-7 Tobacco Fertilizer
Eastern States Open Formula 10-3-8 Tobacco Fertilizer
Eastern States Precipitated Bone
Eastern States Sulphate of Ammonia Eastern States Sulphate of Ammonia Eastern States Sulphate of Potash

Eastern States Urea

#### Edward Eggert, 208 State St., Hartford, Conn.

Diamond E. E. Cotton Seed Hull Ashes

### Essex Fertilizer Company, 39 North Market St., Boston, Mass.

Essex A1 Super 2-10-2 Essex Complete Manure 5-8-7
Essex Fish Fertilizer For All Crops 3-8-4 Essex Market Garden 4-8-4 Essex Peerless Potato Manure 4-6-10 Essex Tobacco Manure 5-3-5 Essex Top Dressing 7-6-5

#### Ford Motor Company, Detroit, Michigan.

Ford Ammonium Sulphate

### Four Seasons Fertilizer Co., Inc., 135 West 29th St., New York City.

Four Seasons Fertilizer

#### The L. T. Frisbie Company, New Haven, Conn.

Frisbie's 5-8-7 Frisbie's Corn and Grain Fertilizer 2-10-2 Frisbie's Fine Bone Meal 4-22 Frisbie's Market Garden 4-8-7 Frisbie's Special 3-8-4 Frisbie's Tobacco Grower 5-3-5 Frisbie's Top Dresser Frisbie's Special Vegetable and Potato Grower 4-8-4

### Gash-Stull Company, Chester, Pa.

Young's Formula (8-7-6)

### The Grasselli Chemical Company, Cleveland, Ohio.

Grasselli Odorless Plant Food

### Humphreys-Godwin Company, Memphis, Tenn.

Bull Brand Cottonseed Meal Danish Brand Cottonseed Feed Dixie Brand Cottonseed Meal

### International Agricultural Corporation, 38 Chauncy St., Boston, Mass.

Cottonseed Hull Ashes 41% Cottonseed Meal 43% I. A. C. Basic Slag Nitrate of Lime Tobacco Special 7-6-5 Vuelta Abajo 7-9-8

#### John Joynt, Lucknow, Ontario, Canada.

Canada Wood Ashes

#### Spencer Kellogg & Sons, Inc., Buffalo, N. Y.

Castor Pomace

#### Lowell Fertilizer Company, 40 North Market St., Boston, Mass.

Lowell 5-10-5
Lowell Animal Brand A High Grade Manure For All Crops 3-8-4
Lowell Bone Fertilizer 2-10-2
Lowell Corn and Vegetable 4-8-4
Lowell Market Garden Manure 5-8-7
Lowell Potato Grower 4-6-10
Lowell Tobacco Manure 5-3-5
Lowell Top Dressing 7-6-5

#### L. B. Lovitt & Company, Memphis, Tenn.

"Lovit Brand" 43% Cottonseed Meal.

#### The Mapes Formula & Peruvian Guano Co., 270 Madison Ave., New York City.

The Mapes Connecticut Valley Special The Mapes Corn Manure The Mapes General Tobacco Manure The Mapes General Truck Manure The Mapes General Use Manure The Mapes Onion Manure The Mapes Potato Manure The Mapes Special Trucker
The Mapes Special Trucker "S. P." The Mapes Tobacco Ash Constituents The Mapes Tobacco Ash and Starter The Mapes Tobacco Manure, Wrapper Brand The Mapes Tobacco Starter Improved The Mapes Top Dresser Nitrate of Soda Pure Fine Ground Bone Sulphate of Potash

#### A. G. Markham & Company, Springfield, Mass.

4-8-4 4-6-10 5-8-7

#### Meech & Stoddard, Inc., Middletown, Conn.

Sheep Manure

# Natural Guano Company, Aurora, Illinois.

"Sheep's Head" Pulverized Sheep Manure

# New England Fertilizer Company, 40A North Market St., Boston, Mass.

REGISTRATIONS

New England Complete Manure 4-6-10 New England Corn Phosphate 2-10-2

New England Market Garden Manure 5-8-7

New England Potato and Vegetable Manure 4-8-4

New England Super, A High Grade Fertilizer For All Crops 3-8-4

New England Tobacco 7-3-7

New England Tobacco Manure 5-3-5

#### Olds & Whipple, Inc., Hartford, Conn.

Favorite Brand Sheep Manure High Grade Carbonate of Potash 96/98

High Grade Sulphate of Potash

O. & W. Acid Phosphate

O. & W. Blue Label Tobacco Fertilizer

O. & W. Castor Pomace

O. & W. Complete Market Garden

O. & W. Complete Tobacco Fertilizer

O. & W. Dry Ground Fish

O. & W. Grain and General Crop Fertilizer

O. & W. Grass Fertilizer

O. & W. High Grade Starter and Potash

O. & W. High Grade Tobacco Starter

O. & W. High Grade Vegetable & Potato

O. & W. Nitrate of Potash 90%

O. & W. Nitrate of Soda

O. & W. Precipitated Bone Meal

O. & W. Pure Bone Meal

O. & W. Sulphate of Ammonia

# Pacific Manure & Fertilizer Co., 429 Davis St., San Francisco, California. Groz-It Brand (Pulverized Sheep Manure)

# Parmenter & Polsey Fertilizer Company, 41 North Market St., Boston, Mass.

"P & P" "AA" Brand 5-8-7
"P & P" Maine Potato Fertilizer 4-6-10

Parmenter & Polsey Top Dressing 7-6-5

### Piedmont Mt. Airy Guano Co., Baltimore, Md.

Harvest Brand 2-8-2 Harvest Brand 3-8-4 Harvest Brand 4-6-10 Harvest Brand 4-8-4 Harvest Brand 5-8-7 Nitrate Soda

### Frank S. Platt Co., 450 State St., New Haven, Conn.

Platt's Concentrated Lawn Fertilizer Platco Special 5-8-7

# Premier Poultry Manure Company, 431 So. Dearborn St., Chicago, Illinois.

Premier Brand Poultry Manure Premier Brand Sheep Manure

15

#### The Pulverized Manure Co., 828 Exchange Ave., Union Stock Yard, Chicago, Illinois.

Wizard Brand Cattle Manure Wizard Brand Pulverized Sheep Manure

#### Rackliffe Bros. Co., Inc., New Britain, Conn.

Acid Phosphate 16% Nitrate of Soda Rackliffe Brand Corn Fertilizer 4-8-4 Rackliffe Brand Potato & Special Vegetable 5-8-7

CONNECTICUT EXPERIMENT STATION

#### The Rogers & Hubbard Company, Portland, Conn.

4-8-4 5-8-7 Acid Phosphate Garden Fertilizer Hubbard's "Bone Base" Fertilizer for Seeding Down Hubbard's "Bone Base" Oats and Top Dressing Hubbard's Pure Raw Knuckle Bone Flour Hubbard's "Bone Base" Soluble Corn and General Crops Manure Hubbard's "Bone Base" Soluble Potato Manure Hubbard's "Bone Base" Soluble Tobacco Manure Hubbard's Strictly Pure Fine Bone Muriate of Potash Nitrate of Soda Rogers & Hubbard's All Soils-All Crops Fertilizer Rogers & Hubbard's Climax Tobacco Brand Rogers & Hubbard's Corn and Grain Fertilizer Rogers & Hubbard's High Potash Fertilizer Rogers & Hubbard's Potato Fertilizer Rogers & Hubbard's Tobacco Grower-Vegetable Formula Sheep Manure

#### F. S. Royster Guano Company, 602 Citizens National Bank Bldg., Baltimore. Md.

Cotton Seed Meal Royster's Comet Guano Royster's Connecticut Tobacco Guano Royster's Fine Ground Bone Meal Royster's 5% Truck Guano Royster's Gem Guano Royster's Nitrate of Soda Royster's Quality Trucker Royster's Sheep and Goat Manure Royster's 16% Acid Phosphate Royster's Top Dresser Royster's Trucker's Delight Royster's Valley Tobacco Guano Royster's Wrapper Brand

#### M. L. Shoemaker and Co., Inc., Delaware Ave. and Venango St., Philadelphia, Pa.

Bone Meal Potato Special "Swift-Sure" Tobacco Starter 4-10-0
"Swift-Sure" 4-8-5 Special Tobacco Formula Tobacco and General Use

# Springfield Rendering Company, Springfield, Mass.

Springfield 3-8-4 Springfield 4-8-4 Springfield 4-8-7 Springfield 5-8-7 Springfield Tobacco Special 5-3-5

# swift & Company Fertilizer Works, Baltimore, Maryland.

# synthetic Nitrogen Products Corp., 285 Madison Ave., New York City.

Calcium Nitrate Basf (Nitrate of Lime) Calurea Nitrophoska

Urea Basf (Floranid)

# Tennessee Copper & Chemical Corp., Lockland, Cincinnati, Ohio.

Loma

#### I. P. Thomas & Son Company, 1000 Drexel Building, Philadelphia. Pa.

Castor Pomace Dairymen's Special 0-10-10 Economy Fertilizer 3-12-3 High Grade Potato Manure 4-8-10 I. P. Thomas 5-8-7 Long Island Special 4-8-7 Muriate of Potash Nitrate of Soda Pure Ground Bone 7% Guano 7-6-5 Sheep & Goat Manure 16% Acid Phosphate Tankage—sold on analysis
Thomas Tobacco Grower 5-4-5 Tip Top Superphosphate 3-10-6 Truckers High Grade Guano 4-8-4 Victor Potash Fertilizer 2-8-5

### Triton Oil and Fertilizer Company, 101 Beekman St., New York City.

Triton 4-8-4 Fertilizer Triton 4-8-7 Fertilizer Triton 5-8-5 Fertilizer Triton 5-8-7 Fertilizer

#### United States Guano Co., care Standard Wholesale Phosphate & Acid Works, Baltimore, Md.

Standard United States Accomac Peninsula King Standard United States Bone Meal 4-40 Standard United States 3 x 50 Bone Meal Standard United States 4½ x 45 Bone Meal Raw Standard United States Evergreen Fish Guano Standard United States Fish Bone and Potash Standard United States General Use Guano Standard United States Grain Grower Standard United States High Grade Phosphate and Potash

Standard	United	States	Jersey Special Mammoth Potato Growe
Standard	United	States	Muriate of Potash
			Nitrate of Soda
			Old Fertility
Standard	United	States	Royal Potato Grower
Standard	United	States	16% Acid Phosphate
Standard	United	States	Special Potato Grower
			Star Brand
Standard	United	States	Sulphate of Ammonia
			Sure Growth
Standard	United	States	6 x 30 Tankage
Standard	United	States	9 x 20 Tankage
			10% Fish
			Truckers Fish Guano
			5 x 10 x 5
Standard	United	States	8 x 6 x 6

#### Virginia-Carolina Chemical Corp., 120 Broadway, New York City.

Bloomaid V-C Aroostook Potato Grower V-C Fish and Potash Compound V-C Phospho-Tobacco V-C 16% Acid Phosphate V-C 20% Acid Phosphate V-C XXXX Fish and Potash

#### Wessel, Duval & Co., 1 Broadway, New York City.

Nitrate of Soda

#### The Wilcox Fertilizer Company, Mystic, Conn.

Acid Phosphate
Ground Steamed Bone
Muriate of Potash
Nitrate of Soda
Wilcox Corn Special 3-10-4
Wilcox Drv Ground Fish
Wilcox H. G. Fish and Potash 4-8-4
Wilcox Potato and Vegetable Phosphate 5-8-7
Wilcox 7-6-5 Top Dresser

#### S. D. Woodruff & Sons, Orange, Conn.

Woodruff's Home Mixed Fertilizer

#### Worcester Rendering Co., Auburn, Mass.

Prosperity Brand Complete Dressing Prosperity Brand Corn and Grain Fertilizer Prosperity Brand Market Garden Fertilizer Prosperity Brand Potato and Vegetable Fertilizer

# INSPECTION OF 1928.

During the past season the Station agent has visited 101 towns and villages in the State and has taken 536 official samples of fertilizers, including all of the registered brands which could be found. These, together with samples submitted by purchasers, and others interested, are classified as follows:

#### CLASSIFICATION OF FERTILIZERS ANALYZED IN 1928.

CLASSIFICATION OF TEXTIBLES III	1010.	
	No. of Samples	Page
I. Containing Nitrogen as the chief active ingredient:  Nitrate of Soda	33 4 5 3 12 55 123 11	18 18 18 19 19 19 20 20
II. Containing Phosphoric Acid as the chief active ingreds Precipitated Bone Phosphate Superphosphate (Acid Phosphate) Basic Slag	ent: 7 18 2	33 33 37
III. Containing Potash as the chief ingredient:  Carbonate of Potash.  Muriate of Potash.  Sulphate of Potash.  Sulphate of Potash-Magnesia.  Cotton Hull Ashes.  Wood Ashes.	16 12 18 2 28 7	37 37 37 38 38 38
IV. Containing Nitrogen and Potash:  Nitrate of Potash and Soda  Nitrate of Potash	2 5	45 45
V. Containing Nitrogen and Phosphoric Acid:  Dry Ground Fish	39 12 52	46 51 53
VI. Mixed Fertilizer:  Containing Nitrogen and Phosphoric Acid Containing Phosphoric Acid and Potash Containing Nitrogen, Phosphoric Acid and Potash	$6\\4\\265$	58 58 59
Special and Home Mixtures.  VII. Miscellaneous fertilizers, amendments, waste products, Sheep Manure, etc Lime, etc Other miscellaneous materials Collaborative check meals and fertilizers	50  etc.: 26 7 35 45	84 84 84 84 84
Total	904	

# I. RAW MATERIALS CHIEFLY VALUABLE FOR NITROGEN.

#### NITRATE OF SODA.

This raw material as used for fertilizer contains from 91 to 97 per cent of sodium nitrate which is equivalent to 15 to 16 per cent of nitrogen or 18.2 to 19.5 per cent of ammonia. The impurities in it are potassium and magnesium chlorides, sodium sulphate and sodium iodide of which there is, collectively, 2 to 3 per cent.

Nitrate of soda is obtained from the west coast of South America, chiefly from Chili. Until recently it has been practically the sole source of this form of agricultural nitrogen but it now encounters competition from synthetic nitrogen products. However, natural deposits of nitrate are not likely to be exhausted for many years to come and improved methods of production will no doubt enable this natural raw material to maintain an important place in the fertilizer trade.

Guaranties have ranged from 14.7 to 15.2 per cent. All samples, excepting 9557, have exceeded guaranties by substantial margins, the average nitrogen content being 15.55 per cent. At the prices quoted nitrogen from this source has cost 22.5 cents per pound. Ton prices ranged from \$63.00 to \$90.00 there being only one quotation at the high figure.

Analyses are given in Table I.

#### CALCIUM NITRATE.

This is one of the artificial nitrates produced in large quantities where air nitrogen is converted into nitric acid and the nitric acid then treated with lime or limestone to form nitrate of lime. Because it absorbs moisture rapidly it is shipped in drums or in paper lined bags.

The four samples examined were all guaranteed to contain 15 per cent of nitrogen and this figure was met excepting sample 9611 which was 0.2% low. The average retail price is about \$65.00 per ton which makes the cost of nitrogen per pound 21.7 cents.

Analyses are given in Table I.

#### CALUREA.

Calurea is one of the synthetic ammoniates and is a combination of calcium nitrate and urea. About 1/5 of the nitrogen is derived from nitrate and the remainder is in organic form from urea.

Five samples were examined, all guaranteed to contain 34 per cent of nitrogen or 41.3 per cent of ammonia. One sample, 9652, was considerably deficient and two others were slightly under the guaranty.

The price quoted is about \$110.00 per ton which, on the basis of 34 per cent nitrogen, makes the cost of nitrogen per pound 16.2 cents

Analyses are given in Table I.

#### UREA.

This synthetic product is made in large quantities in Europe and is obtained by combining synthetic ammonia with pure carbon dioxide. The commercial article offered for fertilizer is generally guaranteed to contain 46 per cent of nitrogen which is equivalent to 55.5 per cent of ammonia. This form of nitrogen is soluble in water and is classed as organic.

Three samples were examined two of which were slightly under guaranties. At \$165.00 per ton which is the retail price quoted to

us the cost of nitrogen per pound is 17.9 cents.

Analyses are given in Table I.

#### SULPHATE OF AMMONIA.

This raw material is made almost entirely in this country from sulphuric acid and by-product ammonia, that is, ammonia obtained in the production of coke and illuminating gas. It may be made, and is made on a large scale in Europe, from synthetic ammonia, gypsum and carbon dioxide.

"Arcadian" sulphate of ammonia is specially treated, dried, and screened to remove lumps and to insure good mechanical condition.

This product is generally guaranteed to contain 20.5 per cent of nitrogen which is equivalent to 25 per cent of ammonia. The twelve samples examined equalled or exceeded their guaranties excepting four cases where slight deficiencies were noted. The average nitrogen found was 20.71 per cent. At \$60.00 per ton the average price quoted, the cost of nitrogen per pound is 14.5 cents. Analyses are given in Table II.

# CASTOR POMACE.

This raw material is the ground residue left after the removal of oil from the castor bean. Caution should be used in storing it as it is very poisonous to farm animals. It is chiefly valuable for its nitrogen although it contains small amounts of phosphoric acid and potash.

Fifty-five samples were examined and only six failed to equal guaranties. The deficiencies were in most cases negligible, however, only two being in excess of 0.1 per cent and none were in

excess of 0.2 per cent.

The average nitrogen found was 4.97 per cent and the average of quoted prices was about \$30.00. Nitrogen from this source has cost about 30 cents per pound as compared with 24 cents last year. Analyses are given in Table III.

#### COTTONSEED MEAL.

One hundred and twenty-three samples of cottonseed meal, most of them submitted by purchasers, were examined. The average of all for which guaranties were given, and exclusive of three sold under odd guaranties, is 6.56 per cent of nitrogen which is a 41% protein grade.

The classification of samples this year in comparison with similar data for several years past is given in the following summary:

Grade	No. of	Average nitrogen		erage nitro	
	samples	per cent	1927	1926	1925
36 per cent protein (5.76 N)	15	5.75	5.78	5.84	5.79
41 per cent protein (6.58 N)	71	6.58	6.56	6.60	6.76
43 per cent protein (6.88 N)	30	6.88	7.05	6.78	7.02
No guaranty or odd guaranty	9	• • •		• • •	
Total	123	6.56	6.61	6.57	6.63

So far as prices are available the range has been from \$48.00 to \$70.00 per ton and the average \$54.00. On this basis nitrogen from this source has cost about 41 cents per pound during the past year.

One hundred and two substantially met or exceeded their guaranties and twenty-one did not.

Analyses are given in Table IV.

#### LINSEED MEAL.

Eleven samples of linseed meal were analyzed and all were of good qualtity. The average nitrogen content was 5.65 per cent. Prices quoted ranged from \$49.00 to \$57.50 and averaged slightly under \$55.00 at which figure the cost of nitrogen per pound was about 48 cents.

Analyses are given in Table IV.

TABLE I. ANALYSES OF NITRATE OF SODA, ETC.

			Per cer Nitroge	
Station No.	Manufacturer or Jobber,	Purchased, Sampled or Sent by	Found.	Guaranteed.
-	Nitrate of Soda			
8869	American Agricultural Chemical Co., New York	Station agent. Stock of Bristol. Grain & Supply, Bristol	15.66	15.22
9530	American Agricultural Chemical Co., New York	Station agent. Stock of E. N. Austin, Suffield	15.56	15.22
8988	Apothecaries Hall Co., Waterbury, Conn	Station agent. Stock of J. A. Glasnap, West Cheshire.	15.52	14.80
8923	Apothecaries Hall Co., Water- bury, Conn	Edwards & Brewer, West Suffield	15.68	14.80
8331	Apothecaries Hall Co., Water-	Hatheway & Steane, Inc., Hart- ford	15.56	14.80
8657	bury, Conn	Hatheway & Steane, Inc., Hartford	15.90	14.80
8835	Apothecaries Hall Co., Waterbury, Conn	ford	15.58	14.80
8941	Apothecaries Hall Co., Water- bury, Conn	Hatheway & Steane, Inc., Hartford	15.64	14.80
9330	Apothecaries Hall Co., Waterbury, Conn	A. N. Shepard & Son, Hartford.	15.48	14.80
8996	Armour Fertilizer Works, New York	Station agent. Stock of J. D. Kelsey, Madison	15.76	14.81
9549	Berkshire Chemical Co.,	Station agent. Stock of Knowles Lombard, Guilford	15.56	15.00
9557	Bridgeport, Conn Everett B. Clark Seed Co., Milford, Conn	Station agent at factory	14.40	15.00
8889	Consolidated Rendering Co., Boston, Mass.	C. R. Burr & Co., Inc., Manchester	15.66	15.22
9121	Consolidated Rendering Co., Boston, Mass	Walter T. Clark, Norwich	15.38	15.22
8882	Consolidated Rendering Co., Boston, Mass	Station agent. Stock of Laden Bros., Wallingford	15.74	15.22
9154	Eastern States Farmers' Exchange, Springfield, Mass.	Station agent. Stock of Ridge-wood Farm, No. Haven	15.60	14.80
8303	W. R. Grace & Co., New York	Bloomfield	15.58	
8169	W. R. Grace & Co., New York	Bloomfield	15.16	
8170	W. R. Grace & Co., New York	American Sumatra Tobacco Co., Bloomfield	15.80	
8171	W. R. Grace & Co., New York	Bloomneld	15.88	
8172	W. R. Grace & Co., New York	American Sumatra Tobacco Co. Bloomfield	15.76	
8914	Mapes Formula & Peruvian Guano Co., New York	Station agent. Stock of Mapes	5	14.81
8911	Olds & Whipple Inc., Hart-			15.00
9482	ford, Conn	Station agent. Stock of C. A		
2			- Later making	I amount

TABLE I. ANALYSES OF NITRATE OF SODA, ETC .- Concluded.

			Per Nitr	cent.
Station No.	Manufacturer or Jobber,	Purchased, Sampled or Sent by	Found.	Guaranteed.
9501 9291 9497	Nitrate of Soda.  Rackliffe Bros. Co. Inc., New Britain, Conn  The Rogers & Hubbard Co., Portland, Conn F. S. Royster Guano Co.,	Station agent at factory Station agent. Stock of W. L. Richmond, New Milford Station agent. Stock of F. B.	15.68 15.56	14.80
9587 9060	Baltimore, Md	Newton, Plainville	15.68 15.66	15.00 15.00
9341	Md. U. S. Guano Co., Baltimore, Md. Wessel Duval & Co., New York	J. A. Barrasso, Andover Station agent. Stock of Frank Libner & Son, Norwalk Station agent. Stock of F. H. Woodruff & Son, Milford, Ct.	15.38 15.24 15.66	14.80 14.80 14.87
9122 9353	Wilcox Fertilizer Co., Mystic, Conn	Walter T. Clark, Norwich Station agent at factory	15.52 14.92	14.72 14.72
9561 8899 9039 9611	Calcium Nitrate (Nitrate of Lime) International Agricultural Corp., Woburn, Mass Synthetic Nitrogen Products Corp., New York	Station agent. Stock of Lyman Farm, Middlefield The Allied Tobacco Co., Hartford Station agent. Stock of J. A. Glasnap, West Cheshire Station agent. Stock of Tobacco Station, Windsor	15.04 15.22 15.06 14.80	15.00 15.00 15.00 15.00
9652 9038 9116 9331 9392	Calurea.  Synthetic Nitrogen Products Corp., New York	J. E. Phelps, Suffield Station agent. Stock of Olds & Whipple, Inc., Hartford Station agent. Stock of Tobacco Station, Windsor  A. N. Shepard & Son, Hartford Eastern States Farmers' Exchange, Springfield, Mass	32.38 34.54 34.40 33.78 33.80	34.00 34.00 34.00 34.00 34.00
9456 9037 9115	Urea.  Eastern States Farmers' Exchange, Springfield, Mass Synthetic Nitrogen Products Corp., New York Synthetic Nitrogen Products Corp., New York	Station agent. Stock of Chester Beeman, Granby	45.40 45.80 45.84	45.24 46.00 46.00

TABLE II. ANALYSES OF SULPHATE OF AMMONIA.

		Per o Nitro	cent. ogen.
Manufacturer or Jobber,	Purchased, Sampled or Sent by	Found.	Guaranteed.
American Agricultural Chemical Co., New York	Station agent at factory, West Haven	20.76	20.56
cal Co., New York	Austin, Suffield	20.42	20.56
bury, Conn	Station agent at factory	21.00	20.58
Armour Fertilizer Works, New York	Bartlett Tree Expert Co., Stamford,	21.00	20.56
The Barrett Co., New York  The Barrett Co., New York	shire Chemical Co., Bridge- port	21.00	20.75
	port	20.70	20.50
Boston, Mass	Frisbie Co., New Haven	20.68	20.50
change, Springfield, Mass	Station agent. Stock of J. A. Sherwood, Bridgeport	20.56	20.55
ford, Conn	J. A. Barrasso, Andover	20.50	20.60
ford, Conn	Station agent at factory	20.54	20.60
Md	Beers, Southport	20.48	20.56
U. S. Guano Co., Baltimore,	J. A. Barrasso, Andover	20.92	20.56
	American Agricultural Chemical Co., New York American Agricultural Chemical Co., New York Apothecaries Hall Co., Waterbury, Conn Armour Fertilizer Works, New York  The Barrett Co., New York  Consolidated Rendering Co., Boston, Mass Eastern States Farmers' Exchange, Springfield, Mass Olds & Whipple Inc., Hartford, Conn Olds & Whipple Inc., Hartford, Conn U. S. Guano Co., Baltimore, Md	American Agricultural Chemical Co., New York American Agricultural Chemical Co., New York Apothecaries Hall Co., Waterbury, Conn Armour Fertilizer Works, New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  Consolidated Rendering Co., Boston, Mass Eastern States Farmers Exchange, Springfield, Mass  Olds & Whipple Inc., Hartford, Conn  Olds & Whipple Inc., Hartford, Conn  U. S. Guano Co., Baltimore, Md  U. S. Guano Co., Baltimore, Md  U. S. Guano Co., Baltimore, Md  Estation agent at factory, West Haven  Station agent. Stock of E. N. Austin, Suffield  Station agent. Stock of F. A. Bartlett Tree Expert Co., Stamford  Station agent. Stock of Berkshire Chemical Co., Bridgeport  Station agent. Stock of J. A. Sherwood, Bridgeport  Station agent at factory  Station agent Stock of J. A. Sherwood, Bridgeport  Station agent. Stock of J. A. Sherwood, Bridgeport  Station agent. Stock of J. A. Sherwood, Bridgeport  Station agent. Stock of H. P. Beers, Southport	American Agricultural Chemical Co., New York American Agricultural Chemical Co., New York Apothecaries Hall Co., Waterbury, Conn Armour Fertilizer Works, New York The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  The Barrett Co., New York  Station agent. Stock of Berkshire Chemical Co., Bridgeport  Station agent. Stock of Berkshire Chemical Co., Bridgeport  Station agent. Stock of L. T.  Frisbie Co., New Haven  Station agent. Stock of J. A.  Sherwood, Bridgeport  Station agent. Stock of J. A.  Sherwood, Bridgeport  Station agent at factory  Station agent. Stock of H. P.  Beers, Southport  20.76  21.00  21.00  Station agent. Stock of Berkshire Chemical Co., Bridgeport  Station agent. Stock of J. A.  Sherwood, Bridgeport  20.70  Station agent. Stock of J. A.  Sherwood, Bridgeport  Station agent. Stock of J. A.  Sherwood, Bridgeport  Station agent. Stock of J. A.  Sherwood, Bridgeport  20.56  Station agent. Stock of H. P.  Beers, Southport  20.68

TABLE III. ANALYSES OF CASTOR POMACE.

			Per c Nitro	
Station No.	Manufacturer or Jobber, Car No. or Mark	Purchased, Sampled or Sent by	Found.	Guaranteed.
8953 9318	The American Agricultural Chemical Co., New York City. Car N. Y. C. 247062	Station agent from factory Spencer Bros., Inc., Suffield	4.78 4.67	4.53 4.53
	Apothecaries Hall Co., Waterbury, Conn.			
8982 9610	waterbury, com.	Station agent from factory Station agent. Stock of Tobacco	4.75	4.52
8897	Car No. 17816	Station, Windsor The Allied Tobacco Co., Hart-	4.48	4.52
8922		ford Edwards & Brewer, West Suf-	5.20	4.52
8353	Car No. 10777	field	4.94	4.52
8361	Car No. 11712	ford	4.86	4.52
8362	Car No. 12075	ford	5.56	4.52
8387	Car No. 20404	ford	5.54	4.52
8388	Car No. 13462	ford	4.72	4.52
8389	Car No. 12952	ford	4.34	4.52
8390	Car No. 20415	ford	4.78	4.52
8391	Car No. 10919	ford	4.58	4.52
8392	Car No. 17915	ford	4.37	4.52
8393	Car No. 11624	ford	5.26	4.52
8451	Car No. 17229	ford	5.16	4.52
8656	Car No. 30368	ford	5.65	4.52
8688	Car No. 20448	ford	4.80	4.52
8689	Car No. 11172	ford	5.34	4.52
8824	Car No. 11104	ford	5.42	4.52
8825	Car No. 10720	ford	4.76	4.52
		ford	5.16	4.52

TABLE III. ANALYSES OF CASTOR POMACE—Continued.

			Per ce Nitrog	
Station No.	Manufacturer or Jobber, Car No. or Mark	Purchased, Sampled or Sent by	Found.	Guaranteed.
	Apothecaries Hall Co.,	The same of the same of		
8826	Waterbury, Conn. Car No. 11815	Hatheway & Steane, Inc., Hartford	5.56	4.52
8827	Car No. 13081	Hatheway & Steane, Inc., Hartford	5.36	4.52
8832	Car No. 10432	Hatheway & Steane, Inc., Hartford	4.32	4.52
8937 9329 9336	Car No. 97143	Hatheway & Steane, Inc., Hartford. A. N. Shepard & Son, Hartford. A. N. Shepard & Son, Hartford.	4.65 4.92 5.43	$4.52 \\ 4.52 \\ 4.52$
9441	Armour Fertilizer Works, New York City.	Station agent. Stock of A. R. Jones, Wallingford	4.58	4.52
8251 8252 9145	Baker Castor Oil Co., New York City. Car No. 17093	American Sumatra Tobacco Co., Bloomfield	4.69 4.85 4.92	4.50 4.50 4.50
8960 9531	Berkshire Chemical Co., Bridgeport, Conn.	Station agent. Stock of Ira Waters, Brookfield Station agent. Stock of E. M.	5.12	4.52
8371 8512 8513 8514 8517 8518 8525 9066 9067 9068 9069 9651	Car No. 213402 Car No. M. C. 91825 Car No. G. N. 33540 Car No. P. R. R. 517613 Car No. P. R. R. 43189	Cullman Bros., Hartford. Cullman Bros., Hartford. Cullman Bros., Hartford. Cullman Bros., Hartford. Spencer Bros., Suffield. Spencer Bros., Suffield. Spencer Bros., Suffield.	5.00 5.24 5.18 4.95 4.91 4.88 5.22 4.97 4.64 4.82 4.90	4.52 4.52 4.52 4.52 4.52 4.52 4.52 4.52

TABLE III. ANALYSES OF CASTOR POMACE—Concluded.

BULLETIN 296

			Per c Nitro	
Station No.	Manufacturer or Jobber, Cai No. or Mark	Purchased, Sampled or Sent by	Found.	Guaranteed.
9600	E. D. Chittenden Co., Bridgeport, Conn.	Station agent. Stock of J. P. Norton, Broad Brook	4.61	4.50
9161	Consolidated Rendering Co., Boston, Mass.	Station agent. Stock of L. T. Frisbie Co., New Haven	5.72	4.52
9233	Spencer Kellogg & Sons, Buffalo, N. Y.	Station agent. Stock of H. H. McKnight, Ellington	4.65	4.52
9485 9393 9394	Olds & Whipple Inc., Hartford, Conn.  No. 1. No. 2.	Station agent. Stock of H. E. Wells, East Windsor Hill H. E. Wells, Warehouse Point H. E. Wells, Warehouse Point	4.94 5.26 5.33	5.00 5.00 5.00
9098	Car C. N. J. 36426	L. Wetstone & Sons, Inc., Hartford.  L. Wetstone & Sons, Inc., Hartford.	5.19 4.98	4.75

TABLE IV. ANALYSES OF COTTONSEED AND LINSEED MEALS.

			Per c Nitro	ent. gen.
Station No.	Manufacturer or Jobber, Car No. or Mark	Purchased, Sampled or Sent by	Found.	Guaranteed.
	American Agricultural Chemical Co., New York City	Cottonseed Meal.		
9519		Station agent at factory	5.88	5.76
	Apothecaries Hall Co., Waterbury, Conn.			
8986 9201	Car No. S. O. U. 155736	Station agent at factory The Allied Tobacco Co., Hart-	6.46	6.58
9202	Car No. S. O. U. 130013	ford The Allied Tobacco Co., Hart-	6.34	6.56
8328	Car No. 57701	ford	6.88	6.56
8329	Car No. 409501	ford	6.83	• • • •
8347	Car No. 26782	ford	6.71	
8348	Car No. 55896	ford	6.87	6.58
		ford	6.71	6.58
8349	Car No. 330185	Hatheway & Steane, Inc., Hartford	6.62	6.58
8350	Car No. 1048	Hatheway & Steane, Inc., Hartford	6.58	6.58
8351	Car No. 172008	Hatheway & Steane, Inc., Hartford	6.39	6.58
8352	Car No. 154294	Hatheway & Steane, Inc., Hart- ford	6.71	6.58
8357	Car No. 159218	Hatheway & Steane, Inc., Hartford	6.60	6.58
8358	Car No. 50232	Hatheway & Steane, Inc., Hart-	6,61	6.58
8359	Car No. 55303	ford	5.92	6.58
8360	Car No. 27478	ford		
8394	Car No. 7052	ford	6.60	6.58
8447	Car No. 10234	ford	6.64	• • • •
8448	Car No. 47475	ford	6.10	6.58
8449	Car No. 55923	ford	5.94	6.58
8450	Car No. 37052	ford	6.48	6.58
8655	Car No. 163453	ford	6.79	6.58
	Cai 110, 105455	Hatheway & Steane, Inc., Hart- ford	6.74	6.58

TABLE IV. ANALYSES OF COTTONSEED AND LINSEED MEALS.—Continued.

			Per Nitr	cent.
Station No.	Manufacturer or Jobber, Car No. or Mark	Purchased, Sampled or Sent by	Found.	Guaranteed.
	Apothecaries Hall Co., Waterbury, Conn.	Cottonseed Meal.		
8684	Car No. 19120	Hatheway & Steane, Inc., Hart-	0. 50	
8685	Car No. 51268	ford	6.79	6.58
8686	Car No. 57978	Hatheway & Steane, Inc., Hart-	6.43	6.58
8687	Car No. 145866	ford	6.99	6.58
8943	Car No. 34009	ford	6.47 6.81	6.58
	Ashcraft-Wilkinson Co., Atlanta, Ga.		0.01	0.00
9598	Helmet	Station agent. Stock of E. A. Root, East Granby	6.54	6.58
9597	Monarch	Station agent. Stock of W. E. Bostwick, New Milford	6.75	6.88
8876 9014 9017	Car A. C. L. 29646	Spencer Bros., Inc., Suffield Spencer Bros., Inc., Suffield	7.00 6.44	6.88 6.56
	37854	Spencer Bros. Inc., Suffield	6.47	6.56
8270 9595	F. W. Brode Corp., Memphis, Tenn. Cold Press Cottonseed Meal. Owl Brand 41%	Ed. Eggert, Hartford Station agent. Stock of Tobacco	4.58	
9650	Owl Brand 36%	Station, Windsor	6.56 6.58	6.56 5.76
	Humphreys-Godwin Co.,			
8896	Memphis, Tenn. Dixie, Car No. 515977	The Allied Tobacco Co., Hart-		
9010	Dixie, Car No. 193844	ford The Allied Tobacco Co., Hart-	6.28	6.58
9147	Dixie, Car No. B. & O. 193844	ford	6.50	6.58
9148	Dixie, Car No. B. & O. 176849	ford The Allied Tobacco Co., Hart-	6.66	6 58
9149	Dixie, Car L. & N. No. 49693	The Allied Tobacco Co., Hart-	6.52	6.58
8944	Dixie, Car No. 41347	ford	6.57	6.58
9232	Danish	fordStation agent. Stock of Amos D.	6.51	6.58
		Bridge's Sons, Hazardville	5.67	5.75

TABLE IV. ANALYSES OF COTTONSEED AND LINSEED MEALS.—Continued.

			Per o Nitro	cent.
Station No.	Manufacturer or Jobber, Car No. or Mark	Purchased, Sampled or Sent by	Found.	Guaranteed.
	Humphreys-Godwin Co., Memphis, Tenn.	Cottonseed Meal.		
3079	Dixie, Car Sou. No. 253049  (Off Color)	A. A. Clark, Windsor	6.56	6.58
8080 8384 8385 8386 88515 88515 88520 88521 28523 88526 88604 88606 88606 88606 88606 88606 88606 88606 88607 88606	Dixie, Car C. & A., No. 36732 (Off Color)  Bull, Car No. 256123.  Bull, Car No. 131981.  Bull, Car No. 99263.  Bull, Car No. 57752.  Bull, Car No. 23062.  Bull, Car No. 305291.  Bull, Car No. 330868.  Bull, Car No. 341282.  Bull, Car No. 165243.  Bull, Car No. 165243.  Bull, Car No. 165472.  Bull, Car No. 166472.  Bull, Car No. 166472.  Bull, Car No. 166496.  Bull, Car No. 166496.  Bull, Car No. 166290.  Bull, Car No. 166290.  Bull, Car No. 166290.  Bull, Car No. 135449.  Dixie, Car No. 30296.  Dixie, Car No. 409299.  Dixie, Car No. 409299.  Dixie, Car No. 409299.  Dixie, Car No. 409299.  Dixie, Car Sou., No. 304709.  Dixie, Car Sou., No. 121643.  Dixie  Bull.  Dixie  Dixie, Car No. 42746.  Danish, Car No. 42746.  Danish, Car No. 42746.  Danish, Car No. 4746.  Danish, Car A. C. L., No. 43846.  Dixie, Car No. Y., No. 7064.  Bull, Car N. Y., No. 164622  and S. P. 28675.  Danish, Car A. C. L. No. 40349.	A. A. Clark, Windsor Cullman Bros., Inc., Hartford L. B. Haas & Co., Inc., Hartford Cullmington Bros., Windsor Huntington Bros., Windsor Huntington Bros., Windsor Huntington Bros., Windsor H. C. Nelson, West Suffield C. A. Peckham, Suffield C. A. Peckham, Suffield C. H. Rollins, Granby A. N. Shepard & Son, Hartford A. N. Shepard & Son, Hartford Spencer Bros., Inc., Suffield Spencer Bros., Inc., Suffield Spencer Bros., Inc., Suffield Spencer Bros., Inc., Suffield	6.56 6.74 6.84 6.65 7.02 6.79 6.80 6.77 6.88 6.66 6.82 7.02 6.96 6.76 6.84 7.07 6.89 6.70 6.84 6.70 6.66 6.66 6.56 6.66 6.56 6.55 6.55 6.5	6.58 6.88 6.88 6.88 6.88 6.88 6.88 6.88

TABLE IV. ANALYSES OF COTTONSEED AND LINSEED MEALS.—Continued.

			Per c Nitro	ent.
Station No.	Manufacturer or Jobber, Car No. or Mark	Purchased, Sampled or Sent by	Found.	Guaranteed.
	Humphreys-Godwin Co., Memphis, Tenn.	Cottonseed Meal.		
9314	Danish, Car A. C. L. No. 41604	Spencer Bros., Inc., Suffield	5.67	5.75
9315	Dixie, Car M. & St. L. No. 20578	Spencer Bros., Inc., Suffield	6.62	6.58
9316	Bull, Car C. N., No. 307465 and S. A. 39754	Spencer Bros., Inc., Suffield	7.00	6.88
9317 9469	Dixie, Car N. Y., No. 168144 and C. Ga. 51071 Danish, C. Ga. No. 57589	Spencer Bros., Inc., Suffield Spencer Bros., Inc., Suffield	6.74 5.62	6.58 5.75
9470	Danish, Car B. & O., No. 268990	Spencer Bros., Inc., Suffield	6.13	5.75
9471	Danish, Car C. & K. W., 141906	Spencer Bros., Inc., Suffield	5.69	5.75
9472	Danish, Car A. C. L. No. 50373	Spencer Bros., Inc., Suffield	5.74	5.75
9473	Danish, Car C. N. J., No. 17672	Spencer Bros., Inc., Suffield	5.70	5.75
9474	Danish, Car S. A. L. No. 12611	Spencer Bros., Inc., Suffield	5.70	5.75
9309	Dixie, Car No. 166221, No. 7.	L. Wetstone & Sons, Inc., Hartford	6.58	6.58
9310	Dixie, Car No. 82911, No. 8.	L. Wetstone & Sons., Inc., Hartford	6.48	6.58
9311	Dixie, Car No. 165007, No. 9.	L. Wetstone & Sons, Inc., Hartford	6.63	6.58
9094	Dixie, Car BO., No. 267964, No. 1	L. Wetstone & Sons, Inc., Hartford	6.49	6.58
9095	Dixie, Car C.R.G., No.57674, No. 2	L. Wetstone & Sons, Inc., Hartford	6.49	6.58
	International Agricultural Corporation,			
8440 8441 9011	Boston, Mass. Car No. C. Ga. 56996 Car No. C. G. 55396	L. B. Haas & Co., Inc., Hartford L. B. Haas & Co., Inc., Hartford	6.61 6.57	6.56
9011	Car No. 66246	The Allied Tobacco Co., Hart- ford	6.91	6.88
	L. B. Lovitt & Co.,			
9016 9071	Memphis, Tenn. "Lovit", Car No. M. & O. 24402	Spencer Bros., Inc., Suffield	7.00	6.88
9011	24299	Spencer Bros., Inc., Suffield	7.18	6.88

TABLE IV. ANALYSES OF COTTONSEED AND LINSEED MEALS.—Continued.

			Per o Nitro	
Station No.	Manufacturer or Jobber, Car No. or Mark	Purchased, Sampled or Sent by	Found.	Guaranteed.
	L. B. Lovitt & Co., Memphis, Tenn.	Cottonseed Meal.		
9073 8367 8368 8369 8370	"Lovit", Car M. & O. No. 23951 "Lovit, Car No. 15242 "Lovit", Car No. 15296 "Lovit", Car. No. 37091 "Lovit", Car. No. 37100	Spencer Bros., Inc., Suffield Cullman Bros., Inc., Hartford. Cullman Bros., Inc., Hartford Cullman Bros., Inc., Hartford Cullman Bros., Inc., Hartford	6.90 6.54 6.30 6.47 6.63	6.88 6.56 6.56 6.56 6.56
	Memphis Cottonseed Products			
8237	Memphis, Tenn. Car M. P. No. 22172	Steane, Hartman & Co., Hart-		
8238	Car C. M. No. 83294	ford Steane, Hartman & Co., Hart-	6.81	6.58
8239	Car M. P. No. 6575	ford Steane, Hartman & Co., Hart-	6.53	6.58
8240	Car M. P. No. 120094	ford Steane, Hartman & Co., Hart-	7.00	6.58
8241	Car M. P. No. 8696	ford	6.72	6.58
8242	Car N. Y. C. No. 96115	ford	6.89	6.58
8306	Car N. H. No. 71455	ford	6.45	6.58
8307	Car M. P. No. 120845	ford	6.75	6.58
8308	Car Sou. No. 155029	ford	6.84	6.58
8309	Car Ga. No. 19135	fordSteane, Hartman & Co., Hart-	6.71	6.58
	Olds_& Whipple, Inc.,	ford	6.76	6.58
8217	Hartford, Conn. Car No. 56539	L. B. Haas & Co., Hartford	6.74	6.58
0===	F. S. Royster Guano Co., Baltimore, Md.			
9568	2011111010, 1111,	Station agent. Stock of Chas. Handel, Glastonbury	6.81	6.88
3712	Simpson, Hendee & Co., Inc., New York	W. T. Dishmar 1 º C. 37		
		W. L. Richmond & Son, New Milford	3.40	3.20

TABLE IV. ANALYSES OF COTTONSEED AND LINSEED MEALS.—Concluded,

			Per c Nitro	ent. gen.
Station No.	Manufacturer or Jobber, Car No. or Mark	Purchased, Sampled or Sent by	Found.	Guaranteed.
9130 9129 9128 9493	Manufacturer Unknown Car A. C. L. No. 35998. Car M. P. No. 120565. Car A. C. L. No. 32569.	Cottonseed Meal.  James W. Shea, Feeding Hills.  James W. Shea, Feeding Hills.  James W. Shea, Feeding Hills.  E. N. & C. C. Austin, Suffield.	5.85 5.70 5.63 6.71	
	Apothecaries Hall Co., Waterbury, Conn.	Linseed Meal		
8354	Car No. 8459	Hatheway & Steane, Inc., Hartford	5.64	
8452	Car No. 60259	Hatheway & Steane, Inc., Hartford	5.79	
	Archer-Daniels-Midland Co.,			
8395	Buffalo, N. Y. Car No. 60493	Hatheway & Steane, Inc., Hart-		
3453	Car No. 9312	ford	5.65	
8895	Car LV, No. 60011	ford	5.64	
3945	Car No. 76685	ford	5.53	5.4
3946	Car No. 114485	ford	5.65	
		Hatheway & Steane, Inc., Hartford	5.75	
9327 9335	(Suffield)(Pelton)	A. N. Shepard & Son, Hartford. A. N. Shepard & Son, Hartford.	5.64 5.50	
	Olds & Whipple, Inc.,			
8216 9096	Hartford, Conn. Car No. 60427 Car N. H. No. 162502(No. 3)	L. B. Haas & Co., Hartford L. Wetstone & Sons, Inc	5.62 5.69	5.8

### II. RAW MATERIALS CHIEFLY VALUABLE FOR PHOSPHORIC ACID.

#### PRECIPITATED BONE.

This raw material is a by-product obtained in the manufacture of gelatin and glue stock from bone. Bones are treated with hydrochloric acid and the acid solution then treated with lime or limestone to precipitate the phosphates.

Seven samples were examined all of which exceeded their guaranties in available phosphoric acid. The average was 38 per cent. Only one price quotation was obtained and this affords no very satisfactory basis for calculating the cost of available phosphoric acid in this material. At \$56.00 per ton available phosphoric acid cost the purchaser 7.3 cents per pound.

Analyses are given in Table V.

### SUPERPHOSPHATE (ACID PHOSPHATE).

This important raw material was formerly called acid phosphate but control officials, agronomists and the fertilizer industry are agreed that the term "acid phosphate" is not properly descriptive and otherwise undesirable and that it should be discontinued. The product is made by treating phosphate rock with sulphuric acid which results in mixture of mono-calcium phosphate and gypsum. The phosphoric acid is in available form and usually present in the proportion of 16 pounds in 100 or 16 per cent.

Eighteen samples were analyzed all of which met or exceeded guaranties. The average for available phosphoric acid found is 17.11 per cent and the average of prices quoted is \$22.40 per ton. Available phosphoric acid from this source has, therefore, cost 6.5 cents per pound.

Analyses are given in Table VI.

Sample 8877, Virginia-Carolina Phospho Tobacco drawn from stock of Stanley-Svea Grain Co., New Britain, was guaranteed to contain 13 per cent of total phosphoric acid. The available found was 12.90 per cent.

TABLE V. ANALYSES OF PRECIPITATED BONE PHOSPHATE.

				Phosph	noric Acid.	
					"Avai	ilable''.
Station No.	Manufacturer or Wholesale Dealer.	Place of Sampling.	Citrate.	Total.	Found.	Guaranteed.
979 609 405 212 908	Sampled by Station.  Apothecaries Hall Co., Waterbury.  Apothecaries Hall Co., Waterbury.  Berkshire Chemical Co., Bridgeport.  Eastern States Farmers' Exchange, Springfield, Mass.  Olds & Whipple, Inc., Hartford.	At factory Tobacco Station, Windsor. At factory H. H. McKnight, Ellington At factory	% 1.68 1.35 2.25 1.10 1.88	% 39.20 39.50 40.60 39.30 40.20	37.52 38.15 38.35 38.20 38.32	36.00 36.00 38.00 38.00 38.00
455 326	Sampled by Purchaser. Apothecaries Hall Co., Waterbury Apothecaries Hall Co., Waterbury	Hatheway & Steane, Inc., Hartford A. N. Shepard & Son, Hartford	0.69 0.75	37.50 39.35	36.81 38.60	36.00 36.00

TABLE VI. ANALYSES OF SUPERPHOSPHATE (ACID PHOSPHATE).

				Phosphor	ic Acid.		
			ble.		"Avail	able".	
Station No.	Manufacturer or Wholesale Dealer.	Dealer or purchaser.	Citrate Insoluble	Total.	Found.	Guaranteed.	Station No.
	Sampled by Station.		%	%	%	%	
8866 8981 9542	American Âgricultural Chemical Co., New York Apothecaries Hall Co., Waterbury Armour Fertz. Works, New York	Bristol: Bristol Grain & Supply Co Sampled at factory, East Windsor Seymour: Seymour Grain & Coal Co.	0.68 1.68 0.40	19.75 17.75 16.50	19.07 16.07 16.10	16.00 16.00 16.00	8866 8981 9542
9548 9451	Berkshire Chemical Co., Bridge- port	Sampled at factory	0.20	18.65 17.05	18.45 16.65	16.00 16.00	9548 9451
8883	Consolidated Rendering Co., Boston, Mass	Wallingford: Laden Bros	1.70	18.15	16.45	16.00	8883
9152 8907	Eastern States Farmers' Exchange, Springfield, Mass Olds & Whipple, Inc., Hartford.	North Haven: Ridgewood Farm Sampled at factory	1.13 1.70	17.30 17.55	16.17 15.85	16.00 16.00	9152 8907
9499	Rackliffe Bros. Co., Inc., New Britain	Sampled at factory	0.23	16.90	16.67	16.00	9499
9036	The Rogers & Hubbard Co., Portland	Sampled at factory	0.15	17.93	17.78	16.00	9036
9298	F. S. Royster Guano Co., Balti- more, Md	Madison: J. D. Kelsey	1.35	17.40	16.05	16.00	9298
9582 8855	I. P. Thomas & Son, Philadelphia, Pa U. S. Guano Co., Baltimore, Md.	Hamden: Ira W. Beers	0.86	18.35 16.55	17.49 16.25	16.00 16.00	9582 8855

1				1			
			.oV noitst		8878	8879 9613	9649
		"Available".	Guaranteed.	%	20.00	16.00	16.00
ed.	Phosphoric Acid.	"Avai	Found	%	19.97	17.02	16.40
-Conclud	Phosph		.jstofr	%	20.30	17.85	18.20 18.50
HATE)—		.əldr	rlosuI ətantiO	%	0.33	0.83	1.80
(SES OF SUPERPHOSPHATE (ACID PHOSP			Dealer or Purchaser		New Britain: Stanley Svea Grain Co.	New Britain: Stanley Svea Grain Co. Woodstock: H. F. Joy.	Suffield: J. B. PhelpsAndover: J. A. Barrasso
TABLE VI. ANALY			Manufacturer or Wholesale Dealer.	Sampled by Station. Virginia-Carolina Chemical Co	New York	New York. Wilcox Fertilizer Co., Mystic	Sampled by Purchaser. Olds & Whipple, Inc., Hartford U. S. Guano Co., Baltimore, Md.
TABLE VI. ANALYSES OF SUPERPHOSPHATE (ACID PHOSPHATE)—Concluded.			.oV noitst	8878	8879	9613	9649

#### BASIC SLAG.

Basic slag is a by-product in the manufacture of steel from phosphatic iron ores. According to the A.O.A.C.¹ definition and standard it shall contain not less than 12 per cent of total phosphoric acid of which not less than 80 per cent shall be "available" by the Wagner method.

Two samples were examined.

Sample 8985 was from stock of Apothecaries Hall Co., drawn at the factory. It was guaranteed 15 available and 17 total phosphoric acid; 16.93 per cent and 18.85 per cent respectively were

Sample 9458, sold by E. D. Chittenden Bridgeport, drawn from stock of Elijah Rogers, Southington, was guaranteed 18 per cent total phosphoric acid and 17.46 per cent was found. No guaranty for available was made but 15.20 per cent was found.

# III. RAW MATERIALS CHIEFLY VALUABLE FOR POTASH.

#### CARBONATE OF POTASH.

In the pure, dry state this salt contains 68.2 per cent of actual potash ( $K_2O$ ), but commercial grades generally contain from 60 to 65 per cent.

Sixteen samples were analyzed of which only one, 8938, was notably deficient. The average potash content was 64.8 per cent and at \$140.00 per ton potash from this source has cost 10.8 cents per pound.

Analyses are given in Table VII.

#### MURIATE OF POTASH.

The grades of this salt which we used for fertilizer contain from 48 to 50 per cent of actual potash ( $K_2O$ ), the potash being largely in the form of chloride.

Twelve samples were examined all of which exceeded 48 per cent of potash and all exceeded guaranties except in case of 9159 which was about 0.5 per cent low. The average potash content was 51.6 per cent and the average ton price quoted was \$41.00. Potash from this source cost about 4 cents per pound.

Analyses are given in Table VII.

#### SULPHATE OF POTASH.

This potash salt contains not less than 48 per cent of potash (K₂O), and not over 2.5 per cent of chlorine according to the tentative definition and standard adopted by the Association of Official Agricultural Chemists.

<sup>&</sup>lt;sup>1</sup> Assoc. Official Agr. Chemists, Proc. of October, 1925.

Eighteen samples were examined. The average potash content found was 49.1 per cent. No considerable deficiencies were found, the greatest being 0.4 per cent. At \$63.00 per ton, the prevailing price quoted, potash from this source cost 6.4 cents per pound. Analyses given in Table VII.

### SULPHATE OF POTASH-MAGNESIA.

Two samples of this material were examined both of which exceeded guaranties by good margins. This salt is generally sold on a guaranty of 26 per cent potash.

#### Analyses are given in Table VII.

#### COTTONHULL ASHES.

This raw material is so extremely variable in composition that there is a great deal of difficulty in arriving at a fair settlement between buyer and seller on the basis of chemical analysis. In the same shipment we have found bags which analyzed about 30 per cent potash and others which tested only about 1/2 as much. Bulk goods arriving in car lots have been in such coarse and lumpy condition in some cases that it was impossible to obtain any adequate sample until the whole lot was ground and mixed.

Twenty-eight samples were examined with the results as shown in Table VII.

#### WOOD ASHES.

Seven samples of wood ashes have been analyzed and all were of good grade except 8381 which was very inferior, due to leaching. Analyses are given in Table VII.

	.oN noites	9194 9439 8912	8927	8840	8841	8938	8940 9324 9333 8445	8258	8259
sh.	Guaranteed.	60.00 63.00 65.00	00.09	00.09	00.09	00.09	60.00 60.00 63.00		
Potash.	Found.	67.52 66.37 64.98	66.62	61.80	64.60	56.541	63.24 65.08 65.43 62.64	66.97	67.23
TABLE VIII. AMALION OF TOTAL	Dealer or Purchaser.	Sampled at factorySampled at factory	Edwards & Brewer, West Suffield	Hatheway & Steane, Inc., Hartford	Hatheway & Steane, Inc., Hartford	Hatheway & Steane, Inc., Hartford	Hatheway & Steane, Inc., Hartford A. N. Shepard & Son, Hartford A. N. Shepard & Son, Hartford T. B. Haas & Co., Hartford	American Sumatra Tobacco Co., Bloom-field	American Sumatra Tobacco Co., Bloom-field
TABLE VII. AMARI	Manufacturer or Wholesale Dealer.	Carbonate of Potash. Sampled by Station. Apothecaries Hall Co., Waterbury. Berkshire Chemical Co., Bridgeport.	Sampled by Purchaser. Apothecaries Hall Co., Waterbury, Conn	Apothecaries Hall Co., Waterbury, Conn., Car No. 137142	Hall	Apothecaries Hall Co., Waterbury, Conn., Car No. 76685.	Apothecaries Hall Co., Waterbury, Conn., Car No. 80671 Apothecaries Hall Co., Waterbury Apothecaries Hall Co., Waterbury	A. Klipstein, New York	A. Klipstein, New York
	.oV noitst2	9194 9439 8512	8927	8840	8841	8938	8940 9324 9333	8445	8259

Moisture, 15.70%

ASH
SALTS,
ETC.

			Pot	tash.	
Station No.	Manufacturer or Wholesale Dealer.	Dealer or Purchaser.	Found.	Guaranteed.	Station No.
9313 9012 9653	Carbonate of Potash.  Sampled by Purchaser. Olds & Whipple, Inc., Hartford.  Manufacturer unknown.  Manufacturer unknown.	L. Wetstone & Sons, Inc., Hartford The Allied Tobacco Co., Hartford J. E. Phelps, Suffield	% 66.66 66.14 64.85	% 65.00 	9313 9012 9653
8955 9529 8980 9029 9544 8887 9159 9091 8852 8856 9590	Muriate of Potash. Sampled by Station.  American Agricultural Chemical Co., New York American Agricultural Chemical Co., New York Apothecaries Hall Co., Waterbury.  Armour Fertz. Works, New York.  Berkshire Chemical Co., Bridgeport.  Consolidated Rendering Co., Boston, Mass. Eastern States Farmers' Exchange, Springfield, Mass.  The Rogers & Hubbard Co., Portland. U. S. Guano Co., Baltimore, Md. U. S. Guano Co., Baltimore, Md. Wilcox Fertilizer Co., Mystic.	Sampled at factory, West Haven E. N. Austin, Suffield Sampled at factory F. A. Bartlett Tree Expert Co., Stamford Sampled at factory Cheshire Reformatory, Cheshire.  H. H. McKnight, Ellington Lyman Farm, Middlefield H. P. Beers, Southport Rippe Bros., Westport Sampled at factory	51.19 51.50 50.85 51.31 51.40 52.30 49.46 60.43 48.44 48.67 54.34	-,50.00 50.00 50.00 48.00 50.00 50.00 50.00 48.00 48.00 50.50	8955 9529 8980 9029 9544 8887 9159 9091 8852 8856 9590
9054	U. S. Guano Co., Baltimore, Md	J. A. Barrasso, Andover	49.78	48.00	9054

TABLE VII. ANALYSES OF POTASH SALTS, ETC.—Continued.

			Pot	ash.	
Station No.	Manufacturer or Wholesale Dealer.	Dealer or Purchaser.	Found.	Guaranteed.	Station No.
8951 9533 8994 9447 9543 8961 8906 9618	Sulphate of Potash. Sampled by Station.  American Agricultural Chemical Co., New York American Agricultural Chemical Co., New York Apothecaries Hall Co., Waterbury. Armour Fertz. Works, New York. Berkshire Chemical Co., Bridgeport. Consolidated Rendering Co., Boston, Mass. Olds & Whipple, Inc., Hartford. Olds & Whipple, Inc., Hartford.	Sampled at factory, West Haven E. N. Austin, Suffield Sampled at factory John Sestakanskas, Granby Sampled at factory L. T. Frisbie Co., New Haven Sampled at factory Sampled at factory Sampled at factory	% 50.30 49.22 49.28 48.94 48.83 48.82 48.45 50.76	48.00 48.00 48.00 48.00 48.00 48.00 48.65 48.65	8951 9533 8994 9447 9543 8961 8906 9618
8925 8837 8838 8839 8939 9325 9332 8065	Sampled by Purchaser.  Apothecaries Hall Co., Waterbury. Car No. 10432 Apothecaries Hall Co., Waterbury, Car No. 86709 Apothecaries Hall Co., Waterbury, Car No. 180284 Apothecaries Hall Co., Waterbury, Car No. 76685 Apothecaries Hall Co., Waterbury (Suffield) Apothecaries Hall Co., Waterbury (Pelton) Hollingshurst Co., New York, Car No. 566200.  Hollingshurst Co., New York, Car No. 3255		49.05 50.01 47.60 49.72 49.58 48.57 47.86 48.59 48.21	48.00 48.00 48.00 48.00 48.00 48.00 48.00	8925 8837 8838 8839 9325 9332 8065 8066

			Pot	ash.	
Station No.	Manufacturer or Wholesale Dealer.	Dealer or Purchaser.	Found.	Guaranteed.	Station No.
900	Sulphate of Potash.  Sampled by Purchaser. Olds & Whipple, Inc., Hartford	American Sumatra Tobacco Co., Bloomfield.	% 49.07	% 48.65	7900
952 995	Sulphate of Potash and Magnesia.  Sampled by Station.  American Agricultural Chemical Co., N. Y  Apothecaries Hall Co., Waterbury	Sampled at factory, West Haven	31.74 27.96	26.00 26.00	8952 8995
573 699 773 774 164 221 113	Cotton Hull Ashes. Sampled by Station. Ed Eggert, Hartford, Car No. R. I. 61821  Ed. Eggert, Hartford, Car R. I. No. 261016  Ed. Eggert, Hartford, Car R. I. No. 261016  Ed. Eggert, Hartford, Car R. I. No. 261016  International Agricultural Corp., Boston, Mass. International Agricultural Corp., Boston, Mass.	American Agricultural Chemical Co., West Haven L. B. Haas & Co., Hartford L. B. Haas & Co., Hartford J. E. Shepard, So. Windsor	26.81 30.41 34.52 35.57 16.03 18.46 18.19		8573 8699 8773 8774 8164 9221 9113

TABLE VII. ANALYSES OF POTASH SALTS, ETC.—Continued.

			Pot	ash.	
Station No.	Manufacturer or Wholesale Dealer.	Found,	Guaranteed.	Station No.	
9220 9114	Cotton Hull Ashes. Sampled by Station. International Agricultural Corp., Boston, Mass. International Agricultural Corp., Boston, Mass.	J. E. Shepard, So. Windsor Spencer Bros., Suffield	$\frac{\%}{12.36^{1}}$ $21.92$	% 	9220 9114
9643 9707 8398 8921 9365 9366 9367 9634 8199 7761 7765 8165 8165 8167 9127 9319 9690	Sampled by Purchaser. F. W. Brode Co., Memphis, Tenn. F. W. Brode Co., Memphis, Tenn. Ed. Eggert, Hartford. Ed. Eggert, Hartford, Car No. 156355. Ed. Eggert, Hartford, Car SL&SF No. 129344. Ed. Eggert, Hartford, Car R. I. No. 350135. Ed. Eggert, Hartford, Car K.C.M.& O., No. 6299 Ed. Eggert, Hartford (Light Gray) Ed. Eggert, Hartford (Dark Gray) Ed. Eggert, Hartford (Brown) Ed. Eggert, Hartford Ed. Eggert, Hartford Olds & Whipple, Inc., Hartford	J. B. Stewart, Windsor. J. B. Stewart, Windsor. The Allied Tobacco Co., Hartford. Edwards & Brewer, West Suffield. P. J. Anderson, Windsor. P. J. Anderson, Windsor. P. J. Anderson, Windsor. Amos D. Bridge's Sons, Hazardville. Ed. Eggert, Hartford. L. B. Haas & Co., Inc., Hartford. Will Hayes, Tariffville. Spencer Bros., Inc., Suffield. The Otee Tobacco Corp., Windsor.	21.69 23.38 16.92 28.61 31.98 29.16 39.27 32.48 25.86 25.70 22.04 11.45 11.78 28.97 16.89 32.25 27.85		9643 9707 8398 8921 9365 9366 9367 9634 8199 7761 7765 8165 8165 8167 9127 9319 9690

<sup>&</sup>lt;sup>1</sup> Contained 21.5% moisture.

45

SALTS, ETC.—Concluded

### IV. RAW MATERIALS CONTAINING NITROGEN AND POTASH.

Seven samples of this group of materials have been examined. four taken by the Station agent and three sent by purchasers.

9504. Nitrate of Soda and Potash. Apothecaries Hall Co... Waterbury. Sampled by Station agent, stock of Barnes & Co... Wallingford.

9769. Nitrate of Soda and Potash. Wilcox Fertilizer Co., Mystic, Conn. Sampled by Station agent at factory.

8910. O & W Nitrate of Potash 90%. Olds & Whipple, Inc. Hartford, Conn. Sampled by Station agent at factory.

9117. Nitrate of Potash (for experiment). Synthetic Nitrogen Products Co., New York. Sampled by Station agent from stock of Tobacco Station, Windsor.

8444. Nitrate of Potash. W. R. Grace & Co., N. Y. Submitted by L. B. Haas & Co., Inc., Hartford.

9089. Nitrate of Potash. Manufacturer unknown. Submitted by Cullman Bros., Hartford.

9648. Nitrate of Potash. Olds & Whipple, Inc., Hartford. Submitted by J. E. Phelps, Suffield.

TABLE VIII. ANALYSES OF NITRATE OF SODA AND POTASH, ETC.

Station No.	9504	9769	9117	8444	9089	9648	8910
Nitrogen:	70	70	70	70	70	10	/0
	14.84	14.63	13.28	15.00	13.44	12.08	12.04
guaranteed		14.72				12.00	12.00
Equivalent to amm	ionia:						
found	18.04	17.79	16.15	18.24	16.34	14.69	14.64
guaranteed	18.00	17.90				14.50	14.50
Potash:							
found	14.08	$11.70^{1}$	45.34	15.29	46.76	41.19	41.00
guaranteed	10.00	11.80				41.00	41.00
guaranteed	10.00	11.80				41.00	41.00

<sup>1</sup> Chlorine 0.46%.

# V. RAW MATERIALS CONTAINING NITROGEN AND PHOSPHORIC ACID.

#### DRY GROUND FISH.

This raw material is made from non-edible fish and from the offal from fish canneries. Oil is removed by steaming and pressing and the residue then dried and ground.

Thirty-nine samples were examined, in most cases samples being

submitted by purchasers.

This material is usually sold on a guaranty of 8.20 per cent of nitrogen and 5 or 6 per cent. of phosphoric acid. The average nitrogen content of all samples was 8.6 per cent and the average

phosphoric acid found was about 5 per cent.

The proportion of nitrogen in the form of ammonia salts was suspiciously high in a number of these samples and microscopic examinations confirmed the conclusion that ammonium sulphate was present. While the purchaser has received the total amount of nitrogenous plant food guaranteed to him he has not received it in the form in which he contracted to purchase it. Nitrogen in the form of ammonium sulphate has been bought this year for about 14.5 cents per pound whereas organic nitrogen in dry ground fish, making due allowance for the phosphoric acid present, has cost the purchaser about 44 cents per pound. The sale of a mixture of ammonium sulphate and dry ground fish, as and for dry ground fish, constitutes the sale of an adulterated article.

There should be no occasion to reinforce high grade dry ground fish which normally contains from 8 to 10 per cent of nitrogen. The increasing demand for fish, bone and tankage for animal and poultry feeding, for which purposes these materials are said to command better prices than for fertilizer purposes, suggests a possible reason for "stretching" the supply of high grade fish as much as possible or for using inferior materials reinforced with inorganic ammoniates to make a mixture appear high grade. However, since our investigation indicates that all of the samples of adulterated fish which we have found this year came from one source, there is no good reason to believe that the sale of spurious

fish fertilizer is at all general.

Dry ground fish will ordinarily contain small amounts of ammonia nitrogen. Thus 115 samples examined over a period of three years showed an average ammonia nitrogen content of 0.30 per cent and rarely exceeded 1 per cent in any case. The average total nitrogen content for the series was 8.6 per cent. In general it may be expected that animal proteins will yield ammonia nitrogen to the extent of from 6 to 10 per cent of the total nitrogen content when completely hydrolyzed by laboratory methods. Fish protein according to analyses by Osborne and Hyle<sup>1</sup> yielded 1.09

per cent of such nitrogen which was 5.85 per cent of the total nitrogen present. The acidulation of fish in commercial practice however, is not at all likely to produce the extent of hydrolysis obtained in laboratory procedure.

In the series of commercial samples examined this year 20 showed substantial or large amounts of ammonium sulphate and in 19 there was no evidence of excessive ammonia nitrogen and no ammonium sulphate could be detected by microscopic examination. Of the 20 samples showing ammonium sulphate the average proportion of ammonia nitrogen was 43 per cent of the total nitrogen and the maximum proportion was 52 per cent. Of the samples presumed to be genuine the average proportion of ammonia nitrogen was 2.6 per cent of the total and the maximum was a little over 10 per cent

We have no information as to whether or not the fish examined this year was acidulated but in any case the data just given are

pertinent to the issue involved.1

Of the samples listed in Table IX, 8155 to 8250 inclusive represented sales by J. W. Wilcox of Mystic, the fish being supplied, according to our information, by the Smith-Douglas Co. of Norfolk, Va. Other samples representing goods obtained through the above named jobber or directly from the Norfolk concern were 9550, E. D. Chittenden Co.; 8898, 9328, 9337, Apothecaries Hall Co.; and 8267, 8268, 8601 and 8603, Berkshire Chemical Co. These samples contained ammonium sulphate sufficient to supply from 10 to 40 per cent of the total nitrogen making due allowance for the ammonia nitrogen ordinarily found in fish fertilizer.

We have no evidence to warrant the conclusion that local dealers were aware of the spurious character of these goods and it is only fair to add that as soon as they were advised of our findings they took immediate steps to make adjustments with their purchasers by means of suitable rebates to cover the difference in

cost between inorganic and organic ammoniates.

<sup>1</sup> The following experiment serves to give some idea of the extent to which nitrogen transformation may occur when fish is acidulated commer-

<sup>&</sup>lt;sup>1</sup> Am. Jour. Physiol., 23, 81, 1908.

Fish containing 9.5 per cent of total nitrogen and .08 per cent of ammonia nitrogen was moistened thoroughly with 50 per cent sulphuric acid and allowed to stand at room temperature for one week. Ammonia nitrogen determined at that time was found to be 0.33 per cent. This experiment represents more vigorous treatment than obtains in commercial practice because the entire sample came into intimate contact with the acid.

			Nitr	ogen.	g.	Phos	phoric cid.	
9379 9528 8983 9545 9550 9452	Manufacturer or Wholesale Dealer.	Dealer or Purchaser.	Total found.	Total Guaranteed	Ammonia equivalent to total nitrogen.	Total found.	Total Guaranteed.	Station No.
9528 8983 9545	Sampled by Station.  American Agricultural Chemical Co., New York	E. N. Austin, Suffield	% 8.72 8.92 9.83 8.26	% 8.23 8.23 8.20 8.22	% 10.60 10.84 11.95 10.04	8.50 8.52 6.55 5.99	% 6.00 6.00 5.00 6.00	9379 9528 8983 9545
	E. D. Chittenden Co., Bridgeport Consolidated Rendering Co., Boston, Mass Olds & Whipple, Inc., Hartford U. S. Guano Co., Baltimore, Md Wilcox Fertilizer Co., Mystic	John P. Mason, Warehouse Point	8.07 8 31 9 48 8 01 8 21	8.00 8 22 8 23 8 22 8.24	9.81 10 10 11.53 9 74 9.98	3.08 8.03 7.48 6.28 6.08	6.00 6.40 5.00 6.00	9550 9452 9022 8857 9354
8898 8926 8330 8396 8454 8690	Sampled by Purchaser.  Apothecaries Hall Co., Waterbury	The Allied Tobacco Co., Hartford. Edwards & Brewer, West Suffield. Hatheway & Steane, Inc., Hartford	8.02 8.48 8.36 7.92 8.26 7.80	8.20 8.20 8.20 8.20 8.20 8.20	9.75 10.31 10.16 9.63 10.04 9.48	3.83 4.53 4.22 4.58 4.59 4.25	5.00 5.00 5.00 5.00 5.00 5.00	8898 8926 8330 8396 8454 8690

TABLE IX. ANALYSES OF DRY GROUND FISH.—Continued.

			Nitro	ogen.		Phosp Aci		
Station No.	Manufacturer or Wholesale Dealer.	Dealer or Purchaser.	Total found.	Total Guaranteed.	Ammonia equivalent to total nitrogen.	Total found.	Total Guaranteed.	Station No.
8817 8818 8834 8936 9328 9337 9526 8267 8268 9097 8155 8156 8157	Sampled by Purchaser.  Apothecaries Hall Co., Waterbury The Berkshire Chemical Co., Bridge- port The Berkshire Chemical Co., Bridge- port The Berkshire Chemical Co., Bridge- port Olds & Whipple, Inc., Hartford J. Waterman Wilcox, Mystic J. Waterman Wilcox, Mystic J. Waterman Wilcox, Mystic	L. B. Haas & Co., Inc., Hartford. L. B. Haas & Co., Inc., Hartford. L. W. Wetstone & Sons, Inc., Hartford.	8.41 8.23 8.56 7.96 8.26 8.77 7.90 9.74 9.75 9.11 8.72 8.30 8.68	8.20 8.20 8.20 8.20 8.20 8.20 8.22 8.22	7% 10.22 10.01 10.41 9.68 10.04 10.66 9.60 11.84 11.85 11.08 10.60 10.09 10.55	4.10 3.60 3.85 3.63 5.00 4.30 5.85 6.84 7.37 8.70 3.05 3.71 3.83	5.00 5.00 5.00 5.00 5.00 5.00 6.00 6.00	8817 8818 8834 8936 9328 9337 9526 8267 8268 9097 8155 8156

E IX. ANALYSES OF DRY GROUND FISH. - Concluded.

		Nitrogen.	gen.		Phosphoric Acid.	horic d.	
Manufacturer or Wholesale Dealer.	Dealer or Purchaser.	.bnuol istoT	Total Guaranteed	Ammonia equivalent to total nitrogen.	Total found.	Total Guaranteed.	Station No.
Sampled by Purchaser.	American Sumatra Tobacca Co	%	%	%	%	%	
M	TODACCO	8.58	:	10.43	3.32	:	8158
Wilcox, Mystic	Sumatra lobacco	8.36	:	10.16	3.51		8159
J. Waterman Wilcox, Mystic		8.36	:	10.16	3.48	:	8160
T With Mileon, Mileon, Marking	Longcco	8.52	:	10.36	2.96	:	8245
J. Waterman Wilcox, Mystic	BloomfieldAmerican Sumatra Tobacco Co.,	8.22		66.6	3.05	:	8246
T. Waterman Wilcox. Mystic	Tobacco	8.91		10.83	2.83	:	8247
Waterman Wilcox Mustic	Tobacco	8.82	:	10.72	2.70		8248
T. Waterman Wilcox, Mystic	Tobacco	8.02		9.75	3.52		8249
	ford	8.64		10.50	3.05		8250
Car No. P. A. 533730	Hartford	9.78	:	11.89	7.48		8603

#### TANKAGE.

This raw material is derived from refuse meat and bone. The refuse is treated with steam and then pressed to remove fat after which it is dried and ground. The composition depends upon the proportions of meat and of bone which are present; high nitrogen and low phosphoric acid indicates a preponderance of meat while the reverse is true if bone predominates. High grade tankage contains 8 to 10 per cent of nitrogen and 5 to 10 per cent of phosphoric acid.

Garbage tankage is less valuable as a fertilizer and generally contains not over 3.5 per cent. of nitrogen and not over 5 per cent

of phosphoric acid.

Twelve samples were analyzed. All met nitrogen guaranties excepting 8683, 8858 and 9606 which were from 0.3 to 0.4 per cent low. Phosphoric acid guaranties were met or exceeded excepting

9135 which was considerably low.

Microscopic examinations were made to discover inorganic ammoniates and mineral phosphates if present. In two or three samples both of those substances were found but the amounts did not appear to be sufficient to materially enhance either the nitrogen or the phosphoric acid.

Analyses are found in Table X.

	.oV noitst8		8956 8987 8993 9612 9438	9135	8962	8963 8683 8858 9606	0688
anical ysis.	Coarser than 1-50 inch.	%	58.0 64.0 48.5 49.5 65.0	0.09	0.79	76.0 62.0 58.0 57.0	74.0
Mechanical Analysis.	Finer than I-50 inch.	%	42.0 36.0 51.5 50.5	40.0	33.0	24.0 38.0 42.0 43.0	26.0
horic d.	Total guaranteed.	%	79.15 20.06 3.00 6.87 6.86	25.00	9.15	4.00 9.15 9.15 9.15	
Phosphoric Acid.	Total found.	2%	9.46 10.53 9.15 4.90 20.25 20.06 0.19 9.25 3.00 9.08 11.13 6.87 8.92 8.78 6.86	4.85 21.10 25.00		4.55 0.15 9.48 0.65	27 14.00
01 1	Ammonia equivalent total nitrogen.	%	9.46 10.53 '9.15 4.90 20.25 20.06 10.19 9.25 3.00 9.08 11.13 6.87 8.92 8.78 6.86	4.85	9.50 10.83	6.29 14.55 14.00 8.53 10.15 9.15 8.55 9.48 9.15 8.63 10.65 9.15	6.27
gen.	Total guaranteed.	%	7.40 7.40 7.40 7.40	3.29	7.41	4.92 7.40 7.40 7.40	<u> </u>
Nitrogen.	.bnuol fstoT	%	7.78 4.03 8.38 7.47 7.34	3.99	7.81	5.17 7.02 7.03 7.10	5.16
	Dealer or Purchaser.		Sampled at factory.  J. A. Glasnap, West Cheshire. Sampled at factory. Geo. R. Davidson, So. Meriden. Sampled at factory.	Sampled at factory	L. T. Frisbie Co., New Haven	L. T. Frisbie Co., New Haven Geo. S. Jennings, Southport H. D. Peters, Highwood J. B. Lewis, Southington, Conn	C. R. Burr & Co., Inc., Man
	Manufacturer	Sampled by Station. American Agricultural Chemical	Co., New York. Apothecaries Hall Co., Waterbury. Apothecaries Hall Co., Waterbury. Armour Fertilizer Works, New York Berkshire Chemical Co., Bridgeport. Connecticut Fat Rendering & Perti.	lizer Co., New Haven.	Mass. Consolidated Rendering Co., Boston,	Mass	Sampled by Purchaser. L. T. Frisbie Co., New Haven
	Station No.	8956	8987 8993 9612 9438	8962	8963		0688

#### BONE MEAL.

Raw Bone Meal or Raw Ground Bone is the product made by drying and grinding animal bones which have not been previously steamed under pressure.

Steamed Bone Meal or Steamed Ground Bone is the product made by steaming bones under pressure after which they are dried and ground.

Steamed bone contains about one-half as much nitrogen as raw

Fifty-two samples were examined, twenty-four of which were official samples drawn by the station agent. There were only three

deficiencies found in nitrogen and four in total phosphoric acid. There has been a scarcity of domestic bone during the past year

and manufacturers have in some cases been obliged to fill their

needs from foreign markets.

Although guaranties in nearly all cases were met or exceeded it was easy to see that some of the samples were not pure bone. This led us to examine microscopically all of the official samples, and many of those submitted by purchasers. All of the samples from stock sold by the U.S. Guano Company contained inorganic ammoniates or mineral phosphates or both. As in the case of dry ground fish, purchasers of this "bone" have received the amounts of nitrogen and of phosphoric acid which were guaranteed to them but it was not in the form in which they expected to obtain it. A mixture of ground bone, ammonium sulphate and mineral phosphate is not ground bone, but a mixed fertilizer containing A mixture of ground bone, ammonium sulphate and mineral phosphate is not ground bone, but a mixed fertilizer containing nitrogen and phosphoric acid. Moreover, there is a wide difference in price between nitrogen in ammonium sulphate and the organic nitrogen

8958. This sample was drawn from stock of the Eldredge Hardware Co. and sold by the Berkshire Chemical Co. A small amount of mineral phosphate and of ammonium sulphate was found. Information furnished by the manufacturer was to the effect that this was a small lot of bone mixture in which, however, no mineral phosphate or ammonium sulphate had been used. Samples of the ingredients used were submitted and the presence of these foreign materials in one of them was found, due, no doubt, to accidental contamination in the storage bins.

Analyses are found in Table XI.

Phosphoric Acid.

Mechanical Analysis.

GF
GROU
UD
B
ONE
-

			Nit.	rogen.	ent		phoric cid.		anical lysis.	
Station No.	Manufacturer.	Dealer or Purchaser.	Total found.	Total Guaranteed.	Ammonia equivalent to total nitrogen.	Total Found.	Total Guaranteed.	Finer than 1-50 inco.	Coarser than 1-50 inch.	Station No.
8864	Sampled by Station. American Agricultural Chemical Co.,		%	%	%	.%	%	%	%	
8984 9190 9000	New York	S. P. Strople, New Britain Sampled at factory C. A. Templeton, Waterbury Silliman Hardware Co., New	$\begin{bmatrix} 4.00 \\ 2.61 \end{bmatrix}$	3.29 2.46	3.14 4.86 3.17	23.10 23.65	$20.00 \\ 22.00$	$\begin{vmatrix} 48.5 \\ 62.0 \end{vmatrix}$	51.5 38.0	8984 9190
9404 9564 8965	Berkshire Chemical Co., Bridgeport. E. D. Chittenden Co., Bridgeport Consolidated Rendering Co., Boston,	Canaan Sampled at factory J. E. Stoddard, Abington	2.66	2.47 $2.47$ $2.47$ $2.47$	2.92 2.82 3.23	24.60 24.00 24.60	22.00 20.00 22.00	66.0	36.0 34.0	9000 9404 9564
9162	Mass Eastern States Farmers' Exchange,	Lightbourn & Pond, New Haven		2.46						
8964 8913	Springfield, Mass		4.37	3.28	3.31 5.31	24.45	22.00	46.0	38.0 54.0	9162 8964
8909 9619 9019 9020 9503	Co., New York Olds & Whipple, Inc., Hartford Olds & Whipple, Inc., Hartford The Rogers & Hubbard Co., Portland The Rogers & Hubbard Co., Portland F. S. Royster Guano Co., Baltimore,		$ \begin{array}{c} 2.15 \\ 2.50 \\ 3.52 \end{array} $	$\begin{bmatrix} 2.50 \\ 2.50 \end{bmatrix}$	2.61 2 3.04 2 4.28 2	26.15 25.70 20.94	$\begin{bmatrix} 22.00 \\ 22.00 \\ 20.50 \end{bmatrix}$	61.5	38.5 35.5	8909 9619
	Md	F. B. Newton, Plainville	2.83	2.47	3.44	23.05	22.90	66.0	34.0	9503

TABLE XI. ANALYSES OF GROUND BONE—Continued.

			Nitro	ogen.	ent	Phosp Ac	horic id.	Mecha Analy		
Station No.	Manufacturer.	Dealer or Purchaser.		Total guaranteed.	Ammonia equivalent to total nitrogen.	Total Found.	Total guaranteed.	Finer than 1-50 inch.	Coarser than 1-50 inch.	Station No.
9589 9580 8949 8950 8854 9343 9607 9360	Sampled by Station.  M. L. Shoemaker & Co., Philadelphia, Pa  I. P. Thomas & Son, Philadelphia, Pa.  U. S. Guano Co., Baltimore, Md  Wilcox Fertilizer Co., Mystic	Geo. T. Soule, New Milford Ira W. Beers, Hamden Frank Libner & Son, Norwalk H. P. Beers, Southport Rippe Bros., Westport E. O. Chapman, North Haven. J. B. Lewis, Southington Jordan Hardware Co., Willimantic	% 4.19 2.81 2.43 2.62 2.69 4.05 3.68 3.11	2.45 2.47 2.47 2.47 3.70 3.30	3.42 2.95 3.19 3.27 4.92 4.47	24.40 18.25 21.45 22.20 19.50 20.87	% 22.88 23.00 22.00 22.00 22.00 18.30 22.00	64.0 41.0 32.0 37.0 38.0 41.0	63.0 62.0 59.0	9580 8949 8950 8854 9343 9607
8924 8456 8457 8652 8653	Sampled by Purchaser. Apothecaries Hall Co., Waterbury Apothecaries Hall Co., Waterbury Apothecaries Hall Co., Waterbury Apothecaries Hall Co., Waterbury, Car No. 71092	Edwards & Brewer, West Suffield Hatheway & Steane, Inc., Hartford.	3.97 4.20 5.50	2.46 3.29 3.29	4.83 5.11 6.69	22.50 22.59 19.30	22.00 22.00 20.00 20.00 20.00	38.0 34.0	62.0 66.0	865

			Nitr	ogen.	ent		phoric eid.		anical lysis.	
Station No.	Manufacturer.	Dealer or Purchaser.	Total found.	Total guaranteed.	Ammonia equivalent to total nitrogen.	Found.	Guaranteed.	Finer than 1-50 inch.	Coarser than 1-50 inch.	Station No.
654	Sampled by Purchaser. Apothecaries Hall Co., Waterbury,	Hatheway & Steane, Inc., Hart-	%	%	%	%	%	%	%	
391	Car No. 92666	ford	4.19	3.29	5.09	23.48	20.00	56.0	44.0	86
	Car No. 10627	ford	4.74	3.29	5.76	20.40	20.00	43.0	57.0	86
19	Apothecaries Hall Co., Waterbury, Car No. 150058	Hatheway & Steane, Inc., Hartford	4.06	3.29		23.45				88
21	Apothecaries Hall Co., Waterbury	Hatheway & Steane, Inc., Hart-				1000				1357
36	Apothecaries Hall Co., Waterbury,	ford	3.55	3.29	4.32	24.60	20.00	62.0	38.0	.88
42	Car No. 10432	ford	4.35	3.29	5.29	19.60	20.00	41.0	59.0	88
.	Car No. 76685	ford	3.85	3.29	4.68	23.50	20.00	49.0	51.0	89
34 52	Apothecaries Hall Co., Waterbury Armour Fertilizer Works, New York,	A. N. Shepard & Son, Hartford American Sumatra Tobacco Co.,	2.48		3.02	27.95		65.0	35.0	93
	Car No. 49435	Bloomfield	2.81	2.47	3.42	23.99	22.00	39.0	61.0	81
53	Armour Fertilizer Works, New York, Car No. 50832	American Sumatra Tobacco Co., Bloomfield	2.54	2.47		24.25	1000			
4	Armour Fertilizer Works, New York,	American Sumatra Tobacco Co.,								81
3	Car No. 60953Armour Fertilizer Works, New York,	Bloomfield	2.60	2.47	3.16	23.80	22.00	50.0	50.0	81
	Car No. 81780	Bloomfield	2.88	2.47	3.50	23.51	22.00	48 0	52 0	81

TABLE XI. ANALYSES OF GROUND BONE-Concluded

			Nitro	ogen.	nt	Phosp Aci	horic id.	Mecha		
Station No.	Manufacturer.	Dealer or Purchaser.	Total found.	Total guaranteed.	Ammonia equivalent to total nitrogen.	Total found.	Total guaranteed.	Finer than 1-50 inch.	Coarser than 1-50 inch.	Station No.
8174	Sampled by Purchaser. Armour Fertilizer Works, New York,	American Sumatra Tobacco Co.,	%	%	%	%	%	%	%	
8253	Car No. 83734Armour Fertilizer Works, New York,	Bloomfield	2.50	2.47	3.04	22.48	22.00			817
8260	Car No. 38146	Bloomfield	2:62	2.47	3.19	26.46	22.00	51.5	48.5	825
	Armour Fertilizer Works, New York, Car No. 31595	Bloomfield	2.42	2.47	2.94	26.69	22.00	42.0	58.0	826
8261	Armour Fertilizer Works, New York, Car No. 299937	American Sumatra Tobacco Co., Bloomfield	2.60			26.51				
8290	Armour Fertilizer Works, New York, Car No. 184241	American Sumatra Tobacco Co., Bloomfield	2.74			25.82				
8291	Armour Fertilizer Works, New York,	American Sumatra Tobacco Co.,								
8304	Car No. 23930	Bloomfield	2.76			25.84				
8333	Car No. 78349Armour Fertilizer Works, New York,	Bloomfield	2.98	2.47		25.70				
9112	Car No. 46186	Bloomfield	2.85	2.47	3.46	25.54	22.00	31.8	68.2	833
8292	Mass Car No. 95779	Suffield	2.69	2.46	3.27	22.50	22.90	62.0	38.0	911
9062	U. S. Guano Co., Baltimore, Md U. S. Guano Co., Baltimore, Md	J. A. Barrasso, Andover Frank Libner & Sons, Norwalk	4.77	3.30	5.80	$24.95 \\ 16.80 \\ 26.53$	18.301	43.4	56.6	906

#### VI. MIXED FERTILIZERS.

# MIXTURES CONTAINING ONLY NITROGEN AND PHOSPHORIC ACID.

Six samples of this group of materials were analyzed.

**9195.** Ammo-Phos. American Cyanamid Co., New York. Sampled by Station agent from stock of Olds & Whipple, Inc., Hartford, Conn.

9191. Liberty Lawn Fertilizer. Apothecaries Hall Co., Waterbury. Sampled by the Station agent at factory.

9502. Liberty Tobacco Starter. Apothecaries Hall Co., Waterbury. Sampled by the Station agent at factory.

**9266.** O & W High Grade Tobacco Starter. Olds & Whipple, Inc., Hartford, Conn. Sampled by the Station agent from stock of E. O. Gates, New Hartford, Conn.

9511. "Swift-Sure" Tobacco Starter 4-10-0. M. L. Shoemaker & Co., Philadelphia, Pa. Sampled by the Station agent from stock of Olds & Whipple, Inc., Hartford, Conn.

9118. Diamonphos (for experiment). Synthetic Nitrogen Products Co., New York. Sampled by the Station agent from stock of the Tobacco Station, Windsor, Conn.

	9195	9191	9502	9266	9511	91181
	%	%	%	%	%	%
Nitrogen, found		4.26	3.80	13.79	3.36	20.66
guaranteed		3.29	3.29	8.23	3.25	
Ammonia equivalent to nitrogen		5.18	4.62	16.77	4.09	25.12
Phosphoric acid, total	19.75	12.68	11.20	2.78	11.98	53.70
available found	19.24	7.60	10.02	2,43	10.73	
guaranteed	19.00	4.00	10.00	3.00	10.00	

# MIXTURES CONTAINING ONLY PHOSPHORIC ACID AND POTASH.

This group is represented by four samples examined.

9386. Tobacco Ash Elements. American Agricultural Chemical Co. sampled from stock of Spencer Bros., Suffield.

9727. Tobacco Ash Elements. This was a second sample of the above but drawn from another source, stock of Howard Bariesford, West Suffield.

**9155.** Eastern States Open Formula 0-16-8. Sampled from stock of J. A. Sherwood, Bridgeport.

9514. Dairymen's Special, 0-10-10. I. P. Thomas and Son, Philadelphia, sampled from stock of Hubert Smith, Milford.

	9386	9727	9155	9514
Available phosphoric acid, found	5.30		16.23	10.95
Potash foundguaranteed.	$\frac{5.00}{14.00}$	15.05	$   \begin{array}{c}     16.00 \\     8.21   \end{array} $	$10.00 \\ 10.95$
guaranteed	15.00	15.00	8.00	$10.00 \\ 7.15$

Samples 9386 and 9727 require particular comment. These samples gave a great deal of trouble in the determination of potash. In spite of great pains taken to insure a uniform mixture very discordant results were obtained. Exchanges of samples with the manufacturers did not clarify the matter, they too having difficulty in securing satisfactory checks, although their guaranty was based upon factory tests which showed uniformly over 15 per cent according to advices received from them.

The second sample drawn by our agent analyzed slightly over the guaranty but in this case also the accepted figure is an average of several results which are not in as good agreement as could be desired.

# MIXED FERTILIZERS CONTAINING NITROGEN, PHOSPHORIC ACID AND POTASH.

In this group of fertilizers 250 official samples have been analyzed in the past season. Analyses of these, and of samples submitted by purchasers, 265 in all, are given in Table XII. The following tabulated statement summarizes the results of the inspection.

Total number of official samples analyzed		250	
one item	58		
two items			
three items	0		
total samples deficient	U	73	
Total items of alast fact and 1 (000 a)			
Total items of plant food guaranteed (250 x 3)		750	
Total items found deficient:			
ammonia	27		
available $P_2O_5$	23		
potash	38	88	
Total guaranties substantially met	90		
Dominant of an interest of the state of the		662	
Per cent of guaranties substantially met or exceeded		88	
Commercial deficiencies exceeding \$1.00 per ton		13	

#### DEFICIENT SAMPLES.

The number of samples which were found considerably deficient in one or more items of plant food is nearly 30 per cent of the total number examined, but the money values of such deficiencies have exceeded \$1.00 per ton in only about 5 per cent of the samples. Nearly 90 per cent of the total guaranties made have been substantially met or exceeded.

The distribution of deficiencies and the percentage of guaranties met is shown in the following summary, Table XIII:

<sup>&</sup>lt;sup>1</sup> Chlorine none.

	Per cent of guaranties met. 99 99 992 880 100 67 100 88 100 992 992 992 992 992 992 992 992 992 9	88
	Ober See in Potash.  Potash.  100  110  111  111  111  111  111  1	00
IES.	-Number deficiencies in Avail PsOs.  1 3 4 1 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	
EFICIENC	Ammonia 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	
SUMMARY OF DEFICIENCIES.	An	
XIII. SUN	Samples Namber 42 133 133 142 142 144 144 174 174 184 184 184 184 184 184 184 18	
TABLE	Manufacturer.  Apothecaries Hall Co. Armour Fettilizer Works. Atlantic Packing Co. Bartlett Tree Expert Co. Bartlett Tree Expert Co. Berkshire Chemical Co. C. & D. Bridge's Sons, Inc. E. D. Chittenden Co. E. B. Clark Seed Co. C. & R. Sales Co. Davey Tree Expert Co. Eastern States Farmers' Exchange. Essex Pertilizer Co. International Agicultural Corp. I. T. Frisbie Co. GashStull Co. International Agicultural Corp. Lowell Fertilizer Co. International Agicultural Corp. Lowell Fertilizer Co. Mapes Formula and Peruvian Guano Co. A. G. Markham & Co. International Agicultural Corp. Lowell Fertilizer Co. International Agicultural Corp. Fowell Fertilizer Co. International Agicultural Corp. Lowell Fertilizer Co. New England Fertilizer Co. Perdmont-Mt. Airy Guano Co. Predmont-Mt. Airy Guano Co. Predmont-Mt. Airy Guano Co. Rackliffe Bros. Co., Inc. The Rogers & Hubbard Co. F. S. Royster Guano Co. Rackliffe Bros. Co., Inc. The Rogers & Hubbard Co. Swift & Co.	The state of the s

#### COMMERCIAL DEFICIENCIES.

MIXED FERTILIZERS

Deficiencies in money value in excess of \$1.00 per ton have been shown in thirteen samples this year. This is an approximate value arrived at by balancing shortages against overages and reckoning ammonia at 20 cents, available phosphoric acid at 51/2 cents, and notash at 5 cents per pound.

TABLE XIV. SAMPLES SHOWING COMMERCIAL DEFICIENCIES.

	Appro	ximate deficiency
Sta. No.	Brand. in mor	ney value per ton.
9200	Bradley's Complete Manure for Potatoes and	
	Vegetables	
9079	National Aroostook Special Fertilizer	
8997	Armour's Big Crop 8-6-6	$1.94^{1}$
9146	Chittenden's Special Top Dresser	
9592	E. B. Clark Tip Top Brand 5-10-5	1.99
9160	Eastern States 8-16-20	1.25
9622	Eastern States 8-16-20	2.07
9454	Eastern States 10-3-8, Open Formula	1.13
9210	Eastern States 10-16-14, Open Formula	$1.99^{1}$
9565	Essex Top Dressing 7-6-5	1.85
9374	I. A. C. Tobacco Special 7-6-5	1.53
8884	Lowell Corn and Vegetable 4-8-4	$1.51^{2}$
9356	Worcester Rendering Co., Prosperity Brand	
	Complete Dressing	

<sup>&</sup>lt;sup>1</sup> Second sample not deficient.

In a system of inspection where many brands are represented by single analyses the products of the several manufacturers may not be fairly evaluated in any one year. Although it is our purpose to examine at least two samples of any brand showing considerable deficiencies, a second sample is not always obtainable. Comparisons are more informing, therefore, if made upon data covering a period of years. A number of different bases might be chosen for such comparisons, but since it has been our practice for many years to cite brands in which commercial deficiencies amount to one dollar or more per ton, these data have been compiled for the 8-vear period 1921-1928, and are tabulated in Table XV. No manufacturer is included unless ten or more samples have been analyzed in the period covered. The compilation shows that of a total of over 1,800 samples analyzed less than 10 per cent have fallen short of guaranties to any considerable extent in commercial value. Or, in other words, purchasers have received commercial values represented by the guaranties in over 90 per cent of purchases made.

<sup>&</sup>lt;sup>2</sup> Two other samples not deficient.

TABLE XV. COMMERCIAL DEFICIENCIES FOR THE PERIOD 1921-1928.

CONNECTICUT EXPERIMENT STATION

	Total	Number equaling or exceeding	Per	No. of	Per
	number	guaranties in money.	for 8 vr	samples	for
Manufacturer.	samples.	value.	period.	1928.	1928.
American Agricultural					
Chemical Co	347	333	96	42	95
Apothecaries Hall Co	74	74	100	13 13	100
Armour Fertilizer Works	86	67	78 92	3	92 100
Atlantic Packing Co	51 69	69	100	9	100
Berkshire Chemical Co	17	17	100	2	100
Bridge's, A. D. & Sons, Co. Chittenden, E. D. Co	52	47	90	7	86
Clark, E. B. Seed Co	37	34	92	5	80
Eastern States Farmers' Ex-					
change	81	66	81	16.	75
Essex Fertilizer Co	53	51	96	7	86
Frisbie, L. T. Co	81	71	88	7	100
International Agricultural					
Corp	61	55	90	2	50
Lowell Fertilizer Co	84	75	89	11	91
Mapes Formula and Peru-		400	0.0		100
vian Guano Co	104	103	99	14 7	100 100
New England Fertilizer Co.	62	59	95	7	100
Olds and Whipple, Inc	51	51	100		100
Parmenter & Polsey Fertili-	28	27	96	3	100
zer Co	48	41	90		100
Piedmont-Mt. Airy Guano	31	22	71	2	100
Rogers & Hubbard Co., The		110	97	16	100
Royster, F. S. Guano Co	63	50	79	9	100
Shoemaker, M. L. & Co	22	22	100	3 3	100
Springfield Rendering Co	32	30	94	3	100
Thomas, I. P. & Sons	32	32	100	8	100
United States Guano Co	24	23	96	8	100
Virginia-Carolina Chemical					
Co	57	53	93	4	100
Wilcox Fertilizer Co	60	56	93	$\frac{4}{7}$	100
Worcester Rendering Co	32	28	88	7	88
Totals	1804	1672	93	232	95

# QUALITY OF THE INSOLUBLE ORGANIC NITROGEN.

In the absence of vegetation tests the character of the nitrogenous material which is insoluble in water is judged by chemical methods. Such methods do not show "availability" of nitrogen but they do distinguish between the better and the poorer sources of water-insoluble nitrogenous material.

Both of the accepted methods for judging the quality of insoluble organic nitrogen depend upon the behavior of such nitrogen when treated with permanganate solutions. When less than 50 per cent of the water-insoluble organic nitrogen is found to be "active" as determined by the alkaline permanganate method, and less than

80 per cent active by the neutral permanganate method, the quality of the insoluble nitrogen is rated as inferior. Both methods are applied before such nitrogen is classed as inferior.

63

By these methods inferior ammoniates have been indicated in eleven samples; but in some cases the amounts of insoluble organic nitrogen have been so small that they were without practical significance and in others the more active forms of nitrogen practically equalled the guaranties. The samples were, therefore, passed without question.

TABLE XII. ANALYSES OF MIXED FERTILIZERS

Station No.	Manufacturer and Brand.	Grade.	Place of Sampling,
9357 9358 9362 8865 8865 9363 9105 9106 9107 9380 9082 8867 9381 9084 9088 9081 9111 9200 9623 9196 9076 9080 8978 9077 8868 9077 8868 9077 8868 9077 8968 9078 9078 9078 9078 9078 9078 9078 907	American Agricultural Chemical Co., New York.  A. A. C. Acme Fertilizer A. A. C. Aroostook Potato Manure A. A. C. Double A Tobacco Fertilizer A. A. C. Gladiator Fertilizer A. A. C. Grass & Lawn Top Dressing A. A. C. Monarch Fertilizer A. A. C. Old Hickory Fertilizer A. A. C. Princess Fertilizer A. A. C. Princess Fertilizer A. A. C. Princess Fertilizer Agrico Fertilizer for Corn Agrico Fertilizer for Truck Complete General Fertilizer Hi Grade Tobacco Bowker's All Round Fertilizer Bowker's Stockbridge Hill & Drill Fertilizer Bowker's Lawn & Garden Dressing Bowker's Potato and Vegetable Phosphate Bowker's Stockbridge Early Crop Manure Bowker's Stockbridge Tobacco Manure Bowker's Sure Crop Fertilizer Bradley's Blood Bone and Potash Bradley's Complete Manure for Potatoes and Vegetables Bradley's Complete Manure for Potatoes and Vegetables Bradley's Complete Tobacco Manure Bradley's Northland Potato Grower Bradley's Northland Potato Grower Bradley's Northland Potato Grower Bradley's Northland Potato Fertilizer National Aroostook Special Fertilizer National Aroostook Special Fertilizer National Arostook Special Fertilizer National Pine Tree State Potato Fertilizer National Pine Tree State Potato Fertilizer National Premier Potato Manure Sanderson's Complete Tobacco Grower Sanderson's Formula A. Sanderson's Formula B.	1-9-4 5-8-7 5-3-5 4-8-7 6-6-4 4-8-4 2-8-10-5 3-10-6 4-8-6 5-10-5 3-8-4 7-3-7 4-8-7 5-8-7 5-8-7 4-8-7 4-8-7 4-8-7 4-8-7 5-8-7 5-3-5 1-9-4 4-8-4 2-9-3 3-8-4 3-8-7 5-8-7 5-3-5 1-9-4 4-8-7 5-3-5 4-8-4 4-8-4 4-8-4	Woodbury. North Haven Gaylordsville. New Britain. New Britain. New Britain. Rockville. Gaylordsville. Danbury. Bethel. Bethel. West Haven. Glastonbury. New Milford. Bristol. West Haven. Collinsville. Thompsonville. Willimantic. Simsbury. East River.  West Haven. Glastonbury. Meriden. Colchester. Meriden. Suffield. Bristol. Middletown. Wallingford. Simsbury. Middletown. Middletown. Middletown. Middletown. Danbury. West Cheshire. Glastonbury. Gullford. Glastonbury. Gullford. Glastonbury.

/		Nitrogen.			to the	Ph	osphoric	Acid.	Po	tash.	
In Nitrates.	In ammonia.	Organic water-soluble.	Organic water-insoluble.	Total.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called "Available."	As muriate.	Total.	Station No.
	%	%	%	%	%	%	%	%	%	%	%
0.06 0.04 0.88 0.42 0.54 0.30 0.30 0.44 0.38 0.06 0.50 0.30 0.44 0.38 0.02 0.65 0.08 0.42 0.58 0.08	0.31 3.11 1.32 2.00 3.79 2.01 1.02 2.56 1.33 1.68 2.63 1.29 1.58 1.94 2.92 2.00 0.95 2.69 1.29 2.16	$\begin{array}{c} 0.39 \\ 0.33 \\ 0.04 \\ 0.46 \\ 0.34 \\ 0.49 \\ 0.52 \\ 0.52 \\ 0.49 \\ 0.66 \\ 0.45 \\ 0.21 \\ 0.31 \\ 0.42 \\ 0.07 \\ 0.47 \\ 0.61 \\ 0.02 \\ 0.46 \\ 0.41 \\ \end{array}$	$\begin{array}{c} 0.27 \\ 0.43 \\ 2.07 \\ 0.47 \\ 0.52 \\ 0.52 \\ 0.30 \\ 0.52 \\ 0.44 \\ 0.50 \\ 0.63 \\ 3.69 \\ 0.30 \\ 0.41 \\ 0.54 \\ 0.44 \\ 0.29 \\ 0.46 \\ 2.18 \\ 0.24 \\ 0.52 \\ \end{array}$	$\begin{array}{c} 1.03\\ 3.91\\ 4.31\\ 3.35\\ 5.19\\ 3.40\\ 1.90\\ 4.10\\ 2.56\\ 3.30\\ 4.12\\ 2.70\\ 5.84\\ 2.27\\ 3.17\\ 4.11\\ 3.29\\ 1.79\\ 4.16\\ 0.99\\ 4.06\\ \end{array}$	1.25 4.75 5.24 4.07 6.31 4.13 2.31 4.98 3.11 5.01 3.28 7.10 2.76 5.500 4.00 2.18 5.00 4.00 2.18 5.06 1.20 4.94	0.68 0.50 0.25 0.35 0.65 0.60 0.65 0.70 0.68 0.75 0.53 0.68 0.78 0.53 0.68	9.48 8.75 3.18 8.63 8.65 8.74 10.55 10.85 8.78 10.68 8.79 4.38 8.90 9.00 8.68 9.43 8.58 9.43 8.58 9.43 8.78	8.80 8.25 2.93 8.08 6.18 8.00 8.14 9.90 10.15 8.08 10.00 8.04 3.42 8.35 8.37 8.32 8.00 8.65 8.05 8.18	$\begin{array}{c} 4.14\\ 7.04\\ 0.38\\ 7.18\\ 3.90\\ 4.04\\ 10.17\\ 5.25\\ 5.93\\ 0.16\\ 3.95\\ 0.16\\ 3.98\\ 3.03\\ 6.85\\ 1.73\\ 4.58\\ 7.04\\ \end{array}$	4.14 7.04 5.29 7.18 3.90 4.04 10.17 5.25 5.93 5.19 3.95 8.01 3.98 3.03 6.78 7.15 3.98 3.03 6.85 7.04	9357 9358 9362 8865 8863 9520 9363 9105 9107 9361 9382 9085 9085 9084 9086 9381 9111
0.21	1.77	0.42	0.45	2.85	3.46	0.30	7.85	7.55	8.16	8.16	9200
0.31 0.64 0.03 0.36 0.07 0.00 0.22 0.41 0.42 0.73 0.05 0.21 0.33 0.05 0.39 0.35	2.03 0.91 0.28 1.99 0.86 1.64 1.46 2.40 2.52 1.16 1.60 2.04 1.98 1.63 1.22 1.97	0.53 1.03 0.41 0.47 0.43 0.40 0.24 0.53 0.56 0.10 0.51 0.51 0.41 0.40 0.53 0.26	0.46 2.28 0.28 0.51 0.42 0.38 0.59 0.48 0.44 2.23 0.40 0.45 0.48 0.35 2.49 0.47 1.30	3.33 4.86 1.00 3.33 1.78 2.42 2.51 3.82 3.94 4.22 2.56 3.29 2.44 4.50 3.32 3.44	4.05 5.91 1.22 4.05 2.16 2.94 3.05 4.64 4.79 5.13 3.11 4.00 4.01 2.97 5.47 4.04 4.18	0.38 0.27 0.60 0.75 0.80 0.75 0.90 0.43 0.48 0.30 0.65 0.50 0.73 0.30 0.65 0.90	8.95 4.00 9.48 8.73 9.43 8.85 10.70 8.45 8.43 3.55 8.93 8.53 8.98 11.11 3.35 8.75 8.43	8 57 3 73 8 88 7 98 8 63 8 10 9 80 9 80 8 02 7 95 3 25 8 23 8 48 10 38 3 05 8 10 8 34	7.40 0.36 4.52 3.86 2.94 4.06 6.78 7.26 0.09 4.01 3.95 7.07 4.24 0.39 4.00 0.78	7 40 6 73 4 52 3 86 2 94 4 06 6 78 7 26 4 57 4 01 3 95 7 07 4 24 5 22 4 00 7 36	9623 9196 9076 9080 8978 9077 8868 9079 9518 9390 9078 9512 9083 9087 9198 9593 9197

TABLE XII. ANALYSES OF MIXED FERTILIZERS

	TABLE XII. ANALYSES OF MIXED	ERS		CONTA	INING 1	Nitrog	EN, PH	OSPHOR	c Acii	AND ]	POTASH-	—Conti	nued.		
				-		Nitrogen.				M 100 100	phoric Ac		Pota		
Station No.	Manufacturer and Brand.	Grade.	Place of Sampling.	In Mitrates.	In Ammonia.	Organic water-soluble.	Organic water-insoluble.	Total.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called "Available".	As muriate.	Total.	Station No.
	Sampled by Station: American Agricultural Chemical Co.,			~	%	%	%	%	%	%	%	%	%	%	%
9384 9383	New York. Sanderson's Corn Superhphosphate Sanderson's Potato Manure	2-9-3 3-8-4		0.08 0.08	0.90 1.48	0.50 0.40	0.43 0.51	1.91 2.47	2.32 3.00	0.80 0.95	9.88 9.23	9.08 8.28	3.17 3.98	3.17 3.98	9384 9383
9102 9103 9110 9101 9387 9385 9100	Apothecaries Hall Co., Waterbury, Conn. Liberty Corn and All Crops Liberty Corn, Fruit and All Crops 2-12-4. Liberty Fish, Bone and Potash 3-8-3. Liberty H. G. Market Gardeners 5-8-7. Liberty H. G. Tobacco Manure 7-3-7. Liberty Home Vegetable Garden Fertilizer. Liberty Onion Special (Potash as Sulphate)	3-8-3 5-8-7 7-3-7	Greenville Greenville Waterbury West Cheshire East Windsor Waterbury	0.02 0.05 0.05 0.22 0.00 0.17	1.28 1.75 1.33 3.58 2.43 0.07	0.22 0.28 0.30 0.26 0.54 0.72	0.54 0.16 1.40 0.34 2.88 2.94	2.06 2.24 3.08 4.40 5.85 3.90	2.50 2.72 3.74 5.35 7.11 4.74	1.05 0.80 0.66 5.68	9.05 8.70 6.33 10.50	7.90 11.20 8.00 7.90 5.67 4.82	2.64 5.61 3.73 7.22 0.20 6.05	2.64 5.61 3.73 7.22 7.77 6.05	9102 9103 9110 9101 9387 9385
9193 9541	Liberty Potato and General Crop 4-8-10 Liberty Potato and Market Garden Special	4-8-7	East Windsor East Windsor	0.34 1.11	$\begin{bmatrix} 2.00 \\ 2.02 \end{bmatrix}$	0.57	0.88 0.13	3.79 3.45	4.61 4.19	0.85	14.68 8.90	11.03 8.05		10.03	9193
9109 9104	4-8-4. Liberty Potato and Vegetable 2-8-10. Liberty Top Dresser for Grass and Grain 10-3	4-8-4	Cheshire Waterbury	$0.04 \\ 0.73$	2.10 1.21	0.44 0.24	0.94	$\frac{3.52}{2.52}$	4.28 3.06	0.85 0.58	9.30 9.18	8.45 8.60	4.49 8.64	4.49 8.64	9109
9108 9551	1/2-8 Liberty Special Fertilizer for Fruit 7-8-6 Liberty Tobacco Special 5-3-5	10-3½-8 7-8-6 5-3-5	New London Waterbury New Milford	2.22 1.01 0.10	5.50 3.91 1.50	$0.10 \\ 0.56 \\ 0.26$	$0.08 \\ 0.36 \\ 2.34$	7.90 5.84 4.20	9.60 7.10 5.11	$0.28 \\ 1.88 \\ 0.36$	$ \begin{array}{r} 4.28 \\ 10.73 \\ 5.50 \end{array} $	4.00 8.85 5.14	$\begin{bmatrix} 8.53 \\ 6.52 \\ 0.77 \end{bmatrix}$	8.53 6.52 6.27	9104 9108 9551
9443 9002 9446 8888 9442 9389 9448 9137 9144 9388 8997 9621 9445	Armour Fertilizer Works, New York.  Armour's Big Crop Fertilizer 2-12-4.  Armour's Big Crop Fertilizer 3-8-4.  Armour's Big Crop Fertilizer 4-6-10.  Armour's Big Crop Fertilizer 4-8-7.  Armour's Big Crop Fertilizer 4-16-4.  Armour's Big Crop Fertilizer 4-16-4.  Armour's Big Crop Fertilizer 5-8-7.  Armour's Big Crop Fertilizer 5-8-7.  Armour's Big Crop Fertilizer 7-11-10.  Armour's Big Crop Fertilizer 7-11-10.  Armour's Big Crop 8-6-6.  Armour's Big Crop 8-6-6.  Armour's Big Crop Tobacco Fertilizer.	2-12-4 3-8-4 4-6-10 4-8-4 4-8-7 4-16-4 5-3-5 5-8-7 7-11-10 7-12-7 8-6-6 8-6-6 7-3-7	Madison	0.32 0.21 0.07 0.81 0.57 0.45 1.33 0.70 0.99 1.21 1.14	1.50 2.55 2.26 2.54 2.41 2.91 0.13 2.91 4.55 4.48 4.61 5.00 0.18	0.05 0.01 0.83 0.00 0.12 0.04 0.30 0.15 0.13 0.26 0.33 0.38 0.05	0.12 0.18 0.08 0.15 0.10 2.32 0.21 0.10 0.11 0.06 0.04 3.66	1.99 2.95 3.24 3.50 3.25 4.08 3.97 5.87 5.84 6.21 6.56 5.77	2.42 3.59 3.94 4.26 4.96 4.83 7.14 7.55 7.98 7.02	0.58 0.73 0.50 0.60 0.58 0.58 0.18 0.40 0.80 0.35 1.13 1.00 0.20	12.00 8.73 6.70 8.65 8.85 17.00 3.65 8.63 12.02 12.48 7.55 7.04 3.33	11.42 8.00 6.20 8.05 8.27 16.42 3.47 8.23 11.22 12.13 6.42 6.04 3.13	5.09 4.58 10.56 3.81 6.76 3.84 0.21 6.77 9.65 6.02 5.40 0.07	5.09 4.58 10.56 3.81 6.76 4.17 4.84 6.77 9.65 6.92 5.40 5.84 5.98	9143 9002 9446 8888 9442 9389 9448 9137 9144 9388 8997 9621 9445
9443 9602	Atlantic Packing Co., New Haven, Conn. Atlantic 5-8-7. Atlantic 5-4-16.	5-8-7 5-4-16	So. Windsor	2.65 1.82	0.07		0.97	4.23 4.78	5.14 5.81	1.29	9.40 7.15	8.11 6.60	0.36	7.25 18.24	9443 9602

TABLE XII. ANALYSES OF MIXED FERTILIZERS

Station No.	Manufacturer and Brand.	Grade.	Place of Sampling,
)437	Sampled by Station: Atlantic Packing Co., New Haven, Conn. Atlantic Special Vegetable 4-8-4	4-8-4	Waterbury
136	F. A. Bartlett Tree Expert Co., Stamford, Conn. Bartlett's Green Tree Food	6-8-4	Stamford
9440 9444 9138 9003 9139 9001 9376 9142	Berkshire Chemical Co., Bridgeport, Conn. Berkshire Complete Tobacco. Berkshire Economical Grass Fertilizer. Berkshire Grass Special. Berkshire Long Island Special. Berkshire Market Garden Fertilizer. Berkshire Potato and Vegetable Fertilizer. Berkshire Tobacco Special. Berkshire Tobacco Starter. Berkshire Truck Fertilizer.	5-3-5 10-3-8 7-6-5- 5-8-7 4-8-4 2-9-3 7-3-7 5-8-10 5-8-5	Suffield. Broad Brook. Waterbury. Brookfield. Torrington. New Canaan. Hockanum. Ellington. Ellington.
)140 )461	Amos D. Bridges' Sons, Inc., Hazardville, Conn. Corn, Onion, Potato and General Purpose Ferts ilizer Special Tobacco Fertilizer.	4-8-4 5-3-5	Hazardville Hazardville
3211 9159 9192 9505 9146 9459	E. D. Chittenden Co., Bridgeport, Conn. Chittenden's Complete Grain Chittenden's High Grade Potato Chittenden's Potato Special 4% Potash Chittenden's Tobacco Special Chittenden's Special Top Dresser Chittenden's Valley Wrapper Brand Chittenden's Vegetable and Onion Grower	2-9-3 5-8-7 4-8-4 5-4-5 8-6-6 6-3-5 3-8-4	Bloomfield
205 207 206	E. B. Clark Seed Co., Milford, Conn. Special Mixture for General Use. Special Mixture with 6% Potash. Super Phosphate.	4-8-4 4-8-6 5-8-7	Branford Branford

		Nitrogen.				Pho	osphoric A	cid.	Pot	ash.	
In Nitrates.	In ammonia.	Organic water-soluble.	Organic water-insoluble.	Total.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called "Available".	As muriate.	Total.	Station No.
%	%	%	%	%	%	%	%	%	%	%	%
0.41	2.12	0.44	0.53	3.50	4.26	1.32	9.45	8.13	3.97	3.97	9437
0.00	4.10	0.18	1.19	5.47	6.65	3.36	10.96	7.60	4.33	4.33	9136
1.09 6.98 2.19 0.87 0.55 0.06 2.20 1.56 0.09	0.07 0.04 1.42 2.51 1.94 1.54 0.10 2.37 3.12	0.46 0.60 1.97 0.29 0.18 0.17 0.51 0.22 0.32	2.86 0.80 0.70 0.76 0.83 0.14 3.30 0.63 1.00	4.48 8.42 6.28 4.43 3.50 1.91 6.11 4.78 4.53	5.45 10.24 7.64 5.39 4.26 2.32 7.43 5.81 5.51	0.56 3.90 1.49 0.45 0.18 0.53 0.15 0.93 0.45	3.63 9.68 7.55 8.78 8.87 10.23 3.80 10.25 8.70	3.07 5.78 6.06 8.33 8.69 9.70 3.65 9.32 8.25	0.52 2.07 6.59 7.60 4.55 4.79 0.70 0.82 5.94	5.68 8.37 6.59 7.60 4.55 4.79 6.88 8.77 5.94	9440 9444 9138 9003 9139 9001 9376 9142 9141
0.80 1.19	1.75 0.07	0.12	0.80 2.96	3.47 4.66	4.22 5.67	0.68 0.45	9.38 4.60	8.70 4.15	4.28	4.28 5.45	9140 9461
0.07 0.08 0.00 0.00 0.00 0.00 0.07	1.44 3.64 2.68 2.48 4.47 2.19 2.29	0.13 0.12 0.23 0.28 0.53 0.22 0.23	0.29 0.30 0.35 1.64 0.32 2.65 0.30	1.93 4.14 3.26 4.40 5.32 5.06 2.89	2.35 5.03 3.96 5.35 6.47 6.15 3.51	0.53 0.18 0.53 0.20 0.23 0.56 0.23	10.05 8.40 8.66 5.20 7.88 4.96 8.88	9.52 8.22 8.13 5.00 7.65 4.40 8.65	2.74 7.09 4.02 1.06 5.34 0.32 3.69	2.74 7.09 4.02 5.75 5.34 5.19 4.19	9211 9199 9192 9505 9146 9459 9214
0.00 0.05 0.14	2.27 2.38 3.46	0.44 0.37 0.18	0.73 0.80 0.44	3.44 3.60 4.22	4.18 4.38 5.13	$\begin{bmatrix} 0.55 \\ 0.61 \\ 0.45 \end{bmatrix}$	8.65 9.08 8.68	8.10 8.47 8.23	4.44 6.45 6.31	4.44 6.45 6.31	9205 9207 9206

TABLE XII. ANALYSES OF MIXED FERTILIZERS

No.	Manufacturer and Brand.	Grade.	Place of Sampling.
Station No.			
	Sampled by Station: E. B. Clark Seed Co.,		
9725 9592	Milford, Conn. Super Phosphate Tip Top Brand	5-8-7 5-10-5	
9562	C. & R. Sales Co., Worcester, Mass. C. & R. Lawn and Shrub Fertilizer	5-6-5	Norwich
8998	Davey Tree Expert Co., Kent, Ohio Davey Tree Food	7-8-3	Sound Beach
	Eastern States Farmers' Exchange, Springfield, Mass.		
9457	Buckland's Formula A 6-2-7 Special Tobacco Mixture	6-2-7	So. Manchester
9209 9208	Eastern States 3-12-3 Open Formula	3-12-3 4-8-10	Bridgeport
9151	Eastern States 4-8-10 Open Formula Eastern States 4-16-4 Open Formula	4-16-4	North Haven
9153 9158	Eastern States 5-8-7 Open Formula Eastern States 5-10-5 Open Formula	5-8-7 5-10-5	North Haven Ellington
9156 9213	Eastern States 8-6-6 Open Formula Eastern States 8-16-8 Open Formula	8-6-6 8-16-8	
9160 9622 9455	Eastern States 8-16-20 Open Formula. Eastern States 8-16-20 Open Formula. Eastern States Open Formula 9-3-7 Tobacco	8-16-20 8-16-20	Ellington
9454	Fertilizer	9-3-7	Granby
9737	Fertilizer	10-3-8	Ellington
9004 9210	Open Formula	10-16-14	
9608	Eastern States 10-16-14 Open Formula	10-16-14	Branford
	Essex Fertilizer Co., Boston, Mass.		
9450 9224	Essex A1 Super 2-10-2 Essex Complete Manure 5-8-7	2-10-2 5-8-7	Wallingford Wallingford
9204	Essex Fish Fertilizer for All Crops 3-8-4	3-8-4	Wallingford
9203 9453 9567 9565	Essex Market Garden 4-8-4. Essex Peerless Potato Manure 4-6-10. Essex Tobacco Manure 5-3-5.	4-8-4 4-6-10 5-3-5	Wallingford Warehouse Point New Milford
2000	Essex Top Dressing 7-6-5	7-6-5	So. Manchester.

CONTAINING NITROGEN, PHOSPHORIC ACID AND POTASH—Continued.

_		Nitroge	n.			Pho	sphoric A	Acid.	Pot	ash.	
In Nitrates.	In ammonia.	Organic water-soluble.	Organic water-insoluble.	Total.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called "Available".	As muriate.	Total.	Station No.
%	%	%	%	%	%	%	%	%	%	%	%
j. i j	2.70	0.39	0.44	4.05 3.72	4.92 4.52	0.75 0.43	9.35 10.78	8.60 10.35	4.54	6.47 4.54	9725 9592
0.25	2.18	1.49	2.28	6.20	7.54	0.95	7.35	6.40	4.47	4.47	9562
0.00	3.50	0.54	2.02	6.06	7.37	5.15	13.68	8.53	3.21	3.21	8998
0.22 0.73 0.16 0.89 1.00 0.64 2.18	0.88 1.25 2.23 1.98 2.70 2.52 3.68	0.92 0.32 0.64 0.30 0.53 0.82 0.66	3.76 0.30 0.29 0.38 0.24 0.35 0.23	5.78 2.60 3.32 3.55 4.47 4.33 6.75	7.03 3.16 4.04 4.32 5.43 5.26 8.21	0.38 0.45 0.39 0.48 0.53 0.18 0.23	4.84 12.18 8.68 15.00 8.83 11.00 6.60	4.46 11.73 8.29 14.52 8.30 10.82 6.37	0.68 3.24 10.05 5.90 7.03 5.44 6.47	8.00 3.24 10.05 5.90 7.03 5.44 6.47	9457 9209 9208 9151 9153 9158 9156
1.26 1.16 1.32 2.50	4.67 4.58 4.22 0.60	0.73 $0.48$ $0.12$ $1.34$	0.34 0.50 0.86 3.61	7.00 6.72 6.52 8.05	8.51 8.17 7.93 9.79	0.38 0.55 0.68 0.08	16.05 17.20 16.05 4.85	15.67 16.65 15.37 4.77	11.91 17.35 18.90 0.74	11.91 17.35 18.90 9.36	9213 9160 9622 9455
2.08	0.85	1.44	2.88	7.25	8.81	0.50	6.25	5.75	1.83	8.61	9454
1.66 1.64 1.00 1.36	5.92 5.44 5.51 5.85	0.26 0.26 1.02 0.65	0.94 0.80 0.42 0.44	8.78 8.14 7.95 8.30	10.67 9.90 9.67 10.09	0.30 0.58 0.63 0.85	16.55 17.38 17.20 18.65	16.25 16.80 16.57 17.80	1.38 14.38 12.70 13.21	14.00 14.38 12.70 13.21	9737 9004 9210 9608
0.00 0.18 0.03 0.31 0.35 1.65 0.00	0.92 2.97 1.36 1.80 1.70 0.14 5.08	0.43 0.66 0.52 0.47 0.57 0.41 0.29	0.30 0.52 0.58 0.74 0.55 2.30 0.09	1.65 4.33 2.49 3.32 3.17 4.50 5.46	2.01 5.26 3.03 4.04 3.85 5.47 6.64	0.65 1.20 0.75 0.88 0.58 0.50 0.33	11.43 9.53 8.35 11.10 6.60 4.57 6.35	10.78 8.33 7.60 10.22 6.02 4.07 6.02	1.92 7.36 4.13 4.41 9.93 0.61 4.57	1.92 7.36 4.13 4.41 9.93 5.29 4.57	9450 9224 9204 9203 9452 9567 9565

73

TABLE XII. ANALYSES OF MIXED FERTILIZERS

	Course for an example of the property		/		Nitroger	1.		t t	Pho	sphoric A	Acid.	Pot	ash.		
Station No.	Manufacturer and Brand.	Grade.	Place of Sampling,	In Nitrates.	In Ammonia	Organic water-soluble.	Organic water-insoluble.	Total.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called ". Available".	As muriate.	Total.	Station No.
	Sampled by Station: L. T. Frisbie Co.,			%	%	%	%	%	%	%	%	%	%	%	%
9225 9227 9229 9223 9226	New Haven, Conn. Frisbie's 5-8-7. Frisbie's Corn and Grain Fertilizer 2-10-2. Frisbie's Market Garden. Frisbie's Special 3-8-4. Frisbie's Special Vestable and Daylor Constitution of Daylor Constitution of Daylor Constit	5-8-7 2-10-2 4-8-7 3-8-4	Wethersfield	0.27 $0.04$ $0.40$ $0.01$	2.90 0.78 1.92 1.52	0.51 0.51 0.50 0.51	0.64 0.58 0.77 0.69	4.32 1.91 3.59 2.73	5.25 2.32 4.36 3.32	0.93 1.08 1.15 1.03	9.53 11.33 9.23 9.20	8.60 10.25 8.08 8.17	7.25 2.26 6.90 4.27	7.25 2.26 6.90 4.27	9225 9227 9229 9223
9460 9230	Frisbie's Special Vegetable and Potato Grower 4-8-4 Frisbie's Tobacco Grower 5-3-5. Frisbie's Top Dresser 7-6-5.	4-8-4 5-3-5 7-6-5	Danbury So. Windsor Watertown	$0.29 \\ 1.41 \\ 1.04$	2.22 0.13 3.52	0.55 0.43 0.68	$0.64 \\ 2.74 \\ 0.66$	3.70 4.71 5.90	4.50 5.73 7.17	1.05 0.57 0.65	9.45 $4.30$ $7.20$	8.40 3.73 6.55	4.12 0.16 5.19	4.12 5.11 5.19	9226 9460 9230
9488	Gash-Stull Co., Chester, Pa. Young's Formula 8-7-6	8-7-6	Mfg.'s sample	2.16	3.18	0.48	0.50	6.32	7.68	0.58	8.15	7.57	5.73	5.73	9488
9234	Grasseli Chemical Co., Cleveland, Ohio Grasselli Odorless Plant Food	5-13-4	Fair Haven	0.00	4.24	0.00	0.11	4.35	5.29	1.08	16.90	15.82	4.42	4.42	9234
9374 9375	International Agricultural Corp., Boston, Mass. I. A. C. Tobacco Special 7-6-5 Vuelta Abajo Special Tobacco Formula	7-6-5 7-9-8	Hockanum	1.43 0.82	2.01 1.76	0.11 0.83	1.97 2.29	5.52 5.70	6.71 6.93	0.28	6.40 9.43	6.12 8.90	$0.54 \\ 0.72$	4.50 8.21	9374 9375
8871 9563 8873 8884 9521 8885 8872 9228 9243 9244 9245	Lowell Fertilizer Co., Boston, Mass.  Lowell Animal Brand for All Crops 3-8-4. Lowell Bone Fertilizer 2-10-2. Lowell Corn and Vegetable 4-8-4. Lowell Corn and Vegetable 4-8-4. Lowell Corn and Vegetable 4-8-4. Lowell Market Garden Manure 5-8-7. Lowell Market Garden Manure 5-8-7. Lowell Potato Grower 4-6-10. Lowell Tobacco Manure 5-3-5. Lowell Top Dressing 7-6-5. Lowell 5-10-5.	3-8-4 2-10-2 4-8-4 4-8-4 4-8-4 5-8-7 5-8-7 4-6-10 5-3-5 7-6-5 5-10-5	Southington	0.00 0.00 0.39 0.36 0.31 0.38 0.35 0.34 1.69 1.09	1.41 1.04 1.76 1.87 2.10 2.55 2.74 1.86 0.11 3.49 1.16	0.61 0.51 0.50 0.40 0.49 0.50 0.53 0.52 0.35 0.65 0.58	0.38 0.30 0.55 0.60 0.70 0.67 0.52 0.64 2.44 0.65 0.70	2.40 1.85 3.20 3.23 3.60 4.10 4.14 3.36 4.59 5.88 4.24	2.92 2.25 3.89 3.93 4.38 4.98 5.03 4.09 5.58 7.15 5.15	0.93 1.20 0.85 1.13 1.08 0.65 0.70 0.60 0.63 0.78 1.13	8.68 10.48 8.85 8.08 8.40 9.03 8.75 6.85 4.85 7.03 11.15	7.75 9.28 8.00 6.95 7.32 8.38 8.05 6.25 4.22 6.25 10.02	1.78 2.15 3.94 3.93 4.43 7.11 6.82 10.48 0.61 5.13 5.40	3.98 2.15 3.94 3.93 4.43 7.11 6.82 10.48 5.46 5.13 5.40	18871 19563 18873 18884 19521 18885 18872 9228 9243 9244 9245
9247 8972 9027 9246	Mapes Formula and Peruvian Guano Co., New York.  Mapes Conn. Valley Special. The Mapes Corn Manure. The Mapes General Truck Manure. The Mapes General Tobacco Manure.	6-4-7 3-8-3 5-6-5 5-4-5	East Granby Hartford Hartford East Granby	2.06 0.10 0.00 1.48	0.00 1.84 3.50 0.04	0.20 0.00 0.23 0.16	2.83 0.50 0.58 2.80	5.09 2.44 4.31 4.48	6.19 2.97 5.24 5.45	1.28 1.18 1.18 1.35	6.03 10.40 8.78 5.40	4.75 9.22 7.60 4.05	0.54 3.46 5.48 0.36	7.67 3.69 5.48 5.34	9247 8972 9027 9246

TABLE XII. ANALYSES OF MIXED FERTILIZERS

	TABLE ATT. ANALYSES OF MIXED FERTILIZERS				Nitrogen.						osphoric A	Lio A			
					-	l	1.	1	int	Pno	I Sphorte	Tera.	P0	tash.	
Station No.	Manufacturer and Brand.	Grade.	Place of Sampling.	In Nitrates.	In ammonia.	Organic water-soluble,	Organic water-insoluble.	Total.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called "Available".	As muriate.	Total.	Station No.
	Sampled by Station: Mapes Formula and Peruvian Guano Co.,			%	%	%	%	%	%	%	%	%	%	%	%
8966 9241 8967 9240 9242 9248 9486 9251 9026 8968	New York  The Mapes General Use Manure. The Mapes Onion Manure. The Mapes Potato Manure. The Mapes Special Trucker "S.P." The Mapes Special Trucker. The Mapes Tobacco Ash Constituents. The Mapes Tobacco Ash and Starter. The Mapes Tobacco Manure Wrapper Brand. The Mapes Tobacco Starter Improved. The Mapes Top Dresser.	3-6-4 4-6-4 4-7-5 5-8-7 5-8-7 1-4-15 4-6-15 7.5-2-10.5 5-6-1 10-4-2	Windsor Locks. Hartford Windsor Locks. Hartford Bloomfield East Granby East Granby Windsor Locks. Windsor Locks.	$\begin{array}{c} 0.00 \\ 0.54 \\ 0.00 \\ 0.00 \\ 0.00 \\ 0.01 \\ 3.10 \\ 2.97 \\ 1.51 \\ 2.43 \end{array}$	2.40 2.40 3.00 3.36 3.07 0.07 0.00 0.16 1.70 5.89	0.11 0.18 0.00 0.21 0.17 0.18 0.51 0.48 0.29 0.19	0.53 0.37 0.43 0.54 0.53 0.93 0.32 3.15 1.02 0.34	3.04 3.49 3.43 4.11 3.77 1.19 3.93 6.76 4.52 8.85	3.70 4.24 4.17 5.00 4.58 1.45 4.78 8.22 5.50 10.76	0.83 0.75 0.93 1.03 1.20 1.38 1.70 1.40 3.28 0.40	7.38 8.70 9.18 11.30 10.30 6.45 7.33 5.20 10.03 6.13	6.55 7.95 8.25 10.27 9.10 5.07 5.63 3.80 6.75 5.73	0.86	4.91 4.48 4.91 7.38 6.93 15.67 15.12 11.51 1.61 2.42	8966 9241 8967 9240 9242 9248 9486 9251 9026 8968
9249 9250 9490	A. G. Markham & Co., Springfield, Mass. 4-8-4. 4-6-10. 5-8-7.	4-8-4 4-6-10 5-8-7	Mansfield Depot Mansfield Depot Stafford Springs	$0.43 \\ 0.43 \\ 0.71$	1.87 1.89 2.13	0.49 0.67 0.74	0.65 0.63 0.62	3.44 3.62 4.20	4.18 4.40 5.11	0.85 0.65 1.13	9.00 6.78 9.30	8.15 6.13 8.17	4.24 10.39 7.30	4.24 10.39 7.30	9249 9250 9490
9260 8971 8970 9259	New England Fertilizer Co., Boston, Mass.  New England Complete Manure 4-6-10  New England Corn Phosphate 2-10-2  New England Market Garden Manure 5-8-7  New England Potato and Vegetable Manure	4-6-10 2-10-2 5-8-7 4-8-4	Meriden Rockville Rockville	0.26 0.01 0.44 0.34	1.86 1.40 2.50 1.90	0.58 0.33 0.60 0.50	0.64 0.36 0.66 0.60	3.34 2.10 4.20 3.34	4.06 2.55 5.11 4.06	0.58 1.40 0.30	6.63 11.55 8.88 9.85	6.05 10.15 8.58 8.92	10.45 2.25 6.90 4.35	10.45 2.25 6.90 4.35	9260 8971 8970
8969 9265 9599	4-8-4.  New England Super, A High Grade Fertilizer for all Crops.  New England Tobacco Manure 5-3-5.  New England Tobacco Manure 7-3-7.	3-8-4 5-3-5 7-3-7	Rockville Warehouse Point	0.04 1.37 1.59	1.50 0.14 0.51	0.36 0.23 0.40	0.58 2.60 3.28	2.48 4.34 5.78	3.02 5.28 7.03	1.00 0.45 0.33	9.05 4.10 4.55	8.05 3.65 4.22	3.78 0.41 0.20	3.78 5.17 7.64	8969 9265 9599
9483 9023 9489 9591 9024 9484 9025	Olds & Whipple, Inc.,  Hartford, Conn.  O & W Blue Label Tobacco Fertilizer.  O & W Complete Market Garden.  O & W Complete Tobacco Fertilizer.  O & W Grain and General Crop Fertilizer.  O & W Grass Fertilizer.  O & W H. G. Starter and Potash.  O & W High Grade Vegetable and Potato	2-9-3 6-6-4 5-4-15	So. Manchester. Hockanum. So. Manchester. So. Manchester. Burnside	1.12 0.64 1.11 0.54 2.18 0.76 1.09	0.06 1.76 0.07 0.90 2.50 0.88 2.72	0.46 0.35 0.51 0.03 0.33 0.40 0.16	3.42 0.73 2.75 0.53 0.18 2.04 0.53	5.06 3.48 4.44 2.00 5.19 4.08 4.50	6.15 4.23 5.40 2.43 6.31 4.96 5.47	0.25 1.08 0.25 0.88 0.75 0.45	4.43 9.40 4.05 10.53 7.35 5.13 8.93	4.18 8.32 3.80 9.65 6.60 4.68 8.28	0.49 4.16 0.54 3.33 4.00 0.98 8.76	6.38 4.16 5.52 3.33 4.00 16.36 8.76	9483 9023 9489 9591 9024 9498

TABLE XII. ANALYSES OF MIXED FERTILIZERS

1	THE THE PROPERTY OF WHALE	1.000000	
Station No.	Manufacturer and Brand.	Grade.	Place of Sampling,
9481 9487 9261	Sampled by Station: Parmenter & Polsey Fertilizer Co., Boston, Mass.  P & P Top Dressing	7-6-5 5-8-7 4-6-10	Wallingford Plainville Wallingford
9263 9262	Piedmont Mt. Airy Guano Co., Baltimore, Md. Harvest Brand 4-8-4	4-8-4 5-8-7	Plantsville
9558 9559	Frank S. Platt Co., New Haven, Conn. Platt's Concentrated Lawn Fertilizer Platco Special 5-8-7	16-5-5 5-8-7	New Haven
9500 9480	Rackliffe Bros. Co., Inc., New Britain, Conn. Rackliffe Brand Corn Fertilizer 4-8-4 Rackliffe Brand Potato and Spec. Vegetable 5-8-7	4-8-4 5-8-7	New Britain
	The Rogers and Hubbard Co.,		
9034 9617 9035 9620 9255	Portland, Conn. 4-8-4. 4-8-4. 5-8-7. 5-8-7. Hubbard's Bone Base Fertilizer for Seeding	4-8-4 4-8-4 5-8-7 5-8-7	Hartford Wallingford Hartford Branford
9256	Down	3-5-6 10-3-8	Portland Willimantic
9258 9297 9624 9301 9045 9506 9042 9498 9041 9300	Hubbard's Bone Base Soluble Corn and General Crops Manure.  Hubbard's Bone Base Soluble Potato Manure.  Hubbard's Bone Base Soluble Potato Manure.  Hubbard's Bone Base Soluble Tobacco Manure.  R & H All Soils All Crops Fertilizer	3-8-6 6-8-5 6-8-5 6-8-10 4-10-4 5-3-5 1-10-3 3-8-10 2-10-4 6-3-5	Branford. Branford. Glastonbury. New Milford. New Milford. Granby. Willimantic. Branford. Willimantic. Suffield.

/		Nitrogen	<b>.</b>			Pho	sphoric A	cid.	Pot	tash.	
In Mitrates.	In Ammonia.	Organic water-soluble.	Organic water-insoluble.	Total.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called "Available".	As muriate.	Total.	Station No.
%	%	%	%	%	%	%	%	%	%	%	%
0.06 0.25 0.34	5.18 2.65 1.82	0.29 0.81 0.59	0.08 0.52 0.56	5.61 4.23 3.31	6.82 5.14 4.02	0.35 0.78 0.65	6.53 9.05 6.55	6.18 8.27 5.90	4.66 7.15 10.19	4.66 7.15 10.19	9481 9487 9261
0.53 0.04	1.63 3.08	0.42 0.47	0.66 0.36	3.24 3.95	3.94 4.80	1.28 0.58	9.20 8.58	7.92 8.00	4.11 7.22	4.11 7.22	9263 9262
0.00	13.57 2.95	0.11	0.48 0.62	14.16 4.44	17.22 5.40	0.03 0.98	5.15 9.65	5.12 8.67	0.28 7.17	5.18 7.17	9558 9559
0.32	2.19	0.46	0.59	3.56	4.33	1.30	9.55	8.25	4.05	4.05	9500
1.54	1.98	0.44	0.16	4.12	5.01	0.50	9.05	8.55	5.57	6.41	9480
0.17 0.86 0.07 0.05	2.09 1.50 2.93 2.93	1.01 0.63 1.08 1.07	0.18 0.32 0.13 0.13	3.45 3.31 4.21 4.18	4.19 4.02 5.12 5.08	1.05 0.38 0.70 0.63	8.65 8.53 8.40 8.45	7.60 8.15 7.70 7.82	4.26 4.26 7.14 7.00	4.26 4.26 7.14 7.00	9034 9617 9035 9620
0.00 4.52	0.80 0.14	0.40 3.26	1.27 0.32	2.47 8.24	3.00 10.02	6.78 3.28	12.63 8.43	5.85 5.15	6.49 2.88	6.49 8.70	9258 9256
0.07 1.08 0.86 0.79 0.09 1.23 0.12 0.06 0.12 1.43	0.95 2.16 2.08 2.24 2.04 0.05 0.23 1.27 0.63 0.09	0.93 0.83 1.37 1.19 0.69 0.23 0.36 0.67 0.66 0.66	0.58 0.95 0.66 0.57 0.72 2.62 0.44 0.70 0.56 2.86	2.53 5.02 4.97 4.79 3.54 4.13 1.15 2.70 1.97 5.04	3.08 6.10 6.04 5.82 4.30 5.02 1.40 3.28 2.40 6.13	1.45 1.98 1.30 1.13 0.65 0.39 0.70 0.78 0.60 0.30	10.30 10.64 10.25 10.38 10.63 3.85 11.50 9.38 10.33 3.50	8.85 8.66 8.95 9.25 9.98 3.46 10.80 8.60 9.73 3.20	5.98 0.73 0.70 2.75 4.20 0.64 3.36 10.53 4.93 0.46	5.98 4.49 .4.95 9.92 4.20 5.01 3.36 10.53 4.93 5.43	9258 9297 9624 9301 9045 9506 9042 9498 9041 9300

78

TABLE XII. ANALYSES OF MIXED FERTILIZERS

	Manufacturer and Brand	Grade.	Place of Sampling.
Station No.			
9513 9294 9296 9293 9295 9290 9299 9614 9516	Sampled by Station:  F. S. Royster Guano Co.,  Baltimore, Md.  Royster's Conn. Tobacco Guano.  Royster's Gem Guano.  Royster's Ouality Trucker.  Royster's Top Dresser.  Royster's Truckers Delight.  Royster's 5% Truck Guano.  Royster's Valley Tobacco Guano.  Royster's Wrapper Brand 7-3-7.	5-3-5 3-10-6 2-12-4 4-8-7 8-6-6 4-8-4 5-8-7 5-4-5 7-3-7	Granby
9588 9601 9560	M. I. Shoemaker & Co., Philadelphia, Pa. Potato Special	5-8-7 4-8-5 3-10-3	New Milford New Milford Granby
9510 9517 9515	Springfield Rendering Co., Springfield, Mass. Springfield, 3-8-4. Springfield 4-8-4. Springfield 5-8-7.	3-8-4 4-8-4 5-8-7	Suffield
9065	Swift & Co., Baltimore, Md: Vigoro	4-12-4	Waterbury
9605	Synthetic Nitrogen Products Co., New York. Nitrophoska		Southington
9342 9585 9044 9345 9584 9586 9344 9581	I. P. Thomas & Sons, Philadelphia, Pa.  Economy Fertilizer 3-12-3 I. P. Thomas 5-8-7 Long Island Special 4-8-7 Thomas Tobacco Grower 5-4-5 Tip Top Superphosphate 3-10-6 7% Guano 7-6-5 Truckers High Grade Guano 4-8-4 Victor Potash Fertilizer 2-8-5	3-8-7 4-8-7 5-4-5 3-10-6 7-6-5 4-8-4	Milford

CONTAINING NITROGEN, PHOSPHORIC ACID AND POTASH—Continued.

-		Nitrog	gen.			1	Phosphori	ic Acid.	Por	tash.	
In Nitrates.	In ammonia.	Organic water-soluble.	Organic water-insoluble.	Total.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called "Available."	As muriate.	Total.	Station No.
%	%	%	%	%	%	%	%	%	%	%	%
$\begin{array}{c} 0.44 \\ 0.00 \\ 0.06 \\ 0.00 \\ 2.49 \\ 0.00 \\ 0.00 \\ 0.32 \\ 0.60 \end{array}$	0.93 1.52 1.16 2.12 3.34 2.10 2.46 0.88 1.20	0.22 0.33 0.28 0.45 0.39 0.56 0.81 0.29 0.14	2.52 0.62 0.34 0.69 0.33 0.70 0.80 2.86 3.72	4.11 2.47 1.84 3.26 6.55 3.36 4.07 4.35 5.66	5.00 3.00 2.24 3.96 7.96 4.09 4.95 5.29 6.88	0.20 0.83 0.70 0.63 0.20 0.95 0.78 0.65 0.08	4.08 10.73 13.82 8.60 6.85 9.08 9.68 4.73 3.65	3.88 9.90 13.12 7.97 6.65 8.13 8.90 4.08 3.57	0.90 6.24 4.06 7.13 6.33 4.30 6.66 1.29 0.47	6.09 6.24 4.06 7.13 6.33 4.30 6.66 5.11 7.29	9513 9294 9296 9293 9295 9290 9299 9614 9516
0.33 0.00 0.08	2.52 2.10 1.50	0.32 0.12 0.17	1.07 1.32 0.90	4.24 3.54 2.65	5.15 4.30 3.22	1.88 0.60 1.20	9.45 9.00 10.68	7.57 8.40 9.48	5.90 0.28 0.27	5.90 5.45 3.17	9588 9601 9560
0.03 0.88 1.20	1.45 1.38 1.68	0.48 0.54 0.74	0.60 0.78 0.54	2.56 3.58 4.16	3.11 4.35 5.06	0.79 0.73 1.15	9.18 8.75 10.40	8.39 8.02 9.25	4.12 4.09 7.17	4.12 4.09 7.17	9510 9517 9515
0.32	2.74	0.05	0.39	3.50	4.26	0.28	13.00	12.72	4.28	4.28	9065
0.00	13.30	1.44	0.26	15.00	18.24	0.00	30.30	30.30	15.21	15.21	9605
0.08 0.46 0.30 0.18 0.36 0.34 0.46 0.08	1.96 2.96 2.29 0.90 1.64 4.06 2.08 1.18	0.23 0.31 0.23 0.50 0.27 0.42 0.35 0.12	0.44 0.69 0.58 2.52 0.39 0.72 0.71 0.40	2.71 4.42 3.40 4.10 2.66 5.54 3.60 1.78	3.29 5.37 4.13 4.98 3.23 6.74 4.38 2.16	$   \begin{array}{c c}     1.05 \\     0.78 \\     3.20   \end{array} $	$ \begin{array}{c c} 9.90 \\ 9.45 \\ 12.00 \end{array} $	12.17 8.85 8.67 8.80 10.91 7.05 8.87 8.33	2.35 6.43 7.05 0.72 4.98 4.48 3.38 4.19	3.25 6.43 7.05 5.85 6.16 5.29 3.78 4.68	9342 9585 9044 9345 9584 9586 9345 9581

TABLE XII. ANALYSES OF MIXED FERTILIZERS

	TABLE XII. ANALYSES OF MIXED	FERTILIZ	ZERS		CONT	AINING	NITRO	GEN, PI	ноѕрної	RIC ACI	D AND	POTASH	-Cont	inued.	
						Nitroger	1.		t t	Phos	phoric A	cid.	Pot	ash.	
Station No.	Manufacturer and Brand.	Grade.	Place of Sampling,	n Nitrates.	In Ammonia.	Organic water-soluble.	Organic water-insoluble.	Total.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called "Available".	As muriate.	Total.	Station No.
9339 9349 9348 9340	Sampled by Station: Triton Oil and Fertilizer Co., New York. Triton 4-8-4 Fertilizer. Triton 4-8-7 Fertilizer. Triton 5-8-5 Fertilizer. Triton 5-8-7 Fertilizer.	4-8-4 4-8-7 5-8-5 5-8-7	New London New London	% 0.73 0.73 0.85 1.26	% 1.30 1.53 1.98 1.99	% 0.48 0.50 0.56 0.46	% 0.64 0.59 0.76 0.92	% 3.15 3.35 4.15 4.63	% 3.83 4.07 5.05 5.63	% 0.37 0.30 0.48 0.23	% 8.52 9.03 9.50 9.28	% 8.15 8.73 9.02 9.05	% 4.69 7.34 5.02 7.21	% 4.69 7.34 5.02 7.21	% 9339 9349 9348 9340
9186 9188 9185 9187 8847 8851 8848 9184	U. S. Guano Co., Baltimore, Md.  Accomac Peninsular King. General Use Guano. Jersey Special. Royal Potato Grower. Standard United States Evergreen Fish Guano. Standard United States Fish, Bone and Potash. Standard United States Old Fertility. Standard United States 5-10-5.	5-8-5 2-12-4 4-8-10 4-8-7 4-8-4 5-8-7 2-8-3 5-10-5	No. Haven No. Haven No. Haven Norwalk Southport Norwalk	1,28 0,70 2,52 1,30 0,00 0,22 0,18 0,60	0.34 1.50 0.22 1.22 2.84 3.50 1.00 2.72	2.18 0.12 0.29 0.28 0.06 0.01 0.04 0.36	0.15 0.30 0.22 0.52 0.52 0.54 0.72 0.22	3.95 2.62 3.25 3.32 3.42 4.27 1.94 3.90	4.80 3.19 3.95 4.04 4.16 5.19 2.36 4.74	0.55 0.68 0.19 0.83 0.40 0.48 0.73 0.48	8.77 11.12 8.63 9.07 8.85 8.85 9.85 11.00	8.22 10.44 8.44 8.24 8.45 8.37 9.12 10.52	4.13 4.03 10.16 6.73 3.93 6.43 3.08 4.76	4.86 4.32 10.16 6.73 4.81 6.43 3.08 4.76	9186 9188 9185 9187 8847 8851 8848 9184
9350 8881 8880 8874	Virginia-Carolina Chemical Co., New York.  Bloomaid	6-10-4 5-8-7 2-9-3 4-8-4	New Britain	1.11 0.00 0.15 0.93	2.75 2.78 1.07 2.20	0.22 0.60 0.39 0.00	1.16 0.69 0.50 0.34	5.24 4.07 2.11 3.47	6.37 4.95 2.57 4.22	0.65 1.10 1.08 0.73	11.45 9.30 9.93 8.30	10.80 8.20 8.85 7.57	0.76 6.66 4.54 4.29	4.96 6.66 4.54 4.29	9350 8881 8880 8874
9347 9346 9359 9355	Wilcox Fertilizer Co., Mystic, Conn.  Wilcox Corn Special 3-10-4. Wilcox H. G. Fish and Potash 4-8-4. Wilcox Potato and Vegetable Phospate 5-8-7 Wilcox 7-6-5 Top Dresser.	3-10-4 4-8-4 5-8-7 7-6-5	Willimantic Willimantic Willimantic Mystic,	1.34 1.73 1.26 2.55	0.10 $0.26$ $1.36$ $1.71$	0.55 0.91 0.43 0.72	$0.70 \\ 0.78 \\ 0.71 \\ 0.32$	2.69 3.68 3.76 5.30	3.27 4.47 4.57 6.44	0.73 0.58 0.54 0.20	10.98 8.16 9.00 7.05	10.25 7.58 8.46 6.85	3.40 3.90 7.40 2.98	4.11 4.66 7.40 5.25	9347 9346 9359 9355
9356 56 9040 58 9352 57 9043	Worcester Rendering Co., Auburn, Mass.  Prosperity Brand Complete Dressing.  Prosperity Brand Corn and Grain Fertilizer.  Prosperity Brand Corn and Grain Fertilizer.  Prosperity Brand Market Garden Fertilizer.  Prosperity Brand Market Garden Fertilizer.  Prosperity Brand Potato and Vegetable Fertilizer.	6-6-4 6-6-4 2-10-2 2-10-2 5-8-7 5-8-7 4-8-4	Groton	0.67 1.21 0.77 0.82	2.83 0.76 2.23 	0.49 0.58 0.67 0.64	0.50 0.35 0.64 	4.49 4.87 2.90 2.29 4.31 4.22 3.26	5.46 5.92 3.53 2.78 5.24 5.13	0.83 0.64 0.95 1.05 0.95 0.84 1.10	7.30 7.07 10.70 12.04 8.48 7.88 9.43	6.47 6.43 9.75 10.99 7.53 7.04 8.33	4.01 2.07 7.75  3.90	4.01 4.34 2.07 2.50 7.75 6.85 3.90	9356 56 9040 58 9352 57 9043

TABLE XII. ANALYSES OF MIXED FERTILIZERS

The state of the latest section in			
Station No.	Manufacturer and Brand.	Grade.	Place of Sampling,
9527	Sampled by Purchaser: Berkshire Chemical Co., Bridgeport, Conn. Complete Tobacco Fertilizer	5-3-5	Suffield
	Lowell Fertilizer Co.,		
8085	Boston, Mass.	5-10-5	Storrs
8916	The Rogers & Hubbard Co., Portland, Conn. Tunacre Fertilizer		Middletown
	Swift & Co.,		
9018	Vigoro	4-12-4	Norwich
	Standard Wholesale Phosphate & Acid Works,		
8254 8255 9052	Baltimore, Md. 5-8-7 Fertilizer. 4-8-4 Fertilizer. Standard United States Buyers Mixture Fertil-	5-8-7 4-8-4	New Britain New Britain
9053	izer Standard United States Buyers Mixture Fertil-		Andover
9058 9492 9252 9253	Standard United States Fish, Bone and Potash.	5-8-7 5-8-7 5-8-7 5-8-7	Andover
9119 9120	Wilcox Fertilizer Co., Mystic, Conn. Wilcox Fertilizer 4-8-4 Wilcox Fertilizer 5-8-7	4-8-4 5-8-7	Norwich
9055	Wilcox Fertilizer 7-6-5.	7-6-5	Andover

		Nitrogen				Ph	osphoric .	Acid.	Pot	ash.	
In Nitrates.	In ammonia.	Organic water-soluble.	Organic water-insoluble.	Total.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called "Available".	As muriate.	Total.	Station No.
%	%	%	%	% 4.61	% 5.60	% 0.20	% 3.65	% 3.45	% 0.93	% 4.78	% 9527
			••••	4.22	5.13	0.97	11.19	10.22		5.27	8085
	••••			8.15	9.91	0.18	4.73	4.55	••••	10.54	8916
	••••	••••	·	3.42	4.16	0.33	13.05	12.72	••••	4.07	9018
****	••••		••••	4.38 3.38	5.33 4.11	0.84 0.71	8.88 8.94	8.04 8.23	5.07 4.21	5.07 4.21	8254 8255
2004 2004 2004 2004 2004 2004	••••			5.20 4.25 4.14 4.00 4.37 4.25	5.17 5.03 4.86 5.31 5.17	1.13 0.60 0.43  0.55 0.48	8.28 11.00 8.50 9.05 8.65 8.50	7.15 10.40 8.07 8.10 8.12	7.04 4.31 6.66 6.13 6.45 6.43	7.04 4.94 7.01 6.13 6.79 6.60	9052 9053 9058 9492 9252 9253
	••••			3.37 4.31 5.66	4.10 5.24 6.88	0.18 0.05 0.23	9.03 8.53 6.88	8.95 8.48 6.65	3.18 7.45 4.59	4.24 7.45 5.61	9119 9120 9055

85

#### SPECIAL MIXTURES AND HOME MIXTURES.

Fifty samples of mixed fertilizers have been examined for individuals such samples for the most part being drawn by the persons interested. The Station is responsible only for the analysis of these samples as received.

Analyses are given in Table XVI.

#### VIII. MISCELLANEOUS FERTILIZERS, AMEND-MENTS, WASTE PRODUCTS, ETC.

#### SHEEP MANURE, ETC.

Twenty-six samples of sheep manure and other farm manures were analyzed. The identity of two of these products is uncertain. Nos. 8959 and 9495 did not have the appearance of sheep manure and did not contain appreciable amounts of phosphoric acid or potash.

Analyses are given in Table XVII.

#### LIME.

Lime is not classed as a fertilizer in this State but seven samples have been analyzed for purchasers. The results are given in Table XVIII.

#### OTHER MISCELLANEOUS MATERIALS.

Other miscellaneous materials, 35 in number, have been examined during the year and the results, with comments where necessary, are given in Table XIX.

#### COLLABORATIVE WORK.

The laboratory has collaborated in the check meal program of the American Oil Chemists' Society, and in the check fertilizer program sponsored by the F. S. Royster Guano Co. Mr. Nolan has collaborated with the Referee of the Association of Official Agricultural Chemists upon methods for determining activity of insoluble organic nitrogen in fertilizers. This work has involved the examination of 45 samples.

TABLE XVI. ANALYSES OF SPECIAL AND HOME MIXTURES.

,	.oV noitet2	9667 9596 9534 9534 9603 9770 9770 9770
ısh.	Total.	% 6.75 7.35 7.35 8.68 8.05 7.19 7.19
Potash	.91sirum zA	% 11.28 0.40 0.40 0.40 6.07 7.36 7.36 7.19
cid.	So-called "Available".	% 8.18 8.18 8.23 7.05 6.27 10.20 6.38
Phosphoric Acid	.lstoT	8.43 8.43 8.73 8.73 8.35 8.58 8.58 7.48 7.48
Phos	Citrate-insoluble.	% 0.25 0.25 0.08 0.15 0.10 0.10 0.13 1.10
4	Ammonia equivalento to total nitrogen.	% 4.7 4.4 6.8 8.2 4.4 6.0 6.0 6.0 6.0 6.0 6.0 6.0 6.0 6.0 6.0
	Total Nitrogen.	% 65.33 % 65.33 % 65.33 % 75.22 % 75.23 % 7
	Place of Sampling.	John B. Laydon, North Haven. Chas. Maag, Manchester. E. N. Austin, Suffield. John Luginbuhl, Ellington. John Luginbuhl, Ellington. John Luginbuhl, Ellington. W. A. Howard, Woodstock. H. P. Joy, Woodstock. S. D. Woodruff & Sons, Orange
	Manufacturer or Brand.	Sampled by Station. Sanderson's Formula B (Old Stock) Special Mixture for Tobacco Home Mixture for Corn Home Mixture for Corn Home Mixture for Grass Wilcox 4-8-4 Fertilizer Wilcox 5-8-7 Fertilizer Wilcox 5-10-5 Fertilizer Woodruff Home Mixture.
	Station No.	9667 9596 9534 9534 9603 9604 9771 9770 9770

				4	Pho	sphoric A	Leid.	Pot	ash.	
Station No.	Manufacturer or Brand.	Place of Sampling.	Total Nitrogen.	Ammonia equivalent to total nitrogen.	Citrate-insoluble.	Total.	So-called "." Available".	As muriate.	Total.	Station No.
	Sampled by Purchaser.		%	%	%	%	%	%	%	
9013	Special Mixture Fertilizer	The Allied Tobacco Co., Hartford	6.41	7.79	0.33	1.85	1.52		11.87	9013
8263	Formula A	American Sumatra Tobacco Co., Bloomfield	6.00	7.29	0.65	5.24	4.59		4.76	8263
8337	Formula A	American Sumatra Tobacco Co.,								
8262	Formula B	Bloomfield	6.02	7.32	0.48	4.69	4.21		5.08	8337
8302	Formula B	Bloomfield	5.36	6.52	1.11	6.30	5.19		6.54	8262
		Bloomfield	5.46	6.64	1.06	6.05	4.99	0.25	6.22	8302
8343	Formula C	American Sumatra Tobacco Co., Bloomfield	5.90	7.17	0.82	4.73	3.91		3.91	8343
8458	Formula E	American Sumatra Tobacco Co., Bloomfield	4.98	6.05	0.45	3.48	3.03		5.79	8458
8816	Formula F	American Sumatra Tobacco Co.,								
8920	Formula G	Bloomfield American Sumatra Tobacco Co.,	5.70	6.93	0.28	5.23	4.95	• • • •	4.56	8816
		Bloomfield	5.36	6.52	0.38	3.63	3.25		5.79	8920
8344	Drill Fertilizer	American Sumatra Tobacco Co., Bloomfield	9.77	11.88	0.22	2.04	1.82		1.07	8344
8006	Fall Top Bed Fertilizer	American Sumatra Tobacco Co., Bloomfield	5.44	6.61	0.65	4.72	4.07		5.80	8006

TABLE XV ANALYSES OF SPECIAL AND HOME MIXTURES.

				t to	Phos	phoric A	cid.	Pota	ash.	
Station No.	Manufacturer or Brand.	Place of Sampling.	Total Nitrogen.	Ammonia equivalent total nitrogen.	Citrate-insoluble.	Total.	So-called ".	As muriate.	Total.	Station No.
8692 9629 9351	Sampled by Purchaser. Apothecaries Hall Co., Waterbury, Car No. 168774. Home Mixture Nature's Own Fertilizer.,	Hatheway & Steane, Inc., Hartford	% 4.12 4.78	% 5.01 5.81	% 0.30 	% 8.58 4.60	% 8.28 ····	%	% 7.05 5.45	8692 9629
9463 9048	Bestivall Mfg. Co., Philadelphia, Pa Special Mixture Fertilizer Fertilizer	Mfr.'s Sample	7.50 6.37 4.68	9.12 7.74 5.69	0.43 0.50	1.80 4.53 10.53	4.10 10.03	0.62	0.99 6.91 6.95	9351 9463 9048
9572 9654 9535	Eastern States Farmers' Exchange, Springfield, Mass. Special Mixture Fertilizer Home Mixed Fertilizer	Leslie W. Newberry, So. Windsor J. E. Phelps, Suffield The Hartman Tobacco Co., Hartford	6.06 4.86 6.96	7.37 5.91 8.46	0.05 1.90 0.40	4.73 6.78 3.44	4.68 4.88 3.04	$0.32 \\ 0.52 \\ 0.21$	7.43 5.04 6.47	9572 9654 9538
7786	Platt's Concentrated Lawn Fertz., Frank S. Platt Co. New Haven	Mfr's Sample	15.86	19.28						778

CONNECTICUT EXPERIMENT STATION

BULLETIN 296

- 540 - 28 P

_										
				t to	Pho	sphóric A	Acid.	Pot	ash.	
Station No.	Manufacturer or Brand.	Place of Sampling.	Total nitrogen.	Ammonia equivalent total nitrogen.	Citrate-insoluble.	Total.	So-called "Available".	As muriate.	Total.	Station No.
9395	Sampled by Purchaser. Special Mixture Fertilizer		%	%	%	. %	%	%	%	
9396	No. 3 Special Mixture Fertilizer	H. E. Wells, Warehouse Point	6.25	7.60	0.10	7.30	7.20	1.04	15.89	9395
	No. 4	H. E. Wells, Warehouse Point.	6.22	7.56	0.15	7.38	7.23	0.80	16.43	9396
9397	Special Mixture Fertilizer No. 5	H. E. Wells, Warehouse Point	5.00	6.08	0.55	6.00	5.45	6.67	6.85	9397
9398	Special Mixture Fertilizer No. 6	H. E. Wells, Warehouse Point	3.60	4.38	0.83	8.13	7.30			
8297	Complete Fertilizer Grower	The Rogers & Hubbard Co						5.25	5.43	9398
8376	"Climax" Fertilizer	Portland	4.91	5.97	0.32	3.41	3.09		5.19	8297
9716	5-8-7 Fertilizer No. 1	Portland The Rogers & Hubbard Co.,	4.27	5.19	0.50	4.02	3.52		5.08	9376
9717	5-8-7 Fertilizer No. 2	Portland	4.18	5.08	0.33	8.70	8.37	7.27	7.27	9716
9666	Special Mixture Fertilizer	Portland Paul P. Rostek, Melrose	4.21 6.04	5.12 7.34	0.35	8.50 4.05	8,15	7.28 1.20	7.28 6.55	9717 9666
										104

TABLE XV. ANALYSES OF SPECIAL AND HOME MIXTURES-Concluded.

				t to	Phosph	oric Acid		Pota	ash.	
Station No.	Manufacturer or Brand.	Place of Sampling.		Ammonia equivalent total nitrogen.	Citrate-insoluble.	Total.	So-called "Available".	As muriate.	Total.	Station No.
9709	Sampled by Purchaser. Special Home Mixture (Ro-	Consolidated Cigar Co., Hart-	%	%	%	%	%	%	%	
	senblum)	ford	5.83	7.09	0.68	3.65	2.97	0.74	6.35	9709
9710	Special Home Mixture (Myers)	Consolidated Cigar Co., Hart- ford.	0.10	F 40						
9711	Special Home Mixture	Consolidated Cigar Co., Hart-	6.16	7.49	0.45	4.40	3.95	0.03	4.43	9710
9712	(Shaker) Special Home Mixture	ford	6.40	7.78	0.55	4.40	3.85	0.04	4.21	9711
	Special Home Mixture (Winton)	Consolidated Cigar Co., Hart- ford.	5.92	7.20	0.90	3.95	3.05	0.33	6.19	9712
9713	Special Home Mixture	Consolidated Cigar Co., Hart-								
9714	(Kanter) Special Home Mixture	ford	6.02	7.32	0.38	3.65	3.27	0.49	4.12	9713
9715	(Huntting)	ford	6.14	7.47	0.58	4.58	4.00	0.11	4.47	9714
9119	Special Home Mixture (Barton)	Consolidated Cigar Co., Hart- ford	5.80	7.05	0.40	4.18	3.78	0.24	6.21	9715
7997 7998	Fertilizer No. 1	Truman H. Hale, Gildersleeve	5.62	6.83	0.97	4.39	3.42	0.21	5.39	7997
8096	Fertilizer No. 2	Truman H. Hale, Gildersleeve S. D. Woodruff & Sons, Orange	5.78	7.03 1.76	$0.91 \\ 1.35$	4.27 8.30	3.36 6.95		5.72	7998
8341	Clay's Fertilizer	J. P. Johnson, Entomology Dept.	4.87	5.92	6.65	9.66	3.01		0.16	8341

TABLE XVII. ANALYSES OF

		· · ·
Station No.	Manufacturer or Brand.	Place of Sampling.
	g .,, ,, , , , , ,	
8870	A. A. C. Pulverized Sheep and Goat Manure. American Agricultural Chemical Co., New York	Bristol Grain & Supply Co.,
9189	Sheep Manure. Apothecaries Hall Co.,	F. T. Blish Hardware Co., So
9028	Armour's Sheep and Goat Manure 1½-1-2. Armour Fertilizer Works, New York	F. A. Bartlett Tree Expert Co.,
8959	Berkshire Sheep Manure. Berkshire Chemical Co., Bridgeport	Eldredge Hardware Co., Nor.
9540	Par Plus Brand Pulverized Sheep Manure. A. H. Case & Co., Buffalo, N. Y	S. D. Woodruff & Son, Orange
8886	Corenco Sheep Manure. Consolidated Rendering Co., Boston, Mass	Cheshire Reformatory, Cheshire
8999	Davey Shredded Cattle Manure, Davey Tree Expert Co., Kent, Ohio	R. E. Landis, Sound Beach
9495	Sheep Manure. Meech & Stoddard, Inc., Middletown	Meech & Stoddard, Middletown
9021	"Sheep's Head" Pulverized Sheep Manure. Natural Guano Co., Aurora, Ill	Cadwell & Jones, Hartford
9479	Favorite Brand Sheep Manure. Olds & Whipple, Inc., Hartford	Sampled at factory
9257	Groz-It Brand (Pulverized Sheep Manure). Pacific Manure & Fertilizer Co., San	bampied at factory
8903	Francisco, Cal	F. F. Hitchcock Co., Woodbury
8904	Poultry Manure Co., Chicago, Ill	Lightbourn & Pond, New Haven
8905	Premier Brand Sheep Manure. Premier Premier Poultry Manure Co., Chicago, Ill.	Lightbourn & Pond, New Haven Comstock-Ferre & Co., Weth-
	Wizard Brand Pulverized Sheep Manure. Pulverized Manure Co., Chicago, III	ersfield
8989	Wizard Brand Pulverized Sheep Manure. Pulverized Manure Co., Chicago, III	H. E. Meeker, Danbury
9264	Wizard Cattle Manure. Pulverized Manure Co., Chicago, Ill	S. P. Strople, New Britain
9292	Portland	W. L. Richmond, New Milford.
9496	Royster's Sheep & Goat Manure. F. S. Royster Guano Co., Baltimore, Md	F. B. Newton, Plainville
9583	Sheep and Goat Manure. I. P. Thomas & Son, Philadelphia, Pa	Ira W. Beers, Hamden
7656 9755 9655 9056 8324	Sampled by Purchaser.  Poultry Manure.  Poultry Manure! Chicken Manure Favorite Brand Sheep Manure. Olds & Whipple, Inc., Hartford Chicago Stock Yard Sheep Manure. Pulverized Manure Co., Chicago, Ill.	Z. N. Beach, Wallingford. Charter Bros., Stafford Springs Raymond C. Bugbee, Groton  J. A. Barrasso, Andover.  Geo. C. Meachen, Stratford.
8325	Venezuelan Goat Manure Summer Ferti-	Coo E Monet - Ct - tford.
8008	lizer Co., Baltimore, Md	Geo. E. Meachen, Stratford  Mfr.'s Sample
1	16.93% moisture.	

#### SHEEP MANURE, ETC.

	Ammoni	a equiva-		Phospho	oric acid.		Pot	ash.	
	nitro	ogen.	Avail	lable.	To	otal			
Total Nitrogen.	Found.	Guaranteed.	Found.	Guaranteed.	Found.	Guaranteed,	Found,	Guaranteed.	Station No.
%	%	%	%	%	%	%	%	%	
1.47	1.79	1.50	1.67	0.50	1.90		3.10	2.00	8870
2.93	3.56	1.80			2.93	1.00	2.52	2.00	9189
1.64	1.99	1.50	0.80		1.45	1.00	3.36	2.00	9028
3.02	3.67	3.00				••••			8959
2.29	2.78	2.25			3.08	1.50	1.65	1.50	9540
1.67	2.03	1.50			1.20	0.50	3.26	2.00	8886
1.86	2.26	1.27	1.40	1.00	1.78	1.00	2.71	1.00	8999
3.08	3.74	3.52	0.15	0.23	0.30		0.12	0.12	9495
1.84	2.24	2.73	1.08	1.00	1.93	1.25	2.32	2.00	9021
1.36	1.65	1.65			1.00	0.75	2.75	2.50	9479
1.47	1.79	1.82	0.75	0.75	0.95	1.25	3.04	3.00	9257
4.86	5.91	6.10	2.38	2.50	2.53	2.75	1.20	1.30	8903
1.87	2.27	2.00	0.93	0.80	1.08	1.00	2.09	2.00	8904
1.78	2.16	2.43	1.03	1.25	1.28		2.77	2.00	8905
2.23	2.71	2.43	1.68	1.25	1.78		2.72	2.00	8989
2.04	2.48	2.10	1.30	1.00	1.60		1.58	1.00	9264
2.50	3.04	2.25			1.28		1.72	1.50	9292
1.76	2.14	2.00	1.20	1.00	1.40		3.68	2.00	9496
1.24	1.51	1.50			2.00	1.00	2.24	2.00	9583
2.19 2.58 1.84	2.66 3.14 2.24				3.92 3.05 3.88		1.38 1.30 1.69	••••	7656 9755 9655
1.50	1.82	1.65			1.10	0.75	3.28	2.50	9056
2.03	2.47				1.47		2.35		8324
1.66	2.02				1.28		3.45		8325
4.26	5.18				0.26		0.12		8008

<sup>&</sup>lt;sup>1</sup> 16.93% moisture.

92

TABLE XVIII. ANALYSES OF

BULLETIN 296

No.	Manufacturer or Brand.	Sampled by.
Station No.		
	American Lime & Stone Co., Bellefonte, Pa.	Late the second
8610	Lime	Blue Hills Farm, Wallingford
9051	The Connecticut Agstone Co., Danbury.	J. A. Barrasso, Andover
9222 9254	Grangers Mfg. Co., West Stockbridge, Mass. Grangers Agricultural Limestone Grangers Lime	Mehmel & Sarvi, Plantsville
8081	Lee Lime Corporation, Lee, Mass. Limestone, unground	W. L. Wallace, Canaan
8266 9573	Manufacturer Unknown. Lime Limestone	E. C. P. Sanger, Falls Village John Swanson, Bolton

#### LIMESTONE, ETC.

Chemical Analysis.							Mech	anical An	alysis.		
Lime	(CaO).	Ma; (M	gnesia gO).		oid.						
Found.	Guaranteed.	Found.	Guaranteed.	Total Oxides.	Insoluble in acid.	20 mesh.	40 mesh.	50 mesh.	80 mesh.	100 mesh.	Station No.
%	7%	%	%	%	%	%	%	%	%	%	
71.38		0.94		72.32					••••		8610
39.67		4.51		44.18		93.0	83.0	73.0	58.0	28.0	9051
41.39		6.98		48.37		97.0 100.0	87.0 100.0	75.0 99.8	61.0 93.5	57.0 85.0	9222 9254
55.08		0.41		55.49	•••••				••••		8081
29.85 46.91	••••	20.45 0.33		50.30 47.24	•••••	50.0	38.0	24.0	14.0	i2.0	8266 9573

#### TABLE XIX. MISCELLANEOUS MATERIALS.

			Phosphoric			
No.	Material Material	Nitrogen	acid	Potash		Remarks
		%	%	%		
8041	Ashes from corn cobs		6.98	13.96		
9050	Cottonhull bran	0.48	0.20	1.23		
7972	Cottonseed hulls	0.74	0.11	1.11		
8007	Cottonseed hulls	0.60	0.24	1.14		
8078	Cottonseed hulls	0.79	0.05	1.05		
9571	Cottonseed meal	6.21			Check to	est for another laboratory.
9046	Fertilizer	••••			Identifie	ed as precipitated bone.
9047	Fertilizer	••••				of limestone and burned
0000	B .:::				lime.	
9088	Fertilizer	••••	••••		Identifie	ed as muriate of potash.
9406	Fertilizer				Nitrate	of potash or nitrate of
8042	Doutilines (outtonnes de 1-1)	2.04			soda and	d potash.
8625	Fertilizer (cottonseed cake)	3.94				
0020	"Fertilizer" (Lawn)	••••				ne. CaO 49.12%, MgO
9508	Fertilizer, mixed				2.08%	
9536	Fertilizer mixed		2/5:1:2			ed as acid phosphate.
9990	Fertilizer, mixed	••••	••••	• • • •	Though	t to have caused injury to
					tobacco	plants. No borax found
9687 )	Fertilizer, mixed				and only	y 0.16% of chlorine.
9688	retuinzer, mixeu				Microsc	opic examination only.
9763	Fertilizer, mixed				Mionaga	onio omomination - 1
9724	Fertilizer (tankage)	••••		••••	Microsco	opic examination only.
	Los Commerces ( )	••••		••••	mont	opic examination. Only d bone detected. Chlo-
					ine 0.49	1%, sulphuric acid, none.

#### TABLE XIX. MISCELLANEOUS MATERIALS—Concluded.

			Phosphoric		
No.	Material	Nitrogen	acid	Potash	Remarks
0045	Ti-1 (-1 1 )	%_	%	%	
8845 9734	Fish (whole)	2.75	3.30		Water and oil 72.70%.
9134	Humus	••••	****		Moisture 27.69%, ash 1.71%, organic matter 70.60%.
9302	Lime, hydrated				Passed 300 mesh, 94.1%.
8110	Manganese sulphate				Contained 33.92% MnO.
9689	Mineral				Identified by Dept. of Miner-
					ology, Yale University, as blast furnace slag.
8286	Molasses waste	0.15	0.03	0.40	Analysis on material as sub-
					mitted. Solids 7.51%. Plant
9638	Nova Scotia Plaster				food in a ton worth about \$1.50.
9733	Sheep Manura		••••		CaO 33.25%. MgO none or trace.
	Sheep Manure				Moisture 12.00%, ash 39.35%, organic matter 48.65%.
9049	Soy bean meal	7.95	1.43	2.11	
9063	Sulphur, dusting.:				All passed 300 mesh.
9064	Sulphur, dusting				73% passed 300 mesh.
8043	Tobacco dust	1.36	0.05	1.72	Nicotine 0.56%.
8068	Tobacco dust				Nicotine 1.62%.
7941	Tobacco leaves, deposit on.		••••		Partly silica. Not completely identified.
7699	Tobacco stems	2.21	0.75	5.11	
9219	Tobacco stems	2.58	0.80	9.66	
9491	Volcanic deposit	none	trace	trace	Water 0.71%, ash 95.16%, organic and volatile 4.13%.
					8 1.10 /0.

# Connecticut Agricultural Experiment Station

Nem Haven, Connecticut

# THE EFFECT OF TOPPING AND SUCKERING ON HAVANA SEED TOBACCO

BEING A REPORT

OF THE

TOBACCO SUB-STATION

AT

WINDSOR

#### NOTE TO LIBRARIANS

The separate series, "Tobacco Station Bulletins" has been discontinued, No. 10 being the last. Hereafter, reports of the Tobacco Substation will be included in the regular Station series, this bulletin, No. 297 being the first to so appear.

#### OFFICERS AND STAFF

#### BOARD OF CONTROL

Charles R. Treat, George A. Hopson Wm. L. Slate, <i>Dire</i> Joseph W. Alsop. Elijah Rogers Edward C. Schnei	r, Governor John H. Trumbull, ex-officio President Vice-President
	STAFF.
	E. H. JENKINS, Ph.D., Director Emeritus.
Administration.	WM. L. SLATE, B.Sc., Director and Treasurer. MISS L. M. BRAUTLECHT, Bookkeeber and Librarian. G. E. GRAHAM, In charge of Buildings and Grounds.
Chemistry: Analytical Laboratory,	E. M. BAILEY, PH.D., Chemist in Charge. C. E. SHEPARD OWEN L. NOLAN HARRY J. FISHER, A.B. W. T. MATHIS DAVID C. WALDEN, B.S. FRANK C. SHELDON, Laboratory Assistant. V. L. CHURCHILL, Sampling Agent. MRS. A. B. VOSBURGH, Secretary.
Biochemical Laboratory.	T. B. OSBORNE, PH.D., Consulting Biochemist. H. B. VICKERY, PH.D., Biochemist in Charge. GEORGE W. PUCHER, PH.D., Research Assistant MISS HELEN C. CANNON, B.S., Dietitian.
Botany.	G. P. CLINTON, Sc.D., Botanist in Charge. E. M. STODDARD, B.S., Pomologist. MISS FLORENCE A. McCORMICK, PH.D., Pathologist. HAROLD B. BENDER, B.S., Graduate Assistant A. D. McDonnell, General Assistant. MRS. W. W. KELSEY, Secretary.
Entomology.	W. E. Britton, PhD., Entomologist in Charge: State Entomologist.  B. H. Walden, B.Agr. M. P. Zappe, B.S. PHILIF GARMAN, Ph.D. ROGER B. FRIEND, Ph.D. JOHN T. ASHWORTH, Deputy in Charge of Gipsy Moth Work R. C. Botsford, Deputy in Charge of Mosquito Elimination. J. P. Johnson, B.S., Deputy in Charge of Asiatic and Japanese Beetle Quarantines.  Mrs. GLADYS BROOKE, B.A., Secretary.
Forestry.	WALTER O. FILLEY, Forester in Charge. H. W. HICOCK, M.F., Assistant Forester. J. E. RILEY, JR., M.F., In Charge of Blister Rust Control. MISS PAULINE A. MERCHANT, Secretary.
Plant Breeding.	Donald F. Jones, S.D., Geneticist in Charge. W. R. Singleton, S.M., Assistant Geneticist. H. R. Murray, B.S., Graduate Assistant. Mrs. R. A. Hunter, Secretary.
Soil Research.	M. F. Morgan, M.S., Agronomist. H. G. M. Jacobson, M.S., Assistant. Herbert A. Lunt, M.S., Research Assistant. Dwicht B. Downs, General Assistant.
Tobacco Sub-station at Windsor.	PAUL J. Anderson, Ph.D., Pathologist in Charge. T. R. Swanback, M.S., Scientific Assistant.

MISS DOROTHY LENARD, Secretary.

# Topping and Suckering Practices As Related to the

## Yield and Quality of Havana Seed Tobacco

N. T. NELSON<sup>1</sup>

The maturity of a tobacco leaf probably determines quality to a oreater extent than any other single factor. The accepted practice among tobacco growers is to go through the crop some time during the blossoming period and break off the tops. The purpose of this operation is to hasten the development of the leaves by retarding or preventing the formation of seeds. Since rapid and important changes are occurring in all plants during the blossoming period, it would seem that the grower had a powerful means of either retarding or accelerating these chemical changes by topping. Similar results (1) have been found to hold true for such plants as alfalfa. timothy, bluegrass and redtop. Important effects (2) also have been obtained by removal of young fruits from the tomato, and much work on the effects of pruning fruit trees have been reported indicating similar trends. All of these experiments indicate that pruning has an important influence on plant growth. Topping and suckering of tobacco is in fact a pruning operation and therefore should exert a marked influence on the metabolic changes occuring in the plant during the period of ripening.

Among tobacco growers there is considerable variation in the time when topping is done and the size of the portion removed from the plant. The time of topping varies from the bud to the full bloom stages. Also, the number of leaves allowed to remain on the plant varies from twelve to eighteen. In view of the results obtained on other plants this operation, as well as the frequency of suckering, should be standardized to such a time and manner so as to permit the most profitable production of leaf by the plant.

Considerably greater difficulty is encountered in determining the most favorable degree of ripeness with the stalkcut than with those varieties which are primed. The primed leaves are picked when the grower judges they are ripe; but the stalk cut varieties, such as Havana seed and Broadleaf, are harvested when the plant as a whole represents the best quality. Therefore, although rules may be laid down as to when certain operation are to be done, the grower must exercise good judgment to obtain best results.

<sup>&</sup>lt;sup>1</sup>Until July 1928 Plant Physiologist at Connecticut Tobacco Substation, Windsor, Conn. Now Chief, Tobacco Division, Central Experimental Farm, Ottawa, Canada.

#### REVIEW OF LITERATURE

BULLETIN 297

In reviewing the work of other investigators on topping and suckering of tobacco, one is impressed with the apparent disagreement in conclusions reached. For instance, Olson (3) of Pennsylvania, obtained increased yields from high topping as compared with low topping. Johnson (4) of Wisconsin, found that high topping did not necessarily increase the yield. Low topping is advocated by some workers as the most satisfactory; others maintain that low topping produces coarse, low-quality tobacco. In regard to suckering, Olson (3) finds that two suckerings improved the quality of Pennsylvania tobacco, whereas Kentucky workers (5) (6) find that a lighter, thinner leaf is produced with less suckering. In the Pennsylvania experiments covering a period of ten vears, the results on height of topping were measured in terms of yield only, nothing specific being stated regarding the effect on quality. In their suckering experiments, however, during 1912 and 1913 the best yield and also the best quality was obtained on the plots suckered twice. The plots suckered once not only gave inferior quality but the yields averaged over 300 pounds less per

These illustrations are typical of the diversity of results on experiments in this country relative to the effects of topping and suckering on the yield and quality of tobacco. Experimental workers are more or less agreed that low topping reduces the variability in the size of the leaves and tends to hasten maturity. The evidence also indicates that a thick, heavy-bodied leaf, better adapted for use as a filler or a cheap binder is produced by frequent suckering.

#### PLAN OF EXPERIMENTS.

**Purpose.**—An attempt was made to determine the relation between common practices of topping and suckering and the subsequent yield and quality of the crop. Experiments were begun at the Connecticut Tobacco Substation in 1925 and continued for three years.

Stages of topping.—Four stages of plant growth were selected, namely: bud, early bloom, full bloom and seed pod. The bud stage was when the top of the plant had elongated to a considerable extent, but the upper stem portion was still succulent. This somewhat immature stage was about three or four days before blossoming commenced. The early bloom stage was when the first blossom opened. The upper stem was more rigid than it was in the bud stage but was still somewhat succulent and could be broken easily. The full bloom stage was represented by an advanced maturity of three or four days when fifteen to twenty blossoms were open. At this stage the stem had become somewhat stiff and woody. The seed pod stage was not topped, nor suckered, until the day of

harvest. This stage represents the normal growth and development of the plant. This stage was used as a standard, any deviations from which indicate the extent to which the plant can be changed.



FIGURE 1.

Top portion of a tobacco plant showing the bare, leafless, spikelike stem of the lowest "spike" sucker. This is sometimes called the "bald" sucker. The ordinary or common suckers growing lower on the stalk have two or more leaves. High topping, as used in the text, refers to plants topped at the internode immediately above the "spike" sucker; low topping is four leaves below this.

Height of topping. All references to high topping as used in this work refer to plants at such a height that approximately eighteen leaves were allowed to remain on the plant after topping. The

point selected was the internode above the so-called "spike" sucker, sometimes called the lowest "bald" sucker. (See Figure 1.) The "spike" sucker is the lowest sucker having a bare, leafless, spike-like stem. All suckers below this have two or more leaves. Topping in the above manner is referred to in this study as high topping. Plants topped four leaves lower are referred to as low topping.

**Suckering.** Shortly after topping, under favorable conditions of growth, the axillary shoots, called suckers, begin to grow. Their removal naturally accentuates the effects induced by topping. Plots suckered once were so treated just before harvesting. On tobacco suckered two or more times, the axillary shoots were removed at topping; about ten days later, and just before harvest.

Replication and size of plots. All yields and sorting records are based on several replications of 1/200 acre plots. The plots of each treatment were in duplicate in 1925; in quadruplicate in 1926; and in triplicate in 1927. All abnormal, stunted and diseased plants were eliminated at the time of harvest. Only plants which were representative of the treatment were included in determining the yield and quality.

**Fertilizers.** The plots had a uniform treatment of a standard tobacco fertilizer at the rate of two tons of a 5–4–5 mixture.

Transplanting and harvest. All plots were set on the same day to insure uniformity in starting. Growth was uniform on all of the plots until the bud stage was reached. The differences in yield and quality, therefore, were a direct effect of topping and suckering treatments. All plots were harvested at the same time and cured in the same shed to insure like treatment after harvest. Since the tobacco was uniform in size, maturity and vigor of growth, on all of the plots at the time the bud stage was reached, important trends on the effects of topping and suckering are indicated in the results obtained.

#### SPECIAL METHODS.

**Stripping.** Instead of stripping the tobacco in the ordinary manner by which all the leaves are mixed together, the leaves from different portions of the plant were kept separate. The upper six leaves in high topping and the upper two leaves in low topping were called "tops". The next four leaves were called "upper leaves"; the next four were designated as "middle leaves"; and the remaining leaves at the bottom of the plant (three to five leaves) constituted the "lower leaves." In using this method the effects of the different topping treatments could be traced to specific portions of the plant.

**Sorting.** The tobacco from these several treatments and plant regions was carefully sorted into the various commercial grades, which were immediately weighed and the results computed on a percentage basis. Also, from the total weight of tobacco from each plot the acre yields were determined.

#### EXPERIMENTAL RESULTS.

#### STAGE OF TOPPING.

**Yield.** As previously mentioned, the treatments were in duplicate, quadruplicate, and triplicate for the years 1925, 1926, and 1927, respectively. These results are presented in Table 1.

Table 1. Yields of Plots Topped at Different Stages of Growth, 1925-1927.

	Average of	f yield per acre of	cured leaf.	
Stage of topping	1925	1926	1927	3 year average
Bud	1638	1398	1148	1395
Early bloom	1800	1478	1192	1490
Full bloom	1710	1476	1182	1456
Seed pod	1512	1322	900	1245

These results indicate that the early bloom period is the best time to top Havana seed tobacco. The results are particularly pronounced in a favorable year. When the season is exceptionally dry (as in 1926) or exceptionally wet (as in 1927) the difference between early bloom and full bloom is not significant. Abnormal years like these tend to smooth out quantity of growth differences, due to treatment, because the seasonal conditions are the limiting factors. In general, it may be said that an early blossom stage of topping gives better yields than when this is done at too immature or succulent stage or at too woody or ripe stage. If topping is done when the plant is too young growth is checked to such an extent that its adverse effect is reflected in the yield. If topped too late. after seed production has progressed to a considerable degree, topping will have less effect in activating the plant to further vegetative growth. One should not top the plant when it is too immature and succulent, not wait until it becomes old and woodv

**Quality.** The effects of stage of topping on the quality as determined by sorting records is computed to a single figure called the grade index and presented in Table 2. The grade index is a single

Table 2. Grade Index of Tobacco Topped at Different Stages, 1925-1927.

	Average	grade index of toba	cco topped at diff	ferent stages.
Year	Bud	Early bloom	Full bloom	Seed pod
1925	.393	.451	.409	.326
1927 3 year average	.471	.459	.432	.226

number, expressing the quality of a particular lot of tobacco. It is based on the percentage of carefully assorted grades and the relative price values. Although market prices for grades vary

from year to year the relative ratio of prices remains fairly constant. The price relationship as used in these experiments is as follows: Light wrappers, 1.00; mediums .75; seconds .50; darks .30; fillers and brokes .10. The grade index is obtained by multiplying the percentage of each grade by the prices indicated above and adding the products.

BULLETIN 297

These results show that the early blossom stage is the best time to top Havana seed tobacco when quality is desired. It is decid-

edly better than topping at more mature stages.

There are other substantial reasons for topping the tobacco at an early bloom stage. It is desirable to do so early in order to reduce the hazard of winds blowing the crop down. Winds may do serious damage at this time. However, if the plants are topped before the tops become too large they are less liable to be damaged. Also, at this early stage the tops are easily broken because the stems are still succulent. If topped earlier the tops do not develop sufficiently to indicate the proper place to top. Hence, for these additional reasons, it seems better to top tobacco when it begins to blossom rather than at the bud or full bloom stages.

#### HEIGHT OF TOPPING

Yield. The height of topping does not have as much influence on the yields as one might anticipate. The results in 1925 did not give any decrease in yield due to low topping, but on the contrary an increase. The low topping yielded 1,850 pounds; high topping 1,810 pounds per acre.

The average results for 1926 and 1927 are given in Table 3.

TABLE 3. EFFECTS OF HEIGHT OF TOPPING AT VARIOUS STAGES ON THE YIELDS, 1926-1927.

	Yield	per acre (lbs.)	at different topping	stages.
Height of topping	Year	Bud	Early bloom	Full bloom
Low	1926 1927 Average	1412 1145 1278	1467 1172 1319	$1462 \\ 1192 \\ 1327$
High	1926 1927 Average	1385 1148 1266	1490 1192 1341	1462 1182 1322

Within the limits of these experiments, the height of topping did not affect the yields. When the tobacco was topped low the remaining leaves increased in size and weight to counterbalance the loss incurred at topping. It is conceivable, however, that still lower topping might reduce the yield. Four leaves below the spike sucker apparently is as low as a plant can be topped without causing a loss in yield.

Quality. Low topping had a marked beneficial effect on the quality. The effect of height of topping on the percentage of topts, mediums and darks is given in Table 4.

TABLE 4. SHOWING THE EFFECTS OF HEIGHT OF TOPPING IN EARLY BLOOM ON THE PERCENTAGE OF DARKS, MEDIUMS AND LIGHTS FOR THREE YEARS, 1925-1927.

		% Lig	ht		9	% Med	diums			% D	arks	
Height of topping	1925	1926	1927	Ave.	1925	1926	1927	Ave.	1925	1926	1927	Ave.
- 1 nod (high)	0	4	0	1	0	0	1	0	37	28	32	32
TT: wh	1.5	15	9	13	13	11	4	9	39	26	31	32
Low	23	31	20	25	16	18	11	15	32	18	28	26

Nearly twice as high a percentage of lights and mediums was produced by low topping as by high topping. The tobacco when allowed to go to seed without topping, until the day of harvest, resulted in very poor quality. Chemical conditions within the plant brought about by seed formation are not correlated with high quality leaf. High topping does not counteract this as effectively as low topping. The percentages of lights and mediums is practically midway between those of low topping and those resulting from the normal development of the plant. Severe pruning, i.e., low topping, in an early bloom stage produces high quality tobacco.

**Grade index.** The simplest way of indicating the relative qualities is probably by expressing it in terms of the grade index. These averages are presented in Table 5.

Table 5. Showing Effects of Height and Stage of Topping on the Quality as Represented by the Grade Index, 1926-1927.

	Average grade i	ndex in relation	to height and	stage of topping
Stage of topping	Year	Low	High	Not topped
Bud	1926 1927 Average	.524 .514 .517	.442 .439 .440	.437 $.276$ $.331$
Early bloom	1926 1927 Average	.523 .524 .523	.455 .395 .425	.437 $.276$ $.331$
Full bloom	1926 1927 Average	.491 .497 .494	.447 .368 .407	.437 $.276$ $.331$
General average		. 512	.426	.331

The best quality tobacco was produced by low topping in the early bloom stage. This tobacco averaged 8.6 cents more a pound than high topped tobacco for the two years, 1925 and 1926. In every instance the tobacco excelled the corresponding plots topped high. The evidence is conclusive that low topping of Havana seed tobacco results in better quality than high topping.

**Regional distribution of grades.** Since the quality is improved to such an extent by low topping, it is of interest to note the particular portions of the plant affected. In table 6, the percentage of lights, mediums and darks, produced at different levels on the plant is given.

Table 6. Showing the Effects of Height of Topping on the Percentage of Darks, Mediums and Lights at Different Levels of the Plant, 1925-1927.

		To	ps	' Upper	leaves	Middle	e leaves	Lower	leaves
Grade Ye	ar	Low.	High	Low	High	Low	High	Low	High
		%	%	%	%	%	%	%	%
Darks195	25	84	93	10	23	0	0	0	0
195	26	66	82	11	5	0	0	0	ő
193	27	100	100	55	51	0	0	0	0
Average		83	92	25	26	0	0	0	0
Mediums 193	25	16	7	3	31	0	5	0	0
195	26	27	8	38	13	0	38	0	0
195	27	0	0	27	12	4	0	0	0
Average		14	5	23	18	1	14	0	0
Lights195	25	0	0	36	25	54	40	5	4
195	26	3	4	42	36	60	16	17	5
195	27	0	0	7	5	51	14	12	0
Average		1	1	28	22	55	23	11	3

There seem to be two opposing tendencies in the plant: first, a decreasing tendency to produce darks extending from the top of the plant toward the base; second, an increasing tendency to produce lights as the lower leaves are approached. There is evidently a point along the stalk where the tendency to produce darks is equal to that of producing lights. This point is higher up on plants that are topped low. Accordingly, the tendency to produce the less desirable darks does not extend as low down on the plant when the plant is topped low. This is indicated by the lower percentage of mediums found in middle leaves under low topping, 1% compared with 14%. The difference between the percentage of darks and lights in the different regions under the two methods is consistently in favor of the lower topping. The tendency to produce inferior quality is consistently associated with high topping, particularly in the middle portion of the plant.

**Grade index.** To further illustrate the effects of topping on the quality of tobacco in different regions of the plant, comparisons of the grade index are given for two years 1926 and 1927. These results are given in Table 7.

TABLE 7. SHOWING EFFECTS OF HEIGHT OF TOPPING ON THE GRADE INDEX IN DIFFERENT REGIONS OF THE PLANT WHEN TOBACCO IS TOPPED AT DIFFERENT STAGES OF GROWTH.

	Height of									
topping	topping	1926	1927	Ave.	1926	1927	Ave.	1926	1927	Ave.
Bud	Low High	$\begin{array}{c} 651 \\ 664 \end{array}$	$.487 \\ .463$	$.569 \\ .564$	$.829 \\ .672$	$.806 \\ .646$	.817 .659	$.498 \\ .408$	.333	. 415 . 349
Early bloom.	Low High	717 659	$.508 \\ .479$	. 613 . 569	$.900 \\ .672$	.756 .614	$.828 \\ .643$	.546 .418	$.382 \\ .305$	$.464 \\ .362$
Full bloom	Low High	753 801	$.455 \\ .507$	. 604 . 654	.776 .649	.740 .547	.758 .598	. 494 . 453	$.351 \\ .218$	. 423 . 336
Seed pod		587	.421	.502	. 630	. 437	. 534	. 450	. 221	. 336
Average	Low			. 595			801			. 434
Average*	High			. 596			. 633			.349
Average not t	opped			. 501			534			. 336
*Not include	ding seed p	od p	lots.							

Tobacco topped later than the early bloom stage rapidly deteriorates in quality. The middle and lower leaves are affected to a greater extent than any other portion of the plant. This undesirable effect is characterized by an overripe condition of these leaves which is associated with yellow, variegated, and mottled colors, when the tobacco is cured

It will be noted in Table 8 that the quality of the middle and lower leaves of the plant is improved the most by low topping. It has been observed that this is the region immediately below the maximum sucker development.

Table 8. Average Relative Quality on Basis of 100 in Different Plant Regions as Effected by Topping.

Topping treatment	Relative quality,	(1926-1927)	ortions of the plant
	Upper leaves	Middle leaves	Lower leaves
Seed pods†	100.0	100.0	100.0
topping	118.7	118.5	100.4
Low topping	118.7	150.0	129.1

†Not topped until harvest.

High and low topping improved the upper leaves 18.7%, but the middle leaves were improved 50%, by low topping and only 18.5% by high topping. The lower leaves were improved 29.1% by low topping. High topping had practically no effect, only .4%, on the bottom leaves.

Burn tests. The fire holding capacity of the unfermented leaves as affected by the height of topping, was tested by counting the number of seconds a leaf continued to glow after it had been

ignited by an electric match. To facilitate this work a metronome was standardized so that there were exactly sixty beats a minute. Each figure recorded was the average of a burn test on each side of at least twenty leaves taken at random from the grades represented. These burn tests indicated that low topping improved burn. The average burn of darks was 6.2 seconds for high and 9.0 seconds for low topping. The medium averaged 10.7 seconds for high topping and 11.7 for low topping. This difference between the lights was not great. Low topping averaged 13.9 and high 13.4 seconds. There also was a consistent relationship between the duration of burn and the portion of the plant from which the leaves were taken. The tobacco became progressively poorer in burn as the top of the plant was approached. The top, upper, and middle leaves averaged 7.5, 11.5, 13.2 seconds respectively.

These figures indicate that the burn of tobacco is intimately associated with the chemical transformations taking place in the plant after the plant is topped. It also shows the farmer has some degree of control over these changes by the manner of topping.

Low topping of Havana seed not only increases the percentage of desirable grades but also results in an improved burn.

#### DEGREE OF SUCKERING

Yield. The degree of suckering exerts a marked influence on yield. These results for 1925 are given in Table 9. The plots suckered once were so treated the day before harvest. Those plots suckered twice had an earlier suckering two weeks before harvest.

Table 9. Yields of Tobacco as Affected by Number of Times Crop was Suckered, 1925.

	Yields per	acre (lbs.)	% increase from
Stage of topping	Suckered once	Suckered twice	two suckerings
Budded*	1377	1674	21.6
Bud	1530	1746	13.5
Early bloom	1746	1854	6.2
Full Bloom	1674	1746	.3
Average	1582	1755	41.4

\*Topped to desired height ten days after the buds were removed.

In 1926, three suckerings in early bloom averaged 1592 pounds per acre as compared with 1,412 pounds when suckered only once.

In all of these trials two or more suckerings consistently yield higher than a single suckering. The average increase per acre was 173 pounds in 1925 and 180 pounds in 1926. These differences are more striking when tobacco is topped in the immature bud stage. This relation is made clear by comparing the percentage increase of the yields resulting from more than one suckering. The average

increase for the bud stages was 17.5% while the more mature plossom stages averaged only 5.2%.

Quality. As previously noted, the yields of the three earliest topping stages were affected more by the number of suckerings than the later stages. These three stages are used to show comparative effects on quality. The percentage of assorted grades from these treatments is given in Table 10.

TABLE 10. PERCENTAGE OF DARKS, MEDIUMS, LIGHTS AND SECONDS ON PLOTS SUCKERED TWICE AS COMPARED WITH THOSE SUCKERED ONCE, 1925.

Number times suckered	Stage of topping	Perce Darks	ntage of indica Mediums	ted grades. Lights	Seconds
Once	Bud E. bloom	25 25 32	3 4 10	5 4 15	39 35 37
	Average	27.3	5.7	8.0	37.0
Twice	Budded Bud E. bloom	37 37 38	7 8 10	7 11 9	27 26 26
	Average	37.3	8.3	9.0	26.3

These data indicate that if the sucker growths on tobacco are removed too diligently, there is a resultant tendency to produce high percentages of heavy dark tobacco. The growth of suckers appear to be a desirable characteristic in the production of light colored, free burning tobacco. These actively growing suckers are instrumental in drawing from the leaf materials which are detrimental to quality. These translocation phenomena are intimately associated with the removal of nitrogenous compounds, whose presence in the leaf is associated with dark colors and poor burn. Considering the tobacco taken from any particular plant, high quality is always correlated with a reduction in the amount of these nitrogenous constituents. These changes are most active during the ripening period between early topping and harvest.

#### SUMMARY

With a given fertilizer treatment, the quality of leaf for any particular season is largely determined by the extent of the changes occurring in the plant between the bud and the harvest stages. The grower can control these changes by judicious topping and suckering.

The best topping stage for Havana seed tobacco is when it begins to blossom. If it is topped at too mature a period of growth the quality deteriorates as evidenced by increased percentages of mottled, yellow-spotted and variegated colors. This over-maturity effect may be avoided by earlier harvest.

Low topping does not necessarily reduce the yield. The remaining leaves grow larger.

High topping increases the percentage of short, low-priced darks. Furthermore, it promotes the production of dark tobacco in regions nearer the base of the plant than is the case with low topping.

Low topping (three to four leaves below the "spike" sucker) consistently gives better quality than high topping (at "spike" sucker).

Low topping stimulates the growth of the lower nine to eleven leaves.

Low topping has a marked beneficial effect on the burn.

There is a regional distribution of grades. The tendency of the plant is to produce darks and mediums in the upper leaves, and lights and seconds in the lower portion of the plant. The lower nine to eleven leaves usually include the bulk of high quality tobacco.

Several suckerings tend toward higher yields but a heavier, darker leaf.

Abundant growth of suckers suggests the removal from the leaf of substances deleterious to quality.

Topping retards the process of seed formation and activates the plant to vegetative growth.

#### LITERATURE CITED

- Graber, L. F.; Nelson, N. T.; Leukel, W. A.; and Albert, W. B. Organic Food Reserves in Relation to the Growth of Alfalfa and other Perennial Herbaceous Plants. Wis. Res. Bul. 80, 1927.
- Murneek, A. E. Physiology of Reproduction in Horticultural Plants.

   Reproduction and Metabolic Efficiency in the Tomato. Mo. Res. Bul. 90. 1926.
- 3. Olson, Otto. Results of Tobacco Experiments in Pennsylvania 1912-1922. Pa. Bul. 179: 13-16. 1923.
- 4. Johnson, James. Tobacco in Wisconsin; Wis. Bul. 337: 24-25. 1921.
- 5. Kinney, E. J.; Farm crops. Ky. Cir. 56: 34-35.
- Scherffius, W. H.; Woolsey, H.; and Mahan, C. A. Cultivation of Tobacco in Kentucky and Tennessee. U. S. D. A. Bul. 343-18-19. 1909.
- 7. Digges, D. D. and Freeman, H. A. Flue-cured Tobacco in Canada (Ottawa), Dom. Can. Dept. of Agr. Bul. 38-22. 1920.
- 8. Garner, W. W. Tobacco Culture. U. S. D. A. Farm Bul. 571.1922.

# Connecticut Agricultural Experiment Station New Haven, Connecticut

# REPORT OF THE DIRECTOR

FOR THE

YEAR ENDING OCTOBER 31

1928

The Bulletins of this Station are mailed free to citizens of Connectiout who apply for them, and to other applicants as far as the editions permit.

#### OFFICERS AND STAFF

As of October 31, 1928

#### BOARD OF CONTROL

His Excellency	, Governor John H. Trumbull, ex-officio President
Charles R. Treat, George A. Hopson, Wm. L. Slate, <i>Dire</i> Ioseph W. Alsop	Vice-PresidentOrange SecretaryMt. Carmel ector and TreasurerNew Haven
Edward C. Schneid	Avon Southington der Middletown Cheshire
	STAFF.
	E. H. JENKINS, Ph.D., Director Emeritus.
Administration.	WM. L. SLATE, B.Sc., Director and Treasurer. MISS L. M. BRAUTLECHT, Bookkeeper and Librarian. G. E. GRAHAM, In charge of Buildings and Grounds.
Chemistry: Analytical Laboratory.	E. M. BAILEY, PH.D., Chemist in Charge. C. E. SHEPARD CWEN L. NOLAN HARRY J. FISHER, A.B. W. T. MATHIS DAVID C. WALDEN, B.S. FRANK C. SHELDON, Laboratory Assistant. V. L. CHURCHILI, Sampling Agent. Mrs. A. B. VOSBURGH, Secretary.
Biochemical Laboratory.	T. B. OSBORNE, PH.D., Consulting Biochemist. H. B. VICKERY, PH.D., Biochemist in Charge. GEORGE W. PUCHER PH.D., Research Assistant. MISS HELEN C. CANNON, B.S., Dictitian.
Botany.	G. P. CLINTON, Sc.D., Betanist in Charge. E. M. STODDARD, B.S., Pomologist. MISS FLORENCE A. McCORMICK, PH.D., Pathologist. Harold B. Bender, B.S., Graduate Assistant. A. D. McDonnell, General Assistant. Mrs. W. W. Kelsey, Secretary.
Entomology.	W. E. Britton, PhD., Entomologist in Charge: State Entomologist B. H. Walden, B.Agr. M. P. Zappe, B.S. Philip Garman, Ph.D. Roger B. Friend, Ph.D. John T. Ashworth, Deputy in Charge of Gipsy Moth Work. R. C. Botsford, Deputy in Charge of Mosquito Elimination. J. P. Johnson, B.S., Deputy in Charge of Asiatic and Japanese Beetle Quarantines. Mrs. Gladys Brooke, B.A., Secretary.
Forestry,	WALTER O. FILLEY, Forester in Charge. H. W. HICOCK, M.F., Assistant Forester. J. E. RILEY, JR., M.F., In Charge of Blister Rust Control. MISS PAULINE A. MERCHANT, Secretary.
Plant Breeding.	DONALD F. JONES, S.D., Geneticist in Charge. W. R. SINGLETON, S.M., Assistant Geneticist. H. R. MURRAY, B.S., Graduate Assistant. Mrs. R. A. HUNTER, Secretary.
Soil Research.	M. F. Morgan, M.S., Agronomist H. G. M. Jacobson, M.S., Assistant Agronomist. Herbert A. Lunt, M.S., Research Assistant in Forest Soils. DWIGHT B. DOWNS, General Assistant.
Tobacco Sub-station at Windsor,	Paul J. Anderson, Ph.D., Pathologist in Charge. T. R. Swanback, M.S., Scientific Assistant. Miss Dorothy Lenard, Secretary.

# Report of the Director

For the Year Ending October 31, 1928

To the Board of Control of the Connecticut Agricultural Experiment Station;

In accordance with the usual custom, I have the honor to submit herewith a statement regarding the work of the Station for the past twelve months, together with other information of a pertinent nature. Our Station, the first to be established in this country, has now completed fifty-three years of service to Connecticut agriculture. With this in mind, the following note on its history and record may be appropriate.

The agricultural experiment station had its origin in Europe, the first having been established at Mockern, Germany, in 1851. By 1873 there were sixty-three such stations in continental Europe supported by agricultural societies or by the states in which they were located. The first privately endowed station was established

at Rothamsted, England.

In 1853, Samuel W. Johnson, a young graduate of Yale University went to Germany to study chemistry, particularly its application to agriculture. While there he visited several of the experiment stations and studied under some of the famous agricultural chemists of that day. Returning to Yale as an instructor in chemistry, he was appointed Chemist to the State Agricultural Society. During the next twenty years he never ceased to urge in his lectures and in the press, the need of agricultural stations in this country, and as a result of his labors, the Legislature of 1875 secured to Connecticut the honor of establishing the first agricultural experiment station in America. Connecticut having pioneered in the movement, several other states followed, until in 1887 Congress passed the Hatch Act, providing for a station in each state.

The duties of the Station as defined in the statute are three-fold: the analysis of fertilizers, feeds, drugs, foods and other products for the protection of the people of the State; the conduct of "scientific investigation and experiments" in agriculture and related sciences: the dissemination of the results.

As an example of the first, the fertilizer inspection of 1928 disclosed the sale of a brand of fish meal to which had been added sulfate of ammonia, thus causing a low grade fish to show the same amount of total nitrogen as does the high grade product. These facts were made known immediately and the purchasers have been reimbursed by the dealers who unwittingly sold the goods.

Other direct services include the inspection of nurseries and orchards, the control of insect pests and plant diseases, the dis-

tribution of forest planting stock, the testing of seeds, advice on the management of land and the like.

Under the head of "experiments" might be listed a long series of investigations covering the fifty-three years of the Station's existence. Notable among them are those dealing with corn breeding, the white pine blister rust, the chestnut bark disease, tobacco wild fire, tobacco root rot, the nature of the vegetable proteins, the discovery of the vitamines and studies on the food value of milk. For this last it is fair to say that most of our recent knowledge of the food value of milk and the resulting increase in its consumption are traceable to the contributions of the Station

The dissemination of results is accomplished through bulletins, lectures, the work of the Extension Service, correspondence and many personal visits to farms and orchards. The Station is not only a research institution—it is and always has been a service agency for the people of the State. The following quotation from the Report for 1902 truly describes the policy that the Station has always tried to follow: "It is the wish of the Board of Control to make the Station as useful as its resources will admit. Every Connecticut citizen who is concerned in agriculture, whether farmer or resident of a city, has the right to apply to the Station for any assistance that comes within its province to render, and the Station will respond as far as lies in its power."

#### REVIEW OF THE YEAR

The Station has been unusually fortunate in the number of outstanding scientists who have been members of its staff since its establishment. Among those deserving highest honor is Dr. Thomas B. Osborne, who at his own request has been relieved of active charge of the Biochemical Laboratory and given the title of Consulting Biochemist. Coming to the Station in 1886, Dr. Osborne soon began his life work, the study of the Nature of the Vegetable Proteins, for which his name is known and honored throughout the scientific world. In collaboration with Dr. Lafayette B. Mendel of Yale University, he has pioneered in the newer phases of nutrition, thus adding to the reputation of the Station for fundamental contributions to agriculture and science. It is a source of satisfaction to all interested in the Station to know that Dr. Osborne is to maintain his active interest in the work he established and carried on so well and that we will continue to have the benefit of his advice and counsel.

The annual reports of the Station now number fifty-two and constitute a record of unusual value. Included in each report is an index but a cumulative index has never been prepared except for two subjects, Entomology and Food and Drugs. Dr. Jenkins volunteered to undertake this task and a complete index of all

matters of permanent value is now ready for the printer. With this will be included several special sections such as Analyses of Unusual Fertilizer Materials, A Complete List of Members of the Staff, and the like. Not only the Station but all those having occasion to consult agricultural literature will be indebted to Dr. Jenkins.

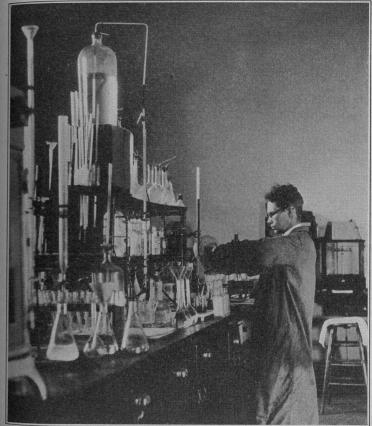


FIGURE 2. Analyzing Foods and Drugs—Analytical Laboratory.

#### CONTROL AND SERVICE WORK

INSPECTION OF FERTILIZERS, FEEDS, FOODS, DRUGS, ETC.

In accordance with the statutes relating thereto, the Analytical Laboratory has analyzed 900 samples of Fertilizers and Fertilizer materials. Of these, 536 were drawn officially by the Station Agent and the balance were received from farmers. The Report

on Fertilizers will appear in December, 1928, as Bulletin 296 of the Annual Report of the Station.

The Feed Inspection has involved the analysis of 800 samples, the report on which will be printed early in 1929. Over 800 brands of feeding stuffs are now registered for sale in Connecticut.

The examination of *Foods and Drugs* occupies a considerable portion of the time of the laboratory. During the past year 1,288 samples were analyzed, including such a wide variety of foods as pickles, sweet chocolate, breads and ice cream; and of drugs, powdered pepsin, Fowler's solution and many proprietary preparations. Special studies were made of cod liver oils (for vitamine A) and of "denicotinized" tobaccos, the latter especially attracting wide attention among physicians and laymen.

Of Babcock Glassware, 2,540 separate pieces have been calibrated and 134 dairy thermometers standardized, in accordance with the statutes.

In addition to the above regular duties, a large amount of detailed analytical work has been done in collaboration with the Tobacco Substation at Windsor, the Storrs Experiment Station and the Association of Official Agricultural Chemists.

#### CONTROL OF INSECT PESTS

The constantly increasing number of insect pests, the establishment and enforcement of quarantines, together with the necessary inspections and scouting, require more attention each successive year.

The State quarantine on account of the Asiatic Beetle has been maintained and enforced, and 426 certificates have been issued covering 2,951 cubic yards of soil and for 4,394 plants to be moved out of the restricted area. About 1,194 plants have been inspected by one of the staff, who has also examined 55 lawns by request and shown the owners how to apply the lead arsenate treatment. No carbon disulphide emulsion was applied in 1928.

The Japanese Beetle has been the subject of much attention. In 1927, the beetles were discovered in rather large numbers in Bridgeport, and on December 1, 1927, the quarantine was extended to cover two rows of shore towns from the New York line to the Housatonic River, in conformity with the Federal quarantine. From June 15 to October 1, all principal highways leading out of the quarantined area were patrolled during the day time, and a 24-hour patrol service was maintained at the eastern end of the Washington Bridge on the Boston-New York Post Road, to prevent the movement of farm products out of the quarantined area except in conformity with the regulations. An inspection stand was established in Bridgeport to facilitate the inspection of truck loads of produce to be shipped out of the area.

During the summer of 1928, Federal men scouted for the beetles in nearly all cities and larger towns of Connecticut outside the quarantined area, resulting in the discovery of infestations in New Haven, New London and Hartford. Only a few beetles were found at each of the first two places but the Hartford infestation was larger and the number of beetles probably ran into the hundreds. At Springfield, Mass., not far from the Connecticut line an infestation was found containing nearly 5,000 beetles. As yet there has been no Federal action regarding quarantines but it is probable that the present quarantine will be extended to include New Haven and adjacent towns, and that the other colonies will be treated as "outside" infestations. This work is conducted in



FIGURE 3. Burning Corn Stalks and Stubble in Stonington—European Corn Borer Control.

co-operation with the Federal Plant Quarantine and Control Administration.

The European Corn Borer has spread markedly during the season. Clean-up measures around the 1927 infestations were conducted partly in the fall and completed the following spring by State and Federal forces working in co-operation. In 1928, 21 new towns were found to be infested. These with the five already under quarantine make 27 towns in which the Corn Borer was found in 1928. A continuation of the clean-up methods practiced in the past seems to be out of the question on account of the expense and legislation compelling the owners or tenants to dispose of their corn stalks is now being considered.

There has been no important spread of the Gipsy Moth in Connecticut during the season. No stripping has occurred within the

State but larger areas than ever before were defoliated in Massa-

chusetts, New Hampshire and Maine.

The annual Inspection of Nurseries has been conducted with . greater care and thoroughness than ever before on account of the new pests to be looked for. The Station has also inspected the fruit and rose stocks imported from foreign countries into Connecticut nurseries for propagation.

The work of Mosquito Elimination has been continued as usual Considerable new ditching work has been done during the year with funds raised by town appropriations and private contributions. The ditching in Hamden, East Haven and Old Saybrook



FIGURE 4. A New Double-width Ditch at Hammonasset Park-Mosquito Elimination.

is not completed, but funds will be provided. The City of New Haven has installed a new tide gate at the Little River bridge on Middletown Avenue, which will make possible the ditching of a considerable area in North Haven. All of this work means a greater cost for maintenance, for according to the law the State is morally, if not legally responsible for maintenance after the areas have been properly ditched and the work approved and accepted. The appropriation for this work must be increased if the Station is to carry out the provisions of the statute.

#### WHITE PINE BLISTER RUST

During the past season 123,385 wild Ribes and 1,151 cultivated Ribes were destroyed on 75,102 acres. The work was conducted in 26 towns. Approximately one and a half percent of the acreage covered this year was a re-eradication of areas worked in previous years. 245,045 acres of non-pine land in the general pine region were eliminated from control.

The nursery sanitation project undertaken in 1927 was extended to include 3,962 acres surrounding commercial nurseries and 1,480 acres of water company plantings. These figures are included in the 75,102 acres reported above.

#### SHMMARY OF WHITE PINE BLISTER RUST CONTROL-1925 to 1928

Year.	Initial Erad. A.	Re-Erad. Acres.	Tot. A. Erad.	Wild Ribes	Cult. Ribes.	Estimated Pine A. Protected.	Nursery Sanitation Acres.
1925 1926	6,688 21,687	40 570	6,728 $22,256$	258,515 182,826	684 330	7.400	0
1927 1928	12,068 68,539	8,836 1,122	20,904 69,661	159,121 123,383	2,235 1,151	10,400 34,800	$\frac{1,000}{3,962}$

All eradication in 1927 and 1928 represents co-operative effort on the part of towns, individuals, pine owners, and the state. There is a noticeable reduction in the number of wild Ribes destroyed each year which indicates that initial eradication is effective in the areas of heaviest Ribes concentration. Blister Rust has been arrested on protected areas, but some re-eradication will be necessary each year if previously eradicated areas are to be kept in a sanitary condition.

#### SEED TESTING

This year a special study was made of the quality of Flower Seed offered on the Connecticut market and the inspection of Vegetable Seed was continued. The following table shows the extent of this work:

Kinds.	Number of samples.	Number of varieties.	Number of strains.
Vegetables	190	25	71
Flowers	62	26	14
Field Crops	33	14	18
Trees	20	8	5

#### SPRAY SERVICE

In continuation of the plan developed some years ago, the Station collaborated with the Extension Service and the Pomological Society in a Spray Service for orchardists. Two members of the Staff visited orchards regularly during the spring and summer, thus keeping in close touch with the appearance and development of insect pests and fungus diseases. The data collected, together with the weather predictions, provided the basis of advice on spraying and dusting.

#### DISTRIBUTION OF PLANTING STOCK

CONNECTICUT EXPERIMENT STATION

One and a quarter million trees were distributed during the past year for forest planting purposes, 333,000 of these going to farmers under the Clark-McNary Act. This is almost twice the number distributed in 1927

#### PROGRESS OF INVESTIGATIONS

Here follow brief notes on those projects of greatest interest or on which definite results have been obtained during the year. No attempt is made to discuss all of the investigations under way a list of which will be found on page 134.

#### BIOCHEMISTRY

Chemistry of the Proteins. As a part of the study of the methods for the separation of the basic amino-acids of proteins, highly purified crystalline samples of several of these substances have been prepared and photomicrographs published. A crystalline preparation of lysine had not been obtained previously.

Convenient methods for the preparation of both arginine and histidine on a large scale have also been developed. Analyses of the basic amino-acids of two proteins, edestin and oxyhemoglobin have been made and a full review of current speculations upon the constitution of proteins has been published in Physiological Reviews. Papers describing the other phases of the work have been published in the Journal of Biological Chemistry.

Nitrogenous Constituents of Plants. Progress has been made in the investigation of the simpler nitrogenous constituents of fresh green tobacco leaf although no papers have as yet been published. This work is being actively continued.

Experiments in Nutrition. Extensive data have been collected on the effect of diets deficient in various respects upon growth and the composition of the bones. A method has been developed for modifying the determining factors, one at a time. Thus the influence of the fat-soluble vitamine, the proportions of the inorganic nutrients and the potential reaction of the diet is being investigated.

In connection with extended earlier observations on the influence of green leaves on nutrition the potency of watercress with respect to some of its constituent vitamines has been investigated. This substance is found to be comparatively rich in vitamine A. The content of vitamine "B" in liver and preparations thereof is also being studied anew.

#### BOTANY

Mosaic. The rogueing experiment for the control of mosaic in Cuthbert Raspberries, in co-operation with the U. S. Bureau of Plant Industry has been continued. Because of unusual spread in 1927, many plants were removed. Plot I, consisting of plants received from Massachusetts, still shows the greatest amount of mosaic, although the disease is now general in all of the five plots.

REPORT OF THE DIRECTOR

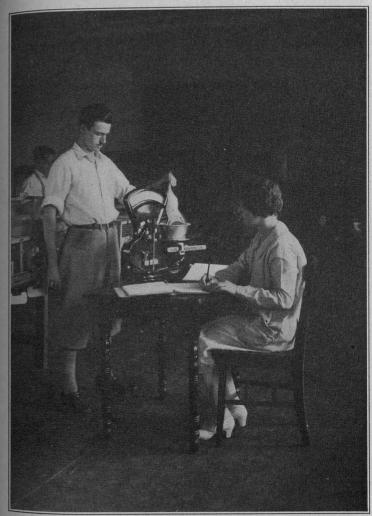


FIGURE 5. Weighing White Rats in the Nutrition Laboratory.

The practicability of this method of control has not yet been demonstrated by this experiment.

Infection of tobacco plants with "white-pickle" mosaic of cucumber produced some leaf mottling, but failed to produce plate

crystals in these mottled leaves and also seemed to disappear in the new growth. Last year somewhat similar results occurred with infection of cucurbits with "white-pickle" that seemed more or less temporary.

Plant Disease Survey. Because of the very wet summer the year. has been unusually favorable for the development of plant diseases As a result we have secured a greater number of diseases new to

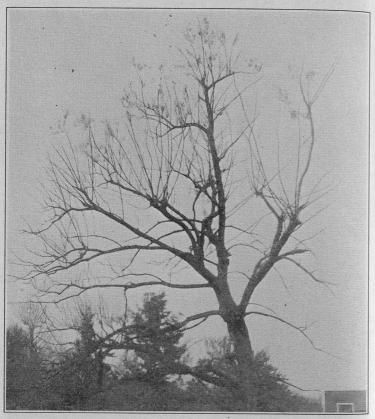


FIGURE 6. A Tree Practically Defoliated by Willow Scab.

the state than for sometime. Among the more important of these are bacterial leaf diseases of corn and horse radish, a Fusicladium disease of poplar similar to the new willow disease of last year, several Helminthosporium diseases of grains and grasses, a fungous leaf-scorch of maple, the Macrosporium blight of carrots, a Phytophthora rot of tomato fruit, and a number of new hosts for the Sclerotium stem-rot of herbaceous perennials. Besides these,

notato blight made its earliest recorded appearance in the state and caused some damage to the vines and rot of the tubers. The Fusicladium scab continued its destruction of willows, being even worse than last year.

Chestnut Blight. Last spring about 1,300 one-year-old chestnut seedlings were planted in two new places, Southport and Redding Ridge. This makes four locations in the State on which over 2,500 seedlings have been set out in recent years to see whether they would survive the blight. So far considerably more than half of the seedlings have died from unfavorable environment but none from blight. It seems to be difficult to carry the seedlings through the first season, especially when set out in the sun. This year's plantings, however, have apparently done better than usual due to the wet summer. A quantity of this year's nuts has been secured from the South for starting more seedlings. Records have also been made of two marked plots of native seedlings and sprouts in the woods to determine the progress of the blight from year to year.

Tree Diseases. The most important and extensive work in this field has been the new willow scab or blightfungus, Fusicladium saliciperdum. Considerable time has been spent in determining the distribution of this fungus, especially in this state and Canada, in noting its damage and the species of willows attacked. Cultures of the fungus were obtained and successful inoculations made in producing the disease similar to that in nature. Successful spraying experiments have also been carried on for its control.

#### ENTOMOLOGY

Asiatic Beetle. The investigations on the life history, habits and methods of control of the Asiatic beetle, Anomala orientalis, are completed and the data has been assembled, analyzed and prepared for publication. During 1928 much attention has been given to control by applications of lead arsenate, both mixed with the top layer of soil before seeding, and washed into the turf where re-seeding was unnecessary.

Plum Cuculio. The six year study of the control of this insect in Connecticut apple orchards has been completed and the data is now nearly ready for publication. It was found that four applications of lead arsenate, preferably the pink, calyx, 7-day and 2weeks sprays will give fair control.

Oriental Peach Moth. No satisfactory method of artificial control has yet been discovered. During the year, the ichneumonid Parasite Glypta rufiscutellaris, was reared for the first time in considerable numbers in Connecticut from the Oriental fruit moth, and it was more abundant than the other ichneumonid parasite, Macrocentrus ancylivora, which has heretofore been the principal Parasite of the Oriental fruit moth in the state. The egg parasite, Trichogramma minuta, was also present in 1928. Nearly 2,500

127

A minor project undertaken during the year in co-operation with the Botany Department, was one to determine the effect of calcium chloride washed from roadways on ornamental Norway spruce trees planted close by. The first results indicate that a solution of more than five percent strength is necessary to seriously damage the trees.

The Relation of Soil Type to Forest Composition and Rate of Growth. The work on this project was continued during the 1928 field season with a study of growth of pine plantations as related to soil type. It was originally planned to carry on the work with both red and white pine, but because of damage done by the weevil to white pine, investigation of this species was temporarily



FIGURE 8. White Pine Plantation after "Cordwood" Thinning.

abandoned and efforts were confined to red pine. Two hundred stations in red pine plantations were established as temporary plots. The data included:

a. The height of ten or more dominant trees.

b. Detailed notes on soil type and other soil factors.

From the tree measurements a "site index curve" for red pine has been prepared and tables derived therefrom. By the use of these tables any plantation or portion thereof may be given an "index number" which denotes the site quality or growing capacity of the land. At an age of fifteen years, the range in height (site index) varied from nine to twenty-two feet. The correlation between soil type and site quality is now being determined.

Using the data taken during the past season it is planned to select some twenty permanent plots where a careful study of soil

larvae of the Oriental fruit moth have been collected and reared for the experimental work of 1929. Over 600 of these were obtained from a bushel of quinces grown in Cheshire.

CONNECTICUT EXPERIMENT STATION

#### FORESTRY

Treatment to Prolong the Life of Timber. In co-operation with the Tobacco Substation at Windsor, the Forestry Department has undertaken to demonstrate the value of preservative treatment for tobacco shade-tent poles and the relative value of various native woods so treated, compared with chestnut. Posts of all the species common in Connecticut were peeled and seasoned last fall. In the spring they were treated with creosote and set in rows

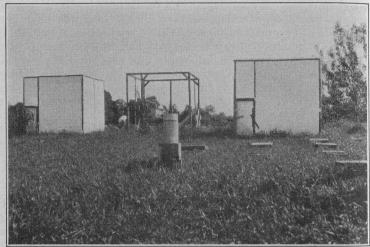


FIGURE 7. Tree Cages at Mt. Carmel Farm Used in Studying the Oriental Peach Moth.

on the Tobacco Station farm. Untreated check posts were included.

Naturally no data will be available for some years.

In the meantime there is a demand for information on methods of treating posts. Tobacco Station Bulletin No. 9 was therefore prepared and distributed last fall. It is a short, illustrated paper describing the methods of treatment adapted to farm use and the results that may be expected therefrom.

The Rainbow Plantations. Begun in 1902, these plantings have reached an age where they are yielding much valuable information. In addition to the comparisons between the several species, pruning and thinning experiments are now possible. Also these plots are furnishing an excellent opportunity for studying the effects of pure and mixed stands on the soil conditions which determine rate of growth.

129

conditions will be begun in red pine plantations to ascertain what soil factors are important in determining the growth of this species and the effect of the forest planting upon the soil.

CONNECTICUT EXPERIMENT STATION

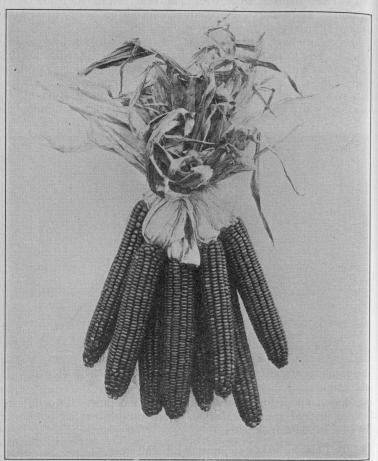


FIGURE 9. Canada Learning. A new type of Crossed Corn. As early as Canada Flint and as productive as Leaming.

#### PLANT BREEDING

Corn Breeding. A new type of "crossed" field corn has been distributed for trial. This is a first generation combination of inbred strains of Learning dent and Canada Yellow Flint and is called Canada Leaming. It unites the high yielding power and stalk growth of the dent type with the early maturity and good grain quality of the flint. This corn is being tested in Southern New

England as a husking corn and in Northern New England for

silage purposes.

About 150 crosses of inbred strains of Whipples Early Yellow sweet corn have been grown in a preliminary trial and some combinations have shown outstanding productiveness and uniformity in size and shape of ear and in time of ripening and at the same time have been as early or earlier in maturity than the original variety.

A new variety of sweet corn, not a first generation cross, has been produced by combining a number of inbred strains of Golden Bantam and Crosby. This new variety, which reproduces itself each year, is called Golden Crosby and in preliminary tests has met with favor from some of the Connecticut seedsmen who are

growing it for canning purposes.

Abnormal segregation of the sugary factor in certain families of corn has given widely deviating ratios, in most cases a large excess of recessive seeds and in other cases a marked deficiency. The ratios show larger variations from normal than have been found previously and indicate a new mode of inheritance that is not clearly understood.

Vegetable Breeding. Crosby Egyptian beets have been completely self-sterile, no seed having been produced from bagged flowers nor from isolated plants. Breeding methods with this plant will necessarily be limited to open pollinated selections and to sib-matings. The effect of environmental conditions on the color

of garden beets is being studied also.

Selections have been made in the second generation of a cross between a late and unproductive straightneck type of summer squash with an early crookneck in an attempt to fix an early straightneck type of good quality.

A new variety of spinach is now in the third generation of selection from a cross of Virginia Savov, Viroflay and King of Denmark. It combines in a large measure size, earliness and long keeping

qualities from its parental types.

A large number of selections from the cross of Alacrity and Earliana tomatoes have been grown in the attempt to produce a strain of the Earliana season, but with better shape and quality of fruit. Some of the selections which are now in the third generation are promising.

A remarkably productive and early-ripening pepper has been developed from a cross of Sweet Spanish by Harris Earliest. Several selections are being put through a careful test and will soon be available for preliminary trial by market gardeners.

#### SOIL INVESTIGATIONS

The Soils of Connecticut. Detailed investigations of the supply and availability of plant nutrients in the important soil types of the State have been continued. During the 1928 season two crops of tobacco and one crop of oats have been grown in the greenhouse on twenty-four soils. Tobacco has proven to be an ideal crop to reveal differences in the availability of the several plant nutrients contained in the soil. The most striking results may be summarized as follows:

Twenty-two of the twenty-four soils showed marked deficiency of potash and twenty exhibited more or less pronounced lack of available phosphorus. Results of greenhouse tests can be correlated very closely with the rapid method for the determination of



FIGURE 10. A Field Party Studying Connecticut Soils.

available phosphorus now used in the soils laboratory. None of the soils were able to supply sufficient nitrogen for normal growth of tobacco, although some contained as high as 10,000 pounds of total nitrogen per acre.

Tobacco showed abnormal growth on two very acid soils (3.9-4.0 pH). On these particular soils, this reaction was correlated with abnormally high concentrations of soluble manganese and aluminum. All the other soils, even with pH values as low as 4.6, showed little or no response to lime for the tobacco crop.

Forest Soils. A study of the characteristics of various types of soil under forested conditions is in progress, with a view to determining the soil factors which have a definite relation to silvicultural practice in Connecticut. Most forest soils that have never been farmed have been found to be considerably more acid than ordinary agricultural soils. Several extremely acid (as low as 3.2 pH) forest soils have been encountered. This phase of the program will be more actively followed than heretofore.

Co-operation with the Storrs Station. During the summer of 1928 one hundred and twenty-five farms distributed over the eastern highland of the State have been surveyed as to soil type in co-operation with the Agricultural Economics Department of the Storrs Experiment Station, which is studying the "Economic Significance of Soil Type." Soil maps of each farm were prepared, and samples of soil from over a thousand fields were tested for acidity. Soil samples from sixteen of these farms have been collected and are to be studied intensively in the greenhouse and laboratory.

Lawn Management. Lawn fertilization experiments have been conducted for the past three years. For average lawn turf on soils similar to that at the Station, top dressing with a readily available nitrogeneous fertilizer like sulfate of ammonia has proven to be the most important requirement. Moderate applications of phosphoric acid and potash help to keep up the fertility of the soil, but without a fertilizer containing available nitrogen, they are of little or no benefit.

Seeding trials with various lawn grasses, particularly the bent grasses, are being conducted. Stolon plantings of several strains of creeping bent have been successful, but are not proving thoroughly satisfactory to maintain under average lawn conditions. Seedings of several strains of bent grass produced commercially have demonstrated that native grown seed of creeping bent, velvet bent and Rhode Island bent are well adapted to local conditions where excellent drainage is provided. Fully as good turf has been produced by seeding creeping bent as by planting the stolons.

Connecticut River Flood Plain Studies. At the request of the Attorney General's office the Soils Department has undertaken a special study of the soils of the Connecticut River Flood Plain and the effects of periodic flooding on their productivity. A detailed soil and land cover survey has been completed of over 20,000 acres adjacent to the Connecticut River north of Middletown and laboratory and greenhouse studies are being made of the characteristics of these soils and of the sediments deposited upon them by floods

#### TOBACCO SUBSTATION

Relation of Fertilizer Ingredients to the Burn of Tobacco. Sulfate of ammonia has seriously lowered the fire holding capacity of the leaf. Dry ground fish in large quantities has had the same effect

but to a less degree. Urea, tankage and nitrate of soda have had no pronounced effect in either direction. Muriate of potash has been extremely injurious to burn. Double sulfate of potash magnesia slightly reduced the fire holding capacity when compared with sulfate of potash. Both carbonatate and nitrate of potash gave a better burn than sulfate. Lime produced a very white ash but lowered the fire holding capacity.

CONNECTICUT EXPERIMENT STATION

The Effect of Various Fertilizers on the Chemical Composition of Tobacco. Different carriers of nitrogen have not changed the total amount of nitrogen absorbed by the plant. The lower leaves of the plant have higher percentages of total ash, potassium, calcium, and nitrate nitrogen than the upper leaves.

The upper leaves have higher percentages of phosphorus, ammonia nitrogen, nicotine and chlorine than the lower leaves.

Plants treated with sulfate of ammonia show increased percentages of manganese, sulfur and aluminum.



Figure 11. Laboratory and Greenhouse at the Tobacco Substation at Windsor.

The potash content of the plant is reduced appreciably by ommitting potash from the fertilizer even though the soil naturally contains a large supply of potash.

Effect of Some Nitrogenous Fertilizers on Soil Reaction. Sulfate of ammonia has had the strongest influence in making soil more acid. Dry ground fish and tankage have had an acidifying tendency but to a less degree. Urea made the soil more acid after the initial alkaline effect had disappeared. Nitrate of soda reduced soil acidity, but nitrate of potash had no pronounced effect.

Urea as a Source of Nitrogen. Results with this fertilizer have been very satisfactory when it was used in moderate amounts. Two other synthetic nitrogen compounds, nitrate of potash and nitrate of lime also have given good results.

Increasing the Organic Matter in the Soil. Manure as a supplement to commercial fertilizer has given somewhat increased yields and better quality on a very sandy soil. A similar effect has been

produced by the use of a commercial "humus" product or by covering the sandy plots with muck soil.

Cover Crops for Tobacco Soils. During wet years on a sandy soil the yield and quality have been increased by the use of certain winter cover crops. Oats, barley and rye have been most effective. Wheat, alfalfa and red top have had a slightly beneficial effect while timothy has been detrimental.

#### FIELD DAYS AND EXHIBITS

The Mt. Carmel Farm Field Day was held on July 18, the New Haven County Farm Bureau joining with the Station in the same



FIGURE 12. Station Exhibit at the State Fair, 1928.

manner as in 1926. The general topic was Plant Pests. Professor W. H. Whetzel of Cornell University and Professor W. C. O'Kane of New Hampshire University were the speakers.

The Field Day at the Tobacco Substation was unique in that some two hundred tobacco farmers from Pennsylvania spent the afternoon inspecting the plots.

A general Station exhibit was made at the State Fair in Hartford and also a special tobacco exhibit.

# LIBRARY

During the year there were added to the Station Library 900 accessions of permanent value. Journals purchased now number 50, in addition to which some 30 farm papers and journals are received as exchanges. For a few of the old farm papers we have complete files. The total number of bound volumes is now 16,800.

#### PHYSICAL EQUIPMENT

Additions to the Station's scientific equipment include a rotary microtome, an apochromatic objective, an electric warming table for slides, a Berkefeld filter, an electric incubator, a steam pressure sterilzer, a moisture equavalent centrifuge, two analytical balances a wide field binocular microscope, two experimental tobacco curing chambers with temperature and humidity controls and an electic computing machine.

An addition to the barn at the Mt. Carmel Farm provides much needed space for the Botany and Entomology departments.

#### CHANGES IN STAFF

Appointments;

Herbert A. Lunt. M.S., Research Assistant in Forest Soils August, 1928.

George W. Pucher, Ph.D., Research Assistant in Biochemistry. September, 1928.

Mrs. Gladys Brooke, B.A., Secretary in Entomology Department, September, 1928.

Harold B. Bender, B.S., Graduate Assistant in Botany, October, 1928.

Resignations:

H. J. Lutz, M. F., Assistant Forester, June, 1928.

George Zundel, Ph.D., Graduate Assistant in Botany, July, 1928. Grace A. Foote, B. A., Secretary in Entomology Department, August, 1928.

# ACTIVE PROJECTS.

#### ANALYTICAL CHEMISTRY.

Dr. E. M. Baily in charge.

- 1. Inspection of Fertilizers.
- Inspection of Feeding Stuffs. 3. Inspection of Foods and Drugs.
- Calibration of Babcock Glassware.
- Inspection of Insecticides and Fungicides.

Analysis of Diabetic Foods.

#### BIOCHEMISTRY.

Dr. T. B. Osborne, Consulting Biochemist. Dr. H. B. Vickery, Biochemist in charge.

(In Collaboration with Dr. L. B. Mendel, Yale University.)

1. Cell Chemistry.

- a. A detailed investigation of the nitrogenous constituents of plant cells including not only the protein components but also the hitherto scantily considered non-protein substances. The methods developed and successfully applied to the green leaf in the case of alfalfa are now being extended to the tobacco leaf.
- b. The investigation of the nature of the simpler nitrogenous constituents of yeast.

Protein Chemistry.

a. The methods for the determination of the basic amino acids of proteins are under investigation with the object of effecting improvements.

h Methods for the preparation of pure proteins on a large scale with the object of obtaining material for chemical and nutri-

tional study.

Nutrition Investigations.

- a. The relation of diet to the rate of growth with especial attention to certain factors which appear to determine rapid growth.
- b. The investigation of the relation of diet to opthalmia.
- c. Experiments on the relation of diet to fertility in cooperation with Dr. Mason of Vanderbilt University.

#### BOTANY.

## Dr. G. P. Clinton in charge.

The Nature and Cause of Mosaic Disease of Plants.

The Ustilaginales of North America.

The Rusts of Connecticut.

Plant Disease Survey of Connecticut.

Thielavia basicola; a Study of the Perfect Stage.

A Study of Pythiums.

- Spraying and Dusting Experiments on Apples and Peaches. (With Entomology.)
- Seed Testing.
  Peach "Yellows."

Chestnut Blight-virulence studies.

Tobacco Diseases—especially Black and Brown Root Rot. (Experiments at Tobacco Substation.)

Tree Diseases.

23. Rogueing as a Control for Raspberry Mosaic. (With U.S. D. A.)

#### ENTOMOLOGY.

# Dr. W. E. Britton in charge.

3. Spraying and Dusting Experiments on Apples and Peaches. (With Botany.)
6. Control of Foul Brood of Bees.
7. A Study of the Asiatic Beetle, Anomala orientalis.

9. Insect Survey of Connecticut.

16. Experiments with the Cabbage Maggot.

17. Life History and Methods of Controlling the Oriental Peach Moth, Laspeyresia molesta.

18. Life History of Imported Currant Worm.

- 20. Life History, Habits and Control of the Imported Birch Leaf-Miner, Fenusa pumila.
- 21. Life History and Control of the Spinach Leaf-Miner. 26. Experiments on the Control of Squash Vine Borer.

#### Control Projects.

10. Inspection of Orchards and Nurseries. Control of Gipsy Moth.

Elimination of the Mosquito Nuisance in Salt Marshes.

Inspection of Apiaries.

Control of the European Corn Borer. Control of the Asiatic Beetle.

Control of the Japanese Beetle.

#### FORESTRY.

#### Mr. W. O. Fillev in charge.

- 1. Experimental Plantations on a Sandy Tract at Rainbow. a. Comparison of a wide variety of conifers and hardwoods.
  - b. Methods of management for those species that have survived

BULLETIN 298

- c. Studies on growth and habits of the several species. Effect of Thinning in White Pine (At Shaker Station)—Three Grades
- Effect of Thinning in Hardwoods (At Quassipaug Lake).
- 5. Distribution of Forest Planting Stock. (Under Clark-McNary Act.)
- 8. Studies of Forest Plantations (State-wide).
  - a. Comparative growth of various species.
  - b. Reasons for success or failure.
- c. Soil and other site factors necessary for success of each species.
- 10. An Investigation of the Distribution and Growth of Forest Trees as Influenced by Soil Conditions and Other Site Factors.
- 11. Coniferous Seed Bed Study to Determine. a. The value of fertilizers in seed beds.
  - b. The value of different amounts of seed.
  - c. The value of dusts and sprays in preventing damping off.

# Control Project

7. Control of White Pine Blister Rust. (With U.S. D. A.)

#### GENETICS (PLANT BREEDING)

## Dr. D. F. Jones in charge.

- 1. A Genetic Study of Hereditary Characters in Corn Involving Their Linkage Relations and Variability, with particular attention to characters directly influencing yield.
- The Effect of Inbreeding and Crossing upon Corn in Relation to Vigor, Rate of Growth, Productiveness and Variability.
- Methods for the Improvement of Naturally Cross-Fertilized Plants by Selection in Self-Fertilized Lines, with particular attention to field corn for grain and ensilage, alfalfa, and to some of the more important Vegetable Crops such as sweet corn for market gardening and canning, beets, cabbage, carrots, cucumbers, melons, onions, radish, rutabagas, squash and some Fruits such as bush fruits and strawberries.
- 4. Methods for the Improvement of Naturally Self-Fertilized Plants, with particular attention to Tobacco and Vegetable Crops such as lettuce, lima beans and tomatoes.

#### SOILS

# Mr. M. F. Morgan in charge.

- 1. What Soil Characters are Factors in Determining the Agronomic Value or Utilization of Land?
- 2. Experiments in Lawn Fertilization.

#### TOBACCO SUB-STATION AT WINDSOR

#### Dr. P. J. Anderson in charge.

- Fertilizer Experiments:
- Various Sources and Rates of Nitrogen, Phosphoric Acid and Potash.
- Experiments with Farm Manure.
- Experiments with Manure Substitutes. Tobacco Nutrition Studies: the Rôle of Nitrogen, Sulfur, Chlorine.
- Potassium, Calcium, Manganese, Boron, Magnesium. Improvement of Havana Seed Tobacco.
- Improvement of Broadleaf Tobacco. Improvement of Cuban Shade Tobacco.
- The Effect of Various Winter Cover Crops used on Tobacco Land. Brown Root Rot of Tobacco (with U. S. D. A.).
- Studies on Black Root Rot of Tobacco. Soil Reaction in Relation to Tobacco. 11.
- Preservative Treatment of Shade Tent Poles. 13.
- Tests of Wires for Shade Tents.
- The Effects of Topping and Suckering at Different Heights and Dates.
- The Effect of Stage of Picking Shade Tobacco.
- The Rôle of Humidity and Temperature in Curing Tobacco.
- Tobacco Diseases—Miscellaneous. Tobacco Insects-Miscellaneous.
- 21. A Study of the Root Development of the Tobacco Plant.

## PUBLICATIONS

#### BULLETINS

- No. 290. Fertilizer Report for 1927.
- No. 291. Report of the Director for 1927.
- No. 292. Some Insect Pests of Nursery Stock in Connecticut.
- The Quality of Vegetable Seed bought by Market Gardeners No. 293. in Connecticut in 1927.
- Report of the State Entomologist for 1927.
- No. 295. Report on Foods and Drugs for 1927.

#### TOBACCO SERIES

- No. 9. Prolonging the Life of Tobacco Shade Tent Poles. No. 10. Report of Tobacco Station at Windsor for 1927.

# CIRCULARS OF IMMEDIATE INFORMATION.

No. 61. Regulations Concerning the Transportation of Nursery Stock in the United States and Canada

# JOURNAL PAPERS.

- Mendel, Lafayette B. and Cannon, Helen C. The relation of the rate of
- growth to diet. II. Jour. Biol. Chemistry, v. 75, (1927), No. 3, p. 779. Vickery, Hubert Bradford and Leavenworth, Charles S. On the separation of Histidine and Arginine III. The preparation of Arginine. Jour. Biol. Chem., v. 75, (1927), p. 115-122. October, 1927.
- Vickery, Hubert Bradford and Leavenworth, Charles S. A note on the crystallization of free Lysine. Jour. of Biol. Chemistry, v. 76, (1928), No. 3, p. 437.

Vickery, Hubert Bradford and Leavenworth, Charles S. A note on the crystallization of free Arginine and Histidine. Jour. of Biol. Chemistry.

v. 76, (1928), No. 3, p. 701.

Vickery, Hubert Bradford and Leavenworth, Charles S. Modifications of the method for the determination of the Basic Amino Acids of Proteins. The bases of Edestin. Jour. Biol. Chemistry, v. 76, (1928), No. 3, p.

Vickery, Hubert Bradford and Leavenworth, Charles S. On the separation of Histidine and Arginine IV. The preparation of Histidine. Jour. Biol. Chemistry, v. 76, (1928), No. 3, p. 627.

Britton, W. E. Insects attacking vegetable crops in 1927. Report Conn. Vegetable Growers' Assoc. 1927. (1928), p. 66-70.

Britton, W. E. Report of Committee on Injurious Insects. Conn. Pomol. Soc. Proc., v. 37, (1927), p. 16-19.
Britton, W. E. Oil sprays and oil injury. Jour. Economic Entomology, v. 21, p. 418-421. April, 1928.

Britton, W. E. The Elm Leaf Beetle. Tree Talk, v. 8, No. 4, p. 32.

Britton, W. E. Some Insect Pests of Cultivated Plants. Garden Guide (Revision A. T. de La Mare Co.), (1928), p. 293-308. Britton, W. E. and Botsford, R. C. Anti-mosquito activities in Con-

necticut in 1927. Proceedings, Fifteenth Annual Meeting New Jersey Mosquito Extermination Assoc. (In press).

Garman, Philip. The European red mite, peach moth and plum curculio.

Conn. Pomol. Soc. Proc., v. 37, (1927), p. 28-34.
Garman, Philip. Dusting menaces fruit growers. Gleanings in Bee Culture, v. 61, p. 293-329. May, 1928.

Zappe, M. P. Fighting the apple maggot and the control of aphids. Connecticut Pomol. Soc. Proc., v. 37, (1927), p. 24-27.

Jones, D. F. Like father like son-in-law. Scientific Monthly, v. 26, (1928), p. 557-560.

# WHAT THE STATION CAN DO

Each mail brings to the station requests for information and service, the range of subjects being almost without limit. Every effort is made to comply with these requests, even though they are outside the fields under investigation. This is one of the purposes for which the library is maintained. However, some of the letters request help that requires an intimate knowledge of live stock management and the like and again we are asked to make laboratory determinations for which we do not have the equipment or staff. Therefore it is helpful to publish from time to time a list of the subjects on which we can furnish information and the kinds of samples we can accept.

The Station can furnish information on:

Fertilizers and fertilization. Soils and their management. The chemical composition of Foods, Drugs, Insecticides and Fungicides. The composition of Diabetic Foods. Insect Pests of plants and their control. Fungus and other Diseases of plants and their control. Sprays and spraying. Fruits and fruit management. Weeds and their control.

Forestry—all phases. Care of Shade Trees. Plant breeding—especially field corn, sweet corn, fruits and vegetables. Lawns, establishment and care. Bees. Mosquito Elimination. Tobacco culture.

camples and specimens that can be analyzed, tested or identified:

Fertilizers. Feeding stuffs. Foods and Drugs. Milk-except for bacterial count. Weeds and other plants. Diseased and injured plants. Soils.

The Station cannot furnish information on:

Live stock feeding and management, including Poultry. Animal diseases. Household management. Clothing. Farm management. Marketing and cooperation.

REQUESTS FOR INFORMATION ON THESE SUBJECTS SHOULD BE SENT TO THE CONNECTICUT AGRICULTURAL COLLEGE AT STORRS.

The Station cannot make analyses and examinations of: Drinking water—apply to the State Board of Health, Hartford. Milk for bacterial content-apply to the Dairy Commissioner, Hartford. Sick or dead poultry should be sent to Poultry Department, Storrs Experiment Station, Storrs, Conn.

All of which is respectfully submitted,

WILLIAM L. SLATE,

Director.

# Connecticut Agricultural Experiment Station

Nem Haven, Connecticut

# REPORT OF

# THE TOBACCO SUBSTATION

AT WINDSOR

FOR

1928

The Bulletins of this Station are mailed free to citizens of Connecticut who apply for them, and to other applicants as far as the editions permit.

# CONNECTICUT AGRICULTURAL EXPERIMENT STATION

# OFFICERS AND STAFF

	BOARD OF CONTROL											
Charles R. Treat, George A. Hopson, Wm. L. Slate, Dire Joseph W. Alsop Elijah Rogers Edward C. Schneid Francis F. Lincoln.	His Excellency, Governor John H. Trumbull, ex-officio President Charles R. Treat, Vice-President. Orange George A. Hopson, Secretary. Mt. Carmel Wm. L. Slate, Director and Treasurer. New Haven Joseph W. Alsop. Avon Elijah Rogers. Southington Edward C. Schneider. Middletown Francis F. Lincoln. Cheshire  STAFF.  E. H. JENKINS, Ph.D., Director Emeritus.											
Administration. Wm. L. Slate, B.Sc., Director and Treasurer. MISS L. M. Brautlecht, Bookkeeper and Librarian. G. E. Graham, In charge of Buildings and Grounds.												
Chemistry: Analytical Laboratory.	E. M. BAILEY, Ph.D., Chemist in Charge. C. E. SHEPARD CWEN L. NOLAN HARRY J. FISHER, A.B. W. T. MATHIS DAVID C. WALDEN, B.S. FRANK C. SHELDON, Laboratory Assistant. V. L. CHURCHLL, Sampling Agent. MRS. A. B. VOSBURGH, Secretary.											
Biochemical Laboratory.	H. B. VICKERY, PH.D., Biochemist in Charge. GEORGE W. PUCHER PH.D., Research Assistant. MISS HELEN C. CANNON, B.S., Dietitian.											
Botany.	G. P. CLINTON, Sc.D., Betanist in Charge. E. M. STODDARD, B.S., Pomologist. MISS FLORENCE A. McCORMICK, PH.D., Pathologist. Harold B. Bender, B.S., Graduate Assistant. A. D. McDonnell, General Assistant. MRS. W. W. Kelsey, Secretary.											
Entomology.	W. E. BRITTON, PHD., Entomologist in Charge: State Entomologist B. H. WALDEN, B.AGR. M. P. ZAPPE, B.S. PHILIP GARMAN, PH.D. ROGER B. FRIEND, PH.D. JOHN T. ASHWORTH, Deputy in Charge of Gipsy Moth Work. R. C. BOTSFORD, Deputy in Charge of Mosquito Elimination. J. P. JOHNSON, B.S., Deputy in Charge of Asiatic and Japanese Beetle Quarantines. MRS. GLADYS BROOKE, B.A., Secretary.											
Forestry.	WALTER O. FILLEY, Forester in Charge. H. W. HICOCK, M.F., Assistant Forester. J. E. RILEY, JR., M.F., In Charge of Blister Rust Control. MISS PAULINE A. MERCHANT, Secretary.											
Plant Breeding.	DONALD F. JONES, S.D., Geneticist in Charge. W. R. SINGLETON, S.M., Assistant Geneticist. H. R. MURRAY, B.S., Graduate Assistant. Mrs. R. A. HUNTER, Secretary.											
Soils.	M. F. MORGAN, M.S., Agronomist H. G. M. JACOBSON, M.S., Assistant Agronomist. HERBERT A. LUNT, M.S., Research Assistant in Forest Scils. DWIGHT B. DOWNS, General Assistant.											
Tobacco Sub-station at Windsor.	PAUL J. ANDERSON, PH.D., Pathologist in Charge. T. R. SWANBACK, M.S., Scientific Assistant. MISS DOROTHY LENARD, Secretary.											

# TABLE OF CONTENTS

POTASH FERTILIZER EXPERIMENTS.  How much fertilizer potash should be applied?.  Wilting due to lack of potash.  Effect of quantity of potash on burn.  Effect on chemical composition of the leaves.  What carriers of potash are best?	147 153 153 154 154
Comparison of high grade sulfate with sulfate of potash- magnesia.  Differences in chemical composition.  Effect on the burn.  Conclusions from the six-year experiment.  Comparison of sulfate, carbonate and nitrate of potash.  Tobacco Stems as a source of potash.  Effect on reaction of the soil.  Effect on chemical composition of tobacco.  Conclusions from all potash experiments to date.	156 156 157 158 159 162 164 164
The potash series.  Nitrogen series.  Lime series.  Unfertilized tobacco.	166 167 171 174 175
INFLUENCE OF THE SEASON ON COMBUSTION	176
Magnesia Hunger or Sand-Drown	178
EFFECT OF SULFUR, AMMONIUM SULFATE AND ALUMINUM SULFATE ON REACTION OF SOIL  Laboratory experiments  Field experiments	181 181 187
Urea and calurea as sources of nitrogen.  Nitrate of lime as a source of mineral nitrogen.  Comparison of single nitrogen sources.  Manure as a supplement to commercial fertilizer.  Fractional application of nitrogen.  Cover crop tests.  Rootrot resistant strains of tobacco.  Topping and suckering experiments.  Fire curing stalk tobacco.  Effect of liming the soil on composition of tobacco.  FIELD EXPERIMENTS WITH CHLORINE, SULFUR AND MAGNESIUM	190 190 191 192 192 193 194 195 195 196
EXPERIMENTS WITH CHLORINE SILLEUR AND MAGNESHIM	108

# Report of the Tobacco Substation 1928

P. J. Anderson and T. R. Swanback.

This, the seventh annual report of the Tobacco Substation is presented to the growers of Connecticut to inform them of the progress of experiments which are being conducted at this station.

On account of excessive rains, the season has not been a favorable one for the grower but in our experiments, definite progress has been registered in several of the lines of investigation and we feel that it

has been a successful year from that standpoint.

The year has again been marked by increased requests from the growers for service in visiting farms, plantations and warehouses, soil testing, seed testing, personal conferences and public talks, correspondence and preparation of articles for the press. This service and contact work is extremely important, is welcomed and will be continued to the limit of our ability but the inroads which it is making on time of the station staff which can be devoted to more fundamental research emphasizes the early necessity of increasing the research staff. Mr. J. S. Owens, Extension Crop Specialist from the Agricultural College at Storrs gave a part of his time to this work and rendered valuable assistance, especially during the curing season, but the limited time which he is able to give falls far short of meeting the needs of the situation. Valuable work along this line is also being conducted by the Hartford County Farm Bureau through the efforts of the county agent, Mr. C. D. Lewis.

Especially significant has been the establishment of the tobacco advisory committee of twelve growers representing the three types of tobacco grown in the state. This committee functions both for the station and for the farm bureau and their advice and suggestions have been helpful in guiding the work of the station and in keeping it in constant touch with the growers. The members of this committee are:

Mr. Ralph G. Tryon, Glastonbury
Chairman
Mr. S. R. Spencer, Suffield
Mr. A. T. Pattison, Simsbury
Mr. W. H. Gowdy, Hazardville
Mr. J. E. Shepard, South Windsor
Mr. R. E. Case, Granby

Mr. Louis L. Grant, Manchester Mr. T. F. Holcomb, West Granby Mr. J. E. Phelps, Suffield Mr. R. D. Steane, Hartford Mr. J. B. Stewart, Windsor Mr. S. F. Brown, Windsor Case, Granby

The annual field day was held at the station on July 30 in cooperation with the New England Tobacco Growers Association.

It was not only largely attended by our own growers but also  $b_y$  ninety growers from Pennsylvania who contributed much to the success of the meeting.

In co-operation with the Connecticut Leaf Dealers Association we also prepared an exhibit for the Connecticut State Fair during

the first week of September.

For some of our lines of investigation this report presents a complete discussion with all pertinent data tabulated. However, the report would be too lengthy if all projects were presented in such detail and it has seemed advisable to merely summarize the points of progress on the others and reserve for future bulletins the more complete presentation.

#### POTASH FERTILIZER EXPERIMENTS

The potash requirement of the tobacco crop is very high when compared with other agricultural plants. A good potash supply is not only essential to the growth of the tobacco plant but also the presence of an abundance of potash in the proper combination in the leaf is the most important factor in producing good combustion.

Potash has at least three functions in the tobacco. 1. It acts as a catalizer or condensing agent in the formation of carbohydrates and proteins; hence the plant would cease to grow if potash were not supplied. 2. It neutralizes acids which develop during the normal metabolism of the cell and removes them to older parts of the plant where they are precipitated and rendered harmless. Otherwise these acids would accumulate in the cells to such an extent as to poison them. The spots which appear on leaves starved for potash may be due to the accumulation of these acids. This may also account for the belief by some that potash makes leaves more resistant to disease. 3. Potash acts as a catalytic agent in combustion. In this role it is absolutely necessary for the type of slow combustion which we wish in tobacco. If it is absent, the leaf burns up like paper. In these important roles potassium does not function properly when combined with the mineral radicals such as chloride or sulfate, but must be present in an organic salt like malate, tartrate or citrate. From the results of several investigations it is now generally agreed that the fire holding capacity is governed largely by the abundance of organic salts of potash which are in the leaves. The potash problem then may be summed up in one question: How can we put into the plant the maximum amount of potash in these desirable organic combinations? Obviously its supply must come from the soil through the roots since there is no other way that the plant can obtain potash. There are two possible sources of potash for the roots, (1) the native potash which is normally in the soil and, (2) fertilizer potash which the grower adds to the soil. With

respect to the first source we should inquire how much is present in our tobacco soils? In what combinations? How available to tobacco plant? How fast does it become available? Are there methods of making it more available? What is the effect of various cultural practices on its availability? With respect to the second source there arise the questions: To what extent can we



Fig. 13. Curing Sheds.

increase the potash content of the plant by increasing the soil supply? What is the optimum quantity of potash to add in fertilizers? In what carriers or combinations of carriers is it supplied to the greatest advantage? Will the plant absorb more potash from one carrier than from another? To what extent are the acid radicals with which potash is combined also absorbed and what is their effect on the tobacco? What are the effects of various potash compounds on the soil and are these effects beneficial or harmful to the crop? To what extent is potash leached? Does unused potash accumulate in the soil?

Some of these questions have been answered by the experiments of the last few years, some have been panially answered. We hope to answer others before the series of experiments is closed. The present report brings together the data which have been obtained to date and shows how far we have progressed toward answering

the questions.

# HOW MUCH FERTILIZER POTASH SHOULD BE APPLIED?

It is a common practice in Connecticut to apply about 200 pounds of potash (K<sub>2</sub>O) to an acre of tobacco although some have applied much more while others have grown good tobacco on less. There are no recorded local experiments on which to base a decision as to whether as good a crop may be produced by a smaller quantity or whether better quality would be produced by a still larger quantity. According to analyses which were made in Connecticut by Jenkins (Conn. Bul. 180:7. 1914) an 1,800 pound crop of tobacco removes from the soil about 133 pounds of potash in

BULLETIN 299

leaves and stalks. If the stalks are returned, the amount removed is only 85 pounds. These figures, however, give us little basis for deciding on the amount which is most advantageous since we know on the one hand that plants may take up quantities of elements which are in excess of their requirements and on the other hand that they may not be able to obtain as large a quantity as they need at a certain period even though that quantity has actually been applied to the soil and is there at the time. The safest way of deciding would seem to be actual field tests where different quantities have been applied to the same plots through a series of years and by making accurate records of the quantity and quality of the tobacco produced.

Such an experiment was begun in 1926 with six one-fortieth acre plots on Field V of the station farm. The soil here is coarse sandy loam of the Merrimac series with coarse sandy, open subsoil subject to rapid leaching. The crop suffers here from lack of moisture during a dry year and from leaching of nitrogen in a wet year. Analyses by the Soils department show that this field contains about 35,000 lbs. of total potash per acre in the upper

8 inches of soil.

Two of the plots (K11, K11-1) received no potash in the fertilizer (except for the small quantity in the cottonseed meal and castor pomace of the formula). Two others (K12, K12-1) received 100

Composition of the three formulas was as follows:

PLOT K11. NO MINERAL POTASH

Carriers	Lbs. plant nutrient per acre				
Name	Lbs. per acre	NH <sub>8</sub>	P <sub>2</sub> O <sub>5</sub>	21.9 5.9	
Cottonseed meal Castor pomace Nitrate of soda	1463.4 588.2 212.7	120 40 40	42.4 10.6		
Precipitated bone	277.9		107.0		
Total	2542.2	200	160.0	27.8	

PLOT K12. ONE HUNDRED POUNDS POTASH

Carriers		Lbs. plant nutrient per acre				
Name	Lbs. per acre	NH <sub>3</sub>	P <sub>2</sub> O <sub>5</sub>	K <sub>2</sub> O		
Cottonseed meal Castor pomace Nitrate of soda Precipitated bone Sulfate of potash Carbonate of potash Nitrate of potash	1463.4 588.2 170.2 277.9 48.0 37.1 53.5	120 40 32 8	42.4 10.6 107.0	21.9 5.9 24.0 24.1 24.1		
Total	2638.3	200	160.0	100.0		

Carriers	Lbs. plant nutrient per acre				
Name	Lbs. per acre	NH3	P <sub>2</sub> O <sub>5</sub>	K <sub>2</sub> O	
Cottonseed meal Castor pomace Vitrate of soda Precipitated bone fulfate of potash Carbonate of potash Vitrate of potash	1463.4 588.2 107.4 277.9 114.8 88.3 132.3	120.0 40.0 20.2	42.4 10.6 107.0	21.9 5.9 57.4 57.4 57.4	
Total	2772.3	200.0	160.0	200.0	

PLOT K9. TWO HUNDRED POUNDS POTASH

ths, of potash per acre. The other two (K9, K9-1) received the standard quantity, 200 lbs. per acre. Potash was supplied in equal amounts from sulfate, nitrate and carbonate because previous experiments had shown most favorable results from this triple combination of potash salts, and also in order to minimize the effect of possible accumulation of any one acid radical.

During the first year of the experiment no differences in growth were observed. The growth on all of these plots, however, was unsatisfactory on account of the dry weather and no conclusion was drawn from the records.

However, when the same plots were treated in the same way the second year (1927), the growth throughout the season appeared smaller on the no-potash plots but no differences between the others were apparent.

When sorted, the tobacco from the no-potash plots, was found to be short, yellow and of poor quality but as between the others the differences were not large. The sorting records are presented in Table 1

TABLE I. QUANTITY OF POTASH SERIES. CROP OF 1927 ON FIELD V

otash lbs. er A.	Plot		Yield	dilit	real (if	Graind	Grade index*						
	No.	Plot	Ave.	L	M	LS	SS	LD	DS	F	В	Plot	Ave.
0	K11 K11-1	1150 1137	1144			18 18	6 11	36 41	7 4	20 16	13 10	.281	.290
200	K12 K12-1	1247 1141	1194	6 6	7 5	15 14	9 8	42 42	2 5	16 16	3 4	.368	.366
.00	K9	1152	1152	10	10	15	8	38	4	10	6	.411	.411

The Grade Index. In comparing the quality of tobacco grown on different plots it is very difficult to keep in mind the percentage of six to eight commercial grades of tobacco from one plot and compare with a like number from another. To simplify these comparisons a grade index was devised. The grade index is a single number expressing the quality of all

REPORT OF TOBACCO SUBSTATION

These data show that there was a slight reduction in yield but a very decided reduction in grading during the second year where no potash was added.

The same treatment was repeated on the same plots in 1928 (3d year). During the summer it was quite apparent that the plots which received no potash were not making as good growth as the others. Distinct symptoms of acute potash hunger, however

were not seen.

In taking down the tobacco from the shed it was found that tobacco from the no-potash plots did not come "into case", i. e., become soft and pliable, during the "damps". Adjacent plots were in excellent condition for handling while this tobacco was hard and dry and never became really ready to take down. The same was true, but to a smaller degree, with tobacco from the plots where only 100 lbs. per acre of potash had been applied.



Fig. 14. Feeding the crop. Shade growers use two tons of fertilizer to the acre.

On the sorting bench, the tobacco from the no-potash plots was found to be yellow, short, thick, "boardy", entirely lacking in elasticity and of such inferior quality generally that it was not fit to sort. Tobacco from the 100 lbs.-potash plots was somewhat

the tobacco grown on a particular plot. It is based on the percentage of carefully assorted commercial grades and the relative price value of the different grades. Although market prices vary from year to year, it was found, after consultation with experienced dealers, that the ratios of prices between the different grades are fairly constant. These adopted price relationships for the different grades are as follows:

(LS)	Short seconds (15" and	.60 .60	(DS) (F)	Long darks (19" up). Dark stemming (17") Fillers Brokes	.30 .20 .10 .10
(00)	17")		(DI)	DIORES	

The grade index of any plot is obtained by multiplying the percentage of each grade by the price in the above schedule and adding the products.

longer but showed much of the same characters as above but to a less degree. Tobacco from the "200 lbs.-potash" plots was rated as satisfactory.

The sorting records for the 1928 crop are presented in Table 2.

TABLE 2. QUANTITY OF POTASH. SERIES OF 1926. ACRE YIELD AND GRAD-ING OF CROP OF 1928

ntity		Acre	Acre Yield Percentage of grades									Grade Index	
Quantity of potash	Plot No.	Plot	Ave.	L	M	LS	SS	LD	DS	F	В	Plot	Ave.
Dolars 0	K11 K11-1	1107 1163	1135				21 21	38	67 15	27 21	5	.194	.214
100	K12 K12-1	1107 1076	1092	3 3	7 5	9 13	25 15	24 31	16 15	16 18		.321	.322
200	K9 K9-1	1178 1221	1199	13 22	9 16	12 5	18 17	31 30	8	9 9		.428	. 463

It appears from these data that the reduction in quality is much more serious than the decrease in yield when potash is omitted from the fertilizer. When 100 lbs. of potash was supplied the quality was somewhat improved but was still much inferior to the tobacco receiving the regular ration. Tobacco from the 200 lbs.-plots was of good quality exhibiting none of the dry, thick, nonelastic, yellow characters of the other plots.

In the Field V experiments just described all treatments were in duplicate. After the first year it was decided to increase the number of replications and enlarge the experiment also by adding 300 lbs.-potash plots. These additional plots were on Field I, a soil of a somewhat different type, being a fine sandy Merrimac loam with a more compact subsoil and less subject to leaching. This field usually grows better tobacco than Field V. The complete lay-out of plots was now as follows:

K11	No po	ntast	1	6	plots
K12	100 lb			5	"
K9	200	"	"	6	"
K13	300	"	"	5	"

During the first year of the new series no differences in growth were observable in the field but the sorting records as presented in Table 3 show that there was a slight reduction both in yield and grading when potash was omitted from the fertilizer.

During the second season (1928) when all treatments were repeated on the same plots on Field I, the effect of the entire omission of potash was apparent in the field. The plants appeared smaller and less "leafy", i. e., they did not seem to fill out the rows as well. The sorting records for this crop, however, do not indicate a reduction in yield. Probably the thickness of the leaves com-



Fig. 15. Transplanter at work under the tent.

Table 3. Quantity of Potash. Field I Series. Sorting Records on Crop of 1927

Potash lbs.	Plot	Acre	Yield .			Pero	centage	e of gr	ades			Grade index	
per A	No.	Plot	Ave.	L	M	LS	SS	LD	DS	F	В	Plot	Ave.
0	K11-2 K11-3 K11-4 K11-5 K11-6	1230 1213 1160 1152 1163	1184	8 7 2 3 8	9 5 4 5 7	17 25 14 19 20	6 6 6 6 7	40 36 39 42 38	3 4 6 2 2	11 11 16 13 13	6 6 13 10 5	.397 .401 .304 .345 .399	. 369
100	K12-2 K12-3 K12-4	1287 1384 1152	1274	7 9 4	8 11 7	18 18 19	4 9 7	38 35 38	4 4 2	13 10 14	8 4 9	.381 .418 .358	.386
200	K9-5 K9-6 K9-7 K9-8	1321 1312 1258 1254	1286	4 6 6 8	8 6 4 6	22 21 18 19	4 5 7 6	40 ·37 37 43	3 3 6 2	14 13 14 11	5 9 8 5	.377 .364 .358 .397	.374
300	K13 K13-2 K13-3 K13-4 K13-5	1230 1316 1325 1205 1246	1264	7 9 11 8 2	4 10 8 7 4	24 19 19 23 25	7 5 7 6 10	36 36 36 34 33	2 4 3 3 5	10 12 13 12 12	10 5 3 7 9	.391 .412 .423 .405 .354	.397

TABLE 4. QUANTITY OF POTASH. FIELD I SERIES. ACRE YIELD AND GRADING OF CROP OF 1928

Quantity	Plot	Acre	Yield		1	Percen	tage of	Grade	s		Grade	index
of potash	No.	Plot	Ave.	L	M	LS	SS	LD	DS	F	Plot	Ave.
0	K11-2 K11-3 K11-4 K11-5 K11-6 K11-7	1259 1114 1152 1185 1312 1216	1206	4 2 7 4 5 1	4 8 3 7 5 3	22 29 11 9 19 17	12 10 24 25 16 22	32 33 29 32 32 32 32	11 6 12 7 7 11	15 12 14 16 16 16	.365 .395 .351 .337 .368 .328	.357
100	K12-2 K12-3 K12-4	1202 1306 1083	1197	7 18 7	6 8 7	30 17 7	11 11 28	34 33 20	2 3 17	10 10 14	.435 .478 .346	.419
200	K9-5 K9-6 K9-7 K9-8	1403 1229 1063 1257	1238	15 8 9 25	13 9 6 10	17 29 17 13	11 9 14 11	31 32 21 26	3 5 18 4	10 8 15 11	.472 .449 .394 .518	.458
300	K13 K13-1 K13-2 K13-3 K13-4	1106 1200 1355 1234 1138	1207	13 24 5 12 11	10 12 5 10 6	22 9 27 19 19	11 12 15 13 16	26 22 36 34 19	7 9 2 2 18	11 12 10 10 10	.464 .498 .409 .449 .412	.446

pensated for their smaller size. There was, however, a decided reduction in quality when no potash was applied, the cured leaves being short, yellow, heavy and non-elastic. This is reflected in the lower grade index (Table 4). Tobacco from the 100 lbs.-potash plots was not quite as good as that from the 200 lbs.-plots. There were no significant differences between the 200 lbs. and the 300 lbs.-plots.

In general, the results from the Field I series confirm those from Field V series.

Wilting due to lack of potash. During hot days, tobacco leaves wilt and flag. During the summer of 1928 it was commonly observed on the above experiments and also in other tests on shade tobacco that the no-potash plots wilt first, and at all times the wilting is more pronounced on these plots.

Effect of quantity of fertilizer potash on burn. Strip burn tests were made on the crops of 1926 and 1927 as recorded on p. 171 These showed only a slight reduction in fire holding capacity the first year after potash is omitted, but serious reduction during the second year. One hundred pounds of potash per acre seems to have been enough to keep up the fire holding capacity, during two years.

Effect on the chemical composition of the leaves. Samples from the crop of 1926 (first year of test) were analyzed by Dr. E. M. Bailey of the chemistry department to see what effect the omission or reduction of fertilizer potash would have on the quantity of potash absorbed. The results presented in Table 5 show a consistent reduction in potash even for the first year and furnish reason to believe that the potash content of the leaves may be very materially affected by the quantity applied to the soil.

CONNECTICUT EXPERIMENT STATION

Table 5. Quantity of Potash in Leaves from Plots with Reduced Fertilizer Potash. Crop of 1926

	Percent of potash in leaves when each acre received								
Grades	No. Fert. potash	100 lbs. Fert. potash	200 lbs. Fert. potash						
Darks Seconds	6.96 7.03	7.20 7.33	7.97 8.32 8.15						
Both	7.00	7.27							

Conclusions. Despite the fact that this soil contains naturally very large quantities of potash it is obvious from these experiments that the availability of the native supply is so low that regular yearly applications in the fertilizer are necessary. When all potash is omitted from the fertilizer, the quality is slightly reduced the first year, seriously reduced the second year and the product is so inferior the third year that it is not worth sorting. The reduction in weight has not been so serious as in quality. One hundred pounds of potash per acre is not enough to keep up the quality for more than one year. As between 200 and 300 pounds, no differences have appeared in two years. Until further data are at hand it would seem best to use 200 pounds per acre although the possibility is not precluded that the minimum application may be somewhat lower or higher.

# WHAT CARRIERS OF POTASH ARE BEST?

There are a number of forms (carriers) in which potash may be added to the fertilizer mixture. Although the element, potassium, is the same from all sources, nevertheless it is a mistake to believe that equal results can be secured irrespective of the form in which it is supplied. New England tobacco growers learned, for instance, many years ago that potash could not be supplied in the form of muriate because it ruined the burn. Such differences and resultant preferences are due to the acid radicals and other undesirable companions of potassium which may either affect the soil adversely or produce unfavorable effects when they enter the plant and change the composition of the leaves. Thus sulfate of potash and double sulfate of potash-magnesia are objectionable because too

much sulfur in the plant reduces the fire holding capacity. Carbonate may cut down the yield. Nitrate may introduce too much nitrogen in the nitrate form. Cottonhull and wood ashes may have an unfavorable effect on soil because of caustic lime, etc. When one tries to make selection among them by theorizing on their possible effects he is confronted with so many conflicting possibilities that he is forced to the final conclusion that the only way of coming to a decision is through actual field tests. It is for this reason that a large number of the experimental plots on the station farm are now devoted to a comparison of different forms of potash. These experiments are being conducted under the most carefully controlled conditions we are able to maintain and every record possible is being made on them. Such experiments must ultimately answer the question proposed, at least as far as this type of soil is concerned.

In Table 6 all the sources of potash which have been used commonly in this section are listed with their essential analyses. Most of these are now included in the tests at the station or elsewhere.

Table 6. Average Analyses of Potash Carriers Which may be used for Tobacco

			Percent	age of pl	ant food		
Name of Carrier	Potash (K <sub>2</sub> O)	Nitro- gen (N)	Phos. Acid (P <sub>2</sub> O <sub>5</sub> )	Lime (CaO)	Magne- sia (MgO)	Sulf. Acid (SO <sub>3</sub> )	Chlor- ine (Cl)
Sulfate of Potash	50	mail (	W.	0.2	1.3	43.6	1.6
Nitrate of Potash <sup>1</sup>	44	12.5	11 202	A United States			0.6
Nitrate of Potash-Soda	12.0	14.5					0.7
Carbonate of Potash	64						0.5
Sulf. Potash-Magnesia	28				11.3	46.7	2.0
Wood Ashes <sup>2</sup>	6.6		2.1	36.6	5.7	1.2	0.5
Cottonhull ashes <sup>2</sup>	25		9.8	5.2	11.2	2.4	0.2
Tobacco Stems <sup>2</sup>	6.4	2.1	0.6	3.8	0.5	0.5	0.5
Cow manure <sup>3</sup>	0.5	0.4	0.3	0.2	0.1	0.1	0.1
Horse manure <sup>3</sup>	0.6	0.4	0.7	0.5	0.2	0.1	0.1

Since the purpose and progress of these experiments have been discussed in previous reports, they will not be repeated here but the present discussion will be confined to results of the last year or two and comparisons with those of the preceding years.

German synthetic or Calcutta.

<sup>&</sup>lt;sup>2</sup> Composition variable. <sup>3</sup> Containing 60-70% water.

# COMPARISON OF HIGH GRADE SULFATE WITH SULFATE OF POTASH-MAGNESIA

BULLETIN 299

The possible benefit to be derived from the substitution of sulfate of potash-magnesia (double manure salts) for the more common sulfate of potash, lies in its content of magnesia which is essential for the growth of tobacco and without which the plant suffers from the malnutrition trouble commonly called sand-drown

Table 7. Comparison of High Grade Sulfate of Potash with Double Manure Salts. Yield and Grades for Crop of 1928

Source	Plot	Acre	Acre Yield				Grade index					
potash	No.	Plot	Ave.	L	M	LS	SS	LD	DS	F	Plot	Ave.
High Grade Sulphate	K1 K1-1	1309 1280	1295	24 24	14 10	13 14	13 12	24 27	4 2	8 11	.529	. 522
Double Man. Salts	K2 K2-1	1313 1318	1315	22 16	13 9	14 21	13 13	28 29	2 3	8 9	.517	. 499
Half from Each	K3 K3-1	1341 1378	1354	19 17	8 10	19 17	11 13	28 27	4 4	01 12	.488	. 480

Just as in the preceding five years (See Tob. Sta. Buls. 5, p. 24; 6, p. 22; 8, p. 36), two of the plots (K1, K1-1) had all of their potash in the form of high grade sulfate, two more (K2, K2-1) all from double manure salts, and the other two (K3, K3-1) had the potash derived equally from the two sources. The growth in 1928 was uniform but light on account of the heavy rains during the growing season. No differences were observed during the summer between the various plots except for a trace of sand-drown on the K1 plots just before harvest but this never became of any importance. It was not observed at any time during the preceding five years. When the tobacco from these six plots was sorted, no signs of magnesia hunger were observed, all the tobacco being rated as satisfactory and of good quality.

Table 7, showing sorting results of the season of 1928, indicates a somewhat higher yield from the use of the two potash carriers in combination but a somewhat higher grade index from the use of high grade sulfate alone. Table 8, showing yields during six years of this experiment, indicates only a very slight difference (about 1%) in favor of the combination. Table 9, showing the grade index for 5 years, shows a very slight higher average for the high grade sulfate.

**Differences in chemical composition.** In order to see whether any chemical changes in the composition of the leaf had been caused by the substitution of double manure salts for high grade

TABLE 8. COMPARISON OF HIGH GRADE SULFATE WITH DOUBLE MANURE SALTS.

ACRE YIELDS FOR SIX YEARS

Source	Plot	THE STATE OF	C	rade ind	ex by yea	rs		Plot	Average 12
of potash	No.	1923	1924	1925	1926	1927	1928	Ave.	replications
High Grade Sulfate	K1 K1-1	2056 2056	1333 1387	2054 2061	1739 1832	1223 1223	1309 1280	1619 1640	1630
Double Man. Salts	K2	1966 1966	1413 1413	1932 1892	1831 1833	1355 1234	1313 1318	1635 1609	1622
Half from each	K3 K3-1	2039 2039	1467 1333	2029 1929	1712 1648	1364 1382	1341 1378	1669 1618	1638

sulfate, samples of seconds and darks for all plots were analyzed by the Station Chemistry Department. Since in double manure salts, considerably more magnesia and sulfur are added to the soil it was anticipated that a larger percentage of these elements would be found in the leaf. In view of the importance of potash in the burn, it also seemed desirable to learn whether the amount of

Table 9. Comparison of High Grade Sulfate with Double Manure Salts. Grade Indices for 5 Years

Source	Plot		Grade	index by	years		Plot	Average 12	
potash	No.	1924	1925	1926	1927	1928	Ave.	replications	
H. G. Sulfate	K1 K1-1	.281 .291	.475 .475	.471 .505	.356 .457	.529 .516	.422	.436	
Double Manure Salts	K2 K2-1	.281 .273	.476 .471	.479 .500	.468	.517 .481	.444	.433	
Half from each	K3 K3-1	.316 .270	.461 .483	.475 .461	.466 .357	.488 .472	.441	.425	

potash absorbed had been affected. Since calcium and magnesium have a somewhat complementary relation in the tobacco plant it was also decided to determine the percentage of calcium. The results of the chemical analyses for crop of 1926 are summarized in Table 10.

From these analyses it is apparent that the use of double manure salts has greatly increased the magnesia content of the leaves and correspondingly reduced the calcium. Both total sulfur and sulfate sulfur have been increased. The percentage of potash absorbed was slightly reduced, especially in the seconds.

Effect on the burn. Since it is generally conceded that burn is roughly proportional to the potash which may form combinations with the organic acids after the mineral acids(sulfuric, hydrochloric, nitric, phosphoric) have been neutralized, it would be

159

anticipated that the small increase in sulfate sulfur and reduction in potash would be reflected in a corresponding reduction in fire holding capacity. Burn tests on these same samples of 1926. published in Bul. 10, Table 5 (before the chemical analyses were made) show that such was the case. This is confirmed by tests for two more years reported on p. 168 of the present bulletin.

TABLE 10. SUMMARY OF CHEMICAL ANALYSES OF TOBACCO FROM HIGH GRADE SULFATE AND DOUBLE SULFATE PLOTS. CROP OF 1926 Averages of duplicate plots

Source	Grade		Percent	age in water	free leaf	
of · potash	of leaf	Potash (K <sub>2</sub> O)	Total Sulfur	Sulfate Sulfur	Lime (CaO)	Magnesia (MgO)
H. G. Sulfate	Darks Sec. Both	7.23 8.07 <b>7.65</b>	0.84 0.72 <b>0.78</b>	0.72 0.58 <b>0.65</b>	5.81 6.84 <b>6.32</b>	1.17 1.41 1.29
Double manure salts	Darks Sec. Both	7.05 7.54 <b>7.30</b>	1.00 0.81 <b>0.90</b>	0.87 0.69 <b>0.78</b>	4.76 5.94 <b>5.35</b>	1.97 2.28 <b>2.13</b>
Half from each	Darks Sec. Both	6.98 7.33 <b>7.16</b>	0.86 0.70 <b>0.78</b>	0.73 0.59 <b>0.66</b>	5.84 6.88 <b>6.36</b>	1.49 1.64 1.56

Character of soil. The soil on which these plots are located is a Merrimac sandy loam with some fragments of red sandstone in the surface. It has never leached seriously nor does it suffer excessively from dry weather. It produces a heavier and better crop on a relatively dry year than on a wet year. It is not the type of soil which suffers excessively from sand-drown.

Conclusions from the six year experiments. The original purpose of this experiment was to find whether any advantage would accrue from the substitution of sulfate of potash-magnesia (25% K2O) for high grade sulfate (48% K2O) as a source of potash in the tobacco mixture. At the end of six years we believe this question has been answered for this particular type of soil as nearly as it can be answered by field and laboratory tests. Two of these years were excessively wet (conducive to sand-drown), one was excessively dry, one just a little too dry, and the other two about optimum in rainfall.

When the records of the six years are averaged, the differences in yield and quality are found to be very small-probably too small to be important. Offsetting a somewhat larger yield from the use of the combination of the two carriers, there is a small advantage in grading and fire holding capacity from use of high grade sulfate. It may be stated definitely that there is no advantage in taking all the potash from double manure salts. On this

type of soil there has been no advantage in getting any of it from that source. In more sandy, "leachy" locations, however, it is conceivable that the use of 100 or 200 pounds of double sulfate per acre might result in some advantage unless there are other sources of magnesia present. If our tobacco mixtures of the future are to include a smaller amount of vegetable organics (which contain magnesia)—as seems likely from present trends it is certain that some carriers of magnesia must be included.

REPORT OF TOBACCO SUBSTATION

The disadvantages attending the use of double manure salts are (1) somewhat higher cost of the potash, (2) handling of a oreater bulk of low grade material, (3) raising the sulfur content of soil and leaf, (4) lowering the potash content, and (5) consequent reduction of fire holding capacity.

#### COMPARISON OF SULFATE, CARBONATE AND NITRATE OF POTASH

This experiment was begun in 1925. Results of first two years are given in Tobacco Sta. Bul. 6, p. 25 and Bul. 8, p. 39. The object of the experiment is to compare the effect of these three carriers of potash on the yield and quality of tobacco. Using our standard formula as a base and with all other ingredients the same in all, the following five sources of potash, or combinations of sources were used:

K1, all potash in H. G. Sulfate.

K5, all potash ir carbonate.

K7, 2-3 of the potash in nitrate\*, the other 1-3 in carbonate. K8, ½ of the potash in sulfate, ½ in carbonate.

K9, potash derived equally from sulfate, carbonate and nitrate.

The exact formula for each of these plots is tabulated in Bul. 6. The quantity of ammonia, phosphorus and potash was the same for all plots. The experiment was begun with 10 plots on Field V where the soil is very sandy and the yield is never large. Each treatment is in duplicate on these original plots.

Since growth on this part of the field is usually not as good as might be wished and since results can be obtained more quickly and with more certainty by using a larger number of replications, the experiment was enlarged in 1927 by repeating the same treatments in triplicate on Field I. These two sets of plots give us five replications of each treatment every year. It is believed that this number is sufficiently large to ensure reliable data in a few years.

It will be noted from the above that the two series of plots (series of 1925 and series of 1927) are on somewhat different types

<sup>\*</sup>In explanation of the K7 formula, it did not seem desirable to derive all the potash from nitrate because this would make a greater proportion of the nitrogen from mineral sources (in nitrate form) than we had in the other formulas and would thus introduce another variable.

of soil, the first being Merrimac coarse sandy loam with rapid drainage and therefore suffering from dry weather and from leaching, the second series on Merrimac sandy loam with a tighter subsoil giving slower drainage and therefore better for a dry year but too slow for a wet year. It is not prone to leaching. It will be most convenient to discuss these two series separately.

CONNECTICUT EXPERIMENT STATION

Series of 1925. Composition of the fertilizer mixtures applied to these plots is described in Bul. 6, p. 27. Significant differences in growth between the various plots were not observed during the summer of any of the four years of this test.

Sorting results for 1928 are presented in Table 11. These data indicate somewhat the best yield for the combination of sulfate. carbonate and nitrate and the best grading for nitrate and carbonate. The differences, however, are quite small. Carbonate alone has the lowest grading, due to the very poor showing of the single plot, K5. Yield on all plots was unusually light due to the very wet season.

TABLE 11. COMPARISON OF SULFATE, CARBONATE AND NITRATE OF POTASH. SERIES OF 1925. YIELD AND GRADING FOR CROP OF 1928

Carriers	D1-4	Acre y	ield lbs.		1		Grade Index					
of potash	Plot No.	Plot	Ave.	L	M	LS	SS	LD	DS	F	Plot	Ave.
Sulphate	K1-2 K1-3	1095 1234	1169	19 17	10 12	5 5	18 18	27 32	13 8	8 8	.449	. 447
Carbonate	K5 K5-1	1077 1190	1134	7 28	10 14	11 3	14 15	25 20	20 13	13 7	.366	. 443
<sup>2</sup> / <sub>3</sub> Nitrate <sup>1</sup> / <sub>3</sub> Carbonate	K7 K7-1	1126 1190	1158	21 17	14 15	9	11 11	28 27	6 9	11 9	.488	. 478
½ Sulphate ½ Carbonate	K8 K8-1	1139 1220	1179	14 21	9 14	14 7	17 17	20 26	17 6	9 9	.432	. 459
1/3 Carbonate 1/3 Sulfate 1/3 Nitrate	K9 K9–1	1178 1221	1199	13 22	9 16	12 5	18 17	31 30	8 1	9 9	.428	.463

In Table 12 the yield and grading data on these plots for 4 years are summarized. The low average for both yield and grading on the sulfate and carbonate plots are due to the poor showing of plots K1-2 and K5. These two plots are in one corner of the field where a building previously stood. As a result, these two plots have not grown as well as the others of this series. If, for the sake of fairness, these two plots were excluded from our calculations the average yield for the K1 plots would be 1,325 lbs. and grade index, .405. For the carbonate plots the corresponding figures would be 1,305 and .425. Carbonate would thus have the highest grade index but lowest yield, which is in accord with our previous observations. The excellent showing which the triple combination (K9) has shown both in yield and grading throughout the series leads us to favor this as the best source of potash.

TABLE 12. COMPARISON OF SULFATE, CARBONATE AND NITRATE OF POTASH. SERIES OF 1925. YIELD RECORDS AND GRADE INDEX FOR 4 YEARS.

		Ac	ere yield	by yea	rs	Ave.	Gra	de inde	x by ye	ars	Ave.
Source of potash	Plot No.	1925	1926	1927	1928	Treat- ment	1925	1926	1927	1928	Treat- ment
Sulfate	K1-2 K1-3	1418 1553	1135 1294	1099 1222	1095 1234	1256	.316	.307	.276	.449	.371
Carbonate	K5 K5-1	1425 1545	1325 1312	1089 1176	1077 1190	1267	.317	.331	. 273	.366	.373
Nitrate Nitrate Carb'ate	K7 K7-1	1434 1695	1350 1393	1100 1155	1126 1190	1305	.407	.328	.337	.488	.392
½ Sulfate ½ Carb'ate	K8 K8-1	1458 1497	1362 1403	1164 1222	1139 1220	1308	.381	.353	.322	.432	.392
1/3 Sulfate 1/3 Carb'ate	K9 K9-1	1563 1524	1372 1424	1152 1093	1178 1221	1316	.369	.364	.411	.428	.399

TABLE 13. COMPARISON OF SULFATE, CARBONATE AND NITRATE OF POTASH. SERIES OF 1927. ACRE YIELD AND GRADING OF CROP OF 1928

Carriers of	D1-4	Acre	yield			Percen	tage of	grade	5		Grade	index
potash	Plot No.	Plot	Ave.	L	M	LS	ss	LD	DS	F	Plot	Ave.
Sulfate	K1-4 K1-5 K1-6 K1-7 K1-8 K1-9	1410 1335 1476 1366 1356 1442	. 1397	13  3 3 10 6	16 3 6 5 9 6	16 39 36 33 26 31	12 11 9 11 11 10	21 35 34 32 31 27	12 3 2 5 2 7	10 9 10 11 11 11 13	.455 .405 .425 .408 .511 .420	. 437
Carbonate	K5-2 K5-3 K5-4	1320 1261 1419	1333	11 16 15	5 10 13	24 21 24	16 8 9	20 32 30	15 2 1	9 11 8	.431 .481 .499	.470
Nitrate Carbonate	K7-4	1394 1330 1416	1380	6 20 0	8 11 4	26 18 37	14 11 15	28 29 24	8 2 9	10 9 11	.416 .507 .392	. 438
Carbonate Sulfate	K8-4	1331 1391 1445	1389	2 3 12	4 5 9	40 28 22	11 19 15	25 25 31	6 9 3	12 11 8	.416 .389 .458	. 421
Carbonate Sulfate Nitrate	K9-2 K9-3 K9-4	1194 1353 1495	1347	3 5 9	3 6 11	28 37 26	21 8 12	27 31 30	7 3 2	11 10 10	.385 .441 .452	. 426

REPORT OF TOBACCO SUBSTATION

163

Series of 1927. This series is a duplicate of the 1925 series (three additional replications of each treatment) on Field 1 as explained above. The fertilizer treatment for 1928 was identical with that for the previous series but ground limestone at the rate of 400 lbs. per acre was added because this soil was considered too acid for best results. Growth was satisfactory and uniform but yield was much reduced by the extremely wet season.

Sorting results on this series, presented in Table 13, indicate slightly highest yield for the sulfate plots and best grading for the carbonate plots.

Summary of results for two years in Table 14 also show the best grading for the carbonate plots but the lowest yield. The triple combination (K9) had somewhat the best average yield.

Table 14. Comparison of Sulfate, Carbonate and Nitrate of Potash. Series of 1927. Summary of 2 Years

Source			Acre yield		Average		Frade inde	ex	Average
of potash	Plot No.	1927	1928	Ave.	for Treatment.	1927	1928	Ave.	for Treatment
Sulfate	K1-4 K1-6 K1-8	1273 1261 1276	1410 1476 1356	1342 1318 1316	1342	.394 .372 .345	.455 .425 .511	.425 .399 .428	. 417
Carbonate	K5-2 K5-3 K5-4	1230 1246 1307	1320 1261 1419	1275 1254 1363	1297	.410 .394 .421	.431 .481 .499	.421 .438 .460	.439
2/3 Nitrate 1/3 Carbonate	K7-2 K7-3 K7-4	1271 1250 1318	1394 1330 1416	1333 1290 1362	1330	.419 .380 .326	.416 .507 .392	.418 .444 .359	. 407
Sulfate	K1-5 K1-7 K1-9	1250 1258 1320	1335 1366 1442	1293 1312 1381	1329	.389 .413 .380	.405 .408 .420	.397 .411 .400	. 403
½ Sulfate ½ Carbonate	K8-2 K8-3 K8-4	1319 1280 1345	1331 1391 1445	1325 1336 1395	1352	.396 .369 .463	.416 .389 .458	.406 .379 .461	.415
1/3 Sulfate 1/3 Carbonate 1/3 Nitrate	K9-2 K9-3 K9-4	1292 1319 1284	1394 1353 1495	1343 1336 1390	1356	.414 .365 .403	.385 .441 .452	.400 .403 .428	.410

#### TOBACCO STEMS AS A SOURCE OF POTASH

Tobacco stems (midribs of the leaf) contain about 6% of potash and for many years have been used by some growers for tobacco land. Besides potash, they also contain around 2.1% nitrogen, .5% phosphoric acid and about 4% of lime. They also add organic matter to the soil.

In order to see whether they can be used to advantage as the only source of potash, three plots were included in the potash series of Field 1 in 1927. The formula for these plots was as follows:

		Nutrients per acre							
Carrier	Pounds per acre	NH <sub>3</sub>	P <sub>2</sub> O <sub>5</sub>	K <sub>2</sub> O					
Stems Cottonseed meal Nitrate of soda	2810 1060 212.7	73.1 86.9 40.0	14 30.7	179.8 20.2					
Total		200	44.7	200					

The stems were applied at the same time as the fertilizer in 1927. The yield and sorting data are presented in Table 15.

TABLE 15. YIELD AND SORTING DATA ON STEMS PLOTS. CROP OF 1927

No.					Percentage of grades								
of plot	Plot	Ave.	L	M	LS	SS	LD	DS	F	В	Plot	Ave.	
K14 K14-1 K14-2	1222 1186 1390	1266	9 4 12	7 4 7	12 25 19	8 5 7	40 39 41	4 3 1	13 12 9	7 8 4	.376 .372 .435	. 394	

The replicates in this series are not very consistent since the K14-2 plot was very much higher than the others both in yield and quality. The average of both yield and grade index is about the same as for the sulfate of potash plots (K1-5, K1-7, K1-9) which were immediately adjacent to them.

This series of three plots on Field 1 was continued in 1928, the fertilizer being the same except for the addition of 400 lbs. of limestone per acre. The stems were plowed under in the spring.

The sorting data for 1928 are presented in Table 16 along with corresponding data from the sulfate and carbonate plots for comparison. From these data it is apparent that there has been as good yield and quality as where other sources of potash have been used. The grading is particularly good.

Table 16. Stems as a Source of Potash. Acre Yield and Grading of Crop of 1928

Source	1	Tiere y		yield Percentage of grades						Grade index		
Potash	Plot No.	Plot	Ave.	L	M	- LS	SS	LD	DS	F	Plot	Ave.
Stems	K14	1292	::::	11	11	24	14	24	8	8	.498	400
	K14-1 K14-2	1369 1378	1346	i.5	iż	ii8	12	31	4	8	.475	.486
Sulfate	6 plots		1397							***		.437
Carbonate	3 plots		1333				11.					.470

EFFECT OF DIFFERENT POTASH CARRIERS ON REACTION OF THE SOIL

In order to see whether the continuous application of any one or any combination of these potash carriers would shift the reaction of the soil, samples were taken from the ten plots in the 1925 series, (1) before sowing the fertilizer each year and (2) just after harvesting the crop. The reaction was determined electrometrically with results as follows:

TABLE 17. SOIL REACTION (pH) OF POTASH PLOTS. SERIES OF 1925

Source	AND THE	19	25	19	26	19	27		1928	
of potash	Plot No.	May	Sept.	May	Aug.	May	Sept.	May	July 13*	Aug. 7
Sulfate	K1-2	6.02	5.84	6.03	5.73	6.08	5.94	5.81	5.74	5.79
	K1-3	5.09	5.00	5.21	4.65	5.10	4.75	5.08	4.67	4.82
Carbonate	K5	5.53	5.45	5.72	5.45	5.89	5.42	5.59	5.51	5.39
	K5-1	5.21	5.25	5.31	5.11	5.32	5.25	5.24	4.99	5.02
% Nitrate % Carbon.	K7	5.31	5.40	5.47	4.93	5.62	5.09	5.31	5.04	5.03
	K7-1	5.30	5.03	5.12	4.41	5.19	4.71	5.10	4.93	4.87
½ Sulfate	K8	5.26	5.27	5.41	4.93	5.54	5.29	5.40	5.05	5.23
½ Carbon.	K8-1	5.10	5.05	5.04	5.19	5.18	4.81	5.08	4.92	4.96
1/3 Sulfate 1/3 Carbon. 1/3 Nitrate	K9 K9-1	5.05 5.05	5.15 5.12	5.19 4.95	5.02 4.84	5.47 5.33	5.06 5.05	5.16 5.02	5.09 4.95	5.15 5.00

<sup>\*</sup> Crop two-thirds grown.

Although the reactions have varied considerably during the four years of this test, there seems to be no definite trend in any one direction for any of the plots. Comparing the reactions of May, 1925 with those of May, 1928, they are found to be almost identical in all cases. The result is the same when the reactions of September, 1925 are compared with those of August, 1928. We may conclude from this that when such potash carriers in these quantities are applied annually to this type of soil there is very little if any permanent change in reaction.

# EFFECT OF POTASH CARRIERS ON THE CHEMICAL COMPOSITION OF THE TOBACCO

For reasons previously mentioned it is desirable to get the plant to absorb as much potash as possible and to keep the sulfur content low. In order to see whether the potash carrier has any effect on the quantity of these elements absorbed, samples from the ten plots of the series of 1925 were analyzed by the Station Chemistry Department. Two grades, seconds and darks of the crop of 1926,

were taken from each plot. Summary of the analyses are presented in Table 18.

TABLE 18. POTASH AND SULFUR CONTENT OF TOBACCO TREATED WITH DIFFERENT CARRIERS OF POTASH. AVERAGES OF DUPLICATES

Source of potash	Grades	Total Sulfur	Potash
Sulfate	Darks	0.57	8.31
	Seconds	0.57	8.66
	Both	<b>0.57</b>	<b>8.49</b>
Carbonate	Darks	0.50	7.76
	Seconds	0.43	8.47
	Both	<b>0.46</b>	<b>8.13</b>
Nitrate Carbonate	Darks Seconds Both	0.43 0.43 <b>0.43</b>	8.38 8.62 <b>8.50</b>
1/2 Carbonate 1/2 Sulfate	Darks Seconds Both	0.53 0.51 <b>0.52</b>	8.09 8.18 <b>8.13</b>
1/3 Sulfate	Darks	0.57	7.97
1/3 Carbonate	Seconds	0.51	8.32
1/3 Nitrate	Both	<b>0.54</b>	<b>8.15</b>

It will be noted in this table that the most sulfur is found in the leaf when sulfate of potash is used as the source of potash. The least amount of sulfur is found when nitrate or carbonate are used. Increase in the sulfur content of leaves following increase in fertilizer sulfur has been noted in our previous experiments at this station (Bul. 10, p. 46) as well as by other investigators there mentioned. The differences in the potash content of the leaves are perhaps too small to be of importance.

#### CONCLUSIONS FROM ALL POTASH EXPERIMENTS TO DATE

- 1. There are about 35,000 pounds of potash per acre in the upper eight inches of soil on the experiment station farm. This soil is typical of a large part of the tobacco section.
- 2. This supply of potash is not sufficiently available to meet the needs of a tobacco crop and yearly additions of fertilizer potash are necessary.
- 3. When all potash is omitted from the fertilizer, the grading of the tobacco is affected, beginning with a decline the first year and rendering the leaves worthless in about three years.

4. Such leaves are thick, dry, non-elastic (boardy) yellow and of very inferior quality generally.

BULLETIN 290

- 5. One hundred pounds of potash per acre is not sufficient  $t_0$  maintain the quality of the leaf.
- 6. Up to the present there has been no benefit from raising the quantity of potash to 300 pounds in the fertilizer.
- 7. Reducing the potash has not seriously affected the acre yield of tobacco.
  - 8. It has had an injurious influence on the fire holding capacity.
- 9. Reducing the quantity of potash in the fertilizer has materially reduced the percentage of potash in the leaf.
- 10. The substitution of sulfate of potash-magnesia for high grade sulfate of potash has been of no advantage on the soil where tested but may be beneficial in preventing sand-drown in lighter soils.
- 11. Its use has lowered the grading and fire holding capacity. It has not increased the yield.
- 12. It increased greatly the magnesia content of the leaves but reduced the calcium and to a less degree, the potash.
  - 13. It increased the sulfur content of the leaves.
  - In comparing sulfate, carbonate and nitrate of potash:
- 14. Carbonate gave the best grading but the lowest yield. It also gave the best fire holding capacity.
- 15. Nitrate was satisfactory from all standpoints but should not be used in excess on account of its high nitrate content.
- 16. Most consistently good results were secured from a combination of the three sources of potash, deriving one-third of the required potash supply from each carrier.
- 17. The quantity of potash absorbed has not been influenced by the carriers used.
- 18. The use of sulfate has increased the sulfur content of the leaves. The use of carbonate or nitrate gave a smaller quantity of sulfur in the leaves.
- 19. The reaction of the soil has not been changed perceptibly by using any of these carriers for four years.
- 20. Tobacco stems have also been a favorable source of potash for tobacco.

# EFFECT OF FERTILIZERS ON THE COMBUSTION OF TOBACCO

In the Report of the Tobacco Station for 1927 (Tob. Sta. Bul. 10) an introductory article on this subject was published and all data on the crops of 1925 and 1926 were presented. The present report is a continuation of that account and gives results of burn tests on the crop of 1927.

When the Havana Seed crop of 1927 was sorted, sample hands of the four grades were taken from each plot, labelled and fermented for six to eight weeks in the sweat room of the W. S. Pinney warehouse in Suffield. The case of samples was left undisturbed to undergo a natural sweat at the warehouse until October, 1928 when it was taken to the Station where strip burn tests were made

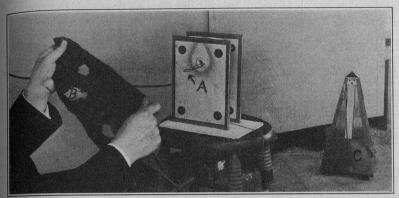


Fig. 16. Making strip tests on combustion. A, Electrically heated coil for ignition. B, Holes burned in leaf. C, Metronome for timing.

with an electrically heated filament. Five leaves were tested from each hand, each in four different places making a total of 20 tests from each hand. The leaves were just moist enough to handle well without breaking. The leaf was ignited near the midrib and held in such position that the flame progressed straight upward between the lateral veins. A metronome was used for counting and 60 seconds was considered the maximum, i. e., the count was stopped at 60 even though the leaf continued to burn.

The crop as a whole had an unusually long fire holding capacity. On most of the plots so many of the tests ran to 60 seconds that the averages showed only very small differences. The unusually long duration of burn—longer than any station crop yet tested—is probably due to the wet growing season of 1927.

#### THE POTASH SERIES

There are five series of potash plots, each of which will be discussed separately.

High Grade Sulfate of Potash vs. Double Sulfate of Potash-Magnesia. The crop of 1927 was the fifth consecutive crop raised on these six plots. Burn records for the two preceding crops are recorded and discussed in Tobacco Station Bul. 10, p. 30. Burn records for the 1927 crop and three year average are presented in Table 19 below.

CONNECTICUT EXPERIMENT STATION

Table 19. Sulfate of Potash vs. Sulfate of Potash-Magnesia. Strip Burn Tests for 1927 Crop

		Duration of burn						4
Plot No.	Source of potash	Darks	Mediums	Lights	Seconds	Ave. of all grades	Both plots	3 year averag
K1 K1-1	High grade 59.5 58.4 sulfate 55.0 59.4		54.6	58.8 57.8	58.9 56.7	57.7	41.2	
K2 K2-1	Sulfate of Pot. Magn.	59.9 52.1	58.8 56.7	57.3	52.6 59.2	57.1 56.3	56.7	36.6
K3 K3-1	One half from each	55.0 56.9	58.8 59.3	59.5 55.8	60.0 57.2	55.8 57.3	56.6	38.7

The fire holding capacity of all grades on all six plots was very high and the differences are probably too small to be significant. Comparing the three year averages (each figure representing 480 tests) there appears to be a small but constant difference in favor of high grade sulfate. It is questionable whether this difference is sufficiently large to offer serious objection to the use of double manure salts.

Table 20. Carbonate, Nitrate and Sulfate of Potash. Series of 1925. Strip Burn Tests for 1927 Crop

				Duration	of burn in	seconds.		98
Plot No.	Source of potash	Darks	Mediums	Lights	Seconds	All grades	Ave. for treat-ment	3 year average
K1-2 K1-3	Sulfate	58.2 51.5	54.3 47.7	56.6 54.3	48.3 49.5	54.4 50.7	52.6	44.2
K5 K5-1	Carbonate	54.9 58.0	56.6 49.8	54.8 59.6	53.5 59.3	55.0 56.7	55.8	49.9
K7 K7-1	2/3 Nitrate 1/3 Carbon.	55.5 56.3	55.5	48.1 49.7	57.2 49.7	54.1 51.9	53.1	45.9
K8 K8-1	½ Sulfate ½ Carbon.	55.3 53.5	56.4 56.9	57.4 56.7	53.4 53.9	55.6 55.3	55.4	45.2
K9 K9-1	1/3 Sulfate 1/3 Carbon. 1/3 Nitrate	59.6 59.2	50.2	50.1 58.0	58.8 52.1	54.7 56.4	55.4	47.8

Comparison of carbonate, sulfate and nitrate of potash. Ten of these plots were started in 1925, the other 22 in 1927. Burn tests on the series of 1925 are recorded in Bul. 10, p. 32. No previous report on the 1927 series has been made. Strip burn tests for the 1927 crop are presented in Table 20 and Table 21.

REPORT OF TOBACCO SUBSTATION

The results agree with those of the two previous years. The carbonate plots have a slightly higher fire holding capacity than the others while the sulfate tobacco is lowest. These differences, however, are quite small. It will be noted that the heavier grades have a somewhat longer fire holding capacity which in general is not in agreement with results of previous years. This is partly due to the fact that the light grade leaves were so thin that the line of fire consumed to the margin of the leaf before 60 seconds elapsed, thus lowering the average.

TABLE 21. CARBONATE, SULFATE AND NITRATE OF POTASH. SERIES OF 1927. STRIP BURN TESTS FOR 1927 CROP

Source			Du	ration of	burn in s	econds	
of potash	Plot No.	Darks	Med'ms	Lights	Seconds	All grades	Average for treatment.
Sulfate	K1-8 55.1 K1-9 58.1 K5-2 56.4		51.7 49.5 55.8 54.3 56.9 57.2	48.3 46.6 47.3 55.6  53.4	56.9 55.9 58.5 57.6 	52.3 46.5 54.8 55.3 56.0 56.6	53.4
Carbonate	K5-2 K5-3 K5-4	56.4 58.6 51.9	60.0 59.2 52.8	58.7 59.8 46.9	58.9 58.0 51.2	58.5 58.9 50.7	56.0
Nitrate Carbonate	K7-2 K7-3 K7-4	59.0 55.8 59.3	57.7	51.7 55.2 55.6	58.0 54.6 55.7	56.2 55.8 56.9	56.2
½ Sulfate ½ Carbonate	K8-2 K8-3 K8-4	56.7 52.7 56.3	54.3 53.8 59.9	57.7 51.7 56.8	59.7 59.2 59.3	57.1 54.3 58.1	56.5
Sulfate Carbonate Nitrate	K9-2 K9-3 K9-4 K9-5 K9-6 K9-7 K9-8	51.2 53.9 51.6 57.3 56.6 52.6 58.7	54.6 51.3 49.6 57.7 52.0 56.1 58.1	47.0 56.9 54.7 55.8 51.3 58.3 54.8	57.5 54.8 56.3 59.5 57.2 58.6 55.5	55.1 54.2 53.1 57.6 54.3 56.4 56.8	55.4

As shown in Table 21, the fire holding capacity of the series of the was somewhat the lowest on the sulfate plots but practically same in all the others.

Tobacco Stems. In the 1927 series of potash plots, there were three plots which received all their potash from tobacco stems.

These were the N4 plots of previous years (1922-26) which received the largest application of sulfate of ammonia and consequently had the poorest burn (Bul. 10, p. 24). It is not unlikely that there was a carry-over effect from previous treatment and this should be kept in mind in judging the tests of at least the first year.

Results of the tests on the 1927 crop are presented in Table 22

TABLE 22. STEMS PLOTS. STRIP BURN TEST ON CROP OF 1927

	Duration of burn in seconds												
Plot No.	Darks	Med'ms	Lights	Seconds	All grades.	Average for treatment.							
K14 K14-1 K14-2	51.1 45.9 56.5	52.3 59.6 55.9	45.8 57.1 52.5	53.4 58.2 57.5	50.7 55.2 55.5	53.8							

The average burn of all (53.8 seconds) is practically the same as for the adjacent sulfate of potash plots.

Quantity of potash. In order to determine the optimum quantity of potash which should be applied in the fertilizer, six plots were started in 1926. The 1927 crop was therefore the second one. Burn tests for this year, presented in Table 23, show that there has been a decided drop in fire holding capacity where no potash has been applied for two years but that otherwise 100 pounds per year has been just as effective as 200 pounds in keeping up the burning quality. Whether this will be true for a longer period of years is yet to be tested.

TABLE 23. QUANTITY OF POTASH, SERIES OF 1926. STRIP BURN
TESTS ON THE 1927 CROP

		Duration of burn in seconds.  Average of												
Quantity of potash	Plot No.													
None	K11 3 K11-1 3				34.2 36.3	35.6 37.0	36.3							
100 lbs.	K12 K12-1	57.1 55.6	58.8 53.8	57.1 55.9	58.4 60.0	57.8 56.3	57.0							
200 lbs.	K9 K9-1	59.6 59.2	50.2	50.0 58.0	58.8 52.1	54.5 56.8	55.6							

Comparison of Table 23a with Table 23 shows that the reduction in fire holding capacity was not serious the first year.

TABLE 23A. QUANTITY OF POTASH; SERIES OF 1926. STRIP BURN TESTS ON THE CROP OF 1926

			Du	ration of	burns in se	econds	
Quantity of potash	Plot No.	Darks	Med'ms	Lights	Seconds	Aver All grades	age of Both plots
None	K11 K11-1		31.9 45.3	49.4 41.4	40.8 52.9	36.3 41.4	38.9
100 lbs.	K12 K12-1	22.0 18.1	46.4 35.9	50.6 57.2	40.6 48.8	39.9 40.0	39.9
200 lbs.	K9 K9-1	11.5 41.5	59.0 42.2	58.9 56.9	53.6 41.7		45.2

In 1927 the series was extended by the addition of 18 plots on Field 1. Burn tests for the first year (presented in Table 24) show only a small reduction in fire holding capacity for the first year where all carriers of potash were omitted.

TABLE 24. QUANTITY OF POTASH; SERIES OF 1927. STRIP BURN
TEST FOR 1927 CROP

Pounds			D	uration o	of burn in	seconds	
of potash	Plot No.	Darks	Med'ms	Lights	Seconds		age of  Treatment
None	K11-2 K11-3 K11-4 K11-5 K11-6 K11-7	K11-3 K11-4 K11-5 K11-5 K11-6 53.6		51.5 56.2 60.0 52.1 58.1 54.3	57.4 58.0 54.4 49.7 55.1 52.2	54.5 51.5 56.5 52.9 54.8 53.1	53.9
100 lbs.	K12-2 K12-3 K12-4	49.7 53.9 58.2	60.0 57.6	58.5 58.6 58.0	58.3 58.8	55.5 57.8 57.9	57.2
200 lbs.	K9-5 K9-6 K9-7 K9-8	57.3 56.6 52.6 58.6	52.7 52.0 56.1 58.0	55.8 51.3 58.3 54.8	59.5 57.2 58.6 55.5	57.6 54.3 56.4 56.7	56.2
300 lbs.	K13 K13-2 K13-3 K13-4 K13-5	60.0 56.9 58.5 56.9 51.0	57.8 57.0 55.0 58.0 58.0	58.7 56.0 55.5 59.0 57.8	52.0 59.4  57.1	57.1 57.6 56.3 58.0 56.0	56.9

#### NITROGEN SERIES

In 1927 we had four sets of plots where nitrogen fertilizers were under comparison. Each is discussed separately below.

Urea plots. Burn tests for previous years on some of these plots were discussed in Bulletin 10, p. 29. Three plots received all their nitrogen from urea, three received one-half of their

nitrogen supply from urea, while three others, which received none of their nitrogen from urea, served as controls. Total quantity of nitrogen as well as the other food elements were the same on all

CONNECTICUT EXPERIMENT STATION

Strip burn tests on the crop of 1927, presented in Table 25 below indicate a slight reduction in fire holding capacity when the entire nitrogen supply was from urea but there was no reduc-

TABLE 25. UREA PLOTS. STRIP BURN TESTS ON CROP OF 1927

Plot				Duratio	on of burn i	n seconds.		
No.	Amount of urea	Darks	Mediums	Lights	Seconds	All grades	Treat- ment	3 yea
N1-5 N1-6 N1-7	None 53.6 58.8 60.0 57.3 58.3 .56.7		55.8  59.1	56.4 60.0 58.9	56.3 59.1 58.2	57.7	33.1	
N8 N8-1 N8-2	½ nitrogen from urea	ogen 58.9 56.3 urea 56.9 55.9 57.3 50.4		57.7 57.4 57.2	59.9 58.7 58.7	58.2 57.2 55.9	57.1	34.1
N9 N9-1 N14	All nitrogen from urea	45.2 53.1 52.6	57.7 55.4 59.8	50.3 55.9 53.5	59.5 58.6 47.3	53.2 55.8 53.3	54.1	31.9

tion when one-half of the supply was from urea. The differences, however, are small and would probably not constitute serious objection to a full urea formula if it were desirable from all other standpoints.

Single sources of nitrogen. On these plots (started in 1926) six different nitrogen carriers, each used in turn as the only source of nitrogen in the formula, are under comparison. Tests of the previous year are discussed in Bulletin 10, pp. 26 and 60. The strip tests of the 1927 crop presented in Table 26, show that sulfate of ammonia gave the lowest fire holding capacity. In this and other

TABLE 26. SINGLE SOURCE OF NITROGEN PLOTS. STRIP BURN TESTS ON **CROP OF 1927** 

				Duration	of burn in	seconds.	
Plot No.	Sources of potash	Darks	Mediums	Lights	Seconds	All Grades	Ave. of treat-ment
N11 N12 N13 N14 N22 N23 N24	C. S. Meal Nitrate Soda Sulfate Am. Urea Nitr. Lime Nitr. Lime Castor Pom.	50.6 56.1 44.3 52.6 55.2 55.4 57.7	59.8	53.5	55.1 59.9 55.6 47.3 59.7 59.7 59.7	52.9 58.0 49.9 53.3 57.5 57.5	52.9 58.0 49.9 53.3  57.5 58.7

tests of the same year it has been noted that the depression of fire holding capacity from the use of sulfate of ammonia was not so pronounced as in the two previous years. The heavy rainfall of 1927 may have had an influence. The fire holding capacity of the nitrate of soda plot is very high again in 1927, as it was in the preceding year. The same was true of the nitrate of lime plots. All of these plots showed distinct signs of nitrogen starvation during the latter part of the wet growing season and the cured tobacco was of poor quality. The long fire holding capacity may be due to the absence of hindering nitrogen compounds in the leaf rather than to any specific effect of the nitrates of lime and soda. Contrary to the popular impression, castor pomace had no unfavorable influence.

Concentrated formula. The trend of recent years is toward a very concentrated fertilizer. This has obvious advantages provided the quality and yield of the crop are not adversely affected. The concentrated formula (18-14-18) described in Bul. 6, p. 16, was slightly modified in 1927 and was composed as follows:

#### CONCENTRATED FORMULA (PER ACRE)

Urea									286	1bs.
Nitrate of potash									250	1bs.
Carbonate of potash.										
Carbonate of magnesi										

This extremely concentrated formula (28.5-0-28.5) contained no phosphorus because experiments on this field indicated no benefit from use of phosphorus.

Since no previous report on the burn tests of tobacco from these plots has been made, records for the crops of three years are presented in Table 27. There is a small reduction in fire holding

TABLE 27. CONCENTRATED FERTILIZER. STRIP BURN TESTS ON CROPS OF 1925-26-27

		STATE OF				Dura	ation o	f burn	in sec	onds					
Plot No.		Darks		Mediums		Lights		Seconds			3 yr. ave.				
		1925	1926	1927	1925	1926	1927	1925	1926	1927	1925	1926	1927	Plot all Gr.	Tre- at- ment
7-4	Stand	$\frac{1}{49.3}$ $\frac{1}{49.0}$	9.7	$54.4 \\ 50.9$	$\frac{1}{48.1}$ $\frac{1}{42.1}$	$23.3 \\ 33.0$	$59.9 \\ 58.4$	$49.4 \\ 41.9$	$56.8 \\ 45.9$	$59.2 \\ 59.4$	56.1 48.1	$52.4 \\ 49.0$	59.0 57.9	$\frac{48.2}{45.5}$	46.8
N10 N10-1		$\frac{34.8}{32.2}$	6.9 $19.1$	57.4 50.7	$\frac{44.0}{21.8}$	$\frac{32.2}{13.0}$	$\frac{56.8}{44.5}$	$\frac{40.5}{44.3}$	47.3 47.3	$53.6 \\ 54.2$	$\frac{46.9}{40.2}$	$\frac{45.0}{45.2}$	59.6 47.1	$\frac{44.0}{38.3}$	41.0

capacity due to the concentrated formula. Since, however, it is not certain just which constituent of this fertilizer is responsible, one would not be justified in concluding that other concentrated mixtures would have the same effect.

Nitrate of soda vs. nitrate of lime. The object of this series of eight plots was to compare nitrate of soda with nitrate of lime as the source of mineral nitrogen of the fertilizer. In the first comparison one-fifth of the nitrogen was in mineral form, while in the second, one-half was in mineral form. The series was begun in 1927. Results of the burn tests of the first crop, presented in Table 28, indicate that the burn was excellent on all plots and there were no significant differences where the two kinds of nitrate were compared.

BULLETIN 299

Table 28. Nitrate of Lime Plots. Strip Burn Tests on 1927 Crop

			Duration of burn in seconds								
Plot No.	Nitrogen Source	Darks	Med'ms	Lights	Seconds	All grades	Average of treatment				
N1-8	½ N. in	47.0	52.7	58.5	55.9	53.5	55.8				
N1-9	Nitr. Soda	58.4	57.3	59.0	57.4	58.0					
N16	% N. in	60.0	58.5	59.2	60.0	59.4	59.2				
N16-1	Nitr. Lime	58.1	59.8	58.7	60.0	59.1					
N2-3	½ N. in	58.8	54.4	56.4	58.9	57.1	57.0				
N2-4	Nitr. Soda	52.0	59.8	59.3	56.7	57.0					
N18 N18-1	½ N. in Nitr. Lime	59.8	59.2 58.3	56.3 55.9	58.9 58.9	58.5 57.7	58.2				

#### LIME SERIES

Field VIII was heavily limed each year up to 1925. Since the soil was then found to have a reaction somewhat above neutral no more was applied. Strip tests on the crops of 1925 and 1926 are recorded in Bulletin 10, p. 33. Tests on the 1927 crop, recorded below in Table 29 agree with those of preceding years in showing a

Table 29. Lime Plots. Strip Burn Tests on Field VIII Tobacco Compared with No-Lime Plots. Crop of 1927

		Duration of burn in seconds								
Plot No.	Lime treatment	Darks	Med'ms	Lights	Seconds	All grades	Treatment average			
L1 L2 L3 L4	Lime	47.1  46.0 55.3	42.6		34.0 54.6 43.5 42.7	40.6 54.6 44.8 46.9	45.7			
N1-5 N1-6	No Lime	53.6 60.0	58.8 57.3	55.8	56.4 60.0	56.3 59.1	57.3			

distinct reduction in fire holding capacity where lime is heavily applied.

On another series of plots on Field I where the effect of different fertilizers on black rootrot was being tested, one-half of each plot was heavily limed each year beginning with 1924. Burn tests for 1925 and 1926 were recorded in Bul. 10, p. 33. Those for the 1927 crop, given in Table 30 below, show some depressing effect of lime

TABLE 30. LIME PLOTS. STRIP BURN TESTS ON LIMED AND UN-LIMED ENDS OF THE BLACK ROOTROT SERIES. CROP OF 1927

			Du	ration of	burn in se	econds	
Plot No.	Lime treatment	Darks	Med'ms	Lights	Seconds	Treat- ment Ave.	Three year Average
Tla Tlax	Lime	34.1			7.5	20.8	
T1b T1bx	No Lime	52.9 53.6	57.5 43.9	57.2	56.8 57.0	54.1	
T2a T2ax	Lime	55.4			34.2 42.3	43.9	
T2b T2bx	No Lime	60.0 60.0	59.4	56.3	58.8 58.7	58.9	
T3a T3ax	Lime	37.4 42.2			42.8 22.9	36.3	
T3b T3bx	No Lime	59.7 59.9	59.5 60.0	58.1 59.7	59.4 58.3	59.2	
Ave	erage of all lime perage of unlimed 1	lots	· · · · · · · · ·			35.4 56.5	26.5 39.8

when measured by the strip test. In this table it will be noted that the acid fertilizer (T1) still further reduced the fire holding capacity. This effect, which is probably due to sulfate of ammonia, was also very evident in the preceding years.

Reasons for the depressing effect of lime are discussed on p. 198.

#### UNFERTILIZED TOBACCO

There are two small plots on the station farm where tobacco has been grown continuously since 1924 but without the addition of any fertilizer. The tobacco grown here has been short and very inferior, in fact, not worth harvesting. In order to see what effect the fertilizer as a whole has on burn, the crops of 1926 and 1927 were tested. The results, presented below in Table 31, show that

the fire holding capacity of this tobacco was just as inferior as the quality. This was probably due to the potash shortage because the plants in the field showed symptoms of shortage of this element and also because other tests have not shown such a depression from shortage of either nitrogen or phosphorus in the fertilizer.

Table 31. Unfertilized Plots Compared With Standard Fertilized Plots.

Burn Tests on Crops of 1926-27

Plot No.		Duration of burn in seconds									
	Fertilizer	Darks		Mediums		Lights		Seconds			
		1926	1927	1926	1927	1926	1927	1926	1927		
Ntr1 Ntr 2	None	4.6 8.0		5.4	42.0		25.8	7.2 5.4	33.		
N1 N1-5	Standard	10.6	34.9	16.2	48.5	39.8	46.6	44.0	55.9		

#### INFLUENCE OF THE SEASON ON COMBUSTION

In making the fire holding capacity tests on the crop of 1927 it was observed that, irrespective of fertilizer treatment, all the tobacco had a much better fire holding capacity than the tobacco of the preceding year. In order to check on our impressions, an average of all burn tests on crops of the two seasons was calculated. The average of 7,460 tests for 1927 was 54.6 seconds as compared with an average of 27.4 seconds for 5,400 tests on the 1926 crop. Since the fertilizer treatment, except for a small number of plots, had been the same and the tests were made in the same way at the same time of year and under the same conditions, the conclusion seemed warranted that the difference was due to the season. Inquiry among tobacco dealers disclosed the general opinion among them that the crop of 1927 was an unusually free burning one.

When the weather records for the two seasons, as recorded at the station farm, were compared, the most apparent difference was found to be in rainfall. During the period from the application of the fertilizer to the last day of harvesting the rainfall in 1926 was 6.33 inches and in 1927 for the same period it was 13.26 inches, i. e., when the rainfall was doubled, the fire holding capacity was also doubled.

In order to see whether this same rule would apply to the preceding years, the rainfall for 1924 and 1925 was tabulated (Table 32) along with the average burn. As indicated in this table the correlation in the extremely dry year 1924 was about as anticipated, viz., extremely poor fire holding capacity. It is a matter of history well known to all dealers that the 1924 crop was the poorest burning crop in many years. The correlation in 1925, however, is not so good, i. e., the fire holding capacity was lower than expected. This may be partially explained by the fact that some of the burn tests for that year were made before sweating the tobacco, and unsweat tobacco never burns as well, but even when allowances are made for this, it is apparent that the 1925



Fig. 17. Rainfall records during tobacco growing season for six years. Height of black column indicates amount of rainfall in one day.

TABLE 32. CORRELATION OF RAINFALL AND FIRE HOLDING CAPACITY

Year	Date of application of fertilizer	Last day of harvesting	Rainfall during this period	Average Fire Holding capacity	Number of test made
1924	June 1(?)	Aug. 12	3.57	9.2	720
925	May 25	Aug. 10	12.02	32.1	4,360
926	May 23	Aug. 16	6.33	27.4	5,400
927	May 20	Aug. 10	13.26	54.6	7,460
928	May 22	Aug. 10	12.66		

crop did not burn so well as that of 1927. The explanation probably lies in the distribution of the rainfall particularly the time of

occurrence of leaching rains and a record of the leaching for each season might be instructive. Distribution of rainfall at the station in Windsor (supplemented where necessary by the U. S. Weather Bureau in Hartford) is presented in Tables 33, and Figure 4.

BULLETIN 299

This same relation of rainfall to fire holding capacity has been observed by Haley and Olson (Rpt. Dir. Pa. Agr. Exp. Station for 1927). They explain it as due to the greater absorption of basic potassium, and the consequent increase in the water soluble alkalinity of the ash of the plant during seasons of high rainfall when the crop is maturing.

Table 33. Distribution of Rainfall in Inches on Tobacco
Experiment Station Farm. 1922-28\*

By 10 day Periods

		Бул	to day r	errous							
	Year										
Period	1922	1923	1924	1925	1926	1927	1928				
May 1-10 May 11-20 May 21-31	3.01 2.20 .21	1.24 .45 .64	1.44 1.35 .91	.08 .73 1.55	.43 1.66 .22	.83 .97 3.24					
June 1-10 June 11-20 June 21-30	.87 1.38 4.67	2.05 .07 .1.72	.71 .01 .90	.49 1.61 1.28	.37 .10 1.39	.39 1.33 .56	1.62 1.57 .97				
July 1–10 July 11–20 July 21–31	2.28 1.39 .89	1.77 .01 3.24	.01 .22 .31	1.02 .27 3.71	.93 1.33	.87 2.51 1.79	2.20 1.64 1.08				
Aug. 1–10	.85	1.17	1.41	3.5	.26	2.57	1.88				

By Months

		Year										
Month	1922	1923	1924	1925	1926	1927	1928					
May	5.42	2.33	3.70	2.36	2.21	5.04	2.30					
June	6.92	3.84	1.62	3.38	1.86	2.28	4.16					
July	5.16	5.02	.54	5.00	2.26	5.17	4.92					
Aug. 20	4.45	2.57	3.40	3.85	2.99	4.34	5.06					

\*Records of 1922-23-24 and May and June of 1925 from the Hartford Station of the U. S. Weather Bureau.

# Magnesia Hunger, or Sand-Drown

This malnutrition trouble has been more prevalent during the season of 1928 than any other year the writers recall. Many

growers who never saw it before are now quite familiar with it In the field it has very distinct symptoms which are easy to diagnose after one has seen it a few times. The leaves fade out between the veins to a light yellow or almost white, contrasting strongly with the dark green pattern of the vein system. They do not crinkle or turn down at the margins, as is the case with potash hunger, but remain smooth and feel thick and stiff between the fingers. The lower leaves are affected first but afterward the symptoms may advance up the plant, even to the top leaves in extreme cases. In very advanced stages the yellow areas between the veins may die and turn brown. The most serious damage, however, comes from checked growth of the plant and the lifeless character of the cured leaves. On the sorting bench we have found these sand-drown leaves characterized by "double colors", a serious defect in tobacco. The areas between the veins which were vellow or white in the field now are either brick red or vellow, contrasting with the greenish brown of the remainder of the leaf.

The trouble occurs only during the years of heavy rainfall and almost always on porous sandy parts of the field where leaching may be expected. Hence the popular name, "sand-drown".

The malady is caused by the inability of the tobacco plant to get from the soil its required amount of magnesium, an element which is just as essential to the proper development of tobacco as nitrogen, phosphorus or potassium. Like salt for the human being, a very small amount will suffice but it is absolutely essential. The fact that in Connecticut we have rarely been troubled with magnesia shortage is due to its presence, in small amounts, in other ingredients of the fertilizer. Unintentionally we have been giving the plant a sufficient ration for ordinary seasons in cottonseed, castor pomace, linseed, stems, manure and the like. All vegetable organic fertilizers contain small quantities of magnesia. Until the present year we have never observed sand-drown on our station plots where we used as much as a ton of cottonseed meal or a ton of combined meal and pomace per acre. But the occurrence of heavy rains for two years in succession has apparently so depleted the available supply of magnesia that in 1928 we have found trouble even under these conditions on light land. Another possible explanation of its prevalence in 1928 may be the frequency of leaching rains during the growing season resulting in constant removal of the magnesia as fast as it came into solution.

The substitution of concentrated synthetic products for a part of the vegetable organics may have made the trouble more wide spread this year, since these materials contain little if any magnesia. At the station we observed sand-drown to be worst where sulfate of ammonia was the only source of nitrogen. It is probable that the sulfate united with the magnesia and in this very soluble form was leached away. This would account for the severity of the trouble immediately after heavy rains. It was also quite common

on the nitrate of soda and nitrate of lime plots. Where a large part of the nitrogen was in urea it was also common in 1928 but not in 1927 and never so severe as on the sulfate of ammonia plots.

BULLETIN 290

The obvious remedy lies in putting more magnesia in the fertilizer ration. Up to the present year we have thought that 15 pounds of magnesia per acre—about the amount in a ton of

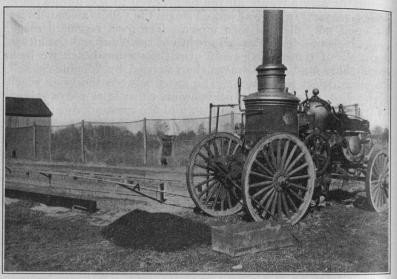


Fig. 18. Steam sterilizing the seed beds with the antiquated town fire engine. Soil can be sterilized to a depth of six inches in 20 minutes with this outfit. Steam supply line at A goes to pan.

cottonseed—was sufficient. As a matter of precaution against the unusual year, however, it may be a good policy for those who have observed this trouble or who have unusually sandy land to add more—perhaps double the dose.

There are a number of materials which may be used. Magnesian lime or limestone (dolomitic limestone) may be applied to lands which are acid enough to stand it. A hundred pounds of double manure salts contain 10 to 12 pounds of magnesia. Cottonhull ashes contain about the same amount. Wood ashes or other vegetable ashes contain a somewhat smaller but valuable supply. A ton of stems contains about 10 pounds.

THE EFFECT OF SULFUR, AMMONIUM SULFATE, AND ALUMINUM SULFATE ON THE REACTION OF SOILS

In previous work at the Tobacco Station in Windsor (Tobacco Station Bulletins 6 and 8) it has been shown that on account of the danger of black rootrot it is desirable to keep tobacco soils fairly acid. The safety point was determined to be near 5.6 pH. Soils less acid than this (above 5.6 pH), especially during cold wet summers, are favorable to black rootrot.

On many tobacco soils where lime and wood ashes have been used too freely the reaction has reached a point far beyond the limit mentioned, i. e., too alkaline, to produce a good crop of tobacco. Hence, it would be beneficial to the grower if some method could be found to increase the acidity of such soils. With this in mind the present investigations were made.

Numerous workers (see Bul. 189 of Rhode Island station with extensive references and Tobacco Station Bul. 10) have found that ammonium sulfate increases the acidity of soils. Doran (2), Hibbard (3), de Long (6), Lipman and co-workers (4, 5), Simon and Schollenberger (7) also found that sulfur caused considerable increase in acidity.

Aluminum sulfate is found to be useful in controlling soil reaction as reported by Amsler (1). To what extent, however, the reaction is affected is not mentioned.

#### LABORATORY EXPERIMENTS\*

In preliminary tests, where 100-gram samples of sandy soil were used, sulfur (fine powder) applied equivalent to 200 pounds per acre, decreased the pH value about one-half unit. The results were corroborated with similar treatments on 500-gram samples of the same soils. The equivalent of 500 pounds of sulfur per acre caused a decrease of about one unit. Increased application of sulfur did not seem to decrease the pH value correspondingly. This preliminary test seemed to indicate that an application at the rate of 500 pounds per acre would be the optimum quantity for an optimum increase in acidity.

Another set of laboratory experiment was made in which three different kinds of soil were used and having the following original pH values:

Clay soil		6.27
Organic soil		6.25
Sandy soil	 	 5.42

<sup>\*</sup>Credit is due to Dr. A. B. Beaumont, Mass. Agr. College, for suggesting the laboratory phases of this work.

As previously, 100-gram samples of these various soils were made up to furnish four non-treated and triplicates of the following treatments:

			Pounds p	er acre	
1. 2. 3.	Sulfur	$   \begin{array}{r}     250 \\     1032 \\     1732   \end{array} $	500 2060 3465	750 3090 5197	$1000 \\ 4120 \\ 6930$
4.	Sulfur	500 1030	500 2060	500 3090	500 4120
5.	SulfurAluminum Sulfate	500 1732	500 3465	500 5197	500 6930

The amount of sulfur represents low, medium, high, and very high applications Ammonium sulfate and aluminum sulfate are taken in chemically equivalent amounts of sulfate, i. e., in the first three treatments the quantities of sulfur is the same in the various applications. In treatments 4 and 5, combinations of sulfur with the other materials, 500 pounds of sulfur has been taken as a medium constant amount of sulfur to use together with the other two materials.

The samples were kept at normal moisture (most suitable for growing conditions) during the course of the experiments. Reactions were determined electrometrically after four and eight weeks and the results are recorded in Tables 34, 35 and 36. Abbreviations used in the tables are:

S = Sulfur
AlS = Aluminum Sulfate
AmS = Ammonium Sulfate
1 2 3 4 = low medium high, and very high at

1, 2, 3, 4, =low, medium, high, and very high applications of the various treatments.

Influence of treatments on clay soil. From Table 34 it is seen that the original pH value of the clay soil was 6.27 (the figure in parenthesis). Receiving nothing but distilled water, the pH had decreased to 5.84 after four weeks and to 5.82 after the lapse of eight weeks, thus showing practically no change after the first decrease. In this case sulfur had apparently been oxidized, although not added. Simon and Schollenberger (7) in sulfofication studies obtained a similar result in a field test, where no sulfur was supplied. They report that the acidity was increased by 0.4 pH as compared with 0.45 obtained in the results above. In computing the actual decrease in pH values, it seemed most reasonable to consider the difference between the value of the checks after eight weeks and those of the various treatments after the same time interval.

Applying sulfur at a rate of 250 pounds per acre decreased the pH by 0.48; 500 pounds of sulfur gave exactly the same results in four weeks, but at the end of eight weeks the difference between the two applications was hardly significant, since the actual decrease in the latter case was 0.54. Applying 750 pounds did not seem to work as quickly as the previous treatment, but had decreased the pH by 0.65 at the end of eight weeks. A similar result was obtained in four weeks by an application of 1,000 pounds per acre, which after eight weeks had caused a decrease of 0.74.

REPORT OF TOBACCO SUBSTATION

TABLE 34. REACTIONS OF CLAY SOIL TREATED WITH SULFUR, ALUMINUM SULFATE AND AMMONIUM SULFATE AFTER FOUR AND EIGHT WEEKS

	Avera	ge pH of tri			Average pH of triplicate treatments			
Treatments	After After 4 weeks		Decrease in pH	Treatments	After 4 weeks	After 8 weeks	Decrease in pH	
Check (6.27) S1 S2 S3 S4	5.84 5.77 5.36 5.50 5.22	5.82 0.45 5.34 0.48 5.28 0.54 5.17 0.65 5.08 0.74		Combined	Treatme nts			
AlS1 AlS2 AlS3 AlS4 AmS1 AmS2 AmS3 AmS4	5.47 5.17 5.15 4.76 6.16 5.72 5.68 5.76	5.15 5.20 5.07 4.84 5.38 5.44 5.41 5.37	0.67 0.62 0.75 0.98 0.44 0.38 0.41 0.45	S+AIS1 S+AIS2 S+AIS3 S+AIS4 S+AMS1 S+AMS2 S+AMS3 S+AMS4	5.23 5.09 4.86 5.24 5.33 5.22 5.31 5.41	4.95 4.96 4.83 5.11 4.97 4.86 4.98 4.89	0.87 0.86 0.99 0.71 0.85 0.96 0.84 0.93	

An application of aluminum sulfate at a rate 1,732 pounds per acre resulted in a decrease of 0.67 after eight weeks. The same result was obtained in four weeks when 3,465 pounds were used. At the end of eight weeks, however, the total decrease of the latter treatment did not quite measure up to the result obtained by the lower application. Applying at a rate of 5,197 pounds showed about the same result as for 3,465 pounds, although the final results showed a decrease of 0.75. The highest application of aluminum sulfate (6,930 pounds) decreased the pH value by more than a unit after four weeks, after which time the maximum decrease apparently was reached as the values increased after the lapse of eight weeks, making the final result 0.98.

Using 500 pounds of *sulfur* in *addition* to the various applications of *aluminum sulfate*, the same relationship seemed to occur between the individual treatments, with exception of the highest application which markedly deviated from the rest. The sulfur added, on the average caused a decrease in pH values of about 0.2 unit below the aluminum sulfate alone at various applications

save the maximum application. Unless some error in the procedure unnoticed crept in, it is hard to explain that the highest application showed a lower result than the lowest amount applied, or 0.87 as compared with 0.71 pH decrease after eight weeks.

BULLETIN 299

With respect to ammonium sulfate the various amounts applied did not show significant differences in results after eight weeks as the average decrease after this time was close to 0.40. Sulfur added to the treatments about doubled the effect, but showed the same general trend.

The ammonium radical in these cases may have had a buffering effect, as very little ammonia could have escaped during the experiment.

Influence of the treatments on the organic soil. In Table 35 the results from the various treatments on the organic soil are listed. It is shown that the original pH value of this soil, 6.25, after four weeks was 6.12 and after eight weeks 5.59. Here the untreated soil had turned more acid than the clay soil.

With respect to the various treatments, 250 pounds of *sulfur* had practically no desirable effect on the reaction; 500 pounds a very slight effect and 750 pounds were able to decrease the pH value only 0.3, and the highest application, 1,000 pounds, 0.55 pH.

On the other hand, aluminum sulfate, applied at a rate of 1,732 pounds per acre decreased the pH value by 0.38 after eight weeks. The double amount, however, was not more effective, neither was the next higher application. Using the rate of 6,930 pounds, caused a decrease of 0.85 after four weeks, but the total decrease after eight weeks amounted to practically the same as for the lowest application.

TABLE 35. REACTIONS OF ORGANIC SOIL, TREATED WITH SULFUR, ALUMINUM SULFATE AND AMMONIUM SULFATE, AFTER FOUR AND EIGHT WEEKS

	Avera	ge pH of tri			Average pH of triplicate treatments			
Treatments	After After Decrease in pH		Treatments	After 4 weeks	After 8 weeks	Decrease in pH		
Check (6.25) S1 S2 S3 S4	6.12 5.59 6.18 5.70 5.92 5.51 5.65 5.29 5.46 5.04		0.66 0.00 0.08 0.30 0.55	Combined	Treatme	nts		
AlS1 AlS2 AlS3 AlS4 AmS1 AmS2 AmS3 AmS4	5.77 5.74 5.65 5.27 5.80 5.67 5.75 5.77	5.21 5.32 5.28 5.18 5.65 5.53 5.04 5.08	0.38 0.27 0.31 0.41 0.00 0.06 0.55 0.51	S+AIS1 S+AIS2 S+AIS3 S+AIS4 S+AMS1 S+AMS2 S+AMS3 S+AMS4	5.44 5.37 5.22 5.79 5.19 5.24 5.30 5.53	5.04 5.20 5.03 4.83 5.18 5.12 4.99 4.94	0.55 0.39 0.56 0.76 0.41 0.47 0.60 0.65	



Fig. 19. Steam pan in position. A, Pan; B, Steam supply line; C, Cross bars for lifting pan; D, Sides of seed beds. Steam beyond pan is rising from soil from which pan has just been removed.

187

BULLETIN 299

It is interesting to note that although sulfur, added to the various applications of aluminum sulfate, considerably increased the acidifying effect, the relation between the individual treatments is somewhat similar to that of aluminum sulfate alone.

CONNECTICUT EXPERIMENT STATION

The lowest two applications of ammonium sulfate had practically no effect on the reaction after eight weeks, while the two higher ones caused a decrease of about half a unit in the same time.

Sulfur added to the various applications of ammonium sulfate caused a decrease of about 0.40, while at the two highest applications the decrease below the results from ammonium sulfate alone was hardly 0.1 of a unit.

Influence of the treatments on sandy soil. Of the three soils included in this study, the sandy soil untreated having an initial pH value of 5.42, had increased in acidity the least, only 0.17 after eight weeks. This soil, being relatively poor in content of buffering substances, showed the largest response to the various treatments. as may be seen in Table 36.

Sulfur at the lowest application resulted in a decrease less than half a unit pH, but doubling the application, more than doubled the decrease. Although further decreases were noted at higher applications, they were not in proportion to the materials supplied.

TABLE 36. REACTIONS OF SANDY SOIL TREATED WITH SULFUR, ALUMINUM SULFATE AND AMMONIUM SULFATE ATFER FOUR AND EIGHT WEEKS

Treatments	Average pH of triplicate treatments				Average pH of triplicate treatments			
	After 4 weeks	After 8 weeks	Decrease in pH	Treatments	After 4 weeks	After 8 weeks	Decrease in pH	
Check (5.42) S1 S2 S3 S4	5.47 5.18 4.87 4.84 4.74	5.25 4.82 4.31 4.17 4.00	0.17 0.43 0.94 1.08 1.25	Combined	Treatme nts			
AlS1 AlS2 AlS3 AlS4 AmS1 AmS2 AmS3 AmS4	4.95 4.66 4.42 4.11 4.86 4.94 5.01 5.11	4.60 4.62 4.37 4.12 3.94 4.08 4.61 4.32	0.65 0.63 0.88 1.13 1.31 1.17 0.64 0.93	S+AIS1 S+AIS2 S+AIS3 S+AIS4 S+AMS1 S+AMS2 S+AMS3 S+AMS4	4.83 4.58 4.29 4.15 4.86 4.94 5.01 5.11	4.33 4.30 4.01 4.00 4.25 4.30 4.29 4.23	0.92 0.95 1.24 1.25 0.99 0.95 0.96 1.02	

The low and medium applications of aluminum sulfate seemed to have about equal effect after eight weeks. With the two higher applications the actual decrease was reached after the first four weeks as after this time the reaction was not markedly changed.

Adding sulfur to the treatment of aluminum sulfate had a marked effect at the low, medium and high applications. The highest application caused a decrease about similar to the high one after eight weeks.

The low application of ammonium sulfate to this soil resulted in the largest decrease in pH value recorded in this experiment as a whole. Doubling the application, however, did not measure up to the effect of the former and trebling the low application had only half of the effect of this one, while the effect of the highest application falls between these two.

On the other hand, when sulfur was added to the various applications of ammonium sulfate the effect was not nearly so great as in the case of the two first applications of ammonium sulfate alone, but instead the sulfur added seemed to balance the reaction, so as to give about equal effect of the four treatments.

Discussion of results from laboratory experiments. In the preliminary study it was found that sulfur applied at a rate of 500 nounds per acre would be about the proper amount to cause an optimum increase of acidity in sandy soils. As a similar result was obtained on the sandy soil in a later experiment, it seems reasonable to assume that in this case 500 pounds of sulfur would best serve the purpose of increasing the acidity.

No optimum application of sulfur could be established for the clay soil or the organic soil as the increased acidity quite uniformly corresponded to the amounts applied. Results from sulfur treatments in Tables 34 and 35 may thus serve as an indication of the difficulties that occur in highly buffered soils. The fact, emphasized by Lipman and co-workers (8), that sulfur is oxidized more rapidly in the absence of organic matter may also have played an important part in case of the organic soil. In general, however, the buffer action of this soil is prominent for all of the treatments used.

The results obtained with ammonium and aluminum sulfate and the combinations of these with sulfur, considering the large quantities used, do not favor a competition with sulfur alone for any of the soils used. The important factor of leaching was entirely eliminated in these experiments. A field study of the treatments discussed is thus needed to justify recommendations for practical purposes.

#### FIELD EXPERIMENTS

Field experiments on soil acidification by the use of sulfate of ammonia and inoculated sulfur were made in co-operation with the Hartford County Farm Bureau. The experiment was begun in the spring of 1927. Three fields were chosen where the reaction was known to be nearly or quite neutral. Eight plots, each  $2 \times 4$  rods (1/20 acre) were measured and treated as follows:

- Plot 1. No sulfate of ammonia or sulfur.
  - 3. 300 lbs. sulfate of ammonia per acre
  - 3. 500 lbs. " " "
  - 4. 800 lbs. " "
  - 5. 200 lbs. inoculated sulfur
  - 6. 400 lbs. "

In applying the fertilizer, the nitrogen of the formula was reduced on the sulfate of ammonia plots so that each of the eight plots received the same amount of nitrogen. On the 800 lb. plot only 500 lbs. were applied at first, the 300 being applied to the crop later.

Location of fields and times of application in 1927 were as follows:

1. Farm of R. E. Distin, Unionville. Field not previously in tobacco. Chemicals applied on June 13th. Land had previously had a heavy coat of manure. The soil showed a content of 7.43% organic matter. Plot 8 received the second application of sulfur on July 11th.

2. Farm of E. H. Sloan, Broadbrook. Field many years in tobacco; content of organic matter, 6.18%. Chemicals applied on June 8th. Second application on July 11th.

3. Farm of W. G. Phelps, Glastonbury. Old tobacco field; had a content of 6.72% organic matter. Chemicals applied on June 2d. Second application July 11th.

Organic matter was determined by Mr. Jacobson of the Station Soils Department.

At the time of the second application of sulfate of ammonia no differences in growth could be observed; neither were consistent

Table 37. Reactions of Soils Treated With Sulfate of Ammonia and Sulfur

	Lbs. chemi	Lbs. chemicals per Acre			Reaction of soil in the fall (pH)					
Plot No.	Sulfate of	Inoculated	Sloan .		Phelps		Distin			
	ammonia	sulfur	1927	1928	1927	1928	1927	1928		
1	0		6.58	6.62	6.21	6.00	6.34	6.0		
2	300		6.73	6.86	5.77	5.77	6.37	6.1		
3	500		6.58	6.79	5.70	5.60	6.28	6.1		
4	800		6.56	6.69	5.61	5.40	6.35	6.1		
4 5		200	6.18	6.18	5.99	6.01	6.11	6.0		
6		400	6.07	6.01	6.10	5.67	5.96	6.0		
7		600	5.83	5.93	5.79	5.60	5.70	5.7		
8		800	5.81	5.77	5.87	5.50	5.66	5.9		

differences noticed later in the season. Therefore, the tobacco was not kept separate at time of harvest.

On May 9th and 10th in 1928 all the plots were treated as in previous year and chemicals applied all at one time before plowing. Also in the growing season of 1928 no differences in growth could be observed between the various treatments.

Both in the late fall of 1927 and 1928 soil samples were taken from the different plots and their reactions determined electrometrically. The results are presented in Table 37.

In regard to the effect of the chemicals on soil reaction a glance at Table 37 shows that sulfate of ammonia did not increase the acidity on the Sloan and Distin farms but had some effect on the Phelps farm. On the other hand, inoculated sulfur in every case increased the acidity. However, it is also apparent that continued applications of sulfate of ammonia and sulfur do not increase the acidifying effect. With respect to sulfur, the optimum effect seems to fall between 400 and 600 pounds per acre which is in fairly good agreement with the laboratory tests, previously discussed. The high content of organic matter in all these soils is probably one reason why they were so slightly affected by the treatments.

In order to observe the effect of aluminum sulfate adapted to field conditions, two 1/80 acre plots were laid out on a field\* at the Tobacco Station in Windsor in the spring of 1928. In addition to general fertilizers the plots each received aluminum sulfate at a rate of 250 lbs. per acre. During the growing season no significant difference in the growth of tobacco was observed between these and adjacent check plots.

Reactions of the soil was determined in the spring before applying sulfate and in the fall after harvest and were as follows:

Adjacent check	Spring	5.77 pH 5.77 "	Fall	5.70 pH
Plot 1.	"	5.77 "	" .	5.70 pH 5.38 "
Plot 2.	"	5.72 "	"	5.17 "

This relatively low application of aluminum sulfate was thus able to increase the acidity considerably.

Discussion of results from field experiments. The use of sulfur, ammonium sulfate and aluminum sulfate has been tried out under field conditions. Sulfur has increased the acidity in all instances, the optimum effect being obtained by applications between 400 lbs. and 600 lbs. per acre. On the average, 500 lbs. of sulfur will cause a decrease in acidity of about 0.5pH. On many soils, however, a decrease of only 0.2 to 0.3 pH would be beneficial in order to approach the safety point from black rootrot. Laboratory as well as field tests have shown that up to 500-600 lbs. of sulfur per acre the acidity increases rather uniformly with the amounts

<sup>\*</sup>Classified by the Soils Dept. as Merrimac loamy coarse sand.

applied. Hence there is no object in applying more sulfur than needed, especially since it is shown that sulfur may impair quality and burn, as discussed in Tobacco Station Bulletin 10.

The same bulletin also contains data on the use of ammonium sulfate in fertilizer for tobacco. An extensive use of sulfate of ammonia injures the burn, hence it should be used cautiously. Its greatest effect as acidifying agent would be on sandy soils, low in organic matter and also where lumps of limestone are not visible in the surface soil.

As for aluminum sulfate, its use still being in the experimental stage, no recommendations can be made at this time. However since this chemical is now put on the market as a comparatively inexpensive product it may be advisable to make trials on a small scale, using up to 500 lbs. per acre.

#### LITERATURE CITED

1. AMSLER, M. The control of soil acidity by the use of aluminum sulfate. Gardner's Chron. (London) 81: 340, 1927.

2. DORAN, WILLIAM L. Relation of the adjustment of soil reaction to black rootrot of tobacco; Science Dec. 1927, Vol. LXVI, No.

3. HIBBARD, P. L. Sulfur on neutralizing alkali soils. Soil Sci. 11: 385-387, 1921.

4. LIPMAN, J. G., PRINCE, A. L. and BLAIR, A. W. The influence of varying the amounts of sulfur in the soil on crop yields, hydrogenion concentration, lime-requirement and nitrate formation. Soil Sci. 12: 197-207, 1921.

5. LIPMAN, J. G., WAKSMAN, S., and JAFFE, J. S. The oxidation of sulfur by soil micro organisms; Soil Sci. 12: 475-489. 1921.

6. DE LONG, W. A. Sulfur and soil acidity. Sci. Agr. (Canada) 3: 354-356; 1923.

7. SIMON, R. H. and Schollenberger, C. J. The rate of oxidation of different forms of sulfur. Soil Sci. 20: 443, 449. 1925.

# SUMMARY OF PROGRESS ON PROJECTS

There are a number of lines of investigation on which distinct progress has been made during the year but which cannot be discussed in detail in the present report. These are briefly summarized below without citation of supporting data. They will be presented more completely in later publications.

## UREA AND CALUREA AS SOURCES OF NITROGEN IN THE FERTILIZER

Results of urea tests during three years have been reported in previous bulletins from this station. Results for 1928 are in line with those of the preceding three years. When one-half of the nitrogen of the formula is derived from urea the resulting yields and quality have been just as satisfactory as for a standard formula in which four-fifths of the nitrogen is from cottonseed meal and castor pomace. When all the nitrogen is from urea the tobacco is

not of such good quality. Sand-drown also affected the all-urea plots. More recently, another urea compound, calurea has appeared on the market and experiments were begun with calurea in 1928. Calurea has 4/5 of its nitrogen from urea and the other 1/5 from calcium nitrate. Tests of one year on calurea are in line with the results from urea. It is a less expensive source of nitrogen than urea because of the high import duty on the latter. Nitrogen in both of these carriers is quickly available and has shown no signs of serious leaching. In this form, nitrogen costs less than half of what it does in cottonseed meal and is much less bulky in handling. Up to the present there has been found no good reason why these cheaper synthetic products should not be used to supplant a part of the more expensive vegetable organic constituents of the fertilizer, but more data are needed before definite recommendations can be made.

REPORT OF TOBACCO SUBSTATION

# NITRATE OF LIME AS A SOURCE OF MINERAL NITROGEN

It is a generally recognized principle that the tobacco fertilizer mixture should have a minor part of its nitrogen in a mineral carrier. The purpose of this is to furnish nitrogen in a quickly available form and thus function as a "starter" when the plants are first set in the field. Nitrate of soda is most extensively used for this purpose because of the immediately available nitrate it contains and because it is cheap and always plentiful on the market. Two objections to the use of nitrate of soda are: (1) during heavy rains the nitrogen in it leaches away very quickly, and (2) the sodium which it contains may raise the soil reaction. We have shown elsewhere that it causes the soil to become alkaline.

Within recent years, another quickly available mineral nitrate, nitrate of lime (calcium nitrate), has come into the market as a competitor of nitrate of soda. This material contains the same amount of nitrogen as nitrate of soda, is just as reasonable in cost (quoted somewhat lower in 1927 and 1928) and otherwise is very similar to nitrate of soda. It has the disadvantage of being somewhat more deliquescent than nitrate of soda. In our experience of 1927-28 we have not found that this property has caused the mixture in which it was used to "cake" or become lumpy even when it was used to supply as much as one-half of the nitrogen and permitted to stand for six weeks after mixing. It is marketed in paper-lined bags to prevent its becoming over-damp or hard during storage. From a theoretical standpoint it should be more suitable for use on tobacco because it contains calcium which is used in large amount by the tobacco crop and because calcium salts are known to give a desirable white color to the ash.

These two materials have been tested side by side for two years on the station farm deriving, (1) 1/5 of the nitrogen from these mineral sources and (2) ½ of the nitrogen from them. Averaging

the results of the two years we find no significant differences between them in yield, grading, or fire holding capacity. There appeared to be serious leaching of the nitrogen from both when they were used to supply one-half the fertilizer nitrogen. There did not seem to be any great difference between them in this respect. When the soil was tested one year after starting the series, the nitrate of soda plots were slightly more alkaline than the nitrate of lime plots. Cigars from these plots have not vet been tested: hence we have no data as to the effect of each on ash color.

CONNECTICUT EXPERIMENT STATION

#### COMPARISON OF SINGLE NITROGEN SOURCES

Results on this series in 1928 were about the same as for 1927 (fully reported in Tob. Sta. Bul. 10, p. 60). The four sources tested at station farm were cottonseed meal, nitrate of soda, sulfate of ammonia and urea. All fertilizer was applied at one time before setting. The nitrate of soda leached so badly that the tobacco made less than half a crop and on the sorting bench was of such inferior quality it was not fit to sort. Besides nitrogen starvation it also showed severe sand-drown. Fire holding capacity, however, was excellent—possibly because of absence of hindering nitrogen compounds. Nitrogen also leached seriously from the cottonseed meal plots, growth was checked and the cured tobacco was yellow and lifeless. It was considerably better than the nitrate of soda tobacco, however. Fire holding capacity was good but was surpassed by nitrate of soda. The sulfate of ammonia plot showed no signs of nitrogen starvation but remained rank and green except for sand-drown which was especially prevalent here. The yield was highest of all the plots but the quality was poor because of the double colors previously mentioned. Fire holding capacity was less than that of the other plots. No leaching of nitrogen was apparent on the urea plot but there was considerable sand-drown. Growth was good, yield being only slightly less than that of sulfate of ammonia. The quality was best of all these four plots.

Monthly soil tests which have been made for a year on these plots show little change in the cottonseed meal plot; the nitrate of soda plot is progressively more alkaline, the sulfate of ammonia plot is the most acid, the urea plot is progressively slightly more acid than the cottonseed meal plot.

## MANURE AS A SUPPLEMENT TO COMMERCIAL FERTILIZER

The 1928 tests confirm those of 1927 (fully discussed in Tob. Sta. Bul. 10, p. 62). The tobacco remained greener and somewhat larger in the field than on adjacent unmanured plots. On the sorting bench it was found to have a better grade index and higher

vield. This was true both for stable manure and for artificial Adco manure. The differences, however, were not large.

REPORT OF TOBACCO SUBSTATION

# FRACTIONAL APPLICATION OF NITROGEN

This experiment was continued on the same six plots as in 1927 (Tob. Sta. Bul. 10, p. 57). The extremely wet year was favorable for this kind of a test. One-half of the quickly available nitrogen (in nitrate of lime and nitrate of potash) was deducted from the

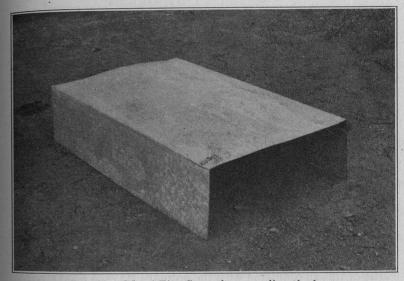


Fig. 20. Metal Fire Cover for spreading the heat.

first (broadcast) application and divided between two later applications. The second application (delayed by mistake) was made on July 2. On July 5 there was a heavy rain (over 2 inches) which caused serious leaching. The last application was therefore made on July 7. Within a few days the yellow color of the check plots showed that they did not have sufficient nitrogen. The fractional plots remained green. On the sorting bench the tobacco from the check plots was found to be yellow and lifeless, much inferior to the fractional plots. The yield was also less. The benefit from fractional application was unquestionable in 1928. This is the first year this has been true. It may be due to the fact that the later applications were made at exactly the critical time with respect to the heavy rains. The only objectional quality noted In the tobacco from the fractional plots was that the veins were

195

somewhat prominent. This soil is very sandy and subject to easy leaching.

These results and the observations on many other fields in 1928 show that if later nitrogen applications are to be used, they should be made *immedately* after the heavy rains *before* any fading of the leaves appears. Delay until the leaves appear yellow results in checked growth from which the plant never entirely recovers and reduced yield results.

#### COVER CROP TESTS

Records for 1928 show that every cover crop used both increased the yield and improved the grading of the tobacco grown on it. Rye gave the highest yield closely followed by vetch and oats. These three also had the best grade index. Timothy, barley, redtop, alfalfa and wheat were not so beneficial as the others but were better than no cover.

#### ROOTROT RESISTANT STRAINS OF TOBACCO

Experiments have been continued with resistant strains of Havana Seed, Broadleaf and Shade Cuban.

Wisconsin Havana No. 142 has again been shown to be highly resistant to Black Rootrot and it produces a heavier yield of leaf than any of the ordinary Connecticut Havana Seed strains. On account of the thinner, larger leaves, set closer together on a larger stalk this appears to suffer more than the others from pole sweat. In setting this type it would be advisable to increase the distance between plants in the row by at least two inches. Packers and manufacturers do not agree among themselves as to the merits of this as compared with the ordinary Havana Seed strains.

Broadleaf tobacco is not as susceptible to rootrot as the other two types but still the reduction in yield from this cause is so serious on some fields that a resistant strain is much needed if one that is desirable from other standpoints could be found or developed. One promising strain has been under test on the station farm for the last two years. Compared with the John Williams strain it yielded more leaf and the quality appeared as good. It seems to be more subject to pole sweat, however, possibly on account of thinner leaves and unusually large succulent stalks. Tests have not been conducted long enough to state whether it will meet the requirements of the trade. It is undoubtedly much more resistant than the John Williams type.

In 1927, in a shade field which was badly dwarfed from black rootrot, fifteen scattered plants were found which were making perfectly normal growth. Seed was saved separately from each of these. After harvest, the roots of these and many adjacent plants were examined and relatively few lesions were found on these while the others about them were severely rotted. A row was planted from the seed of each one of these plants in 1928 and compared with alternate rows from common Cuban seed. Rootrot was not very severe on this field in 1928 and the contrast was not as marked as could be desired. Nevertheless it was apparent both from the growth and from the condition of the roots that some of these had considerably more resistance than the common Cuban plants. Further selections and tests are necessary and it is hoped that a satisfactory Cuban resistant strain may be secured from this chance find. Other resistant Cuban strains from another source are also under test but it is too early to predict what the results will be.

#### TOPPING AND SUCKERING EXPERIMENTS

Results of three years experiments at the station on these practices have just been published by N. T. Nelson as Bul. 297 of the Connecticut Agricultural Experiment Station.

#### FIRE CURING OF STALK TOBACCO

The importance of curing tobacco by charcoal fires is so fully recognized by the growers of shade tobacco that the practice is universal among them. The stalk tobacco growers, however, are inclined to "take a chance" on the weather and only a few of them practice charcoal firing.

Therefore when weather conditions are favorable for pole sweat—as they were in 1928—the amount of loss from this source is very large. A conservative estimate of the loss to the Connecticut growers this year is over a million dollars. With an expenditure of one-fifth of that amount, most of this could have been prevented.

Experiments which have been conducted in the experiment station sheds during the last four years and in the sheds of practical growers of both broadleaf and Havana Seed lead us to believe that fire curing should be practiced universally by stalk growers at least during seasons when the weather is conducive to sweat. In a later separate bulletin, full data on these experiments and more complete discussion of the practice will be published.

At this time, for the benefit of those who have not been accustomed to fire curing, the following recommendations are made, based on our experiments:

For every acre of tobacco a minimum of 50 bushels of charcoal should be on hand, before harvesting starts.

A larger number of small fires is better than a small number of large fires. The air drafts which are created by the fires are as important, if not more so than the actual heat produced. These drafts, therefore, should be distributed as much as possible over

the shed. A large number of fires prevents "dead air pockets'. The same result is obtained by the use of metal "spreaders" or covers over the fires. They also prevent scorching the tobacco directly over the fires.

The temperature should be kept between 85° and 95° F in the second tier. In very warm weather it may be necessary to raise this to 100° at times.

Firing to wilt, i. e., within a few days after filling the shed and while the leaves are still green is a good practice but is not always necessary for prevention of sweat. Pole sweat never attacks leaves when they are in the green stage. The late yellow and early brown stages are the danger stages.

If the tobacco is in those stages and wet weather sets in, with high humidity preventing evaporation of moisture from the leaves, it is time to start firing. Don't wait until the leaves begin to

"puff" and the midribs "strut".

The condition of the tobacco should be the guide in firing. The minimum period of firing should be thirty-six hours. Frequently two or three times as long is necessary. When the leaves become dry in the green or yellow stage it is a sign you are firing too much.

The temperature should then either be reduced or the firing stopped entirely for a day and then started again if weather conditions demand. Intermittent firing is better than continuous firing except where pole sweat has already started and rapid drying is required.

All brown parts of the leaf should be dry. This is the best guide

by which to know that you are firing enough.

During firing, the ridge and gable ventilators should be open

but the side ventilators (boards) should be closed.

Some have the impression that fire cured tobacco is not so elastic as that which is cured naturally. We believe that this is due to curing too rapidly. Tobacco properly fire-cured is just as good as naturally cured tobacco in this respect.

#### EFFECT OF LIMING THE SOIL ON COMPOSITION OF TOBACCO

Tobacco from limed soils exhibits characters of combustion which are different from those of tobacco grown on soils which have not been limed. Some of these characters, particularly the whiteness of ash and closeness of burn are desirable; others, such as the "flaking" of ash and reduction in fire holding capacity, are objectionable. These differences in burn are probably correlated with changes which liming produces in the chemical composition of the leaf. From the standpoint of good growth of tobacco, heavy liming of the soil is not desirable but it is not beyond the range of possibility that the good effects on combustion may be produced by application of some material other than lime. Whether or not there is such a possibility can be intelligently determined

TABLE 38. ANALYSES OF TOBACCO FROM LIMED AND UNLIMED PLOTS.

CROP OF 1926. WATER FREE BASIS

REPORT OF TOBACCO SUBSTATION

			Percentage of minerals.						
Plot No.	Grade	Lime	Total ash	P <sub>2</sub> O <sub>5</sub>	K <sub>2</sub> O	CaO	MgO	Mn <sub>2</sub> O <sub>3</sub>	A1 <sub>2</sub> O <sub>3</sub>
					F	I IELD VI	II		
L	M	Lime	23.89	0.78	6.52	4.72	3.79	.01	.09
L1	M	Lime	23.05	0.73	6.38	4.82	3.58	.01	.08
L2	M	Lime	22.63	0.73	5.92	4.73	4.21	.01	.06
L3	M	Lime	22.90	0.74	6.06	4.79	3.89	.01	.08
L38	M	Lime	23.49	0.75	6.87	4.52	3.61	.01	.11
L39	M	Lime	21.83	1.05	6.37	4.13	3.12	.01	.08
C3-1	M	No lime	26.92	0.85	8.90	5.67	1.16	.10	.09
F6-1	M	No lime	28.15	0.87	8.36	6.55	1.19	.06	.21
N1-5	M	No lime	25.20	0.88	7.51	6.53	1.38	.06	.07
N1-6	M	No lime	25.22	0.93	7.62	6.13	1.29	.07	.07
					Po	QUONO	FIELD		
4A 15A 6A	A11 A11 A11	Lime No lime No lime	24.57 27.26 28.19	0.77 0.77 0.85	6.79 7.23 7.80	4.98 5.30 5.84	3.11 1.55 1.18	.03 .21 .12	.17 .23 .31
			, In Sec	Тн	IELA VIA	PLOT	S		
T1A	M	Lime	22.83	0.72	5.12	5.53	3.49	.01	.08
T1B	M	No lime	24.78	0.84	6.67	6.11	1.56	.29	.14
T1A	S	Lime	25.33	0.62	4.18	6.38	4.54	.02	.11
T1B	S	No lime	27.43	0.76	6.62	7.04	1.57	.28	.31
T2A	M	Lime	23.86	0.82	6.10	5.37	3.13	.01	.05
T2B	M	No lime	24.63	0.84	7.37	5.62	1.18	.04	.11
Г2А	S	Lime	25.70	0.60	5.52	6.00	3.15	.02	.14
T2B	S	No lime	26.82	0.75	7.10	6.42	1.18	.04	.13
T3A	M	Lime	23.83	0.69	5.73	5.15	3.58	.01	.11
T3B	M	No lime	25.43	0.85	6.81	6.19	1.62	.06	.06
T3B	S	No lime	27.49	0.68	7.17	6.89	1.59	.09	.17

199

only after we have found out what changes lime has produced in the composition of the leaf.

BULLETIN 299

Tobacco (of the crop of 1926) from limed and adjacent unlimed plots—which otherwise were identical in character of soil and previous treatment—was therefore analyzed by the Station Chemistry Department with respect to those elements which it was believed might be affected by liming. Three different series of plots were used in these Analyses as follows:

Limed plots on Field VIII. Beginning with 1922 these plots were limed heavily each year. With the last application in the spring of 1925 they had received during these four years five tons of hydrated lime per acre and the reaction of the soil was slightly above 7.0 pH.

Thielavia series. These consisted of three plots. One-half of each received an application of one ton of hydrated lime per acre each spring, beginning in 1924.

Poguonoc field. Plots on this field were limed at the same rate as the Thielavia series in the spring of 1925 and 1926.

The lime was not analyzed each year but it was commercial lime from western Massachusetts and unquestionably all of it had some magnesia in it. Some samples from that section have as high as 30% MgO.

Results of the analyses presented in Table 38 show that:

1. In every comparison, liming reduced the percentage of total ash, of calcium, phosphorus, manganese and potash. Aluminum was reduced in some cases but not in all.

2. In every case, liming approximately doubled the percentage of magnesium.

In experiments with tobacco in Ohio. Ames and Boltz (Ohio Sta. Bul. 285, 1915) also found that liming a soil reduced the percentage of calcium, potassium, manganese, phosphorus and sulfur but increased the magnesium.

Garner (U. S. D. A., Bur. Pl. Indus., Bul. 105) found that magnesium salts are injurious to fire holding capacity of tobacco (more so than the calcium salts). He also found that all magnesium salts produce a white ash.

Apparently the white ash and reduced fire holding capacity which we find on our limed plots are due to magnesium rather than to calcium.

THE EFFECTS OF MAGNESIA, SULPHUR AND CHLORINE ON THE GROWTH AND QUALITY OF TOBACCO1

### H. F. Murwin<sup>2</sup>

With the coming use of more concentrated fertilizers and syn-

thetic nitrogen products in the tobacco growing industry it would be well to keep in mind the nutritional disturbances which may result. The trend today is toward the use of more concentrated mixtures which necessitates the use of more chemicals to replace some of the cottonseed meal or the like. We usually consider the value of a tobacco fertilizer in terms of nitrogen, phosphoric acid and potash. While these three elements are absolutely essential. the tobacco plant requires more than these three for normal growth. In this connection fertilizer tests have been conducted over a period of six years at the Tobacco Station in an effort to determine some specific effects of magnesia, sulphur and chlorine on the growth, quality and burn of Havana seed tobacco.

These fertilizer tests consisted of six treatments in duplicate on 1/40 acre plots. A basal ration which furnished only nitrogen, phosphoric acid and potash was applied on all plots. During the first three years of the experiment 40 pounds of nitrogen, 64 pounds of phosphoric acid and 80 pounds of potash were furnished annually in the basal ration. Starting with the 1925 season the quantities of nitrogen and potash applied annually were increased from year to year and in 1927 the basal mixture furnished about 200 pounds of nitrogen, 64 pounds of phosphoric acid and 200 pounds of potash per acre. The exact amounts and types of carriers used in this mixture are given in Table 39.

TABLE 39. COMPOSITION OF BASAL RATION. MAGNESIA, SULPHUR AND CHLORINE PLOTS-1927

Materials	Lbs. per acre	$NH_3$	$P_2O_5$	$K_2O$
Nitrate of potash	460.0	72.63		199.6
Precipitated bone	166.5		64.1	
Dried blood	200.0	31.56		
Urea	119.0	65.45		
Nitrate of soda	324.0	60.58		
Total	1269.5	230.22	64.1	199.6

TABLE 40. PER ACRE AMOUNTS OF PURE CHEMICALS APPLIED IN MAGNESIA. SULPHUR AND CHLORINE TREATMENTS

Plot	Chemicals applied	Nutrients supplied
1 & 1+	Control	Control
2 & 2+	185 lbs. Magnesium Sulphate	30 lbs. MgO; 60 lbs. SO <sub>3</sub>
	88 lbs. Sodium Chloride	53 lbs. C1; (50 lbs. Na <sub>2</sub> O)
3 & 3+	185 lbs. Magnesium Sulphate	30 lbs. MgO; 60 lbs. SO <sub>3</sub>
4 & 4+	150 lbs. Magresium Chloride	30 lbs. MgO; 53 lbs. Cl
5 & 5+	185 lbs. Magnesium Sulphate	60 lbs. MgO; 60 lbs. SO <sub>3</sub>
0 00 0 +	150 lbs. Magnesium Chloride	53 lbs. Cl
6 & 6+	107 lbs. Sodium Sulphate	60 lbs. SO <sub>3</sub> ; 53 lbs. Cl;
0 40 0+	88 lbs. Sodium Chloride	$(96 \text{ lbs. Na}_2\text{O})$

<sup>&</sup>lt;sup>1</sup>The plots were conducted on a cooperative arrangement between the United States Department of Agriculture and the Connecticut Agricultural Experiment Station.

<sup>&</sup>lt;sup>2</sup>Agent, Office of Tobacco and Plant Nutrition, Bureau of Plant Industry, United States Department of Agriculture.

200

In addition to the basal ratio, magnesia, sulphur and chlorine were supplied annually as shown in Table 40. No magnesia has been applied on plots 1, 1+, 6 and 6+, no sulphur on plots 1, 1+ 4 and 4+, and no chlorine on plots 1, 1+ 3 and 3+ during the six years period. All other plots received applications of magnesia sulphur and chlorine at the rates specified in the preceding table Part of these fertilizers was uniformly drilled in the row a few days before transplanting the tobacco. The remaining portion was applied as top dressing at the time of the first hoeing. There were no great differences in field growth, but plots 1, 1+, 6 and 6+ were smaller and plots 4, 4+, 5 and 5+ were somewhat larger than the average.

BULLETIN 299

The tobacco on all plots that received no magnesia in the fertilizer exhibited the light colored mottling characteristic of magnesium starvation during the seasons of 1922, '24, '25 and '27. Nearly one hundred percent of the leaves on some plots were mottled in 1927. This condition we term sand-drown. It might be well to briefly describe this malady. Magnesia deficiency results in a very characteristic mottling which ordinarily develops first on the lower leaves of the plant and usually begins at the tip and progresses inward toward the base of the leaf, more particularly between the veins and along the margins. When magnesia is not sufficiently available in the soil these symptoms appear and as magnesia is very mobile in the plant that which is already taken into the lower leaves is transferred to meet new needs in the growing region. Thus the malady progresses from the lower leaves upward as the season advances. The leaves do not become deformed as in the case of potash hunger because symptoms ordinarily develop on mature leaves.

It will be noted that no sand-drown was present in 1923 and '26. Sand-drown has not occurred in the Valley to any extent in the past except on sandy soil during seasons with heavy rainfall. This explains in part the irregularity of occurrence on these plots. Both surface and subsoil samples were analyzed for magnesia, sulphur and calcium in the Washington laboratories. These data show that the total magnesia content of this sandy soil was nearly equal to the content of calcium which is much higher than would be expected. There were no significant differences in treatments 1, 3 and 5 where 0, 30, and 60 pounds of magnesia, respectively, have been supplied annually for six years. Neither were there any differences in sulphur content but the total amount of sulphur was so low that accurate data could not be expected. With as high a total magnesia content as shown in these analyses evidently enough became available to the plant during the dry years to promote a normal growth. If the magnesia content of this soil had been depleted the absence of magnesia in the fertilizer would have greatly lowered both yield and quality. This has been shown by experiments in other sections. It has been reported by

the Tobacco Station that fifteen pounds of magnesia per acre has prevented the occurrence of sand-drown on the Station fertilizer plots. A number of cases of sand-drown were reported during the past season resulting from a combination of two things, an insufficient amount of magnesia applied, together with the heavy rainfall. This is just an indication of what may happen in the future when less nitrogen is supplied from organic sources which carry magnesia. The point is, if a shortage does occur, it must be corrected in one way or another if yield and quality are to be expected. But if the rate of application is excessive there is a tendency to decrease the quality of leaf. These tests have shown that 30 pounds of magnesia has been sufficient to prevent sand-drown in any year during the experiment but a number of instances can be cited where 15 pounds was not sufficient during the past season. No effects of sulphur or chlorine were discernible in the field except for slight differences in growth.

Plants were selected at harvest in 1925 from treatments 1, 3 and 5. The cured leaf and stalk were analyzed for magnesia in the laboratories at Washington. The results are given in Table 41.

TABLE 41. MAGNESIA CONTENT OF CURED TOBACCO

Plot	Magnesia (MgO) supplied in Fert. per acre	Condition of leaf	% Magne in water-fre Leaf	sia (MgO) ee material Stalk
1 3	None 30 lbs.	Sand-drown Normal	.30 1.40	.25
5	60 lbs.	Normal	2.07	.45
1+A 1+B	None None	Normal Sand-drown	.57 .27	.27 .23

These data show that the amount of magnesia taken up by the plant is influenced by the amount applied to the soil. In other words, if sufficient magnesia is available in the soil sand-drown will not occur. They also show that the leaf is influenced much more than the stalk because of the greater metabolic activity in the leaf as compared with the stalk.

TABLE 42. SUMMARY OF THE YIELDS PER ACRE

Plot	1924	1925	1926	1927	Average	Plots	Ave. of 8 replications.
1	880	1456	1368	1266	1242	1&1+	1192
1+	780	1290	1392	1105	1142	LEE TOBE	
2	1140	1465	1392	1364	1340	2 & 2+	1302
2+ 3	1000	1425	1342	1288	1264	1	
	920	1490	1488	1362	1315	3 & 3+	1293
3+	900	1490	1379	1314	1271		
4	1000	1537	1414	1347	1324	4&4+	1345
4+ 5+ 5+	1000	1582	1534	1348	1366		
5	960	1443	1466	1448	1329	5&5+	1344
9+	1000	1540	1534	1368	1360		
6+	900	1321	1388	1250	1215	6&6+	1286
0+	1067	1477	1533	1352	1357		

Yield and quality data were obtained from these plots the last four years of the experiment. A summary of these yields are presented in Table 42. We could not expect great differences in . yield of cured leaf from such treatments unless either magnesia or sulphur was so depleted or unavailable to the plant as to be a limiting factor in growth processes. From the data presented this. appears to be the case on plots which received no magnesia. The yields from the control plots which received no magnesia, sulphur or chlorine are considerably lower than the rest. Where only one of the elements was lacking the effect was not as marked. It is quite evident from a comparison of treatments 3 and 4, however, that somewhat larger yields were consistently obtained from the application of chlorine in absence of sulphur than from the application of sulphur in the absence of chlorine.

The cured leaf was assorted into the various commercial grades at the Tobacco Station shop. As it is rather difficult to keep a number of figures in mind, such as the percentages of assorted grades, when comparing two treatments, a single figure was devised to represent the quality of cured leaf from the entire plot. This single figure is termed the average price per pound and was computed from arbitrary values given to assorted grades. The average prices over a four year period are summarized in Table 43.

TABLE 43. A COMPARISON OF THE AVERAGE PRICE PER POUND

Plot	1924	1925	1926	1927	3 yr. ave. 1925-1927	Plots	Ave. of 8 replications.
1	.179	.277	.395	.316	.329	1&1+	
1+	.158	.262	.427	.285	.325		.021
2	.148	.258	.329	.398	.328	2 & 2+	.340
$\frac{2}{3}$	1111	.298	.383	.378	.353		
	.155	.320	.408	.403	.377	3 & 3 +	.377
3+		.340	.406	.388	.378 •	E)(0.000)	
4	.169	.310	.368	.372	.350	4 & 4+	.369
4+	1111	.342	.404	.420	.388		
5	.151	.288	.401	.383	.357	5 & 5+	.350
5+	1111	.282	.369	.380	.344		
6	.175	.313	.408	.329	.350	6 & 6+	.356
6+		.357	.346	.385	.362		

If quality is considered on this basis we obtained the lowest average price per pound when no magnesia, sulphur or chlorine were supplied in the fertilizer, and the highest average price when magnesia and sulphur were applied in the absence of chlorine. It will be noted that the quality of cured leaf improved from year to year with the increased annual application of nitrogen and potash but the relative effects of magnesia, sulphur and chlorine remained the same aside from effects on quality when sand-drown occurred.

Burn tests were conducted on samples of sweated tobacco from the four principle commercial grades in each treatment. An

electric resistance coil was used to initiate the burns in these tests. Two burns, one on either side of the mid rib and approximately in the center of the leaf were made on ten leaves from each grade.

REPORT OF TOBACCO SUBSTATION

TABLE 44. BURN TESTS ON THE 1925 SWEATED TOBACCO Dir. 1. 11: - - - - sites of the loof (Coconda)

		Fire holding c	apacity of th	e leaf (Secor	ids)	Average of
Plot	Light Wrappers	Medium Wrappers	Light Seconds	Darks	Plot Average	duplicate plots
1 1+·	13.7 10.5	9.9	$\frac{14.5}{9.6}$	11.8 5.9	13.3 9.0	11.2
2 2+	3.0 3.3	$\frac{3.2}{2.6}$	$\frac{3.1}{3.9}$	$\frac{3.1}{2.7}$	$\frac{3.1}{3.1}$	3.1
3 3+	12.4 17.5	16.0 13.9	15.5	$9.5 \\ 11.2$	$13.4 \\ 14.2$	13.8
4	2.8 5.9	2.4 2.8	2.9 3.0	2.9	2.8 3.9	3.4
4+ 5 5+	3.2 3.1	3.4 2.9	5.2 6.1	3.5 2.6	3.8 3.7	3.8
6 6+	3.5 3.2	3.0 4.2	3.9 3.7	3.4	3.5 3.7	3.6

Each plot average represents eighty burn tests.

The decidedly harmful effect of chlorine is demonstrated in Table 44 where the average length of burns is recorded on the 1925 crop. While the fire-holding capacity, as determined by leaf tests, was poor throughout the experiment in 1925 the burn was three to four times as great in the no chlorine treatments. At the rates supplied in these tests sulphur and magnesia failed to show any harmful effects on the burn.

#### SUMMARY

The necessity for including magnesia in the fertilizer mixture is demonstrated by these data. This is of special importance at this time when the trend is toward more concentrated fertilizers.

If magnesia is deficient in the soil and is not supplied in the fertilizer both yield and quality of cured leaf may be greatly lowered.

The amount of magnesia taken up by the plant is influenced by the amount applied to the soil.

Both chlorine and magnesia increased the yields to some extent in these tests.

The treatments do not show any great differences in quality, although the poorest quality resulted from the control plots where no magnesia, sulphur or chlorine were supplied in the fertilizer.

Chlorine almost destroyed the fire-holding capacity of the Cured leaf in these tests. Sulphur or magnesia failed to show any harmful effects on the burn at the rates supplied.

# Connecticut Agricultural Experiment Station

Nem Haven, Connecticut

# THE COMPOSITION OF SOME COMMERCIAL INSECTICIDES, FUNGICIDES, BACTERICIDES, RODENTICIDES AND WEED KILLERS

A COMPILATION

H. J. FISHER AND E. M. BAILEY

# CONNECTICUT AGRICULTURAL EXPERIMENT STATION

#### OFFICERS AND STAFF

#### BOARD OF CONTROL

His Excellency, Governor John H. Trumbull, ex-officio	
Charles R. Treat, Vice-President.	Orange
George A. Hopson, Secretary	Mt. Carmel
Wm. L. Slate, Director and Treasurer	New Haven
Joseph W. Alsop	Avon
Elijah Rogers	Southington
Edward C. Schneider	Middletown
Francis F. Lincoln	Cheshire
진 보기보다 공료하게 보통하다 보는 이 경기 때문에 대표했다. 이 사람이 가지 않는 것이 하나는 것이 되는 것이 되었다. 그리고 있다면 없는 것이 없는데 없다면 없다.	

#### STAFF.

E.	H.	TENKINS.	Рн. D	Director	Emeritus.
14.	44.	1 Trivering,	T 11. 1 . 9	Duccion	Linet www.

Administration.	WM. L. SLATE, B.Sc., Director and Treasurer.
	MISS L. M. BRAUTLECHT, Bookkeeper and Librarian.
	G. E. GRAHAM. In charge of Buildings and Grounds.

Chemistry:	E. M. BAILEY, Ph.D., Chemist in Charge.
Analytical	C. E. SHEPARD
Laboratory.	OWEN L. NOLAN

Botany.

OWEN L. NOLAN	1
HARRY J. FISHER, A.B. W. T. MATHIS	Assistant Chemists
DAVID C. WALDEN, B.S.	

FRANK C. SHELDON, Laboratory Assistant. V. L. CHURCHILL, Sampling Agent. MRS. A. B. VOSBURGH, Secretary.

#### T. B. OSBORNE, PH.D., Consulting Biochemist. H. B. VICKERY, PH.D., Biochemist in Charge. GEORGE W. PUCHER, PH.D., Biochemist Assistant MISS HELEN C. CANNON, B.S., Diettian. Biochemical Laboratory.

G. P. CLINTON, Sc.D., Botanist in Charge.
E. M. STODDARD, B.S., Pomologist.
MISS FLORENCE A. McCORMICK, Ph.D., Pathologist.
H. B. BENDER, M.S., Graduate Assistant
A D M-D C 1 4 ··· ·

. D. McDonnell, General Assistant. MRS. W. W. KELSEY, Secretary.

#### Entomology. W. E. BRITTON, PHD., Entomologist in Charge: State Ento-

B. H. WALDEN, B.AGR. M. P. ZAPPE, B.S.	Assistant	E
PHILIP GARMAN, PH.D.		
ROGER B. FRIEND, PH.D.		

JOHN T. ASHWORTH, Deputy in Charge of Gipsy Moth Work R. C. BOTSFORD, Deputy in Charge of Mosquito Elimination. J. P. JOHNSON, B.S., Deputy in Charge of Asiatic and Japanese Beetle Quarantines.

MRS. GLADYS BROOKE, B.A., Secretary.

ntomologists.

Forestry.

WALTER O. FILLEY, Forester in Charge. H. W. HICOCK, M.F., Assistant Forester. J. E. RILEY, JR., M.F., In Charge of Blister Rust Control. MISS PAULINE A. MERCHANT, Secretary.

Plant Breeding. Donald F. Jones, S.D., Geneticist in Charge. W. R. SINGLETON, S.M., Assistant Geneticist. H. R. MURRAY, B.S., Graduate Assistant. MRS. R. A. HUNTER, Secretary.

M. F. Morgan, M.S., Agronomist in Charge. H. G. M. JACOBSON, M.S., Assistant Agronomist. HERBERT A. LUNT, M.S., Assistant in Forest Soils. DWIGHT B. DOWNS, General Assistant. Soil Research.

PAUL J. ANDERSON, PH.D., Pathologist in Charge. T. R. SWANBACK, M.S., Agronomist. Tobacco Sub-station at Windsor. MISS DOROTHY LENARD, Secretary.

# The Composition of Some Commercial Insecticides, Fungicides, Bactericides, Rodenticides and Weed Killers. A Compilation.

H. J. FISHER AND E. M. BAILEY.

Frequent requests for information as to the composition of commercial insecticides, fungicides and similar materials suggested the desirability of assembling the data on this subject to be found not only in various bulletins of this station but in those of institutions in other States where similar work is carried on. Designed at first for our own convenience, it was later felt that the information would be useful also to others interested. No pretense is made that the compilation is complete, but it is, at least, fairly comprehensive. For the most part the data were obtained from regular or special bulletins submitted by experiment stations or other state departments in response to our request for information on this subject. If any important sources of information are omitted it is because our requests were not referred to, or did not reach, the proper authorities.

In the case of insecticides which are definite compounds, sold under their own or well recognized names, such as arsenate of lead, Paris green, sulphur, etc., the publication of very old analyses did not seem to be warranted, and only those analyses reported from 1918 to the present are included in this index. In the case of products sold under proprietary names which do not make their composition readily evident, all analyses found are included how-

ever long ago they may have been published.

Only one analysis is given for each product except in those cases where different analyses have shown widely different composition. In such cases several analyses are included. An attempt was made to insert the latest analysis found in each case. When the latest Publication found containing analyses of a certain insecticide or other product has given several analyses of that product, the figures in the index are usually averages.

It should be noted that the percentages given under "guaranteed" in the index represent maximum amounts in the case of water, inert matter and water-soluble arsenic, while the guaranties for other constituents are usually stated by the manufacturers as "not

less than" the amount specified.

The index is arranged in alphabetical order by names of the products. Names beginning with arabic numerals are indexed as they would be if the numeral were spelled, - i.e., "20" is listed under "T". Certain common classes of insecticides are grouped together in tables alphabetically by the names of the manufacturers, but they are also cross-indexed by the names of the brands. An exception to this rule is made in the case of arsenate of calcium, arsenate of lead, Bordeaux mixture, lime-sulphur, London purple, Paris green and sulphur, which will not be found listed separately by the brand names, but only under the tables for these respective classes of insecticides.

CONNECTICUT EXPERIMENT STATION

Reference is made after each analysis to the source from which it was obtained. Some of the analyses are those made at this Station and not previously published. After such analyses the sample number and "Conn. Agr. Expt. Sta." is given in place of a publication reference.

No responsibility is assumed for the correctness of any analyses other than those made by this Station.

The following publications were abstracted:

California Department of Agriculture, Special Publications 34

(1923); 51 (1925); 58 (1925); 66 (1926); 75 (1927).

Canada Laboratory of the Inland Revenue Department, Bulletins 158 (1908); 205 (1910); 284 (1914); 303 (1915); Department of Agriculture, Division of Chemistry, Report of the Dominion Chemist (1928).

Colorado Sixteenth Annual Report of State Entomologist,

Circular 47 (1925).

Connecticut Agricultural Experiment Station, Bulletins 157

(1907); 242 (1922); 258 (1924); 272 (1925).

Maine Agricultural Experiment Station, Bulletin 154 (1908); Official Inspections 110 (1923); 114 (1924); 118 (1925); 122 (1926); 126 (1927).

Michigan Agricultural College Experiment Station, Special Bulletin 74 (1915).

New Jersey Agricultural Experiment Stations, Bulletins 214 (1908); 222 (1909); 262 (1913); 273 (1914); 286 (1915); 301 (1916); 407 (1924); 424 (1925); 441 (1926); 459 (1927).

New York Agricultural Experiment Station, Bulletin 348 (1912);

384 (1914).

North Dakota Office of the State Food Commissioner and Chemist, Bulletin 17 (1927).

Oregon Agricultural College Experiment Station, Circular 64 (1925); 84 (1927)

Pennsylvania Department of Agriculture, Bureau of Chemistry, Bulletins 166 (1910); 192 (1909).

University of California College of Agriculture, Agricultural Experiment Station Bulletin 151 (1903).

U. S. Department of Agriculture, Bureau of Chemistry, Bulletin 68 (1902).

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

U. S. Department of Agriculture, Department Bulletin 1439 (1926).

U.S. Department of Agriculture Farmers' Bulletin 146 (1902).

# Ace-Hy.

(GENERAL CHEMICAL CO., NEW YORK, N. Y.)

Guaranteed: Inert matter (water), not more than 40 per cent. Found: The preparation is an emulsion in which a cyanide, equivalent to 2.29 grams CN per 100 cc was the chief active ingredient detected. The ash, 3.58 per cent, consisted chiefly of iron and copper oxides. Water and volatile matter (at 100° C), made up 77.3 per cent of the material.—Conn. Agr. Expt. Sta. Bull. 242, 161 (1922).

#### Acid Carbolic.

See "Phenol."

#### Acid Hydrocyanic.

See "Hydrocyanic Acid.

#### Acme Garden Guard.

(ACME WHITE LEAD & COLOR WORKS, DE	TROIT, MICH.)	
	Guaranteed.	Found.
Copper Total arsenic, metal. Water-soluble arsenic, metal. Copper aceto-arsenite. "Copper of bordo". Sulphur.	$1.70 \\ 0.25 \\ 4.50$	1.77 1.98 0.50  3.90

Cal. Dept. Agr., Spec. Pub. 75, 44 (1927).

# Acme Sheep and Cattle Dip.

See "Phenol-Soap Solutions."

# Acme 2-way Spray.

See "Bordeaux Mixture-Lead Arsenate."

Acto.

See "Oils, Mineral."

Mois

Tota

Wat

#### Adheso Green Label.

(ANSBACHER INSECTICIDE CO., INC., NEW YORK, N. Y.)

	Guaranteed.	Found.
sture		58.39
al arsenic, metal	5.50	5.56
ter-soluble arsenic, metal	0.50	0.05
per	4 00	4 15

N. J. Agr. Expt. Sta., Bull. 459, 10 (1927).

#### Adheso 7 O-Blue Dust.

	(ANSBACHER	INSECTICIDE	CO., INC.,	NEW YORK, N. Y.)	
0				Guaranteed.	Found.
Copper				7.00	7.25
N. T	Aar Fret Sto	Bull 450	10 (1927)		

#### Adheso-Yellow Label.

See "Concentrated Adheso."

#### A D S Rat and Roach Paste.

See "Phosphorus Preparations."

#### Aero Brand Hydrocvanic Acid.

See "Hydrocyanic Acid."

#### Aero Rodent Exterminator.

See "Calcium Cyanide."

#### Agricultural Sulphur Compound.

(TOYAH VALLEY SULPHUR CO., NEW ORLEANS, LA.)

Found: Sulphur 21.50 per cent.— N. J. Agr. Expt. Sta., Bull. 441 7 (1926).

#### Agri-Pax.

(PAX MFG. CO., NEW YORK, N. Y.)

Declared: "A contact spray insecticide based on the extract of pyrethrum. Inert material water 77 per cent."

Found: Chloroform extract 20.54 per cent; fatty acids 18.83 per cent; moisture 74.2 per cent. Calculated composition: soap 21.37 per cent; pyrethrum extract 1.71 per cent; moisture 74.2 per cent; undetermined 2.72 per cent.—Conn. Agr. Expt. Sta., Sample 9305.

#### Albatross Puritol.

(GENERAL BASIC PRODUCTS CO.)

	Guaranteed.	Found.
Active Ingredients:	3.00	de
Oils		1.20
Phenols		0.40
Inert Ingredients:		
Whiting	91.50	90.50
Glue	5.00	5.40
Volatile at 100° C	0.50	1.50
Cal. Dept. Agr., Spec. Pub. 75, 66 (1927).		

#### Alcopol Oil.

## Aleph Oil

See "Oils, Mineral."

Alhambra Nico-Soap.

See "Nicotine Soaps."

Alhambra Spray.

See "Oil Emulsions, Mineral."

#### Allan's Lightning Roach Paste.

See "Phosphorus Preparations."

#### Allen's Kilto, Paste.

(W. A. ALLEN, PITTSTOWN, N. J)

	Guaranteed.	Found.
Moisture		55.17
Total arsenic, metal	6.52	4.73
Water-soluble arsenic, metal	0.50	0.08
Copper	4.00	5.02
N. J. Agr. Expt. Sta., Bull. 441, 10 (1926).		

#### All in One.

See "Niagara All in One" and "No. 6 All in One Dust."

#### Altair Oil.

See "Oils, Mineral."

#### American Jazz Spray.

See "Oil Emulsions, Mineral."

#### Ammonia Spray.

(BEAR AMMONIA CO., RIVERSIDE, CAL.)

Found: Moisture 18.33 per cent; sodium carbonate 63.50 per cent: sodium chloride 3.20 per cent; soap 12.85 per cent; calcium carbonate 2.09 per cent.—Eal. Dept. Agr., Spec. Pub. 51, 58 (1925).

#### Anchor Brand Hellebore.

See "Hellebore."

#### An-Fo Disinfectant.

(AN-FO MFG. CO., OAKLAND, CAL.)

See "Phenol-Soap Solutions."

#### An-Fo Louse Powder.

(ANIMAL FOOD CO., OAKLAND, CAL.)

Not chemically analyzed.—Cal. Dept. Agr., Spec. Pub. 58, 48 (1925).

#### An-Fo Nicotine Spray.

(AN-FO MFG. CO., OAKLAND, CAL.)

See "Nicotine Soaps."

#### An-Fo Round Worm Capsules.

(AN-FO MFG. CO., OAKLAND, CAL.)

	Guaranteed.	Found.
Nicotine	15.00	16.36
Cal. Dept. Agr., Spec. Pub. 66, 26 (1926).		

#### An-Fo Sheep Dip.

See "Phenol-Soap Solutions."

#### Ansbor Green.

(ANSBACHER INSECTICIDE CO., INC., NEW YORK, N. Y.)

	Guaranteed.	Found.
Total arsenic, metal	12.50	16.14
Water-soluble arsenic, metal	3.00	0.27
Copper		17.62
N. I Agr Erect Sta Bull 450 10 (1927)		

#### Ant Destroyer.

(PETER HENDERSON, NEW YORK, N. Y.)

Found: Borax 54.12 per cent; cane sugar 44.80 per cent.— U. S. D. A., Bur. Chem., Bull. 68. 43 (1902).

#### Ant Eater.

(AN-FO MFG. CO., OAKLAND, CAL.)

	Guaranteed.	Found.
Total arsenic	0.17	0.16
Cal. Dept. Agr., Spec. Pub. 75, 24 (1927).		

Found.

0.12

#### Ant Foil Ant Poison

FO MEG.	CO	OAKLAND	CAT )

,,	11111	
	Guaranteed.	Found.
Arsenious oxide		0.16
Arsenic, metal	0.10	0.12
Cal. Dept. Agr., Spec. Pub. 66, 19 (1926).		

#### Anthracine Oil Emulsion.

(THE SHERWIN-WILLIAMS CO., CLEVELAND, OHIO.)

Guaranteed: Anthracene oil 75 per cent; fish oil soap 3 per cent; water 22 per cent.

Found: Total oil 72.63 per cent; water 25.08 per cent; soap and undetermined 2.29 per cent.—Conn. Agr. Expt. Sta., Bull. 272, 148 (1925).

#### Anti-Ant Argentine Ant Poison.

(BRUNSWIG DRUG CO., LOS ANGELES, CAL.)

Guaranteed: Arsenic metal 0.20 per cent.

Found: Arsenious oxide 0.76 per cent; arsenic metal 0.057 per cent; dextrose 45.36 per cent; maltose 5.87 per cent; dextrin 0.54 per cent; moisture 46.61 per cent.—Cal. Dept. Agr., Spec. Pub. 58, 16 (1925).

#### Antiseptic Sheep Dip.

See "Phenol Soap Solutions."

#### Ant Poison.

(ALDERMAN CO., INC., PASADENA, CAL.)

Found: Arsenious oxide 0.19 per cent; arsenic metal 0.14 per cent.— Cal. Dept. Agr., Spec. Pub. 58, 16 (1925).

#### Antrol Argentine Ant Syrup.

	(ANTROL	LABORATORIES,	INC.,	LOS	ANGELES,	CAL.)	
					Gu	aranteed.	
-	motal					0.10	

Cal. Dept. Agr., Spec. Pub. 75, 24 (1927).

#### Ant Syrup.

(SMALLS SEED CO., RIVERSIDE, CAL.)

	Guaranteed.	Found.
Arsenic, metal	0.10	0.17
Cal. Debt. Agr., Spec. Pub. 75, 24 (1927).		

#### A-1 Dust Mixture.

See "Nicotine Dusts."

#### Aphid Spray.

(RADIUM SPECIALTIES MFG. CO., SAN FRANCISCO, CAL.)

Found: Light rosin oil 95.4 per cent; water 4.6 per cent.—Cal. Dept. Agr., Spec. Pub. 34, 58 (1923).

#### Aphine.

(APHINE MFG. CO., MADISON, N. J.)

Guaranteed: Nicotine 0.90 per cent.

Found: Nicotine 1.07 per cent. Cedar and pine oils and potassium hydroxide present.—N. J. Agr. Expt. Sta., Bull. 407, 10 (1924)—N. Y. Agr. Expt. Sta., Bull. 384, 301 (1914).

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

#### Aphis Getter.

(WALNUT CREEK SPRAY CO., WALNUT CREEK, CAL.)

	Guaranteed.	Found.
Water	92.00	69.20
Cal Debt Agr Spec Pub 66 36 (1926)		

#### Aphistrogen.

(CHEMICAL PRODUCTS DIVISION, ROSE MFG. CO., PHILADELPHIA. PA.)

Guaranteed: C35H46O10 about 3/4-1 per cent; "trioxymethaline" about 34-1 per cent; available nitrogen about 0.4 per cent; "inert remedial substances" about 94 per cent. "Aphistrogen is the active glucoside content of a vegetable substance. Aphistrogen being highly concentrated goes 60 times farther than the amount you pay for. Aphistrogen is prepared after the famous Rosenbluth formula, combined with a fertilizer as used with unprecedented success at the nationally known Rose Gardens at Wallingford, Pa."

Found: The preparation is a soap emulsion. Total nitrogen 0.14 per cent; water 80.9 per cent; formaldehyde recovered by steam distillation

0.05 per cent.—Conn. Agr. Expt. Sta., Sample 9306.

#### Apricot Oil.

See "Oil Emulsions, Mineral."

#### Arcol.

See "Phenol Soap Solutions."

#### Arctic Whale Oil Soap.

See "Soaps."

#### Argentine Ant Poison.

(COFFIN-REDINGTON CO., SAN FRANCISCO, CAL.)

Guaranteed: Arsenious oxide 0.20 per cent; arsenic metal 0.15 per cent. Found: Arsenious oxide 0.26 per cent; arsenic metal 0.20 per cent; invert sugar 10.79 per cent; cane sugar 17.47 per cent; dextrin, etc., 11.53 per cent. water 59.94 per cent.—Cal. Dept. Agr., Spec. Pub. 58, 16 (1925).

#### Argentine Ant Poison

(FEDERAL DRUG CO., OAKLAND,	CAL.)	
	Guaranteed.	Found.
Arsenious oxide		0.11
Arsenic, metal	0.25	0.08
Cal. Dept. Agr., Spec. Pub. 66, 19 (1926).		

#### Argentine Ant Poison.

(C. W. HILL CHEMICAL CO., LOS ANGELES, CAL.)

	Guaranteed.	Found.
Total arsenic, metal	0.10	0.097
Water-soluble arsenic, metal	0.10	0.097
Cal. Dept. Agr., Spec. Pub. 34, 18 (1923).		

#### Argentine Ant Poison.

(LOS ANGELES CHEMICAL CO., LOS ANGELES, CAL.)

Guaranteed: Arsenic metal 0.10 per cent.

Found: Arsenious oxide 0.13 per cent; arsenic metal 0.099 per cent; invert sugar 8.64 per cent; cane sugar 39.66 per cent; dextrin, etc. 0.21 per cent; water 51.64 per cent.—Cal. Dept. Agr., Spec. Pub. 58, 17 (1925).

Arsenious oxide...

Cal. Dept. Agr., Spec. Pub. 58, 17 (1925).

#### Argentine Ant Poison.

(PHILIP & PHILIP, OAKLAND,	CAL.)	
	Guaranteed.	Found.
	ö.ii	0.16
	0.11	0.12

BULLETIN 300

# Argentine Ant Poision.

# (ROSE WATERMAN DRUG CO.)

	Guaranteed.	Found.
Arsenious oxide	0.20	0.20 0.15
Cal. Dept. Agr., Spec. Pub. 66, 19 (1926).		

#### Argentine Ant Poison.

#### (TOWNE ALLISON DRUG CO., SAN BERNARDINO, CAL.)

Found: Total arsenious oxide 0.60 per cent; water-soluble arsenious oxide 0.60 per cent.—Cal. Dept. Agr., Spec. Pub. 34, 18 (1923).

#### Argentine Ant Poison.

#### (WEBB & SEWARD, PASADENA, CAL.)

	Guaranteed.	Found.
Arsenious oxide Arsenic, metal	0.20	0.09 0.07
Cal. Debt. Agr., Spec. Pub. 58, 17 (1925)		

#### Argo Ant Poison.

#### (BOWMAN DRUG. CO.)

The Latino Alba Continue and the	Guaranteed.	Found.
Arsenic, metal	0.11	0.15
Cal Debt Agr Spec Pub 75 24 (1927).		

#### A. R. M. Liquid Lice Killer.

#### (ARTHUR R. MAAS CHEMICAL CO.)

Guaranteed: Water 5.00 per cent. Found: Phenols 14.50 per cent; coal-tar oils 84.80 per cent; water 0.40 per cent.—Cal. Dept. Agr., Spec. Pub. 75, 56 (1927).

#### Arsenate of Calcium. Arsenate of Lead.

See Tables I and II.

#### Arsenate of Barium.

See "White Arsenoid."

#### Arsenite of Lead.

See "Pink Arsenoid."

Arsenite of Zinc.

See Table III.

Arsenoid, Pink.

See "Pink Arsenoid."

Arsenoid, Soluble.

See "Watson."

Arsenoid White.

See "White Arsenoid."

# Arsite.

#### (MORRIS HERRMANN & CO., NEW YORK, N. Y.)

	Guaranteed.	Found.
Total arsenic, metal	40.0	$\frac{31.05}{31.05}$
Mich. Agr. College, Expt. Sta., Spec., Bull. 74,	10 (1915).	

#### At-Ko Brand Sheep Dip.

See "Phenol Soap Solutions."

#### Atlas "A" Weed Killer.

#### (CHIPMAN CHEMICAL CO., NEW YORK, N. Y.)

Guaranteed: Sodium arsenite 45 per cent. Found: Total arsenic metal 23.67 per cent; sodium arsenite (Na<sub>2</sub>HAsO<sub>3</sub>) 53.64 per cent.—Conn. Agr. Expt. Sta., Bull. 272, 149 (1925).

#### Atlas Weed Killer.

#### (CHIPMAN CHEMICAL ENGINEERING CO., INC. NEW YORK, N. Y.)

Found: Sodium chlorate 19.22 per cent; sodium chloride 18.73 per cent; water 57.59 per cent.—Cal. Dept. Agr., Spec. Pub. 75, 65 (1927).

#### Avenarius Carbolineum.

#### (CARBOLINEUM WOOD PRESERVING CO.)

	Guaranteed.	Found.
Water	1.50	1.43
Carbon	0.50	0.97
Cal. Dept. Agr., Spec. Pub. 75, 65 (1927).		

#### Avon Brown Neutral.

See "Oils, Mineral."

#### Axfixo.

See "Nicotine Dusts."

#### B.

#### B. A. Cartridges.

See "Key Brand B. A. Cartridges."

#### Bacili-Kil.

See "B-K Bacili-Kil."

#### Barium Arsenite.

See "White Arsenoid."

TABLE I. ARSENATE OF CALCIUM.

		Arsenic O	xide, As <sub>2</sub> 6	)5			
	To	otal.	Water	Water-soluble			
Manufacturer or Distributor and Brand	Guaranteed	Found	Guaranteed	Found	Calcium Arsenate, Cas(As04)2	Publication	
General Chemical Co., New York, N. Y. Orchard.  The Glidden Co., Reading, Pa.  Interstate Chemical Co., Jersey City, N. J. Key.  Leggett & Bro., Inc., New York, N. Y.	39.87 39.87  40.00 41.00 45.84 39.87	41.51 42.03	1.53 1.15 1.15 1.15 0.75  0.75 1.53 1.15	0.67 0.15 0.24 0.57 0.08 0.54 0.52 2.41 0.18	72.39 67.34 72.02 71.30 72.60  69.01 72.42 71.89 72.79	<ul> <li>N. J. Agr. Expt. Sta., Bull. 459, 6 (1927).</li> <li>Maine Agr. Expt. Sta., Official Inspection 126, 84 (1927).</li> <li>Maine Agr. Expt. Sta., Official Inspections 114, 86 (1924).</li> <li>N. J. Agr. Expt. Sta., Bull. 459, 6 (1927).</li> <li>N. J. Agr. Expt. Sta., Bull. 407, 8 (1924).</li> <li>Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).</li> <li>Cal. Dept. Agr., Spec. Pub. 75, 21 (1927).</li> <li>Cal. Dept. Agr., Spec. Pub. 75, 21 (1927).</li> <li>N. J. Agr. Expt. Sta., Bull. 407, 8 (1924).</li> <li>N. J. Agr. Expt. Sta., Bull. 459, 6 (1927).</li> </ul>	
Anchor	39.87 39.87	$\begin{vmatrix} 42.76 \\ 40.97 \end{vmatrix}$	1.15		74.05 70.95	N. J. Agr. Expt. Sta., Bull. 424, 9 (1925). N. J. Agr. Expt. Sta., Bull. 424, 9 (1925).	

Lucas-Kil-Tone Co., Vineland, N. J., Green Cross	39.87 20.14	40.61	0.77	$0.21 \\ 0.82$	70.32	N. J. Agr. Expt. Sta., Bull. 459, 6 (1927). Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).
Mechling Bros. Chemical Co., Camden, N. J	39.87 40.00	39.10 43.47	1.00 1.15		67.72 75.20	N. J. Agr. Expt. Sta., Bull. 459, 6 (1927). Cal. Dept. Agr., Spec. Pub. 75, 21 (1927).
Centro, Cal. Flag	40.00		0.77		71.85	Cal. Dept. Agr., Spec. Pub. 75, 21 (1927). Cal. Dept. Agr., Spec. Pub. 75, 21 (1927).
Nitrate Agencies Co., New York, N. Y.	40.02		0.75		74.05	N. J. Agr. Expt. Sta., Bull. 424, 9 (1925).
Pittsburgh Plate Glass Co., Newark, N. J. Corona Calsenate	40.00		1.15		70.42	Cal. Dept. Agr., Spec. Pub. 75, 21 (1927)
Ceres	39.87 40.00	$\begin{vmatrix} 44.70 \\ 41.49 \end{vmatrix}$	$ \begin{array}{c} 1.15 \\ 0.77 \end{array} $		77.42 71.78	N. J. Agr. Expt. Sta., Bull. 424, 9 (1925) Cal. Dept. Agr., Spec. Pub. 75, 22 (1927)
N. J. Electro	39.87	42.33	1.15 0.77		73.30	N. J. Agr. Expt. Sta., Bull. 441, 6 (1926) Cal. Dept. Agr., Spec. Pub. 75, 22 (1927)
Oai. Dovoc	10.00	11.01	0.	1.00	1.0.00	Cur. Dopo. 11g1., opec. 1 ub. 10, 22 (1021)

	N	

	Water		Arsenic Oxide, As <sub>2</sub> O <sub>5</sub>					Oxide,	
			То	Total		Water-Soluble		bO	
Manufacturer or Distributor and Brand ·	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	Publication
Acme White Lead & Color Works,	%	%	%	%	%	%	%	%	Conn Age Frot Sto Dull 949 151
Detroit, Mich. Acme			30.00	33.12	1.00	0.14		63.65	Conn. Agr. Expt. Sta., Bull. 242, 151 (1922).
N. Y			30.00	32.22	1.00	0.59		65.16	N. J. Agr. Expt. Sta., Bull. 459, 5 (1927).
The James Blanchard Co., New York, N. Y				30.80		0.23		66.90	Ore. Agr. Expt. Sta., Cir. 64, 8 (1925).
Bowker Insecticide Co., Boston, Mass. Bowker's	50.00	49.45	15.00	15.60	0.60	0.09		32.37	Conn. Agr. Expt. Sta., Bull. 272, 145 (1925).
Bowker Insecticide Co., Boston, Mass. Bowker's			30.00	30.59	0.77	0.38		66.31	Conn. Agr. Expt. Sta., Bull. 242, 150 (1922).
California Rex Spray Co., Benicia, Cal. NuRexform			30.00	31.51	0.77	0.45		64.50	Cal. Dept. Agr., Spec. Pub. 75, 14 (1927).
California Rex Spray Co., Benicia, Cal. Rex Basic			22.00	22.23	0.77	0.21		74.93	Cal. Dept. Agr., Spec. Pub. 75, 17 (1927).
California Rex Spray Co., Benicia, Cal. Rex Neutral Dry			24.00	32.77	0.77	0.46		64.70	Cal. Dept. Agr., Spec. Pub. 66, 16 (1926).
California Spray Chemical Co., Watsonville, Cal. Ortho Spray Basic.			22.00	22.62		0.12		75.19	Cal. Dept. Agr., Spec. Pub. 75, 17 (1927).
California Spray Chemical Co. Wat- sonville, Cal. Ortho Spray Standard			30.00			0.23		65.25	Cal. Dept. Agr., Spec. Pub. 75, 14 (1927).
Chipman Chemical Engineering Co., New York, N. Y. Chipman			30.00		1.07	0.21		63.95	Conn. Agr. Expt. Sta., Bull. 272, 145 (1925).
Corona Chemical Co., Milwaukee, Wis. Corona Dry			30.00		0.77	0.21		64.25	Conn. Agr. Expt. Sta., Bull. 242, 150 (1922).

Deloro Chemical Co., Deloro, Ont., Canada Detroit White Lead Works, Detroit,		0.28		31.71		0.08	 64.80	Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928). Conn. Agr. Expt. Sta., Bul. 242, 150
Mich. Rogers  Devoe & Raynolds Co., Inc., New			30.00	31.51	1.00	0.18	 63.23	(1922).
York, N. Y. Devoe Standard		A COLUMN	31.00	The second second	1.00	0.97	63.65	Conn. Agr. Expt. Sta., Bull. 242, 150 (1922).
John Dorland, Cobalt, Ont., Canada <sup>1</sup>							 	Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).
The Dow Chemical Co., Midland, Mich. Dow			30.00	31.59	0.99	0.72	64.03	Cal. Dept. Agr., Spec. Pub. 75, 14 (1927).
General Chemical Co., New York, N. Y. Orchard			30.00	31.74	1.50	0.18	65.07	Conn. Agr. Expt. Sta., Bull. 242, 150 (1922).
General Chemical Co., New York, N. Y. Orchard Basic			22.00	22.35	0.38	0.18	72.73	Cal. Dept. Agr., Spec. Pub. 75, 17 (1927).
General Chemical Co., New York, N. Y. Orchard Paste			15.00		0.38	0.17	31.16	Cal. Dept. Agr., Spec. Pub. 75, 14 (1927).
General Chemical Co., New York, N. Y. Orchard Tri Plumbic.			25.00	18 A A A A A A	0.38	0.34	69.90	Cal. Dept. Agr., Spec. Pub. 75, 20 (1927).
General Chemical Co., New York,	52.00				0.38	0.34	33.51	Cal. Dept. Agr., Spec. Pub. 66, 16
The Glidden Co., Cleveland, Ohio.								(1926). Conn. Agr. Expt. Sta., Bull. 242, 150
The Grasselli Chemical Co., New			31.00		0.77	0.44	63.46	(1922). Conn. Agr. Expt. Sta., Bull. 272, 145
York, N. Y. Grasselli The Grasselli Chemical Co., New	1.11		31.00		1.15	0.16	62.65	(1925) Cal. Dept. Agr., Spec. Pub. 66, 16
York, N. Y. Basic Powder The Grasselli Chemical Co., Cleve-			23.00		0.75	0.17	 74.10	(1926). Conn. Agr. Expt. Sta., Bull. 242, 150
land, Ohio. Grasselli Paste C. W. Hill Chemical Co., Los Angeles,	50.00	43.73	15.00	17.80	0.50	0.29	 36.47	(1922). Cal. Dept. Agr., Spec. Pub. 34, 15
Cal. Mission		0.18		31.28		0.33	 64.84	(1922). Conn. Agr. Expt. Sta., Bull. 242, 150
N. J. Key-DryInterstate Chemical Co., Jersey City,			30.00	30.88	0.75	0.23	 63.18	(1922) and Bull. 272, 145 (1925). N. J. Agr. Expt. Sta., Bull. 424, 8
N. J. Key Brand Paste			15.50	17.04 30.24	1.15	0.60	 35.16	(1925).
The isit-10th Co., vinciand, IV. J			00.00	00.24	1,10	0.10	63.55	Conn. Agr. Expt. Sta., Bull. 272, 145 (1925).
Control of the Contro		1				COLUMN TO SERVICE	1 200	SPECIAL PROPERTY OF THE PROPER

<sup>&</sup>lt;sup>1</sup>Found to be ground gypsum.

			Arsenic O	xide, As <sub>2</sub> C	)5	Lead	Oxide,			
	Water		Total		Water-Soluble		PbO			
Manufacturer or Distributor and Brand	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	Publication	
Latimer-Goodwin Chemical Co., Latimer Leggett & Bro., New York, N. Y	50.00	% 44.23	% 28.38 14.00	% 30.94 18.03	% 0.77 0.75	% 0.32 0.11	30.00	% 64.07 36.43	Cal. Dept. Agr., Spec. Pub. 75, 15 (1927). Conn. Agr. Expt. Sta., Bull. 242, 150	
Leggett & Bro., New York, N. Y			30.00	32.66	1.00	0.28	61.00	64.50	(1922). Conn. Agr. Expt. Sta., Bull. 242, 150 (1922).	
John Lucas & Co., Inc., Philadelphia, Pa Lucas Kil-Tone Co., Vineland, N. J. Green Cross R. G. Maxtone-Graham, Berkeley, Cal.		0.40	30.00		1.15	0.40 0.58 0.38		64.40 66.25 63.50	Cal. Dept. Agr., Spec. Pub. 75, 15 (1927). Cal. Dept. Agr., Spec. Pub. 75, 20 (1927). Cal. Dept. Agr., Spec. Pub. 34, 15	
Mechling Bros. Chemical Co., Camden, N. J. Paste and Water  Mechling Bros. Chemical Co., Camden, N. J. Powder  Merrimac Chemical Co., Boston, Mass. Swift's			10.00 31.00 12.50		0.75 0.75 0.75	0.07 0.21 0.26		20.94 65.20 33.48	(1922). N. J. Agr. Expt. Sta., Bull. 459, 6 (1927). N. J. Agr. Expt. Sta., Bull. 459, 5 (1927). Conn. Agr. Expt. Sta., Bull. 242, 150 (1922).	
Montgomery Ward & Co., Portland, Ore			30.00 30.00 22.00	31.39	1.00 0.77 0.77	0.20 0.27 0.11		65.10 64.26 74.36	Ore. Agr. Expt. Sta., Cir. 84, 8 (1927). Cal. Dept. Agr., Spec. Pub. 75, 15 (1927). Cal. Dept. Agr., Spec. Pub. 75, 18 (1927).	

Niagara Sprayer Co., Middleport, N. Y. Niagara			30.00	31.54	1.15	0.18		64.35	Conn. Agr. Expt. Sta., Bull. 272, 145 (1925).
Niagara Sprayer Co., Middleport, N. Y. Niagara Basic Niagara Sprayer Co., Middleport,			22.00	22.52	0.77	0.11		75.28	Cal. Dept. Agr., Spec. Pub. 75, 18 (1927).
N. Y. No. 1 Niagara Entodust Nitrate Agencies Co., New York,			30.00	31.17	0.77	0.29		66.20	Cal. Dept. Agr., Spec. Pub. 75, 15 (1927). Conn. Agr. Expt. Sta., Bull. 242, 150
N. Y. Naco		1.1	30.00		1.53		62.00		(1922). Conn. Agr. Expt. Sta., Sample 420.
Oderberg, Czechoslovakia Pittsburgh Plate Glass Co., Newark				27.06		2.16		59.96	Conn. Agr. Expt. Sta., Bull. 272, 145
N. J. Corona Dry			30.00	31.86	2.30	0.14		65.65	(1925). Conn. Agr. Expt. Sta., Bull. 242, 150 (1922).
Riches, Piver & Co., New York, N. Y.	50.00		15.00		0.50		31.00		Conn. Agr. Expt. Sta., Bull. 272, 145 (1925).
Riches, Piver & Co., New York, N. Y. R. P. & Co.		0.12	30.00	30.80	1.53	1.20	62.00	64.61	Cal. Dept. Agr., Spec. Pub. 51, 15 (1925).
Schiefelin & Co., New York, N. Y. Powder			30.00		1.00 1.00	$0.20 \\ 0.29$		63.50 63.50	N. J. Agr. Expt. Sta., Bull. 441, 6 (1926). Cal. Dept. Agr., Spec. Pub. 34, 15
The Sherwin-Williams Co., Cleveland,					1.00	0.23		00.00	(1922.) Conn. Agr. Expt. Sta., Bull. 242, 150
Ohio. Sherwin-Williams				31.46	1.00	0.23		64.48	(1922) and Bull. 272, 145 (1925). Cal. Dept. Agr., Spec. Pub. 75, 19
Ohio. Sherwin-Williams, Basic D. B. Smith & Co., Utica, N. Y. Lightening			30.00	21.95	$0.77 \\ 0.77$	0.13		73.38	(1927). Conn. Agr. Expt. Sta., Bull. 272, 145
Standard Chemical Works, Reading, Pa. Standard			31.00		0.46	0.18		64.47	(1925). N. J. Agr. Expt. Sta., Bull. 459, 6 (1927).
Toledo Rex Spray Co., Toledo, Ohio. Rex			15.00		1.15	0.73		34.20	N. J. Agr. Expt. Sta., Bull. 424, 8 (1925).
U. S. Smelting, Refining & Mining Co., Redding, Cal		0.20	30.00	32.95	0.75	0.36	63.00	63.45	Cal. Dept. Agr., Spec. Pub. 34, 15 (1922).
Falls, N. Y. Electro			30.00	30.24	0.77	0.18		64.40	Conn. Agr. Expt. Sta., Bull. 272, 145 (1925).

407, 13 (1924). 34, 16 (1923) 6 (1927). Bull. 459, Spec. Pub. Bull. Sta., Sta., Expt. Expt. Cal. Dept. Agr., Agr. Expt. Agr. Agr. ż ż ZINC 0.16 24 20 20 Water-Soluble Found 0 0 OF 0.75 Metal 1.00 Guaranteed ARSENITE 0 30.75 80 86 Lound Arse 30. 29. Total 50 30.00 30.31 00 Guaranteed 30. 29. TABLE Cross J. Dry Powdered Watsonville, Cal. Green City, ż ż California Spray Chemical Co., Ortho.... Vineland, Brand The Kil-Tone Co., Vineland, Manufacturer Lucas Kil-Tone Co.,

CONNECTICUT EXPERIMENT STATION

Barium Tetrasulphide.

223

B. T. S. Manufacturer not stated. Cal. Dept. Agr., Spec. Pub. 34,34 (1923).Orchard Brand B. T. S. General Chemical Co., San Francisco, Cal. Cal. Dept. Agr., Spec. Pub. 34, 34 (1923).
Solbar. The Bayer Co., New York, N. Y. Conn. Agr. Expt. Sta., Bull. 258, 371 (1924).

		Orchard	Brand	
Analyses:	B.T.S.	B.T.	S.	Solbar.
WATER-SOLUBLE.	Found.	Guaranteed.	Found.	Found.
Total	96.10		58.95	59.58
Sulphide sulphur	40.05		29.10	15.04
Thiosulphate	3.50		1.42	2.55
Total sulphur	43.80		30.52	17.59
Barium				43.42
Barium sulphide		38-44	57.38	
Barium thiosulphate		1-4	5.54	
WATER-INSOLUBLE.				
Total			40.51	
Silica $(S_1O_2)$			9.77	
Iron and Aluminum Oxide				
$(Fe_2O_3+Al_2O_3)$			1.98	
Ferrous sulphide (FeS)			0.15	
Barium sulphite (BaSO <sub>3</sub> )			1.01	
Barium sulphate (BaSO <sub>4</sub> )			7.60	
Free sulphur (S)			13.61	
Barium oxide (BaO)			6.39	
Carbon dioxide, carbon, etc			traces	
Total sulphur				6.13
Total barium				9.35

# Barnes Worm Emulsion.1

(S. O. BARNES & SON).

Guaranteed, Found, Guaranteed, Found, 50.00 94.00 94.00 93.60 Cal. Dept. Agr., Spec. Pub. 75, 64 (1927).

#### Bayer Dipdust.

Guaranteed: Hydroxymercurichlorphenol sulphate 6.00 per cent; hydroxymercurinitrophenol sulphate 2.00 per cent. Found: Mercury 5.17 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 62 (1927).

# Bean's Bug-Go; Crude.

See "Oil Emulsions, Mineral."

# Beck's Flea Powder.

(CHAS. H. BECK MFG. CO., LOS ANGELES, CAL.)

Guaranteed: Talc. 17.50 per cent. Found: Ash 10.43 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 63 (1927).

Beetle Mort.

See "Green Cross".

<sup>1</sup> Two grades.

#### Be-Health.

(GENERAL LABORATORIES, MADISON, WIS.)

Guaranteed: Active ingredients 8.50 per cent: inert matter 91.50 per

Found: (grams per 100 cc.) Available chlorine 3.77; total chlorine 3.81. sodium hypochlorite 3.96. Trace of sulphate, no calcium.—Conn. Agr Expt. Sta., Bull. 258, 376 (1924).

#### Bejo Orchard Spray.

See "Oil Emulsions, Mineral."

#### B-K Bacili-Kil.

(GENERAL LABORATORIES, MADISON, WIS.)

Guaranteed: Inert matter: calcium chloride 1.71 per cent; water 89.64 per cent.

Found: Available chlorine 3.20 per cent. Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).

#### Black Death.

(BLACK DEATH CO., BINGHAMTON, N. Y.)

Found: Carbon 17.11 per cent; sand 5.86 per cent; moisture 10.18 per cent; carbon dioxide 12.39 per cent; sulphur trioxide 23.15 per cent: arsenic trioxide 0.97 per cent; cupric oxide 0.59 per cent; ferric oxide 0.46 per cent; calcium oxide 24.97 per cent; magnesium oxide 5.03 per cent. U. S. D. A., Bur. Chem., Bull. 68, 28 (1902).

#### Black Leaf 40.

See "Nicotine Sulphate Solutions."

#### Black Leaf F-2 Nicotine Dust.

See "Nicotine Dusts."

#### Bleaching Water.

See "Wescco."

#### Bleach-It.

(GILT EDGE PACKING CO., SAN FRANCISCO, CAL.)

Guaranteed. Found. 4.00 4.99 Available chlorine..... Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).

#### Bliss's Bedbug Exterminator.

(FRANK BLISS, ST. LOUIS, MO.)

Found: Borax 20.00 per cent; ether extract 6.79 per cent. A mixture of borax and pyrethrum. - U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

#### Bliss's Cockroach Exterminator.

(FRANK BLISS, ST. LOUIS, CHICAGO AND CINCINNATI.)

Found: Borax 81.89 per cent. The balance is pyrethrum and pink coloring matter. — U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

#### Blue Dust.

See "Adheso" and "Sanders."

# Blue Label Tonicide.

#### (MORRIS HERRMANN & CO., NEW YORK N. V.)

	Guaranteed.	Found.
Water		57.25
Total arsenic, metal	3.70	5.48
Water-soluble arsenic, metal	0.25	0.15
Lead oxide		13.51
Copper oxide		5.65
N. J. Agr. Expt. Sta., Bull. 301, 13 (1916).		2.00

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

# Blue Ribbon Orchard Spray.

See "Oil Emulsions, Mineral."

#### Borco.

See "Bordeaux Mixture"

# Bordeaux Cal-Arsenate, Powdered.

# (RICHES, PIVER & CO., NEW YORK, N. Y.)

	Guaranteed.	Found.
Total arsenic, metal	15.00	16.07
Water-soluble arsenic, metal	0.50	0.18
Copper	5.00	4.99
N. J. Agr. Expt. Sta., Bull. 286, 13 (1915).		

#### Bordeaux Dust with Poison.

# (MECHLING BROS. CHEMICAL CO., CAMDEN, N. J.)

	Guaranteed.	Found.
Total arsenic, metal	6.00	7.46
water-soluble arsenic, metal	1.00	0.20
Copper	6.00	7.50
N. J. Agr. Expt. Sta., Bull. 459, 12 (1927).		

#### Bordeaux-Green Powder.

# (ANSBACHER INSECTICIDE CO., NEW YORK, N. Y.)

	Guaranteed.	Found.
Total arsenic, metal	11.00	13.82
water-soluble arsenic, metal	2.25	2.06
Copper	16.00	19.44
N. J. Agr. Expt. Sta., Bull. 286, 11 (1915).		

#### Bordeaux Mixture.

See Table IV.

# Bordeaux Mixture-Lead Arsenate.

Tee Table V.

#### Bordeaux-Paris Green.

#### (LIGGETT & BRO., NEW YORK, N. Y.) (See also "Naco-Bordeaux-Paris Green.")

	Guaranteed	Found.
Copper oxide, CuO	17.50	17.37
otal arsenious oxide. As <sub>2</sub> O <sub>2</sub>	16 50	17.69
water-soluble arsenious oxide, As <sub>2</sub> O <sub>3</sub>	2.64	2.72
Conn. Agr. Expt. Sta., Bull. 242, 153 (1922).		

Guaranteed	Found	Calcium Hydroxide— Found	Calcium Carbonate— Found	Inert Calcium Compounds— Guaranteed	Publication
%	%	%	%	%	
1.00 3.00	14.43 12.73				Cal. Dept. Agr., Spec. Pub. 75, 43 (1927). N. J. Agr. Expt. Sta., Bull. 459, 7 (1927). Cal. Dept. Agr., Spec. Pub. 58, 21 (1925). N. J. Agr. Expt. Sta., Bull. 441, 7 (1927). Cal. Dept. Agr., Spec. Pub. 75, 43 (1927).
5.00	26.06				Cal. Dept. Agr., Spec. Pub. 66, 20 (1926).
		91.91	4.30		Cal. Dept. Agr., Spec. Pub. 66, 20 (1926).
2.75	13.08				Cal. Dept. Agr., Spec. Pub. 66, 20 (1926).
2.75	13.58				Cal. Dept. Agr., Spec. Pub. 75, 43 (1927).
					Ore. Agr. Expt. Sta., Cir. 84, 12 (1927). Ore. Agr. Expt. Sta., Cir. 84, 12 (1927).
				75.00	Cal. Dept. Agr., Spec. Pub. 34, 22 (1922). Cal. Dept. Agr., Spec. Pub. 58, 21 (1925).
3.50	4.48			90.00	Cal. Dept. Agr., Spec. Pub. 34, 23 (1922).
2.54	2.84			90.00	Cal. Dept. Agr., Spec. Pub. 34, 23 (1922).
4.00	24.49				Cal. Dept. Agr., Spec. Pub. 58, 21 (1925).
4.50	24.64				Cal. Dept. Agr., Spec. Pub. 58, 21 (1925).
6137 E 2 28 EE 3 2 4	% 2.75 3.00 3.00 5.00 5.00 5.00 6.30 6.30 6.30 6.30 6.30	%     %       %     %       3.75     12.72       16.27     10.01       14.43     12.73       7.00     19.63       5.00     26.06       2.75     13.08       2.75     13.58       2.50     12.50       3.00     11.15       5.00     24.99       6.30     4.48       2.54     2.84       4.00     24.49	%     %       8.75     12.72       .00     16.27       .00     14.43       3.00     12.73       7.00     19.63       5.00     26.06        91.91       2.75     13.58       2.50     12.50       3.00     11.15       5.00     24.99       5.30     14.86       3.50     4.48       2.54     2.84       4.00     24.49	%         %         %         %           3.75         12.72             3.00         16.27             3.00         12.73             7.00         19.63             5.00         26.06             2.75         13.08             2.75         13.58             3.00         11.15             5.30         11.86             3.50         4.48             3.54         2.84             4.00         24.49	%     %     %     %       2.75     12.72         3.00     16.27         3.00     12.73         5.00     26.06         2.75     13.08         2.75     13.58         2.50     12.50         3.00     11.15         5.30     14.48      90.00       2.54     2.84      90.00       4.00     24.49

F. A. Frazier Co., Pt. Richmond, Cal. Frazier's Bordo "B".  General Chemical Co., New York, N. Y. Orchard. The Glidden Co., Cleveland, Ohio. Glidden The Glidden Co., Cleveland, Ohio. Glidden Dry	12.50 16.00	15.24	93.96	* * * * *	 Cal. Dept. Agr., Spec. Pub. 58, 21 (1925). Cal. Dept. Agr., Spec. Pub. 75, 43 (1927). Cal. Dept. Agr., Spec. Pub. 58, 22 (1925).
Powdered	22.00 13.00				 Cal. Dept. Agr., Spec. Pub. 58, 22 (1925). Conn. Agr. Expt. Sta., Bull. 242, 153 (1922).
Hammond's Paint & Slug Shot Works, Beacon, N. Y. Pulp <sup>1</sup>	24.00 12.50	4.15 24.60 19.80 14.58			 <ul> <li>N. J. Agr. Expt. Sta., Bull. 441, 7 (1926).</li> <li>Ore. Agr. Expt. Sta., Cir. 84, 12 (1927).</li> <li>Cal. Dept. Agr., Spec. Pub. 34, 23 (1922).</li> <li>N. J. Agr. Expt. Sta., Bull. 459, 7 (1927).</li> <li>N. J. Agr. Expt. Sta., Bull. 407, 9 (1924).</li> </ul>
The Kil-Tone Co., Vineland, N. J. Modified Kil-Tone, Dry Form. Kirk, Geary & Co., Sacramento, Cal. Paste <sup>4</sup> . Leggett & Bro., New York, N. Y. Anchor <sup>5</sup>	22.50				 N. J. Agr. Expt. Sta., Bull. 407, 9 (1924). Cal. Dept. Agr., Spec. Pub. 51, 19 (1925). Conn. Agr. Expt. Sta., Bull. 242, 153 (1922).
John Lucas & Co., Inc., Philadelphia, Pa. Bordo Mixture Dry Powder 25%	25.00	27.14			 N. J. Agr. Expt. Sta., Bull. 407, 9 (1924).
John Lucas & Co., Inc., Philadelphia, Pa. Lucas Dry Powdered <sup>4</sup> Lucas-Kil-Tone Co., Vineland, N. J. Green Cross	16.00 21.50				 Cal. Dept. Agr., Spec. Pub. 75, 43 (1927). N. J. Agr. Expt. Sta., Bull. 459, 7 (1927).
Miller Products Co., Portland, Ore. <sup>2</sup>		12.73 17.80 25.20			 N. J. Agr. Expt. Sta., Bull. 459, 7 (1927). N. J. Agr. Expt. Sta., Bull. 459, 7 (1927). N. J. Agr. Expt. Sta., Bull. 441, 7 (1926). Lre. Agr. Expt. Sta., Cir. 84, 12 (1927). Ore. Agr. Expt. Sta., Cir. 84, 12 (1927).
National Chemical Co., Pittsburg, Pa. National Bordo "A"	24.00	24.90			 Cal. Dept. Agr., Spec. Pub. 75, 43 (1927).

<sup>&</sup>lt;sup>1</sup> 68.64% water.

<sup>2</sup> A two powder preparation. One package contains partially dehydrated copper sulphate, the other hydrated lime. Copper is reported as found in the copper sulphate package.

<sup>3</sup> 50.71% water.

<sup>4</sup> Decomposed.

<sup>5</sup> 58.03% water. 168.64% water.

# TABLE IV. BORDEAUX MIXTURE—Concluded.

	Col	Copper				
Manufacturer or Distributor and Brand	beetnarand	Found	Calcium Hydroxide— Found	Calcium Carbonate— Found	Inert Calcium Compounds— Guaranteed	Publication
Chemical Co Dittelyung Do Mational	%	%	%	%	%	
: 7			93.706			Cal. Dept. Agr., Spec. Pub. 75, 43 (1927).
	12.75	10.79				Cal. Dept. Agr., Spec. Pub. 58, 22 (1925)
To look	24.00	24.35				Cal. Dept. Agr., Spec. Pub. 58, 22 (1925).
Home-mixed Bordeaux Package No. 1	24.00	24.26				Cal. Dept. Agr., Spec. Pub. 58, 22 (1925).
	13.00	14.14	93.91			Cal. Dept. Agr., Spec. Pub. 58, 22 (1925) N. J. Agr. Expt. Sta., Bull. 459, 7 (1927)
	18.00 15.00 12.75	20.34 17.50 11.51				Cal. Dept. Agr., Spec. Pub. 75, 43 (1927) N. J. Agr. Expt. Sta., Bull. 407, 9 (1924) Cal. Dept. Agr., Spec. Pub. 58, 22 (1925)
in Jose Spray Mig. Co., San Jose, Cal. 1 Win Bordeaux, Package 1	24.00	25.09				Cal. Dept. Agr., Spec. Pub. 66, 20 (1926)
Dardeaux, Package 2.  Shorterin Williams Co. Clareland Object Brand				83.32	12.70	Cal. Dept. Agr., Spec. Pub. 66, 20 (1926)
Cleveland, Omo.	12.75	12.90				Cal. Dept. Agr., Spec. Pub. 75, 43 (1927)
way Dry. Sterling Chemical Co., Cambridge, Mass. Sterling-	12.75	13.13	:	:		Cal. Dept. Agr., Spec. Pub. 66, 20 (1926).
worth Vreeland Chemical Mfg Co 14ttle Balls N T	10.00	11.96				(1922).
100	18.00 19.78	19.78	:	:		N. J. Agr. Expt. Sta., Bull. 424, 9 (1925)

#### Bordeaux Zinc.

(THOMSEN CHEMICAL CO., BALTIM	ORE,	MD.
-------------------------------	------	-----

	Guaranteed.	Found.
Water Total arsenic, metal	7.63	53.02 7.53
Water-soluble arsenic, metal	0.57	0.08
Copper	4.50	4.57
N. J. Agr. Expt. Sta., Bull. 286, 13 (1915).		

#### Bordeth.

#### (KILTONE CO., NEWARK, N. J.)

	Guaranteed.	Found.
Total arsenic oxide	20.00	18.01
Water-soluble arsenic oxide	1.00	0.28
Copper hydroxide	15.00	15.44
N. J. Agr. Expt. Sta., Bull. 273, 12 (1914).		

#### Borecide.

#### (MASON DRUG & CHEMICAL CO., HANCOCK, MD).

Found: Ash 58.14 per cent; loss on ignition 41.86 per cent; insoluble in boiling water 65.35 per cent; melting-point of sublimate, 31.0° C. The substance had the odor of naphthalene, and the sublimate combined with picric acid, but the melting-point was nearer that of methyl naphthalene.— Conn. Agr. Expt. Sta., Bull. 242, 160 (1924).

#### Boxal.

#### (BOWKER INSECTICIDE CO., BOSTON, MASS.)

Found: Copper 12.08 per cent; arsenious oxide 13.52 per cent.— N. Y. Agr. Expt. Sta., Bull. 384, 289 (1914).

#### Brazilian Insecticide Powder.

#### (I. MAJORA CO. ALA MARA, BRAZIL.)

Found: Borax 8.84 per cent; sodium carbonate 37.52 per cent; sulphur 13.57 per cent. Balance pyrethrum and ultramarine. U.S. D. A., Bur. Chem., Bull. 68, 43 (1902).

#### Brininstool's Coal Tar Creosote.

#### (THE BRININSTOOL CO.)

Found: Oils 94.80 per cent, phenols 3.90 per cent, trace of water and ash, no rosin. Baumé gravity of separated oil, 17.3; per cent unsulphonated oil, 24.60—Cal. Dept. Agr., Spec. Pub. 51, 45 (1925).

#### Brown's Pink Powder Insecticide.

#### (BROWN'S INSECTICIDE CO., SYRACUSE, N. Y.)

Found: Borax 88.55 per cent. Balance cloves and pink coloring matter.— U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

#### Brunswig Squirrel Annihilator.

See "Strychnine Preparations".

#### Bug and Blight Dust.

#### (LEGGETT & BRO., NEW YORK, N. Y.)

	Guaranteed.	Found.
Total arsenic, metal	16.00	16.21
Water-soluble arsenic, metal	3.00	2.25
Copper	6.00	6.44
N. J. Agr. Expt. Sta., Bull. 273, 12 (1914).		

TABLE V. BORDEAUX MIXTURE—LEAD ARSENATE

	Arsenic Oxide, As <sub>2</sub> O <sub>5</sub>					pper,	Lead Oxide, (PbO)		Water		
Manufacturer or Distributor and Brand	Total		Water-Soluble		(Cu)				11400		
	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	Found	Publication	
1 10 C1 W 1- Detail	%	%	%	%	%	%	%	%	%	Cal. Dept. Agr., Spec.	
Acme White Lead & Color Works, Detroit, Mich. Acme 2-way Spray	4.29	5.11	0.46	0.31	11.00	11.25		9.28		Pub. 75, 44 (1927).	
Bowker Chemical Co., New York, N. Y. Bowker's Dry Powdered Bordeaux Arsenate	7.96	8.11	0.77	0.41	15.00	14.62			2,	N. J. Agr. Expt. Sta., Bull. 407, 12 (1924). Conn. Agr. Expt. Sta.,	
Bowker Chemical Co., New York, N. Y. Pyrox	5.00	5.57	0.46	0.06		6.70		1.28	65.95	Bull. 272, 146 (1925).	
California Sprayer Co. Calispray Dust	13.46	13.92	1.15	0.44	3.80	4.45		27.67		Cal. Dept. Agr., Spec. Pub. 75, 44 (1927).	
Chipman Chemical & Engineering Co., New York, N. Y. Bordo-Lead.	7.29	8.73	0.38	0.18		16.47		15.24		Conn. Agr. Expt. Sta., Bull. 272, 146 (1925).	
General Chemical Co., New York, N. Y. Orchard Bordeaux Lead	9.37	10.25	1.50	0.06	15.00	15.95				N. J. Agr. Expt. Sta., Bull. 459, 10 (1927).	
The Glidden Co., Cleveland, Ohio, Glidden Bordo-Arsenate	15.50	17.54	0.50	0.20		9.32		31.71		Conn. Agr. Expt. Sta., Bull. 242, 154 (1922).	
The Glidden Co., Cleveland, Ohio. Glidden Bordo-Arsenate	5.90	14.77	0.50	0.18		13.31		26.64	****	Conn. Agr. Expt. Sta., Bull. 272, 146 (1925).	
Interstate Chemical Co., Jersey City, N. J. Key Bordo-Lead.	4.45	6.65	0.77	0.60		6.05		13.76	60.85	Conn. Agr. Expt. Sta., Bull. 242, 153 (1922).	
Interstate Chemical Co., Jersey City, N. J. Key Bordo-Lead	7.67	7.82	0.77	0.09		2.99			60.83	Conn. Agr. Expt. Sta., Bull. 242, 153 (1922).	
Interstate Chemical Co., Jersey City, N. J. Key-Cide	4.22	5.75		0.18		10.06		9.90		Conn. Agr. Expt. Sta., Bull. 242, 158 (1922).	

			Ī.			10.054				
Riches, Piver & Co., New York, N. Y. Ceres Bordeaux-Lead Powdered	5 20	8 02	0 77	0.86	11 50	11.65				N. J. Agr. Expt. Sta., Bull. 424, 16 (1925).
Sherwin-Williams Co., Cleveland, Ohio.								(		Cal. Dept. Agr., Spec.
Pestroy Skinner Machinery Co., Dunedin, Fla.	4.28	3.52	0.46	0.32	11.00	11.02		6.61	• • • • •	Pub. 75, 44 (1927). N. J. Agr. Expt. Sta.,
Skinner's Arsenate Bordeaux	26.01	28.25		0.28	4.00	2.64				Bull. 407, 16 (1924).
H. J. Smith Co., Utica, N. Y. Hexpo	6.90	8.37	0.50	0.28		18.07		18.89		Conn. Agr. Expt. Sta., Bull. 272, 146 (1925).
H. J. Smith Co., Utica, N. Y. Hexpo	6.88	7.33	0.50	0.57	15.40	19.27	,	14.33		Conn. Agr. Expt. Sta.,
Vreeland Chemical Mfg. Co., New York,										Bull. 242, 158 (1922) Conn. Agr. Expt. Sta.,
N. Y. Bordo-Lead Mixture	5.58	8.91	0.78	0.11		2.57		18.32	44.23	Bull. 272, 146 (1925)
				1	1 3 3 3 6 3			1		

#### Bug Death.

#### (DANFORTH CHEMICAL CO., LEOMINSTER, MASS.)

Zinc oxide	Guaranteed. 47.00	Found. 54.15
Lead oxide	5.00	14.11

#### Bug Death Alpha.

# (DANFORTH CHEMICAL CO., LEOMINSTER, MASS.)

Nicotine Sulphur. Zinc oxide. Lead oxide.	0.05 20.00 15.00 1.00	Found. trace 18.20 16.50 4.80
Ore. Agr. Expt. Sta., Cir. 84, 11 (1927).		

#### Bug Death Aphis.

#### (DANFORTH CHEMICAL CO., LEOMINSTER, MASS.)

	Guaranteed.	Found.
Nicotine	0.05	present
Sulphur	20.00	19.29
Zinc oxide	15.00	23.63
Lead oxide	1.00	5.69
Maine Agr. Expt. Sta., Official Inspections 110,	55 (1923).	

#### Bug Dope.

#### (H. J. SMITH & CO., UTICA, N. Y.)

	Guaranteed.	Found.
Total arsenic, metal	0.37	0.62
Water-soluble arsenic, metal	2.70	0.23
Copper		0.30
N. J. Agr. Expt. Sta., Bull. 286, 13 (1915).		

#### Buhack Insect Powder.

See "Pyrethrum".

#### Buker's Sheep Dip.

See "Phenol Soap Solutions".

#### By-Sul.

(GEO. H. CORSE, JR. & CO., SAN FRANCISCO, CAL.) Two samples:

#### Sample 1.

Guaranteed: Carbon disulphide 41.00 per cent; inert matter 52.00 per cent

Found: Carbon disulphide 37.98 per cent; water 50.82 per cent; calcium carbonate 4.06 per cent; calcium polysulphide 6.06 per cent.

#### Sample 2.

Guaranteed: Carbon disulphide 41.48 per cent; inert matter 52.59 per cent

Found: Carbon disulphide 46.20 per cent; water 46.00 per cent; clay and oil 6.18 per cent; calcium polysulphide none.

Cal. Dept. Agr., Spec. Pub. 51, 60 (1925).

#### Caascu.

#### (HEMINGWAY & CO., INC., BOUND BROOK, N. J.)

	Guaranteed.	Found.
Total arsenic, metal	16.00	18.23
Water-soluble arsenic, metal	4.00	0.12
Copper		7.73
N T Agr Expt. Sta. Bull. 459, 11 (1927).		

#### Caasen.

# (HEMINGWAY LONDON PURPLE CO., LONDON & NEW YORK). Found: Arsenious oxide 34.13 per cent; copper 8.10 per cent.

# N. Y. Agr. Expt. Sta., Bull. 384, 301 (1914).

Cabot's Gypsy Moth Creosote.
(SAMUEL CABOT, INC., BOSTON, MASS.)

Guaranteed: Not more than 10 per cent inert matter.

Found: Substance is composed of tarry matter like gas-house wastes. There are no oils present.

Maine Agr. Expt. Sta., Official Inspections 114, 88 (1924).

#### Calcium Arsenate.

See "Arsenate of Calcium".

#### Calcium Caseinate.

See Table VI.

#### Calcium Cyanide.

See Table VII.

#### Calcium Fluosilicate Compound.

(VICTOR CHEMICAL WORKS, NEW YORK, N. Y.)

Guaranteed: Calcium fluosilicate not less than 15 per cent; inert matter not over 85 per cent.

Found: Phosphorus pentoxide 28.37 per cent; calcium oxide 21.80 per cent; iron and aluminum oxides 20.60 per cent; silica 11.00 per cent; fluorine 11.95 per cent.

Conn. Agr. Expt. Sta., Bull. 272, 149 (1925).

#### Calco-Green.

(Manufacturer not stated).

Found: Total arsenious oxide 30.0 per cent; water-soluble arsenious oxide 7.0 per cent.

Univ. of Cal., Coll. of Agr., Expt. Sta., Bull. 151, 24 (1903).

#### Calcyco Braun Gas.

#### Calcyco Hydrocyanic Acid.

See "Hydrocyanic Acid".

#### California C & G Brand Bleaching Water.

(CALIFORNIA BLEACHING WATER CO., SAN FRANCISCO, CAL.)

A- ::	Guaranteed.	Found.
Available chlorine	4.00	5.17
Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).		

TABLE VI—CALCIUM CASEINATE

Manufacturer or Distributor and Brand	Nitrogen	Casein N x 6.38	Calcium Oxide	Publication
	%	%	%	
California Central Creameries, San Francisco, Cal. Kayso	3.28	20.93	46.29	Conn. Agr. Expt. Sta., Bull. 258, 369 (1924); 272, 149 (1925).
Casein Mfg. Co., New York, N. Y. A-7-ML	7.92	50.53	27.27	Conn. Agr. Expt. Sta., Bull. 258, 369 (1924).
Rosin & Co., Philadelphia, Pa. Red Diamond	3.31	21.12	50.64	Conn. Agr. Expt. Sta., Bull. 258, 369 (1924).

TRBLE VII. CALCIUM CYANIDE

TABLE VII. CALCIUM CYANIDE										
		nogen,			Calcium Cyanide, Ca (CN) <sub>2</sub>					
Manufacturer or Distributor and Brand	Guaranteed	Found	Calcium, Ca.	Chlorine, Cl.	Guaranteed	Found	Publication			
American Cyanamid Co., New York, N. Y. G Grade	% 40.00	% 40.30	%	%	%	% 71.35	N. J. Agr. Expt. Sta., Bull. 424, 12 (1925).			
American Cyanamid Co., New York, N. Y. Cyanogas					40.00	43.17	Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).			
California Cyanide Co., Southgate, Cal. Calcanide Fumigant Grade No. 1					88.00	91.73	Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).			
California Cyanide Co., Southgate, Cal. Cal-Sy					25.00	27.73	Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).			
California Cyanide Co., Southgate, Cal. Citrofume Citrus Fumigant					53.00	54.30	Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).			
L. E. Neville, Los Angeles, Cal. Aero Rodent Exterminator	20.00	23.71	32.77	43.68		41.98	Cal. Dept. Agr., Spec. Pub. 34, 57 (1923).			

# California Spray.

(UNITED STATES SPRAY CO., SAN BERNARDINO, CAL.)

Guaranteed: Inert matter 65.00 per cent.

Found: Oils 29.00 per cent; inert matter 69.00 per cent. Sodium fluoride and rosin present.

Cal. Dept. Agr., Spec. Pub. 34, 37 (1923).

# Calispray Combined Insecticide and Fungicide Dust No. 3.

See "Nicotine-Sulphur Dusts".

# Calispray Complete Garden Dust No. 83.

(CALIFORNIA SPRAYER CO.) Guaranteed. Foun	id.
1.25	5
Nicotine 16	9
Total arsenic metal	26
Water-soluble arsenic, metal	7
Lead oxide	15
Sulphur	
Cal. Dept. Agr., Spec. Pub. 75, 38 (1927).	

# Calispray Dust No. 1.

#### Calispray Dust No. 2.

See "Nicotine-Sulphur Dusts".

# Calispray Dust No. 12.

See "Nicotine Dusts".

# Calispray Dust No. 86.

See "Bordeaux Mixture-Lead Arsenate".

# Calispray Insecticide Dust No. 11.

# Calispray Insecticide Dust No. 15.

See "Nicotine Dusts".

# Calispray Insecticide Dust No. 81.1

(CALIFORNIA	SPRAYER Guarantee	CO.) d. Found.	Guaranteed.	Found.
Nicotine Total arsenic, metal Water-soluble arsenic, metal Lead oxide	2.00 3.50 0.50	2.21 $3.39$ $0.18$ $13.87$	2.00 3.80 0.50	2.28 4.06 0.29 13.02
Cal. Dept. Agr., Spec. Pub. 75, 38	8 (1927).	F-1-7-	,	

# Calispray Nico-Dust No. 82

(CALIFORNIA SPRAYER CO	Guaranteed.	Found.
Nicotine	$\frac{2.50}{3.50}$	$\frac{1.22}{4.30}$
Total arsenic, metal	0.50	$0.46 \\ 12.76$
Lead oxide	5.00	4.85
Cal. Dept. Agr., Spec. Pub. 75, 38 (1927).		

<sup>1</sup> Two samples.

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

#### Calite.

(MORRIS HERRMANN & CO., NEW	YORK, N. Y.)	
	Guaranteed.	Found.
Total arsenic, metal		$   \begin{array}{c}     11.55 \\     0.45 \\     \hline     \end{array} $
Water		67.07
Mich. Agr. Coll. Expt. Sta., Spec. Bull. 74, 10	0 (1915).	

#### Calox.

See "Oil Emulsions, Mineral".

#### Calpest Garden Dust.

See "Nicotine Dusts".

#### Calpest Sow Bug Killer.

(CALIFORNIA PEST CONTROL CO., BURLINGAME, CAL.) Guaranteed: Copper aceto-arsenite, 8.00 per cent. Found: Arsenious oxide, 11.04 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 26 (1927).

#### Calpest Summer Spray.

See "Oil Emulsions, Mineral".

#### Calpest Weed Killer.

(CALIFORNIA PEST CONTROL CO., BURLINGAME, CAL.)

	Guaranteed.	For	ınd.
Arsenious oxide	30.00	31.	30
Cal Debt Agr Spec Pub 75 25 (1927)			

#### Calpest Whale Oil Soap.

See "Soaps".

#### Calpro Ant Syrup.

(THE CALPRO SALES CO., LOS ANGELS CAL.)

	Guaranteed.	Found.
Arsenic, metal	0.10	0.22
Cal. Dept. Agr., Spec. Pub. 75, 24 (1927).		

#### Cal-Sy.

See "Calcium Cvanide".

#### Cann's Ant Exterminator.

(ARTHUR CANN CO., SAN JOSE, CAL.

Guaranteed: Inert matter 10.00 per cent.

Found: Phenols 12.20 per cent, inert matter 9.60 per cent. Rosin

Cal. Dept. Agr., Spec. Pub. 34, 36 (1923).

#### Cann's Canco.

See "Phenol Soap Solutions".

#### Cann's Carco.

(ARTHUR CANN CO., SAN JOSE, CAL.) Found: Phenols 7.30 per cent. Rosin present.

Cal. Dept. Agr., Spec. Pub. 34, 36 (1923).

# Capital Sheep Dip and Cattle Wash.

See "Phenol Soap Solutions".

# Capital Squirrel Poison.

See "Strychnine Preparations".

#### Carboleine, Pratt's.

See "Oil Emulsions, Mineral".

Carbolic Acid.

See "Phenol".

Carbolicide.

See "Hockwald".

#### Carbolineum.

See "Avenarius Carbolineum".

#### Carbon Disulphide.

See Table VIII.

# Carbon Disulphide Emulsion.

(I. P. THOMAS & SON CO., PHILADELPHIA, PA.) Found: Carbon Disulphide 68.62 per cent. Conn. Agr. Expt. Sta., Sample 3893.

#### Carbo-White Disinfectant.

(R. L. STEVENS BROKERAGE CO., SAN FRANCISCO, CAL.) Found: Phenols 1.40 per cent; organic matter, partly glue and water, 6.50 per cent; mineral matter, mostly calcium carbonate, 92.10 per cent. Cal. Dept. Agr., Spec. Pub. 51, 58 (1925).

(SUNSET SALES CO., TACOMA, WASH.)

#### Carco.

(See also "Cann's Carco".)

Claimed phenols, soap and hydrocarbons present. Ore. Agr. Expt. Sta., Cir. 64, 15 (1925).

#### Carco Natholeum Dip.

See "Phenol Soap Solutions".

#### Caustic Soda Arctic Whale Oil Soap.

See "Soaps".

#### Cedar Blocks.

See "Paradichlorbenzene".

# "Cee-Pee-Dee" Carbolized Petroleum Distillate.

See "Oil Emulsions, Mineral".

# Cenol Ant Destroyer.

(CENTRAL CITY CHEMICAL CO., SAN F	RANCISCO, CA	L.)
	Guaranteed.	Found
Sodium fluoride	25.00	30.82 1.00
Ash		1.00
Cal Debt Agr Spec Pub 75 63 (1927).		

	Carbon Disulphide, CS <sub>2</sub>	bon shide,		əbide	
Manufacturer or Distributor	Cuaranteed	Found	Residue on Evaporation	Hydrogen Sult	Publication
Brunswig Drug Co	99.00 99.56		0.02	% none	Cal. Dept. Agr., Spec. Pub. 51, 47 (1925).
Sun Drug Co., Pasadena, Cal	97.00	97.00 98.50	0.03	none	Cal. Dept. Agr., Spec. Pub. 51, 47 (1925).
Herbert F. Dugan, San Francisco, Cal. Kilmol Squirrel-gophene <sup>1</sup>	99.22 99.23	99.23			Cal. Dept. Agr., Spec. Pub. 66, 36 (1926).
Towne-Allison Drug Co., San Bernardino, Cal.		99.90	$0.09^{2}$	none	Cal. Dept. Agr., Spec. Pub. 51, 47 (1925).
Western Wholesale Drug Co., Los Angeles, 97.00 99.94	97.00	99.94	0.033	none	Cal. Dept. Agr., Spec. Pub. 51, 47 (1925).
Wheeler, Reynolds & Stauffer, San Francisco, Cal.	76.98 89.97	76.66		none	Cal. Dept. Agr., Spec. Pub. 51, 47 (1925).
Wheeler, Reynolds & Stauffer, San Francisco, Cal.	97.00	97.18	2.25	present	97.00 97.18 2.25 present Cal. Dept. Agr., Spec. Pub. 51, 47 (1925).

in oil, guaranteed 0.70 <sup>1</sup> Free sulphur guaranteed 0.08 per cent; found 0.10 cent; found 0.67 per cent.

<sup>2</sup> Chiefly ferric oxide.

<sup>3</sup> Guaranteed not more than 3.00 per cent.

34.30

240	CONNECTICUT EXPERIMENT STATION	BULL	ETIN 30
Arsenic	Cenol Argentine Ant Poison (CENTRAL CITY CHEMICAL CO., SAN FF, metal	RANCISCO, CAI Guaranteed. 0.11	Found. 0.17
	Cespi Poisoned Barley.		
See '	Cespi Poisoned Wheat. 'Strychnine Preparations''.		
	Cespi Rat Poison. (SCOTT & GILBERT CO., SAN FRANCISC	O, CAL.) Guaranteed.	Found
Barium Cal.	n carbonate	95.00	98.06

# Chemco Insect & Germ Destroyer.

See "Oils, Mineral".

#### Chlorafectant.

(AMERICAN OIL CO., NEW YOR	RK.	
Sodium hypochlorite	Guaranteed. 4.0	Found. 3.55
Conn. Agr. Expt. Sta., Sample 516.		
Chloride of Lime.		
(GREAT WESTERN ELECTRO CHEMI Available chlorine	CAL CO.) 30.00	35.26
(WESTERN WHOLESALE DRUG COM	MPANY) Guaranteed.	Found.

	Guaranțe
Available chlorine	
Cal. Dept. Agr., Spec. Pub. 51, 53 (1925).	

Chloro Naptholeum. (WEST DISINFECTING CO., NEW YORK, N.Y.)

A creosote oil and napthalene soap emulsion. U. S. D. A., Bur. Chem., Bull. 68, 58 (1902).

#### Citro-Mulsion.

See "Oil Emulsions, Mineral".

#### Citrospray.

(CITROSPRAY CHEMICAL CO. LTD., LOS ANG	ELES, CAL.)	
(CIIROSIRMI GAZZE	Guaranteed.	Found.
Total arsenic, metal	4.50	4.31
Total arsenic, metal	1.00	1.48
Water-soluble arsenic, metal	12.50	12.89
Lead oxide	20.00	19.17
Sulphur	13.00	
Sodium oxide		27.56
Sodium carbonate	10.50	8.20
Sodium oleate	16.50	7.16
Sodium phosphate	10.00	THE RESERVE OF THE CORP.
Water		19.58
Inert matter	23,50	
Inert matter		
Cal. Dept. Agr., Spec. Pub. 58, 19 (1925).		

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC. Citrus Washing Powder.

See "Soaps".

#### Clay.

(COLLODITE MFG. CO., LOS ANGELES, CAL.)

The clay used in the manufacture of "Collodite" sprays. Found: Water 9.60 per cent; silica 60.90 per cent; aluminum oxide 19.77 per cent; ferric oxide 1.15 per cent; calcium oxide 0.22 per cent; magnesium oxide 0.57 per cent; sulphur trioxide 0.22 per cent; sodium and potassium oxides 0.77 per cent; water of constitution 6.80 per cent. Cal. Dept. Agr., Spec. Pub. 34, 61 (1923).

#### Clensel.

See "Soaps".

#### Clorox Liquid Cleaning Compound.

(CLOROX CHEMICAL CORPORATION, OAKLAND, CAL.)

	Guaranteed.	Found.
Available chlorine	4.50	5.50
Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).		

#### Clumina.

(L. S. B. A. I., ROME, ITALY).

Found: Free chlorine 0.004 per cent; total chlorine 1.08 per cent. Cal. Dept. Agr., Spec. Pub. 66, 35 (1926).

#### C. N.

See "Coro Noleum".

#### Coal Tar Insecticide Oil.

(SCOTT & GILBERT CO., SAN FRANCISCO, CAL.)

	Guaranteed.	Found.
Phenols	40.00	39.00
Coal-tar oils	55.00	59.30
Water	2.00	1.60
Cal. Debt. Agr., Spec. Pub. 75, 56 (1927)		

# Collodite Arsenical Spray.

(COLLODITE SPRAY MFG. CO., LOS ANGELES, CAL.)

Found: Total arsenic oxide 8.87 per cent; water-soluble arsenic oxide 1.89 per cent. Contains calcium arsenate.

# Cal. Dept. Agr., Spec. Pub. 34, 61 (1923).

#### Collodite Copper Spray.

(COLLODITE MFG. CO., LOS ANGELES, CAL.)

Found: Copper 5.26 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 62 (1923).

# Collodite Nicotine Spray.

(COLLODITE MFG. CO., LOS ANGELES, CAL.)

Found: Nicotine 2.93 per cent; sodium carbonate 37.10 per cent. Cal. Dept. Agr., Spec. Pub. 34, 62 (1923).

# Collodite Rosin Paste.

(COLLODITE MEG. CO., LOS ANGELES, CAL.)

Guaranteed. Found. 36.00

29.40

Water Cal. Dept. Agr., Spec. Pub. 58, 46 (1925).

242

#### Collodite Scale Spray.

(COLLODITE MFG. CO., LOS ANGELES, CAL.)

Found: Sodium carbonate 28.19 per cent: soap 4.50 per cent; sulphur 2.16 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 61 (1923).

#### Collodite Sulfur Spray.

(COLLODITE MFG. CO., LOS ANGELES, CAL.)

Found: Sulphur 25.44 per cent; soda ash large. Cal. Dept. Agr., Spec. Pub. 34, 26 (1923).

#### Columbian Insecticide.

(COLUMBIAN INSECTICIDE CO., BOSTON, MASS.)

Found: Borax 94.74 per cent: sand 2.52 per cent: cloves 3.28 per cent. Blue coloring matter present.

U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

#### Common Sense Rat Exterminator.

See "Phosphorus Preparations".

#### Concentrated Adheso Yellow Label.

	(ANSBACHER	INSECTICIDE	CO., INC., N	EW YORK, N.Y.)	
				Guaranteed.	Found
Copper				10.38	13.84
Water					52.02

N. J. Agr. Expt. Sta., Bull. 441, 10 (1926).

#### Condensed Rosin Spray.

See "Soap".

#### Conkey's Noxicide.

See "Phenol Soap Solutions".

#### Conkey's Poultry Worm Remedy.

(THE G. E. CONKEY CO., CLEVELAND, OHIO.)

Guaranteed: Pelletierine 0.200 per cent; arecoline 0.020 per cent; nicotine 0.100 per cent; oil of chenopodium 0.120 per cent.

Found: Nicotine 0.22 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 65 (1927).

# Cooper's Sheep Dipping Powder.

(S. COOPER & NEPHEWS,	CHICAGO, ILL.)
	Guaranteed. Found.
Total arsenic, metal	16.75 14.00
Water-soluble arsenic, metal	17.48 14.18
Water Solder Eacht Sta Official Inspect	ions 114, 88 (1924).

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

Copodust.

(NIAGARA SPRAYER CO., MIDDLE)	PORT, N. Y.)	
	Guaranteed.	Found.
Copper sulphate	14.0	12.0

Ore. Agr. Expt. Sta., Cir. 64, 11 (1925).

#### Conotex.

See "Niagara Copotox".

#### Copper-Calcium Arsenate Dust. 13-8-79.

(DORCH CHEMICAL CO LOUISVILLE KY)

(	,/	
	Guaranteed.	Found.
Total arsenic oxide	3.00	1.39
Water-soluble arsenic, metal	0.50	
Copper	4.40	5.20
Conn. Agr. Expt. Sta., Bull. 242, 152 (1925).		

Copper Carbonate.

See Table IX.

Copper Sulphate.

See Table X.

#### Coreco Argentine Ant Poison.

(COFFIN-REDINGTON CO., SAN FRANCISCO, CAL.)

Guaranteed. Found. Arsenic, metal.... 0.15 0.21 Cal. Dept. Agr., Spec. Pub. 75, 24 (1927).

#### Coro Natholeum Din.

(WEST DISINFECTING CO., SAN FRANCISCO, CAL.)

Found: Phenols 9.00 per cent: water 11.20 per cent. Cal. Dept. Agr., Spec. Pub. 51, 45 (1925).

#### Coro Noleum and C. N. Disinfectant.

(WEST DISINFECTING CO., SAN FRANCISCO, CAL.)

Found: Phenols 28.00 per cent; water 8.80 per cent. Cal. Dept. Agr., Spec. Pub. 51, 44 (1925).

Cot Oil.

See "Oil Emulsions, Mineral".

Coulson's Poultry Spray.

See "Oil Emulsions, Mineral".

#### County Farm Bureau Mixture.

#### County Squirrel Poison (Barley).

See "Strcyhnine Preparations".

C. P. Lice and Mite Liquid.

See "Lime-Sulphur".

# C. P. Lice and Mite Tablets.

(CHICKEN PHARMACY, PETALUMA, CAL.)

Found: Calcium sulphide 16.13 per cent; silica 7.38 per cent; iron and aluminum oxides 0.14 per cent; gypsum 6.47 per cent; sugar 57.80 per cent; starch 11.63 per cent.

Cal. Dept. Agr., Spec. Pub. 51, 11 (1925).

TABLE IX. COPPER CARBONATE

	Copper	, Cu.	radius!
Manufacturer or Distributor and Brand	Guaranteed	Found	Publication
The single of the second second	%	%	O A Frint Sto Cir 84
Braun-Knecht-Heimann Co., San Francisco, Cal	50.00	52.10	Ore. Agr. Expt. Sta., Cir. 84, 12 (1927).
California Spray Chemical Co., Watsonville, Cal. Ortho	51.92	52.40	Cal. Dept. Agr., Spec. Pub. 51, 19 (1925). Ore. Agr. Expt. Sta., Cir. 84,
Miller Products Co., Portland, Ore.	53.00	55.50	12 (1927).
Montgomery, Ward & Co., Portland, Ore	54.00	54.00	Ore. Agr. Expt. Sta., Cir. 64, 12 (1925).
The Mountain Copper Co., San Francisco, Cal	54.00	54.10	Ore. Agr. Expt. Sta., Cir. 84, 12 (1927). Ore. Agr. Expt. Sta., Cir. 84,
Nichols Copper Co., New York, N.Y.	53.00	54.00	12 (1927).
Pittsburgh Plate Glass Co., Newark, N. J. Corona Copper-carb Dry.	18.00	20.09	Cal. Dept. Agr., Spec. Pub. 75, 43 (1927).
Roessler & Hasslacher Chemical Co., New York, N. Y	2000	49.30	Ore. Agr. Expt. Sta., Cir. 64, 12 (1925).
San Jose Spray Mfg. Co., San Jose Cal	,	53.301	
Sherwin-Williams Co., Emeryville Cal	50.00	51.30	Ore. Agr. Expt. Sta., Cir. 64, 12 (1925).
Wheeler, Reynolds & Stauffer, San Francisco, Cal. Stauffer's	50.00	50.13	Cal. Dept. Agr., Spec. Pub. 58 22 (1925).

<sup>199.90</sup> per cent passes a 200 mesh sieve.

	Copper, Cu.	, Cu.	Co1 Sulp	Copper Sulphate	ri, Jo
Manufacturer or Distributor and Brand	Guaranteed	bnnoA	beetasrand	Found	Publication
California Spray Chemical Co Wetconvilla	%	%	%	%	
	23.00 24.41	4.41	:		Cal. Dept. Agr., Spec. Pub. 66, 20 (1926).
California Spray Chemical Co., Watsonville, Cal. Ortho Powdered Bluestone, Package B.	24.00 23.06	3.06		alegi il Bayla	Cal. Dept. Agr., Spec. Pub. 58, 21 (1925).
Manufacturer unknown. Monohydrated Copper Sulphate!	20052-38 24 20 4 24 20 40				Conn. Agr. Expt. Sta., Sample 6913 (1928).
Mountain Copper Co., San Francisco, Cal. Mococo Bluestone	<u>51</u>	5.30	25.30 99.00 99.38	99.38	Cal. Dept. Agr., Spec. Pub. 66, 20 (1926).
Nichols Copper Co., New York, N. Y			25.20	24.90	25.20 24.90 Ore. Agr. Expt. Sta., Cir. 84, 15 (1927).

94 per cent passed a 300-mesh sieve.

#### Edward I. Creeley's Insect Powder.

CONNECTICUT EXPERIMENT STATION

(ARBURUA & MC INNES, SAN FRANCISCO, CAL.)

	Guaranteed.	Found.
Nicotine	0.002	0.04
Naphthalene	3.00	3.20
Sulphur	5.00	5.60
Phenols	1.00	0.40
Pyrethrum	6.00	

Cal. Dept. Agr., Spec. Pub. 75, 40 (1927).

# Creo Fenol Sheep Dip.

See "Phenol Soap Solutions".

#### Creolineum.

(H. S. FAWCETT, RIVERSIDE, CAL.)

Found: Light oils 17.10 per cent; medium oils 27.20 per cent; heavy oils 49.00 per cent; total oils 93.30 per cent; phenols 2.90 per cent; inert matter 3.80 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 39 (1923).

#### Creosote.

See "Brininstool".

#### Creosote Oil.

(AN-FO MFG. CO., OAKLAND, CAL.)

	Guaranteed.	Found.
Phenols	25.00	26.70
Coal-tar oils		72.10
Water	10.00	1.20

Cal. Dept. Agr., Spec. Pub. 75, 56 (1927).

#### Cresol.

(WESTERN WHOLESALE DRUG CO., LOS ANGELES, CAL.) Found: Phenols 45.30 per cent; water 21.80 per cent. Cal. Dept. Agr., Spec. Pub. 51, 43 (1925).

#### Cresolite Sheep Dip.

See "Phenol Soap Solutions".

#### Crestall Dip.

(BAIRD & MC GUIRE, HOLBROOK, MASS.) Found: Phenols 51.90 per cent; water 12.40 per cent. Cal. Dept. Agr., Spec. Pub. 51, 43 (1925).

#### Cresvlic Acid. Pure.

(PURITY CHEMICAL PRODUCTS CO., SANTA ROSA, CAL.)

	Guaranteed.	Found.
Phenols	98.00	96.85
Water	2.00	3.20

Cal. Dept. Agr., Spec. Pub. 75, 56 (1927).

#### Cresy-Lol.

See "Phenol Soap Solutions".

#### Crude Oil.

See "Oils, Mineral".

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

#### Crude Oil Emulsion.

See "Oil Emulsions, Mineral".

#### Cut Worm and Grub Destroyer.

	(CARPENTER-UDELL CHEM. CO., GRAND	RAPIDS, MICH.)	
		Guaranteed.	Found.
Arsenic,	metal	0.95	0.83
Mich.	Agr. Coll. Expt. Sta., Spec. Bull. 75, 9 (	1915).	

#### Cvanide-Chloride Marl Dust Mixture.

(J. S. MUNROE, SAN JOSE, CAL.)

Found: Pounds per ton: sodium cyanide, 4.24; sodium chloride 4.7. Cal. Dept. Agr., Spec. Pub. 34, 57 (1923).

#### D.

#### D. C. Dust No. 3

See "Niagara D. C. Dust No. 3".

#### Dead Shot Squirrel and Gopher Killer.

See "Strychnine Preparations".

#### Death Dust for Insects.

See "Pyrethrum".

#### D 18 Dust Mixture.

See "Niagara D 18 Dust Mixture".

#### De Lapp's Improved Lice Powder.

(W. C. DE LAPP).

	Guaranteed.	Found.
Nicotine	0.10	0.11
Naphthalene	3.00	1.02
Sulphur	15.00	12.40
Phenois	0.50	0.60
Hydrocarbons	7.00	4.20
Sodium Fluoride	7.00	3.44
Cal. Dept. Agr. Spec. Pub. 75 40 (1927)		

#### Del Monte Bleaching Fluid.

(DEL MONTE BLEACHING FLUID CO., SAN	FRANCISCO,	CAL.)
	Guaranteed.	Found.
Available chlorine	5.00	4.82
Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).		

#### Delousing Grease.

(PETALUMA AVIAN	PATHOLOGY, LABORATORY	, PETALUMA,	CAL.)
		Guaranteed.	Found.
Sodium fluoride		50.00	55.93
Cal. Dept. Agr. Spec.	Pub 75 62 (1927)		

#### Dendrol.

See "Oils, Mineral".

#### Derrisol

(WILLIAM COOPER & NEPHEWS, INC., CHICAGO, ILL.)

Guaranteed: Derris extract 5 per cent; fatty acid 50 per cent; inert matter 45 per cent.

Found: Non-volatile chloroform extract 60.45 per cent; water 18 per cent; ash 8.08 per cent; combined fatty acids (calculated from alkalinity of soluble ash) 0.62 per cent. The fatty acids if present must be present in the free form and not as soap.

Conn. Agr. Expt. Sta., Sample 8931.

#### Derris Root.

#### (MANUFACTURER UNKNOWN)

Found: Moisture 6.48 per cent; ether extract 8.79 per cent; methoxyl content 14.90 per cent; alcohol extract after ether extraction 14.25 per cent; passing 20 mesh, 100 per cent; 40 mesh, 100 per cent; 60 mesh, 91 per cent; 80 mesh, 79 per cent; 100 mesh, 72 per cent.

Canada Dept. Agr., Div. Chem., Rept., Dominion Chemist (1928).

#### Destroyer, Argentine Ant Syrup.

(TOWNE-ALLISON DRUG CO., SAN BERNARDINO, CAL.)

Found: Arsenious oxide 0.11 per cent; total arsenic, metal 0.083 per cent; invert sugar 32.10 per cent; cane sugar 20.86 per cent; water 47.25

Cal. Dept. Agr., Spec. Pub. 58, 17 (1925).

#### Destruxol.

(PARAGON CHEMICAL CO.)

	Guaranteed.	Found.
Nicotine	12.00	18.54
Sodium cyanide	8.00	15.67
Carbolic acid	2.00	2.39
Wood creosote oils	20.00	17.27
Water		27.36

Cal. Dept. Agr., Spec. Pub. 75, 42 (1927).

#### Dethol Fly Spray.

(DETHOL MFG. CO., RICHMOND, VA.)

Not analyzed.

Cal. Dept. Agr., Spec. Pub. 58, 48 (1925).

#### Di-Fli Home Spray.

See "Oils, Mineral".

#### Dr. Baker's Liquid Death Drops.

(JAMES AINSLIE, BROOKLYN, N.Y.)

Found: The substance is gasoline.

U. S. D. A., Bur. Chem., Bull. 68, 57 (1902).

#### Dipdust.

See "Bayer Dipdust".

Dips.

See "Phenol Soap Solutions"

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

#### Dr. David Robert's Poultry Louse Powder.

(DR. DAVID ROBERTS VETERINARY	CO., INC.)	
	Guaranteed.	Found.
Nicotine	0.04	0.28
Naphthalene	9.00	9.98
Sulphur	18.00	19.80
Sodium fluoride	5.00	0.54
Cal. Dept. Agr. Spec. Pub. 75, 40 (1927).		

Dr. G. Z. Wait's Sheep Dip.

Dr. Hess Dip & Disinfectant.

Dr. Le Gear's Dip & Disinfectant.

See "Phenol Soap Solutions".

#### Dr. Pierce's Bug Killer.

(THE KELLEY ISLAND LIME & TRANSPORT CO., CLEVELAND, OHIO.)

	Guaranteed.	Found.
Total arsenic (metallic)	0.56	1.33
Water-soluble arsenic (metallic)	0.03	0.30
Copper		0.53
N. J. Agr. Expt. Sta., Bull. 286, 13 (1915).		

#### Dormant Dust.

(NIAGARA SPRAYER CO., SAN FRANCISCO, CAL.)

	Guaranteed.	Found
Sodium polysulphide:	40.00	22.3
0 4 7 6. 6. 6. 6. 6. 6.		

Ore. Agr. Expt. Sta., Cir. 84, 11 (1927).

#### Dormant Soluble Oil.

See "Oil Emulsions, Mineral".

#### Dosch B-12 Green Copper Arsenic Dust.

(DOSCH CHEMICAL CO., LOUISVILLE, KY.)

	Guaranteed.	Found.
Copper	5.75	5.79
Arsenic, metal	2.75	3.39
The arsenic is present as calcium arsenate.		

Conn. Agr. Expt., Sta., Bull. 258, 370 (1924).

#### Dosch 85-15 Sulfur-Lead Arsenate.

(DOSCH CHEMICAL CO., LOUISVILLE, KY.)

Found: Moisture 0.50 per cent; free sulphur none; lead oxide 8.60 per cent; arsenic oxide 4.80 per cent; silica 0.50 per cent; sodium sulphate 2.33 per cent; calcium carbonate 0.23 per cent; calcium hydroxide 79.52 per cent; magnesium carbonate 2.41 per cent; iron and aluminum oxides.

Cal. Dept. Agr., Spec. Pub. 34, 56 (1923).

#### Dosch Nicotine-Sulfur Dust.

See "Nicotine-Sulphur Dusts".

Dosch No. 6 Nicotine Dust.

Dosch No. 10 Nico-Dust.

See "Nicotine Dusts".

# Dosch Sulfur-Arsenate Dust.

(DOSCH CHEMICAL CO., LOUISVILI	E. KY.)	
(DOSCH CHEMICAL CO., BOOLEVIEL	Guaranteed.	Found.
SulphurLead Arsenate	$84.00 \\ 14.70$	84.00 14.80
Cal. Dept. Agr., Spec. Pub. 34, 56 (1923).		

#### Dosch Tobacco Dust.

See "Tobacco Dusts".

250

#### Double Nico-Dust.

# Double Nicotine Dust.

See "Nicotine Dusts".

#### Dry Mix Sulfur-Lime.

(MECHLING BROS. CHEMICAL CO., CA See also "New Jersey Dry Mix Sulfur-Lime",	MDEN, N. J.) and "Niagara	Dry Mix."
Sulphur	Guaranteeu.	Found. 62.80
N. J. Agr. Expt. Sta., Bull. 424, 10 (1925).		

#### D 6 Dust.

See "Niagara D 6 Dust".

#### D 20 Dust.

See "Niagara D 20 Dust".

#### D 25 Potato Dust.

See "Niagara D 25 Potato Dust".

#### Du Pont Semesan, Jr.

(E. J. DU PONT DE NEMOURS & CO., INC., WILMINGTON, DEL.) Guaranteed: Hydroxymercurichlorphenol 10.00 per cent. Found: Mercury 6.59 per cent. Cal. Dept. Agr. Spec. Pub. 75, 62 (1927).

Dustall No. 3.

Dustall No. 6.

Dustall No. 8.

Dustall No. 10.

See "Nicotine Dusts".

#### Dust Mixture No. 3, Niagara.

See "Nicotine Dusts".

# Dust Mixture with Sulphur and Nicotine, Niagara.

See "Nicotine-Sulphur Dusts".

# Dyke's Louse Paint.

(RAYMOND & CO., LAWRENCE, KANSAS) Found: Petroleum, coal-tar, and carbon in suspension. U. S. D. A., Bur. Chem., Bull. 68, 58 (1902).

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

#### E.

Earwig Bait.		
(CITY OF PORTLAND, PORTLAND	ORE.)	
	Guaranteed.	Found.
Sodium fluoride	4.3	4.9
Ore. Agr. Expt. Sta., Cir. 64, 14 (1925).		
Earwig Bait.		
(HARDY MFG. CO., PORTLAND, C	ORE.)	
	Guaranteed.	Found.
Sodium fluoride	4.3	4.8
Ore. Agr. Expt. Sta., Cir. 84, 15 (1927).		
Earwig Bait.		

(MILLER PRODUCTS CO., PORTLAN	D, ORE.)	
	Guaranteed.	Found.
Sodium fluoride	4.3	4.7
Ore. Agr. Expt., Sta., Cir. 64, 14 (1925).		

#### Economy Germicide Dip.

See "Phenol Soap Solutions".

#### 85-15 Dusting Mixture.

#### 80-10-10 Mixture.

See "Nicotine Dusts".

#### 80 Sulphur-20 Lime.

(MECHLING BROS. CHEMICAL CO., CAI	MDEN, N. J.)	
	Guaranteed.	Found.
Sulphur	79.00	77.99
N. J. Agr. Expt. Sta., Bull. 441, 7 (1926).		

#### 80-20.

(J. R. GILLAM & BRO., BURLINGTON	N, N. J.)	
	Guaranteed.	Found
Sulphur	78.00	74.60
N. J. Agr. Expt. Sta., Bull. 441, 7 (1926).		

#### 80-20 Mixture.

See "Niagara 80-20 Mixture".

#### 80-20 Sulphur-Lime Dust.

(LUCAS KIL-TONE CO., VINELAND, N. J.)

Guaranteed: Sulphur 80.00 per cent.
Found: Sulphur 76.49 per cent. Fineness: coarser than 100 mesh, 0.80; passes 100 mesh, 99.2; 200 mesh, 77.1; 300 mesh, 24.60.

N. J. Agr. Expt. Sta., Bull. 459, 8 (1927).

#### 80-20 Sulfur-Lime Mixture.

(LUCAS KIL-TONE CO., VINELAND, N. J.)

Guaranteed: Sulphur 80.00 per cent. Found: Sulphur 77.84 per cent. Fineness: passes 100 mesh, 55.50; 200 mesh, 37.80; 300 mesh, 6.70.

N. J. Agr. Expt. Sta., Bull. 459,8 (1927).

#### Electric Rat & Roach Paste.

See "Phosphorus Preparations".

#### Electric Vermin Exterminator.

(F. B. SMITH, CANTON, OHIO.)

Found: Sand 0.69 per cent; carbon dioxide 5.22 per cent; phenol anhydride 2.25 per cent; calcium oxide 58.28 per cent; magnesium oxide 12.77 per cent; water and pink dye, by difference, 20.79 per cent. Material is partially air-slaked lime treated with crude phenol and dyed pink.

U. S. D. A., Bur. Chem., Bull. 68, 51 (1902).

#### Electro Micro.

(VREELAND CHEMICAL MFG. CO., LITTLE FALLS, N. J.)

	Guaranteed.	Found.
Total arsenic, metal		10.73 0.08 46.21
N. J. Agr. Expt. Sta., Bull. 459, 16 (1927).	10.00	10.21

#### Electro Micro $\frac{1}{2}$ and $\frac{1}{2}$ .

(VREELAND CHEMICAL CO., LITTLE FALLS, N. J.)

(VREEDAND CHEMIOID CO.,	Guaranteed.	Found.
Total arsenic oxide	12.60	14.36
Water-soluble arsenic oxide	0.77	0.57
Lead oxide		29.49
Sulphur	32.00	39.71
N. J. Agr. Expt. Sta., Bull. 301, 16 (1916).		

#### Elkay's Rat and Roach Paste.

See "Phosphorus Preparations".

#### El Roy Argentine Ant Poison.

(ROY SPECIALTY CO.

(ROY SPECIALTY CO.)	Guaranteed.	Found.
Arsenic, metal	0.20	0.11
Cal. Dept. Agr., Spec. Pub. 75, 24 (1927).		

#### El Roy Gopher Poison.

See "Strychnine Preparations".

#### El Vampiro.

(ALLAIRE, WOODWARD & CO.)

Guaranteed: Pyrethrum 75.00 per cent. Found: Ash 6.98 per cent; organic matter 93.02 per cent. Cal. Dept. Agr., Spec. Pub. 75, 63 (1927).

#### E. M. F. Orchard Spray.

#### Emulso.

See "Oil Emulsions, Mineral".

#### Entodust.

See "Arsenate of Lead".

#### Entomocide.

(SACRAMENTO CHEM. CO., SACRAMENTO, CAL.)

Claimed to contain phenol, carbon tetrachloride, oil of citronella and mineral oil.

Ore. Agr. Expt. Sta., Cir. 64, 15 (1925).

#### Estes's Roach Powder.

(N. T. ESTES & CO., OMAHA, NEB.)

Found: Borax 93,31 per cent; balance is pink coloring matter. U. S. D., A. Bur. Chem., Bull. 68, 43 (1902).

#### Exelol.

See "Oil Emulsions, Mineral".

#### Extermo.

(L. T. GRAVES FERTILIZER CO., PASADENA, CAL.)

Claimed to be both a fertilizer and an insecticide.

Found: Nitrogen 0.18 per cent; phosphoric acid 0.06 per cent; potassium oxide 0.08 per cent; organic matter 62.19 per cent; mineral matter 32.21 per cent; sodium carbonate 4.66 per cent; soap 1.10 per cent; moisture 5.60 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 60 (1923).

#### E-Z Burgundy.

See "Bordeaux Mixture".

#### E-Z Spray.

(E-Z WAY CO., OAKLAND, CAL.)

	Guaranteed.	Found.
Sodium polysulphide	33.00	33.62
Sodium thiosulphate	3.00	6.97
Soap	5.00	8.08
Cal. Dept. Agr., Spec. Pub. 58, 34 (1925).		

# F.

#### Fancier's Friend.

(JAMES BLANCHARD, NEW YORK, N.Y.)

Found: Moisture 5.03 per cent; ash 7.21 per cent; sulphur 30.77 per cent. Colored with lead chromate. Apparently intended as an imitation of pyrethrum.

U. S. D. A., Bur. Chem., Bull. 68, 53 (1902).

#### Felbro Whale Oil Soap.

See "Soaps".

#### Fertilizer and Insect Exterminator.

(MANUFACTURER NOT STATED).

Found: Moisture 7.60 per cent; organic matter 53.60 per cent; ash 39.80 per cent; sodium carbonate 5.30 per cent; nitrogen 0.69 per cent; phosphoric acid none; potassium trace. Probably a mixture of straw and sawdust, limestone, charcoal and ashes.

Cal. Dept. Agr., Spec. Pub. 34, 60 (1923).

#### Fertilore Soil Stimulant.

(FERTILORE CO., LOS ANGELES, CAL.)

, (-2011	, , , , ,	
WATER-SOLUBLE	Guaranteed.	Found.
Ferric oxide	23.98	19.25
Aluminum oxide	0.52	0.75
Copper oxide	0.40	1.25
Sulphur trioxide	39.75	25.80
WATER-INSOLUBLE		
Acid-insoluble	8.95	30.15
Ferric oxide	7.64	21.65
Aluminum oxide	0.11	0.50
Sulphur trioxide	6.72	45.93
GALGULATED COMPOSITION		
CALCULATED COMPOSITION		
Ferric sulphate	36.64	
Copper sulphate	3.92	
Aluminum sulphate	5.02	
Iron Pyrites	3.50	
Silica		

#### Fibro-Ferro Feeder.

(FIBRO-FERRO-FEEDER CO., GLENROSE, OHIO.)

Found: Moisture 7.80 per cent; sand 5.56 per cent; sulphur trioxide 22.37 per cent; chloride 7.70 per cent; ferric oxide 16.06 per cent; ferrous oxide 8.06 per cent. The substance is a mixture of organic matter (wood fibers) and partially oxidized ferrous chloride and sulphate.

U. S. D. A., Bur. Chem., Bull. 68, 52 (1902).

#### Fidelity Cockroach Paste.

See "Phosphorus Preparations".

Cal. Dept. Agr., Spec. Pub. 51, 58 (1925).

#### 50-50.

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y.)

	Guaranteed.	Found.
Lead arsenate	48.00	48.10
Ore. Agr. Expt. Sta., Cir. 64, 11 (1925).		

#### Fir-Tree Oil Soap.

Fish Oil.

See "Oil, Fish".

Fish Oil Soap.

See "Soaps".

5 X.

See "Nicotine-Sulphur Dusts".

#### Flea-Off-Flea Powder.

(OSGOOD DRUG STORES, OAKLAND, CAL.)

Not analyzed. Cal. Dept. Agr., Spec. Pub. 58, 48 (1925).

#### Fleck's Lice Exterminator.

(J. J. FLECK, TIFFIN, OHIO.)

Found: Insoluble in hydrochloric acid 4.20 per cent; carbon dioxide 9.56 per cent; sulphur trioxide 1.99 per cent; ferric oxide 1.43 per cent; calcium oxide 31.90 per cent; magnesium oxide 21.61 per cent; volatile matter other than carbon dioxide 29.88 per cent. The substance is a mixture of naphthalene, tobacco, partially air-slaked lime and pink dye.

U. S. D. A., Bur. Chem., Bull. 68, 54 (1902).

#### Fli-Mo-Cide.

See "Oils, Mineral".

#### Flit.

(THE STANDARD OIL CO., BAYONNE, N. J.)

Found: Specific gravity, 19° C., 0.810; flash point 60° C.; fire point 69° C.; residue at 100° C., 0.28 per cent; methyl salicylate 0.75 per cent. The base is kerosene.

Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).

#### Floraferro.

(CALIFORNIA COPPER PRODUCERS TRUST.)

	Guaranteed	Found.
Ferrous sulphate	25.00	29.74
Ferric sulphate	20.00	24.34
Copper sulphate	5.00	5.61
Cal. Dept. Agr., Spec. Pub. 75, 45 (1927).		

#### Floral Nicotine.

See "Nicotine Sulphate Solutions".

#### Fluoricide.1

(NITRATE AGENCIES, BAYONNE, N. J.)

	Guaranteed.	Found.	Guaranteed.	Found.
Fluorine	15.00	15.31	7.00	7.88
Conn. Agr. Expt. Sta., Samples		507.		

Fly Croke.

See "Oils, Mineral".

Fly Foil.

Not analyzed.

Cal. Dept. Agr., Spec. Pub. 58, 48 (1925).

#### Fly-Tox.

(CANADA REX SPRAY CO., BRIGHTON, ONT., CANADA)

(See also "Rex Fly Tox").

Found: Specific gravity, 19°C., 0.830; flash point 66° C.; fire point 78° C.; residue at 100° C., 0.31 per cent; methyl salicylate 3.13 per cent; pyrethrum present. Base is kerosene.

Canada Dept. Agr., Div. Chem. Rept. Dominion Chemist (1928).

Fly X.

See "Sapho Fly X".

<sup>1</sup> Two grades.

#### Ford's Ant Powder.

(FORD CHEMICAL CO., OAKLAND, CAL.)

	Guaranteed.	Found.
Insect powder	7.00	
Sodium fluoride	76.00	76.30
Organic matter		10.29
Cal. Dept. Agr., Spec. Pub. 75, 63 (1927).		

#### Formaldehyde (Formalin).

TABLE XI. FORMALDEHYDE

	Formal	dehyde	And the sylventer of th
Manufacturer or Distributor and Brand	Guaranteed	Found	Publication
Tuesd toesamoise	%	%	
Braun-Knecht-Heimann Co	37.00	36.80	Cal. Dept. Agr., Spec. Pub. 51, 54 (1925).
Minneapolis Drug Co		36.21	North Dakota Bull. 17, 52 (1927).
Noyes Bros. & Cutler		37.22	North Dakota Bull. 17, 52 (1927).
Perth Amboy Chem. Wks., P.A.C	37.00	36.82	Cal. Dept. Agr., Spec. Pub. 58, 50 (1925).
Perth Amboy Chem. Wks., U. S. P. Solution	37.30	35.60	Cal. Dept. Agr., Spec. Pub. 51, 54 (1925).
Roessler & Hasslacher Chem. Co., New York, N. Y	37.00	37.70	Ore. Agr. Expt. Sta., Cir. 84, 14 (1927).
Western Wholesale Drug Co., Los Angeles, Cal., U. S. P. Solution	37.30	37.28	Cal. Dept. Agr., Spec. Pub. 51, 54 (1925).

#### "40" Dust.

(VAYCIDE CHEMICAL CORP., BINGHAMTON, ALA.)

	Guaranteed.	Found.
Total arsenic, metal	1.80	2.33
Water-soluble arsenic, metal	0.75	0.37
Copper	1.05	0.55
Sulphur	25.00	22.73
Nicotine	1.00	0.34

N. J. Agr. Expt. Sta., Bull. 424, 16 (1925).

40% Nicotine Sulphate.

See "Nicotine Sulphate Solutions".

#### Four and One Spray.

No. 14 Spray Oil.

See "Oils, Mineral."

Frazier's 5X.

See "Nicotine-Sulphur Dusts."

#### Frazier's Wet-O-Dry Sulphur.

(r. A. FRAZIER CO.)		
	Guaranteed.	Found.
Sodium polysulphide	21.00	18.71
Sodium thiosulphate	12.00	13.15
Free sulphur	60.00	61.43
Cal. Dept. Agr., Spec. Pub. 58, 34 (1925).		

# Frazier's XX Sulphur.

(F. A. FRAZIER CO.)

	Guaranteed.	Found
Sodium polysulphide	21.00	15.84
Sodium thiosulphate	12.00	10.09
Free sulphur	60.00	69.36
Inert matter	7.00	

Cal. Dept. Agr., Spec. Pub. 51, 29 (1925).

#### Free Mulsion.

See "Oil Emulsions, Mineral".

#### Fresno.

(SCHOONMAKER & SONS, CEDAR HILL, N.Y.)

Found: This is an ammoniacal solution of copper carbonate containing  $2.69~\mathrm{per}$  cent. cupric oxide.

N. Y. Agr. Expt. Sta., Bull. 348, 98 (1912).

# Fuller's Carbolic Insecticide Sheep Dip.

Fuller's Sheep Dip and Cattle Wash.

See "Phenol Soap Solutions".

Fumigator.

See "Nicotine Dusts."

Fumispray.

See "Oil Emulsions, Mineral".

Fungi-Bordo.

See "Bordeaux Mixture"

Fungine.

(APHINE MFG. CO., MADISON, N. J.

Found: Soluble sulphur 4.95 per cent.

N. Y. Agr. Expt. Sta., Bull. 384, 296 (1914).

177		
run	RMO	gen.

(CHEMICAL PRODUCTS DIVISION, ROSE MFG. CO	., PHILADELPH	IA, PA.)
	Guaranteed.	Found.
Copper	0.47	0.43 0.32

#### Fungtrogen.

(GERMAIN SEED & PLANT CO., LOS ANGELES, CAL.)

	Guaranteed.	Found.
Copper	10.00	0.86
Cal. Dept. Agr., Spec. Pub. 75, 43 (1927).		

#### G.

#### Garden Dust.

(DOSCH CHEMICAL CO., LOUISVILI	LE, KY.)	
	Guaranteed.	Found.
Nicotine	2.00	2.24
A mixture of lead arsenate, sulphur and nicoting	ne.	
Conn. Agr. Expt. Sta., Bull. 242, 154 (1922).		

# Garden Guard.

See "Acme Garden Guard".

#### Garden Insecticide.

(GEO. H. LEE, OMAHA, NEB.)

Guaranteed: Nicotine, sodium fluoride, sulphur, naphthalene and pyrethrum present.

Found: There is a small deficiency in the sodium fluoride and sulphur.

Ore. Agr. Expt. Sta., Cir. 64, 15 (1925).

#### Germain Crude Carbolic Acid.

See "Phenol."

#### Germain Fish Oil Soap.

#### Germain's Rosin Wash.

See "Soaps".

#### Germain's Soluble Sulfur Compound.

(GERMAIN SEED & PLANT CO., LOS ANGELES, CAL.)

(ODRIMINITY DELLE & TELLE ,		
	Guaranteed.	Found.
Water-soluble		95.85 $37.22$
Polysulphide sulphur	• • • •	
Thiosulphate sulphur		13.57 $0.66$
Sulphate and sulphite sulphur		51.45
Total sulphur		
Sodium polysulphide	56.00	48.95
	25.00	33.50
Sodium thiosulphate	4.00	4.00
Inert matter	15.00	
Cal. Dept. Agr., Spec. Pub. 34, 32 (1923).		

#### Germo.

See "Oils, Mineral."

#### Germo Carboline.

See "Phenol Soap Solutions."

#### Germo Cresolis.

(GERMO MFG. CO., LOS ANGELES, CAL.) Found: Phenols 56.25 per cent; water 10.00 per cent. Cal. Dept. Agr., Spec. Pub. 51, 43 (1925).

#### Germo Cresosote Dip.

See "Phenol Soap Solutions".

#### Germo Magic Lice Powder.

(GERMO MFG. CO., LOS ANGELES, CAL.)

	Guaranteed.	Found.
Nicotine	0.40	0.42
Naphthalene	10.00	11.12
Sulphur	20.00	14.16
Phenols	1.83	1.22
Cal. Dept. Agr. Spec. Pub. 75, 40 (1927).		

#### Germo Rat Death.

(GERMO MFG. CO., LOS ANGELES,	CAL.)	
Barium carbonate	Guaranteed. 96.50	Found. 98.50

#### Germosol Special Disinfectant.

#### Germo Sheep Dip.

See "Phenol Soap Solutions."

#### Germo Worm Powder.

(GERMO MFG. CO., LOS ANGELES, CAL.)

Found: Volatile and organic matter 64.50 per cent; insoluble matter 2.35 per cent; calcium oxide 7.07 per cent; sulphur trioxide 10.81 per cent; ferric oxide 15.30 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 65 (1927).

#### Getsem Gopher Exterminator.

(RICHERT AND ROVER.)

Dhomat.	(MOTER!)	Guaranteed.	Found.
Water	Pub 75 56 (1097)	1.00	4.40 trace

#### Go-For Gopher.

See "Strychnine Preparations."

#### Gold Crown Poison.

# Gold Crown Poison Barley.

See "Phosphorus Preparations."

Found.

5.40

Gopher Death.

Gopher-Get-er.

Gopher-Go.

Gopher-Scent.

See "Strychnine Preparations."

#### Government Formula Argentine Ant Poison.

(F. A. GARDNER & CO., RIVERSIDE, CAL.) Guaranteed. Found. 0.17 Arsenious oxide..... 0.20 0.13

Go-West

(M. J. FORSELL & CO., SEATTLE, WASH.) Guaranteed. 5.00

Calcium arsenate..... Ore. Agr. Expt. Sta., Cir. 84, 15 (1927).

#### Grape Dust.

See "Hammond's Grape Dust."

#### Grasshopper Poison.

(EXCHANGE ORANGE PRODUCTS CO., LOS ANGELES, CAL.) Found. Guaranteed. 16.63 16.66 Arsenious oxide..... 12.00 12.58 Arsenic, metal.....

(All water-soluble).

(All water-soluble).

Cal. Dept. Agr., Spec. Pub. 34, 21 (1923).

#### Grasshopper Poison.

(SMITH-FRANK PACKING CO., SACRAMENTO, CAL.) Found. Guaranteed. 14.41 10.00 Arsenious oxide..... 10.91 Arsenic metal.....

Cal. Dept. Agr.; Spec. Pub. 34, 21 (1923).

#### Gray Arsenoid (Calcium and Copper Arsenite).

(ADLER COLOR & CHEMICAL WORKS, NEW YORK, N.Y.

Found: Water 16.10 per cent; combined arsenious oxide 21.24 per cent. soluble arsenious oxide 13.76 per cent; cupric oxide 15.10 per cent; calcium oxide 27.10 per cent; carbon dioxide, Prussian blue, sodium sulphate, etc., 6.70 per cent.

Univ. of Calif., Coll. of Agr., Expt. Sta., Bull. 151, 25 (1903).

#### Graylawn Farm Louse Chase.

(GRAYLAWN FARMS, INC., NEWPORT, VT.)

Guaranteed: Asagrea officinalis 10 per cent; sulphur 30 per cent; nicotine not less than 0.5 per cent; inert matter 59.50 per cent.

Found: Sulphur 32.21 per cent; nicotine 0.03 per cent; nón-volatile alkaloids 1.88 per cent. Sabadilla alkaloids present. Inert matter mostly calcium hydroxide.

Conn. Agr. Expt. Sta., Sample 3432.

#### Green Arsenoid (Copper Arsenite).

(ADLER COLOR & CHEMICAL WORKS, NEW YORK, N.Y.)

Found: Cupric oxide 28.83 per cent; combined arsenious oxide 53.51 per cent; free arsenious oxide 7.82 per cent; moisture 2.77 per cent; silica 0.40 per cent; organic matter, sodium sulphate, etc. 6.67 per cent. Is dved with "soluble blue".

Univ. of Calif., Coll. of Agr., Expt. Sta., Bull. 151, 26 (1903).

#### Green Cross Beetle Mort.

(LUCAS-KIL-TONE CO., VINELAND	o, N. J.)	
。 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1	Guaranteed.	Found.
Total arsenic, metal	18.00	18.53
Water-soluble arsenic, metal	1.00	0.12
Copper	5.00	5.90
그 경험 집에 보다 하다 하는 사람들이 살아 있다면 하는데 되었다면 하다 하는데 하는데 하는데 하는데 없다.		

#### N. J. Agr. Expt. Sta., Bull. 459, 12 (1927).

#### Green Cross Copper Lime Dust.

(LUCAS-KIL-TONE CO., VINELAND, N. J.)

Found: Copper 10.99 per cent.

N. J. Agr. Expt. Sta., Bull. 459, 11 (1927).

#### Green Cross Nico-Tone.

(LUCAS-KIL-TONE CO., VINELAND	, N. J.)	
	Guaranteed.	Found.
Nicotine	2.75	2.86
N. J. Agr. Expt. Sta., Bull. 459, 7 (1927).		

#### Green Cross P. B. K.

(LUCAS-KIL-TONE CO., VINELAND, N. I.)

	Guaranteed.	Found.
Total arsenic, metal	21.00	20.23
Water-soluble arsenic, metal	1.00	0.16
Copper	6.00	11.72
N. I. Agr. Expt. Sta. Bull. 459 12 (1927)		

# Green Cross Sulpho-Arsenate Powder.

(THE KIL-TONE CO., NEWARK, N. J.)

	Guaranteed.	Found.
Total arsenic, metal	10.10	15.39
Water-soluble arsenic, metal	0.66	0.71
Lead oxide		33.17
Sulphur	48.00	49.20
37 7 4 77 6. 77 77 66		

N. J. Agr. Expt. Sta., Bull. 301, 14 (1916).

#### Green Death.

(FARMERS' INDUSTRIAL UNION, SYRACUSE, N. Y.)

Found: The substance is artificially colored calcium arsenite, containing 35.94 per cent arsenious oxide.

N. Y. Agr. Expt. Sta., Bull. 348, 98 (1912).

#### Green Label Hydroxide Paste.

See "Mechling."

#### Gregory's Special Ant Syrup.

(GREGORY INSECTICIDE CO., BERKELEY, CAL.)

Guaranteed: Arsenic 2.56 per cent. Found: Arsenious oxide 1.01 per cent; arsenic metal 0.77 per cent; cane sugar 18.45 per cent; water 77.72 per cent.

Cal. Dept. Agr., Spec. Pub. 58, 16 (1925).

# Gregory's Standard Ant Destroyer.

(GREGORY	INSECTICIDE	CO., BERKELEY, CAL.)
		Guaranteed.
		2.00

	Guaranteed.	Found.
Invert sugar	2.00	1.57
Canelsugar	38.00	20.78
Water	56.00	76.21
Tartar emetic	4.4.	1.74

Cal. Dept. Agr., Spec. Pub. 58, 18 (1925).

#### Grub and Canker Worm Exterminator.

(G. H. MORRILL & CO., BOSTON, MASS.)

Found: Linseed oil 87.28 per cent; carbon 11.40 per cent; Prussian blue 1.32 per cent. This substance is printers' ink.

U. S. D. A., Bur. Chem., Bull. 68, 54 (1902).

#### H

# Hall's Lightning Squirrel and Gopher Poison.

See "Strychnine Preparations".

# Hall's Nicotine Sulphate Solution.

See "Nicotine Sulphate Solutions."

#### Hammond's Copper Solution.

(HAMMOND'S SLUG SHOT WORKS, BE.	ACON, N.Y.)	
1. 15 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	Guaranteed.	Found.
N. J. Agr., Expt. Sta., Bull. 441, 11 (1926).	3.05	3.70
N. J. Agr., Expl. Sta., Dun. 441, 11 (1920).		

# Hammond's Grape Dust.

# (HAMMOND'S PAINT & SLUG SHOT WORKS, BEACON, N. Y.)

	Guaranteed.	Found.
Sulphur	$64.00 \\ 0.7969$	63.87
Cupric oxide	0	present
Passing 100 mesh sieve		98.00
Passing 200 mesh sieve		88.00

Cal. Dept. Agr., Spec. Pub. 58, 50 (1925).

#### Handy Killer.

(RALPH B. ADAMS, LAKEVILLE, N. B., CANADA).

Found: Arsenious oxide 40.67 per cent. A strongly alkaline solution of sodium arsenite with a slight residue of iron oxide.

Canada Dept., Agr., Div. Chem., Rept. Dominion Chemist (1928).

#### He-Bo.

# (STERLING CHEMICAL CO., CAMBRIDGE, MASS.)

	Guaranteed.	Found.
Barium carbonate	36.00 2.00	34.41 $1.35$ $0.12$
water-soluble alsemic, metal		

Conn. Agr., Expt. Sta., Bull. 242, 159 (1922).

# Heine's Liquid Insect Destroyer and Disinfectant.

(HEINE CHEMICAL CO., HOLLIS, L. I.)

Found: The substance is a mixture of turpentine and kerosene. U. S. D. A., Bur. Chem., Bull. 68, 57 (1902).

Manufacturer or Distributor and Brand	Ч				
1 80 SV2 64 C	sA:	Alka	Alkaloids		
ETOT	Acid-Insoluble	beetnarand	Pound	Nitrogen	Publication
1 1. Hondins & Co Naw Voor's N V WIS:42	%	%	%	%	£ £
Hellebore Root	6.97	0.79	1.07	1.21	Conn. Agr. Expt. Sta., Bull. 242, 156 (1922).
Interstate Chemical Co., Jersey City, N. J. Key	5.03	0.25	1.43 1.29	1.29	Conn. Agr. Expt. Sta., Bull. 242, 156 (1922).
Leggett & Bro., New York, N. Y. Anchor. 7.50 3.23 0.20 1.26 1.59	3.23	0.20	1.26	1.59	Conn. Agr. Expt. Sta., Bull. 242, 156 (1922).
Lehn & Fink, New York, N. Y.1		:			Ore. Agr. Expt. Sta., Cir. 64, 15 (1925).
S. B. Penick & Co., New York, N. Y. Hellebore Root		5.40 1.14 1.27	1.27		Conn. Agr. Expt. Sta., Bull. 242, 156 (1922).
S. B. Penick & Co., New York, N. Y. Hellebore Root	4.46 1.05 1.00 2.70 1.48	1.00	2.70	1.48	Conn. Agr. Expt. Sta., Bull. 242, 156 (1922).

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

contain the alkaloids of powdered hellebore. 1 Guaranteed

#### Herbicide.

(READE MFG. CO., JERSEY CITY, N. J.)

Found: Solids 31.09 per cent; arsenious oxide 24.30 per cent; sodium arsenite (NaAsO<sub>2</sub>), 31.90 per cent. A green alkaline solution.

Conn. Agr. Expt. Sta., Bull. 258, 376 (1924).

#### Herbicide (Concentrate) Arsenic.

(STRAUSS-LASHER LABORATORIES, LOS ANGELES, CAL.)

Found: Baumé gravity 50.40; arsenious oxide none; arsenic oxide 43.99 per cent. The preparation is a solution of arsenic acid.

Cal. Dept. Agr., Spec. Pub. 51, 17 (1925).

#### Hercules Carbolic Acid.

See "Phenol."

#### Herold Poultry Worm Capsules.

	Guaranteed.	Found.
Nicotine	15.00	15.05
Cal Debt Agr Spec Pub 75 35 (1927).		

#### Hexol.

(SANITARY SUPPLY CO., SAN FRANCISCO, CAL.)

Guaranteed: Water 20.00 per cent.

Found: Soap 14.20 per cent; oils 61.80 per cent; water 23.40 per cent. Probably a pine oil preparation.

Cal. Dept. Agr., Spec. Pub. 75, 59 (1927).

#### Hexpo.

See "Bordeaux Mixture-Lead Arsenate."

#### Hirschey's Ant Control.

(HIRSCHEY'S ANT CONTROL)

Found: Arsenic 0.04 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 24 (1927).

# Hirschey's P. D. Q. Argentine Ant Solution.

(HIRSCHEY'S ANT CONTROL).

	Guaranteed.	Found.
Arsenic, metal	0.05	0.04
Cal. Dept. Agr., Spec. Pub. 66, 19 (1926).		

#### Hockwald's Carbolicide.

	(HOCKWALD CHEMICAL CO., SAN FRAN	NCISCO, CAL.)	
		Guaranteed.	Found.
Water		1.00	0.05
11 00001	후 전통 하나 하다 하다 하나 나는 사람들은 사람들이 되었다면 하는 것이 없는데 그렇게 되었다.		

Cal. Dept. Agr., Spec. Pub. 75, 66 (1927).

#### Hockwald's Creosote Oil.

(HOCKWALD CHEMICAL CO., SAN FI	RANCISCO, CAL.)	
	Guaranteed.	Found.
OilPhenolsWater	40.00	43.60 3.00 1.20
Baumé gravity of oil		1.80
Unsulphonated oil		1.00

Cal. Dept. Agr., Spec. Pub. 58, 46 (1925).

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

# Hockwald's Sheep Dip.

See "Phenol Soap Solutions".

#### Hooper's Fatal Food.

(O. HOOPER JADWIN, NEW YORK, N. Y.)

Found: Borax 92.44 per cent. The balance is corn meal and red coloring matter.

U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

#### Hydrated Lime.

See "Lime."

#### Hydro-Cv.

See "Hydrocyanic Acid."

#### TABLE XIII. HYDROCYANIC ACID

		lrogen anide	Constitution of the consti
Manufacturer or Distributor and Brand	Guaranteed	Found	Publication
California Cyanide Co., Cudahy, Cal. Calcyco Braun Gas	80.00	% 82.30	Cal. Dept. Agr., Spec. Pub. 58, 47 (1925).
Morago & Woodhead, Whittier, Cal. Hydro-Prussic Acid	96.98	96.80	Cal. Dept. Agr., Spec. Pub. 34, 24 (1923).
Owl Fumigating Co., Aero	96.00	97.50	Cal. Dept. Agr., Spec. Pub. 66, 34 (1926).
Owl Fumigating Corporation. Owl Prussic Acid		97.60	Cal. Dept. Agr., Spec. Pub. 34, 24 (1923).
The Pacific R & H Chemical Corp. Hydro-Cy	96.00	97.27	Cal. Dept. Agr., Spec. Pub. 66, 34 (1926).
The Pacific R & H Chemical Corp., Pacific R & H	96.98	97.19	Cal. Dept. Agr., Spec. Pub. 51, 20 (1925).

#### Hydro-Prussic Acid.

See "Hydrocyanic Acid."

#### Hydroxcide Paste.

(MECHLING BROS. MFG. CO., CAMDEN, N. J.)

Total -	Guaranteed.	Found.
Total arsenic, metal	6.50	4.90
		0.07
	5.00	4.89
		64.84
N. J. Agr. Expt. Sta., Bull. 407, 14 (1924).		

#### Hydroxcide Powder.

(MECHLING BROS. CHEMICAL CO., CAM	IDEN, N. J.)	
No.	Guaranteed.	Found.
Total arsenic, metal	1.50	16.33 $0.23$
Copper	12.50	12.92

#### I

# Impco Brand Nicotine Sulphur Dust.

See "Nicotine-Sulphur Dusts".

266

# Impco Complete Garden Dust.

(IIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIII	~ - /	
	Guaranteed.	Found.
Nicotine	0.50	0.52
Sulphur	25.00	27.82
Arsenic oxide		4.11
Lead oxide		5.75
Passing 100 mesh		85.00 $66.00$
Passing 200 mesh		00.00

Cal. Dept. Agr., Spec. Pub. 58, 26 (1925).

# Impco Extra Strong Insecticide Dust.

#### Impco Insecticide Dust.

#### Impco Poultry Tobacco Dust Worm Cure.

#### Impco Strong Insecticide Dust.

See "Nicotine Dusts."

#### Imperial Lice and Mite Remedy.

See "Lime-Sulphur."

#### Improved Kil-Tone.

(THE KIL-TONE CO., VINELAND, N	· J·)	
	Guaranteed.	Found.
Total arsenic, metal	5.80	7.48
Water-soluble arsenic, metal	0.50	0.19
Copper	2.45	2.14
N. J. Agr. Expt. Sta., Bull. 407, 13 (1924)		

#### Individual Single Treatment for Round Worms.

(AN-FO MFG. CO., OAKLAND, C	CAL.)	
Nicotine	Guaranteed.	Found. 15.61
Cal. Dept. Agr., Spec. Pub. 75, 35 (1927).		

#### Insecticide.

# (A. ST. LEGER, LAGO MAGGIORE, ITALY.)

A dark-colored odoriferous liquid, strongly alkaline; a sulphonated tar oil neutralized with ammonia and apparently containing "Rotenone", the active principle of derris.

Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).

#### Insectonos.

#### (WILLIAM BRANSON)

Found: Pyrethrum is the active ingredient. The inert ingredients are charcoal and essential oils.

Mich. Agr. Coll. Expt. Sta., Spec. Bull. 74, 10 (1915).

#### Insecto No. 1.

See "Oil Emulsions, Mineral".

#### Insect Powder.

See "Pvrethrum."

#### Insectrogen.

(CHEMICAL	PRODUCTS	DIVISION,	ROSE	MFG.	co.,	PHILADELPH	IIA, PA.)
						Guaranteed.	Found.
-1							

	Guaranteed.	Found.
Total arsenic, metal	1.75	2.69
Water-soluble arsenic, metal	0.15	0.27
Lead arsenate (PbHAsO <sub>4</sub> )	10.53	12 47
Nickel	0.47	0.44
Soap	13.05	10.89
Inert matter	73.95	

Conn. Agr. Expt. Sta., Sample 9308.

#### Instant Louse Killer.

# (DR. HESS & CO., ASHLAND, OHIO.)

Found: Moisture, 1.21 per cent; ash insoluble in hydrochloric acid, 27.10 per cent; ash soluble in hydrochloric acid, 64.78 per cent; carbon dioxide, coal-tar products and tobacco by difference, 6.91 per cent. Large amounts of iron, aluminum, calcium and carbonate, some magnesium, a little sulphate and phosphate. The substance consists of small amounts of calcium phenate and tobacco with large amounts of lime, calcium carbonate and clay.

U. S. D. A., Bur. Chem., Bull. 68, 51 (1902).

#### Iricide.

# (GRING'S IRICIDE CO., BERKELEY, CAL.)

Arcenia	Guaranteed.	Found.
Arsenic	0.10	0.15
Cal. Dept. Agr., Spec. Pub. 75, 24 (1927).		

#### J. Jebl.

# (J. H. RICE, ASHTABULA, OHIO.)

Guaranteed: Nicotine 0.80 per cent; inert matter 99.20 per cent. Found: Nicotine 1.08 per cent; water and volatile matter 79.29 per cent; solids 20.71 per cent; ash 7.85 per cent.

Conn. Agr. Expt. Sta., Bull. 242, 154 (1922).

# Jenning's Soluble Pine Oil.

#### (SANITARY SUPPLY CO.)

Soan	Guaranteed.	Found.
Soap Water. Pine Oil		12.10
Pine Oil.  Cal. Dept. Agr. Sh. D. J. T.		34.80
Cal. Dept. Agr., Spec. Pub. 75, 55 (1927).		53.32

# J. & M. Reliable Insecticide.

See "Soaps".

# K.

# Kalibor.

(NITRATE AGENCIES CO., BAYONNE	, N. J.)	
Total arsenic, metal	Guaranteed. 18.20 0.75 3.50	Found. 18.18 0.12 3.51

#### Kamforite "H".

# (HENEMAN BROS., HORNCASTLE, ENGLAND).

Claimed to be a combined pest destroyer, fumigant and fertilizer. Found: Naphthalene 25.41 per cent; nitrogen 0.25 per cent; phosphoric anhydride 4.33 per cent; potassium none. Ground bone, soot, lime and

Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).

#### Karspray.

See "Phenol Soap Solutions".

#### Kasulime.

(NITRATE AGENCIES CO., BAYONNE, N. J.)

Guaranteed: Sulphur 60.00 per cent.

Found: Sulphur 49.82 per cent; coarser than 100 mesh, 2.15 per cent; passing 100 mesh sieve, 86.35 per cent; 200 mesh 10.25 per cent; 300 mesh 1.25 per cent.

N. J. Agr. Expt. Sta., Bull. 459, 8 (1927).

#### Keresol.

# Kerosene Emulsion.

See "Oil Emulsions, Mineral".

Key Brand B. A. Cartridges.

(INTERSTATE CHEMICAL CO., JERSEY	Guaranteed.	Found.
Total arsenic, metal. Water-soluble arsenic, metal. Copper	0.75	6.39 0.68 10.14

# Key Brand B. Beetle Destroyer.

(INTERSTATE CHEMICAL CO., JERSEY C	ITY, N. J.) Guaranteed.	Found.
Total arsenic, metal	5.00	19.40 0.20 5.02

#### Key Brand Key B.

itty Diama		
(INTERSTATE CHEMICAL CO., JERSEY	CITY, N. J.) Guaranteed.	Found.
Total arsenic, metal Water-soluble arsenic, metal Copper N. J. Agr. Expt. Sta., Bull. 441, 12 (1926).	7.00	18.90 1.21 4.93

Key Brand No. 7 Key Green.

(INTERSTA	TE CHEMICAL	CO., JERSEY	CITY, N. J.) Guaranteed.	Found.
Total arsenic, metal				16.02
Water-soluble arsenic,	metal		3.00	0.34
Copper				17.33
N. J. Agr. Expt. Ste	a., Bull. 441,	12 (1926).		

#### Key-Cide.

See "Bordeaux Mixture-Lead Arsenate".

#### K-G Ant Poison.

(KIRK-GEARY CO., SACRAMENTO, CAL.)

Guaranteed: Arsenic 0.10 per cent.

Found: Arsenic 0.13 per cent; invert sugar 39.50 per cent; cane sugar 19.38 per cent; dextrin, etc., 4.22 per cent; water 35.01 per cent.

#### Cal. Dept. Agr., Spec. Pub. 58, 16 (1925).

# Kill-A-Mite.

(GRANT-ERICSON CO., SAN DIEGO, CAL.)

Guaranteed: Carbolic acid a certain per cent, inert matter none. Found: Oils 94.11 per cent; phenol 1.84 per cent free sulphur 2.17 per cent: water 1.40 per cent.

Cal. Dept. Agr., Spec. Pub. 51, 60 (1925).

#### Kill Hopper.

(GERMAIN SEED & PLANT CO., LOS AN	Guaranteed.	Found.
Arsenic	12.00	12.78
Cal. Dept. Agr., Spec. Pub. 75, 26 (1927).		

#### Kill-Well Worm Expeller. (CHICKEN BHARMACY BETALLIMA CAL)

(CHICKEN THARMACI, TETALUM	Guaranteed.	Found.
Water	54.00	76.60
Cal. Dept. Agr., Spec. Pub. 75, 64 (1927).		

#### Kilmol Squirrel-zophene.

See "Carbon Disulphide."

Kilto.

See "Allen's Kilto Paste."

Kil-Tone, Improved.

See "Improved Kil-Tone."

Kil-Tone, Modified.

See "Bordeaux Mixture."

# Kirk Geary Dipsol Sheep Dip.

See "Phenol Soap Solutions."

#### Kisants.

(BRUNSWIG DRUG CO., LOS ANGEL	ES, CAL.)	
	Guaranteed.	Found.
Arsenious oxide	11.00	7.78
Arsenic, metal	8.60	5.89
Cal. Dept. Agr., Spec. Pub. 58, 16 (1925).		

#### Kleenup Oil.

See "Oils, Mineral."

#### Kloral.

(A.	R.	MAAS	CHEMICAL	co.)
				Guarante

	Guaranteed.	Found.
Available chlorine	5.00	4.18
Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).		

#### K. M. G. Kills Morning Glory.

(WEED CONTROL CO. OF CALIFORNIA, BERKELEY, CAL.)

	Guaranteed.	Found.
Arsenic chloride (AsCl <sub>5</sub> )	5.00	8.02
Arsenic acid (H <sub>3</sub> AsO <sub>4</sub> )		4.79
Cal. Dept. Agr. Spec. Pub. 66, 18 (1926).		

Kolodust.

See "Sulphur".

Koloform.

Kolotax.

See "Nicotine Dusts."

Krelos.

See "Phenol Soap Solutions."

#### Krenol Dip.

See "Phenol Soap Solutions."

#### Kreo Phene.

(PURITY CHEMICAL PRODUCTS CO., SANTA ROSA, CAL.)

Guaranteed: Coal-tar oils 97.00 per cent.

Found: Phenols 41.40 per cent; coal-tar oils 57.60 per cent; moisture 1.00 per cent.

Cal. Dept. Agr. Spec. Pub. 75, 56 (1927).

#### Kreso Dip No. 1

See "Phenol Soap Solutions."

#### K. S. Q.

(HEMINGWAY & CO., INC., BOUND BROOK, N. I.)

	Guaranteed.	Found.
Total arsenic oxide	35.00	31.00
Water-soluble arsenic oxide	0.50	0.27
Cupric hydroxide [Cu(OH) <sub>2</sub> ]		14.16
N. J. Agr. Expt. Sta., Bull. 273, 11 (1914).		

#### Lambert's Death to Lice.

(D. J. LAMBERT, APPONAUG, R. I.)

Found: Moisture 3.97 per cent; ash insoluble in hydrochloric acid 35.91 per cent; ash soluble in hydrochloric acid 31.40 per cent; volatile matter other than moisture 28.72 per cent; a large amount of calcium, some iron, magnesium, potassium, sodium and carbonate, traces of sulphate and phosphate. The substance is a mixture of tobacco, lime, coal-tar products and dirt.

U. S. D. A., Bur. Chem., Bull. 68, 50 (1902).

#### Larvex.

#### (LARVEX CORP., BROOKLYN, N.Y.)

Guaranteed: Sodium aluminum silicofluoride 0.52 per cent: inert matter 99.48 per cent.

Found: Total solids 0.89 per cent; fluorine 0.41 per cent; silica 0.15 per cent; aluminum oxide 0.02 per cent; sodium and organic matter 0.40 per

North Dakota, Bull. 17, 52 (1927).

#### Laurel Green.

# (NICHOLS CHEMICAL CO., NEW YORK, N. Y.)

Found: Gypsum 50.00 per cent: greensand 20.20 per cent: copper arsenite 24.70 per cent; moisture, etc., 5.1 per cent; soluble arsenic compounds, 0.8 per cent.

Univ. of Cal., Coll. of Agr., Expt. Sta. Bull. 151, 24 (1903).

#### Lead Arsenate.

See "Arsenate of Lead."

#### Lead Arsenite.

See "Pink Arsenoid".

#### Lead Phenate.

(These are experimental preparations made by Dr. Friend of the Entomology Department of the Connecticut Agricultural Experiment Station.)

Sample No.	3351	7523	8257
Lead oxide	62.64	62.00	66.23
Phenol	31.61	38.82	33.25
Water	2.65		0.05

Sample 8257 corresponds approximately to the formula 6PbO. 7C6H5OH

#### Lee's Insect Powder.

(G. H. LEE CO., OMAHA, NEB.)

Found: Sulphur 47.93 per cent; arsenious oxide 5.00 per cent. Remainder probably tobacco and pyrethrum.

U. S. D. A., Bur. Chem., Bull. 68, 31 (1902).

#### Lee's Lice Killer.

(G. H. LEE CO., OMAHA, NEB.)

Found: Sample is probably creosote oil.

U. S. D. A., Bur. Chem., Bull. 68, 55 (1902).

#### Lee's Louse Powder.

(GEO. H. LEE CO., OMAHA, NEB.)

	Guaranteed.	Found.
Naphthalene.	2.00	0.60
2 100 Sulphili	12 (10)	43.80
Tuolide	3.00	3.01
Cal. Debt Agy Shee Pub 75 10 (1007)		

#### Agr., Spec. Pub. 15, 40 (1927).

#### Leggett's Killer.

(LEGGETT & BRO., NEW YORK, N. Y.)

Found: The substance is oil of turpentine. U. S. D. A., Bur. Chem., Bull. 68, 56 (1902).

#### Leggett's Roach Destroyer.

(LEGGETT & BRO., NEW YORK, N. Y.)

Found: Borax 22.22 per cent; pyrethrum and blue coloring matter present.

U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

#### Leinen's Mildew Go.

(BAY CITY SEED CO., SAN FRANCISCO, CAL.)

	Guaranteed.	Found
Free sulphur	75.00	73.34
Calcium hydroxide	25.00	23.84
Cal Debt Agr Spec Pub 34 57 (1023)		

#### Leinen's Poisoned Barley.

See "Strychnine Preparations."

#### Leinen's Rat-Go.

(JOHN F. LEINEN CHEMICAL CO.)

Guaranteed: Active ingredients 97.50 per cent; barium sulphate 0.30 per cent; silica 0.40 per cent; water 1.80 per cent.

Found: Barium carbonate 98.33 per cent. Cal. Dept. Agr., Spec. Pub. 75, 66 (1927).

#### Leinen's Sow-Bug Go.

(JOHN F. LEINEN CHEMICAL CO.)

	Guaranteed.	Found,
Total arsenic, metal	37.00	5.05
Water-soluble arsenic, metal	2.80	0.64
Cal. Debt. Agr., Spec. Pub. 75, 26 (1927)		

#### Lemon Oil Company's Standard Insecticide.

(GERMAIN SEED & PLANT CO., LOS ANGELES, CAL.)

	Guaranteed.	Found.
Dry Soap	6.00	11.85
Potassium carbonate	0.50	1.11
Vegetable oil	3.50)	
Terebenthine	5.00	7.45
Water	85.00	79.20
Cal. Dept. Agr. Spec. Pub. 75, 65 (1927).		
1 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		

#### Lennox Tobacco Soap.

See "Nicotine Soaps".

#### Lice-Go-Powder.

(S. O. BARNES & SON, GARDENA, CAL.)

Guaranteed: Talc 70.00 per cent.

Found: Naphthalene 3.71 per cent; sodium fluoride 3.30 per cent; essential oils present; loss at 400° C., 3.78 per cent; water 1.40 per cent; talc 11.16 per cent; calcium carbonate 70.10 per cent; calcium hydroxide 7.62 per cent; iron and aluminum oxides 1.10 per cent.

Cal. Dept. Agr., Spec. Pub. 51, 59 (1925).

#### Licresol.

#### Licresolis.

See "Phenol Soap Solutions."

# COMPOSITION OF COMMERCIAL INSECTICIDES. ETC.

#### Lilacko Spray

See "Oils, Mineral."

Nicot

#### Lilly Pulvules No. 142.

(ELI LILLY & CO., INDIANAPOLIS, IND.)

	Guaranteed.	Found.
Ticotine	13.00	12.88
Cal. Dept. Agr., Spec. Pub. 75, 35 (1927).		

# Lime Hydrated.

Manufacturer or Distributor	Passing 300 mesh sieve
The Miller Co., Stockbridge Mass	84.00%

# Conn. Agr. Exp. Sta., Sample 9509.

#### Limestone, Ground. (MANUFACTURER UNKNOWN.)

Found: Magnesium oxide 30.86 per cent; calcium oxide 45.87 per cent. Conn. Agr. Expt. Sta., Sample 8168.

Lime-Sulphur	. Drv.

See Table XIV.

Lime-Sulphur Solution.

See Table XV.

#### Lion Brand Whale Oil Soap.

See "Soaps."

#### Liquid Insecticide.

(MERRITT CHEMICAL SPECIALTIES, OAKLAND, CAL.)

Petroleum	Guaranteed. 94.00	Found. 92.00
Essential oils	6.00	present
Cal Debt Agr Spec Pub 34 58 (1023)		

# Liquid Lice Killer.

See "A. R. M. Liquid Lice Killer".

#### Liquor Cresolis Compound.

See "Phenol Soap Solutions".

#### Lockman's Worm O-Tone.

(KIRK, GEARY & CO.)

	Guaranteed.	Found
Water	75.00	77.00
Oil		22.76
CIP I C. BIER OF COM		

#### Cal. Dept. Agr., Spec. Pub. 75, 64 (1927).

#### London Purple.

See Table XVI.

#### Long and Gretter's Squirrel Poison.

See "Strychnine Preparations".

			cium ulphide		cium ulphate		ree ohur.	
Manufacturer or Distributor and Brand.	Total Sulphur.	Guaranteed.	Found.	Guaranteed.	Found.	Guaranteed.	Found.	Publication.
Aomo White I all Old Will O	%	%	%	%	%	%	%	
Acme White Lead & Color Works, San Francisco, Cal. Acme	59.10	65.00	65.85	5.00	6.22	10.00	11.36	Cal. Dept. Agr., Spec. Pub. 75 32 (1927) N. J. Agr. Expt. Sta., Bull. 441 8 (1926)
California Rex Spray Co., Benicia, Cal. Horseshoe		63.00	72.62	5.00	9.16	12.00	6.76	Cal. Dept. Agr., Spec. Pub. 66, 23 (1926)
California Spray Chemical Co, Watsonville, Cal. Ortho  W. C. Collins, San Francisco, Cal Devoe & Raynolds Co, New York, N. Y. The Dow Chemical Co., Midland, Mich The Glidden Co., Cleveland, Ohio.		64.00 70.00 64.00	53.91 70.30	5.00	11.10 13.50 12.92	8.00	9.38 11.20 9.94	Cal. Dept. Agr., Spec. Pub. 75, 32 (1927) Cal. Dept. Agr., Spec. Pub. 34, 30 (1923) Ore. Agr. Expt. Sta., Cir. 84, 7 (1927). Cal. Dept. Agr., Spec Pub. 75 32 (1927)
Glidden		63.00	59.27	5.00	7.45	12.00	18.42	Cal. Dept. Agr., Spec. Pub. 58, 32 (1925)
The Glidden Co., Cleveland, Ohio. Glidden.  Grasselli Chemical Co. Hemmingway's, Oakland, Cal.		71.00 65.00	61.10	5.00 5.00	10.64 6.85 6.80	8.00 15.00	21.35 8.23 15.40	Cal. Dept. Agr., Spec. Pub. 58, 32 (1925) Conn. Agr. Exp. Sta., Sample 9806. Ore Agr. Expt. Sta., Cir. 84, 7 (1927).
	62.43 60.79							N. J. Agr. Expt. Sta., Bull. 441, 8 (1926 N. J. Agr. Expt Sta., Bull. 441, 8 (1926
Lucas		70.00	64.44	5.00	6.38	10.00	13.08	Cal. Dept. Agr., Spec. Pub. 75, 32 (1927)

John Lucas & Co., Inc., Philadelphia, Pa.								
Lucas		65.00	64.27	5.00	8.80	15.00	13.00	Cal. Dept. Agr., Spec. Pub. 75, 32 (1927).
Montgomery Ward & Co., Portland, Ore.		70.00	64.00	5.00	8.70	10.00	12.00	Ore. Ag . Expt. Sta., Cir. 84, 7 (1927).
Chas. C. Navlet Co., San Jose, Cal. Navco		65.00	63.98	5.00	6.08	15.00	13.08	Cal. Dept Agr., Spec. Pub. 75, 32 (1927).
Nitrate Agencies Co., Bayonne, N. J	60.88							N. J. Agr. Expt. Sta., Bull 459, 9 (1927).
Sherwin-Williams Co., Cleveland, Ohio.								
SW		65.00	62.31	5.00	6.26	15.00	14.25	Cal. Dept. Agr., Spec. Pub. 75, 32 (1927).
Sherwin Williams Co., Cleveland, Ohio.								
SW		60.00	62.68	5.00	6.08	20.00	13.56	Cal. Dept. Agr., Spec. Pub. 75, 32 (1927).
Sherwin-Williams Co., Cleveland, Ohio.								
SW		68.00	62.40	5.00	6.99	12.00	14.88	Cal. Dept. Agr., Spec. Pub. 75, 32 (1927).
Sherwin-William; Co., Cleveland, Ohio.								1 3 , 1
SW		63.00	66.15	5.00	6.36	12.00	14.60	Cal. Dept. Agr., Spec. Pub. 75, 32 (1927).
Sherwin-Williams Co., Cleveland, Ohio								1 -8-1, 1
SW		65.00	66.43	5.00	6.36	10.00	11.68	Cal. Dept. Agr., Spec. Pub. 75, 32 (1927).
Sherwin-Williams Co., Cleveland, Ohio.								
SW		62.00	64.04	5.00	6.38	13.00	12 04	Cal. Dept. Agr., Spec. Pub. 75, 32 (1927).
E. H. Smith, Berkeley, Cal	25.02		33.05		6.10		35.36	Cal. Dept. Agr., Spec. Pub. 34, 30 (1923).

			To Sul <sub>I</sub>	otal ohur		cium ılphide		cium ulphate		umé r, degrees	
Manufacturer or Distributor and Brand.	Polysulphide Sulphur Found	Thiosulphate Sulphur Found	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	Guaranteed.	Found	Publication.
	%	%	%	%	%	%	%	%			
W. A. Allen, Pittstown, N. J. Allen's Concentrated	,			18.17		21.10		7.90		24.00 30.00	N. J. Agr. Expt. Sta., Bull. 424, 10 (1925). Cal. Dept. Agr., Spec,
Bear Creek Spray Co., Medford, Ore				22.90		24.20			30.00	29.10	Pub. 58, 27 (1925). Ore. Agr. Expt. Sta.,
James A. Blanchard Co., New York, N. Y. Lion	24.07	1.09		25.16	24.00	25.04	1.00	2.13		30.00	Cir. 84, 6 (1927). Cal. Dept. Agr., Spec. Pub. 75, 30 (1927). Cal. Dept. Agr., Spec. Pub. 34, 28 (1923).
California Paint Co., Oakland, Cal	24.32	0.95		25.26						33.90	Cal. Dept. Agr., Spec. Pub. 34, 28 (1923).
California Rex Spray Co., Benicia, Cal. Rex	.,					30.22		1.75	32.00		Cal. Dept. Agr., Spec. Pub. 75, 30 (1927). Cal. Dept. Agr., Spec.
ville, Cal. Ortho					28.00	30.30	2.00	1.75		32.75	Pub. 75, 30 (1927). Cal. Dept. Agr., Spec.
C. P. Lice and Mite Liquid						35.34		1.52		35.00	Pub. 51, 10 (1925).
Chino Valley Apple Growers' Association, Chino, Cal Lowell M. Clark. Home			••••		26.50 28.00	22.22 28.52		6.84	30.00	29.70 31.75	Cal. Dept. Agr., Spec. Pub. 58, 27 (1925). Cal. Dept. Agr., Spec. Pub. 75, 30 (1927).

Cunningham & Thomas, Ukiah, Cal.											Cal. Dept. Agr., Spec.
Farm Bureau Coop. Exchange, Eugene,					30.00	28.32	2.00	2.43	30.00	31.00	Pub. 75, 30 (1927). Ore. Agr. Expt. Sta.,
OreT. A. Felton's, Portland, Ore				$\begin{vmatrix} 18.70 \\ 23.10 \end{vmatrix}$	26.00	$\begin{vmatrix} 19.40 \\ 27.00 \end{vmatrix}$			30.00	29.00	Cir. 84, 6 (1927). Ore. Agr. Expt. Sta.,
F. A. Frazier Co						30.95			32.00		Cir. 64, 6 (1925). Cal. Dept. Agr., Spec.
General Chemical Co., San Francisco,											Pub. 66, 22 (1926). Cal. Dept. Agr., Spec.
Cal. Orchard					29.00	29.07	1.50	1.59	32.50	32.50	Pub. 75, 30 (1927). Cal. Dept. Agr., Spec.
Cal. Orchard						30.00		1.54	32.00	32.38	Pub. 75, 30 (1927). Cal. Dept. Agr., Spec.
Cal.¹ The Grasselli Chemical Co., New York,			25.00	25.54		31.31		1.75		33.50	Pub. 75, 30 (1927). Conn. Agr. Expt. Sta.,
N. Y. Grasselli's					26.35					34.10	Bull. 272, 147 (1925). N. J. Agr. Expt. Sta.,
N. J Hood River Spray Co., Hood River, Ore.		:::	24.00	24.70	25.64	27.10				32.50	Bull. 441, 8 (1926). Ore. Agr. Expt. Sta.,
Hood River Spray Co., Hood River, Ore.			22.00	1000		28.20			30.00		Cir. 84, 6 (1927). Ore. Agr. Expt. Sta.,
Imperial Laboratories, Kansas City,											Cir. 84, 6 (1927). Cal. Dept. Agr., Spec.
Mo. Imperial Lice and Mite Remedy. J. G. Orchards, Newburg, Ore	25.28	1.22		$26.50 \\ 22.70$		$\frac{31.98}{24.00}$		2.90	30.00	33.50 29.10	Pub. 34, 29 (1923). Ore. Agr. Expt. Sta.,
Leffingwell Rancho Co. XXX					30.00	30.62			32.00		Cir. 84, 6 (1927). Cal. Dept. Agr., Spec.
Leggett & Bro., Inc., New York, N. Y.									52.00	33.00	Pub. 75, 30 (1927). N. J. Agr. Expt. Sta.,
Anchor B. Leis, Beaverton, Ore			24.00	$25.82 \\ 24.10$		26.70			30.00	33.00 30.20	Bull. 441, 8 (1926). Ore. Agr. Expt. Sta.,
										55.25	Cir. 84, 6 (1927).

 $<sup>^{\</sup>rm 1}$  Calcium oxide, guaranteed 10.00 per cent; found 9.85 per cent.

			To Sul <sub>I</sub>	otal ohur	Cal Polysu	cium Ilphide		cium ilphate	Ba Gravit <b>y</b>	umé , degrees	
Manufacturer or Distributor and Brand	Polysulphide Sulphur Found	Thiosulpate Sulphur Found	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	Publication
	%	%	%	%	%	%	%	%			
Los Angeles Chemical Co., Los Angeles, Cal. Mission			3		28.00	31.24	4.00	2.12	32.00	32.00	Cal. Dept. Agr., Spec. Pub. 66, 22 (1926). Cal. Dept. Agr., Spec.
CalA. R. Maas Chemical Co., Los Angeles,					28.50	29.92	1.50	2.43		32.50	Pub. 75, 30 (1927).
Cal. <sup>1</sup>	13.80	4.80	20.00	18.60					30.00	35.10	Cal. Dept. Agr., Spec. Pub. 34, 29 (1923).
N. J. Concentrated Miller Products Co., Portland, Ore			$25.00 \\ 22.10$			26.20			30.00	32.50 32.30	N. J. Agr. Expt. Sta., Bull. 459, 8 (1927). Ore. Agr. Expt. Sta.,
Montgomery Ward & Co				24.50	26.40	26.20			30.00	32.40	Cir. 84, 6 (1927). Ore. Agr. Expt. Sta.,
Murphy Oil Co., East Whittier, Cal	12.95	3.39		16.34		19.55		8.05		25.00	Cir. 84, 6 (1927). Cal. Dept. Agr. Spec. Pub. 34, 29 (1923).
National Chemical Co., Pittsburgh, Pa. Flag Chas. C. Navlet Co., Inc., San Jose,					28.50	32.56	1.50	1.25	32.00	34.71	Cal. Dept. Agr., Spec. Pub. 75, 30 (1927). Cal. Dept. Agr., Spec.
Cal. NavcoParker's Laboratories, South Pasadena,			· · · · ·			31.73		1.34	32.00	32.80	Pub. 58, 28 (1925).
Cal. Parker's Magic Discovery <sup>2</sup> Comona Valley Lime & Sulphur Co.						4.71		2.43			Cal. Dept. Agr., Spec. Pub. 51, 10 (1925). Cal. Dept. Agr., Spec.
Gold Buckle					25.00	31.32	1.50	1.71	32.00	32.50	Pub. 75, 31 (1927).

								J. Landiele			
San Jose Spray Mfg. Co., San Jose, Cal. <sup>3</sup>						30.65		1.44	32.00	33.17	Cal. Dept. Agr., Spec. Pub. 75, 31 (1927).
Sherwin-Williams Co., Cleveland, Ohio. Hemmingway's					29.00	30.30		1.52	32.00	32.25	Cal. Dept. Agr., Spec. Pub. 75, 31 (1927). N. J. Agr. Expt. Sta.,
Pa			25.00	$25.72 \\ 23.20$	27.20	26.20			30.00	$\begin{vmatrix} 32.50 \\ 29.90 \end{vmatrix}$	Bull. 459, 8 (1927). Ore. Agr. Expt. Sta.,
Sutherlin Spray Plant, Sutherlin, Ore.				22.40		23.20			30.00	30.00	Cir. 84, 6 (1927). Ore. Agr. Expt. Sta., Cir. 84, 6 (1927).
S. Tinel, Yucaipa, Cal	15.39	3.46		18.35		19.33		8.21		29.50	Cal. Dept. Agr., Spec. Pub. 34, 29 (1923).
Valley Fruit Co., Walla Walla, Wash				25.20	31.80	30.90			30.00	31.90	Ore. Agr. Expt Sta., Cir. 84, 6 (1927).
G. P. Weldon, Ontario, Cal	17.82	1.18		19.00						27.00	Cal. Dept. Agr., Spec. Pub. 34, 28 (1923).
Yucaipa Valley Fruit Co. Mountain Boy					28.00	28.20	2.00	2.43	32.00	31.50	Cal. Dept. Agr., Spec. Pub. 75, 31 (1927).

<sup>&</sup>lt;sup>1</sup> Calcium oxide 11.2 per cent; sodium chloride 0.12 per cent; potassium and sodium sulphates, 0.38 per cent; cyanogen compounds as ferrocyanide 3.13 per cent. Polysulphide sulphur guaranteed 11.69 per cent; thiosulphate sulphur guaranteed 5.21 per cent. Made with refuse gas-house lime.

<sup>2</sup> Sugar 0.46 per cent.

<sup>3</sup> Guaranteed water 70.00 per cent; calcium sulphate 1.00 per cent.

Publication

Found

Lound

Guaranteed.

Guaranteed

(1927). 26 (1927).

27 75, 75,

Agr., Dept. Agr.,

Dept.

Cal.

24.08

Cleveland, Ohio.

Co.,

Sherwin-Williams

Pub. Pub.

Spec. Spec.

0.30 1.50

26.00 00

Works, Detroit,

Acme White Lead & Color Mich. Acme.....

	_	2	_	ч	
L	o	ы	u	1	ĸ

(GARDEN CHEMICAL CO., NEW Y	Guaranteed.	Found.
Water	40.00	12.20
Cal. Dept. Agr., Spec. Pub. 66, 36 (1926).		

## M.

## Maggi's Ant Poison.

(PAUL MAGGI, SANTA CLARA,	CAL.)	
	Guaranteed.	Found.
Sodium arsenate	0.20	0.26
Cal. Dept. Agr., Spec. Pub. 66, 19 (1926).		

	Maggotbait.		
	(CHAS. H. LILLY CO., SEATTI	LE, WASH.)	
		Guaranteed.	Found.
Anthracene o	il	0,80	1.40
Ore. Agr. E	Expt. Sta., Cir. 64, 15 (1925).		

## Magnesium Chloride, Flake.

(DOW CHEMICAL CO., MIDLAND, MICH.)

Found: Magnesium oxide 11.99 per cent; magnesium chloride (MgCl $_2$ -6H $_2$ O) 60.45 per cent. Nearly all water-soluble. Traces of iron, aluminum and calcium present.

Conn. Agr. Expt. Sta., Sample 6831.

## Mag-O-Tite.

(RANDALL-MC LAUGHLIN, SEATTLE	s, WASH.)	
	Guaranteed.	Found.
Total arsenic, metal	0.06	
Water-soluble arsenic, metal	0.002	0.06
Naphthalene	4.39	6.35
Conn. Agr. Expt. Sta., Bull. 242, 159 (1922).		

## Mapco Miscible Oil.

See "Oil Emulsions, Mineral."

# Mapco Nicotine Miscible Oil.

(MICHEL & PELTON CO.)

Found: Nicotine 0.90 per cent; soap 6.50 per cent; phenols 6.00 per cent; oils 79.30 per cent; water 6.20 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 39 (1927).

#### Mapco Nicotine Spray. (MICHEL & PELTON CO.)

(MICHEL & LELION CO.)		
	Guaranteed.	Found.
Nicotine	1.00	1.20
Soap	20.00	20.20
water	78.00	75.20
Pine oil	1.00	

Cal. Dept. Agr., Spec. Pub. 75, 39 (1927).

## Mapco Rosin Soap for Spray.

See "Soaps."

PURPLE	
LONDON	
XVI.	
TABLE	

Total arsenic, metal

## COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

## Mapco Sheep Dip.

See "Phenol Soap Solutions."

Mapco Whale Oil Soap.

See "Soaps."

Marcol.

See "Oils, Mineral".

Marvel Ant Gelatin.

(H. W. MC SPADDEN)		
	Guaranteed.	Found.
Arsenious oxide	0.10	0.11
Cal. Dept. Agr., Spec. Pub. 75, 24 (1927).		

## Mattes Ant Paste.

(L. A. MATTES.)		
	Guaranteed.	Found.
Arsenic, metal	0.75	0.82
Cal Debt Agr Spec Pub 75 24 (1927)		

#### McClellen's Sodium Fluoride.

See "Sodium Fluoride".

## McNear's Poultry House Spray.

See "Oils, Mineral."

## F. W. McNess Krenol Dip & Dininfectant.

See "Phenol Soap Solutions".

## F. W. McNess Lice Powder and Insecticide.

(FURST-MC NESS CO., OAKLAND, CAL.)

	Guaranteed.	Found.
Nicotine	1.00	1.36
Naphthalene	9.00	11.90
Sulphur	9.50	10.30
Cal. Dept. Agr., Spec. Pub. 75, 40 (1927)		

#### Mealy Bug Spray Oil.

See "Oil Emulsions, Mineral."

## Mechling's Dry Mix.

(MECHLING BROS. CHEMICAL CO., CAMDEN, N. J.)

Guaranteed: Sulphur 60.00 per cent.

Found: Sulphur 60.93 per cent; coarser than 100 mesh, 2.23 per cent; passes 100 mesh, 97.77 per cent; 200 mesh, 25.13 per cent; 300 mesh, 3.03 per cent.

N. J. Agr. Expt. Sta., Bull. 459, 8 (1927).

#### Mechling's Green Label Hydroxide Paste.

(MECHLING	BROS.	MFG.	CO.,	CAMDEN,	N.	J.)
				G	112 12	ante

(	2-19 -10 30/	
	Guaranteed.	Found.
Water		56.30
Total arsenic, metal	3.25	2.70
Water-soluble, arsenic, metal	0.25	0.12
Copper	4.50	3.50
Lead oxide		9.95
N. J. Agr. Expt. Sta., Bull. 286, 13 (1915).		

#### Mechling's P. T. B.

See "Paradichlorbenzene."

## Mechling's Yellow Label Hydroxide Paste.

	(MECHLING BROS. MFG. CO., CAMDER	N, N. J.)	
		Guaranteed.	Found.
The same of the sa	Water Total arsenic, metal. Water-soluble arsenic, metal. Copper	5.50 0.50 2.25	59.02 3.90 0.13 2.46 17.55
	Lead oxide		11.00

N. J. Agr. Expt. Sta., Bull. 286, 13 (1915).

#### Medol Emulsion.

See "Oil Emulsions, Mineral."

#### Melrosine.

(THE GARDEN CHEMICAL CO., NEW YORK, N. Y.)

Guaranteed: Inert matter (water) not over 85 per cent.

Found: Chloroform extract 12.94 per cent; water 79.70 per cent; ash 2.93 per cent; fatty acids (oleic) 10.92 per cent. Potassium present, arsenic absent. Small amount of essential oil probably present. The solid matter is mostly potassium oleate.

Conn. Agr. Expt. Sta., Sample 8933.

## Mercury Preparations, Organic.

See "Bayer Dipdust," "DuPont Semesan, Jr," "Semesan," "Semesan Bel" and "Uspulun".

#### Merit Lice Powder.

(TERRY PRODUCTS CO., PASADENA, CAL.)

Two samples:

#### Sample 1.

Found: Calcium hydroxide 59.08 per cent; calcium carbonate 17.64 per cent; calcium sulphate 0.34 per cent; magnesium carbonate 1.06 per cent; silica 12.80 per cent; iron and aluminum oxides 1.98 per cent; organic matter and water of constitution 6.72 per cent. Red coloring matter present.

## Sample 2.

Guaranteed: Talc 67.00 per cent.

Found: Calcium carbonate 1.36 per cent; sodium carbonate 2.91 per cent; ferric oxide 0.66 per cent; water 3.61 per cent; phenols 3.60 per cent; naphthalene 8.44 per cent; talc 78.33 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 57 (1923).

#### Merit Worm Expeller.

(TERRY PRODUCTS CO., PASADENA, CAL.)

Two samples:

#### Sample 1.

Found: Talc 40.62 per cent; magnesium sulphate 12.04 per cent; calcium sulphate 9.05 per cent; calcium silicate 6.85 per cent; Venetian red 7.34 per cent; sand 2.11 per cent; loss on ignition 21.18 per cent; nicotine 0.22 per cent.

13.46

Sample 2.

Found: Calcium hydroxide 1.55 per cent; magnesium sulphate 15.79 per cent; iron and aluminum oxides 6.00 per cent; sand 24.59 per cent; loss on ignition 51.60 per cent; nicotine 0.06 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 58 (1923).

Mildew-Go.

See "Leggett."

Mildew Killer.

See "M. K."

Miko Argentine Ant Poison.

(OSGOOD BROS. DRUG STORES).

	Guaranteed.	Found.
Arsenious oxide	0.10	0.10
Arsenic, metal	0.075	0.08
Cal Debt Agr Spec Pub 66 10 (1026)		

Mirto.

(H. HUNTER).

Found: Calcium hydroxide and calcium carbonate present, reaction alkaline.

Cal. Dept. Agr., Spec. Pub. 75, 66 (1927).

Miscible Oil.

Mission Brand Insecto No. 1.

See "Oil Emulsions, Mineral".

Mission Rat Poison.

See "Phosphorus Preparations."

Mixture 154 A Special.

Mixture S 155 A Special.

See "Nicotine-Sulphur Dusts."

Mixture M 163 Special.

Mixture No. 150.

Mixture No. 274 Special.

Mixture No. 221 Special.

See "Nicotine Dusts."

M. K. Mildew Killer (Unburnt Sulfur Ore).

(CHAS. A. LEE CO., KERMAN, CAL.)

Found: Moisture 2.68 per cent; free sulphur 48.15 per cent; free sulphuric acid 1.67 per cent; silica 43.50 per cent; iron and aluminum oxides 0.26 per cent; loss on ignition 3.12 per cent; passing 200 mesh sieve, 60.0 per cent.

Cal. Dept. Agr., Spec. Pub. 51, 58 (1925).

Modified Kil-Tone.

See "Bordeaux Mixture".

Moore's Prepared Squirrel Poison.

See "Strychnine Preparations".

David C. Moore U. C. Chicken Worm Capsules.

(PALMER DRUG CO.)
Guaranteed. Found.

13.00

Cal. Dept. Agr., Spec. Pub. 75, 35 (1927).

Moor Mans Stock and Poultry Dip and Disinfectant.

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

See "Phenol Soap Solutions."

Morehead's Ant Killer.

(MOREHEAD ANT KILLER CO., SAN FRANCISCO, CAL.)

| Guaranteed. Found, | 1.05 | 1.05 | Glucose. | 23.18 | large amount | Cane sugar. | 46.38 | large amount | Water. | 26.09 | 25.27 | Trace of arsenic present.

Cal Dept. Agr. Spec. Pub. 34, 18 (1923).

Morse's Gopher Poison.

See "Strychnine Preparations."

Morse's Snail and Slug Destroyer.

(C. C. MORSE & CO., SAN FRANCISCO, CAL.)

Moth Exterminator and Ovicide.

(H. LIEBES CO., SAN FRANCISCO, CAL.)

Found: Naphthalene 87.20 per cent; aldehydes 12.00 per cent. Cal. Dept. Agr., Spec. Pub. 34, 59 (1923).

Mulford Phosphorus Paste.

See "Phosphorus Preparations."

Muller's Grand Tobacco.

See "Tobacco Dusts."

Munn's Argentine Ant Poison.

(COFFIN-REDINGTON CO.)

Edgar A. Murray Roach Doom.

(EDGAR A. MURRAY CO.)

Guaranteed: Inert ingredients: curcuma 5.00 per cent; sodium carbonate 2.00 per cent; sodium sulphate 2.00 per cent; sodium chloride 2.00 per cent; iron oxide 1.00 per cent; insoluble in acids, 2.00 per cent.

Found: Sodium fluoride 91.99 per cent. Cal. Dept. Agr., Spec. Pub. 75, 62 (1927).

287

286 CONNECTICUT EXPERIMENT STATIO	N BULL	ETIN 300
Mystic Spray Insecticide		
(MYSTIC MFG. CO., GLENDALE,	CAL.)	
	Guaranteed.	Found.
Water	3.00	0.05
Cal. Dept. Agr., Spec. Pub. 75, 64 (1927).		
See "Oils, Mineral." Mystic XX Spray.		
N.		
Nabob Oil.		
See "Oils, Mineral."		
Naco Bordeaux-Paris Gree		
· (NITRATE AGENCIES CO., NEW YO		
	Guaranteed.	Found.
Cupric oxide	18.00	16.23
Total arsenious oxide	17.00 2.00	17.32
	2.00	0.82
Conn. Agr. Expt. Sta., Bull. 242, 153 (1922).		
Naco Brand Sulfur Dusting Mixture Fo	ormula No. 5	
(NITRATE AGENCIES CO., BAYONN	E, N. J.)	
	Guaranteed.	Found.
Total arsenic, metal	1.95	2.22
Water-soluble arsenic, metal	0.50	0.23
Sulphur	70.00	67.12
N. J. Agr. Expt. Sta., Bull. 441, 16 (1926).		
Naco Brand Sulphur Smoke Dusting	Mixture.	
(NITRATE AGENCIES CO., BAYONNI		
	Guaranteed.	Found.
Sulphur	80.00	79.58
N. J. Agr. Expt. Sta., Bull. 441, 7 (1926).		
Naco Dust Special.		
(NITRATE AGENCIES CO., BAYONNE		
m . 1	Guaranteed.	Found.
Total arsenic, metal	6.50	7.56
Water-soluble arsenic, metal	$0.50 \\ 7.30$	$0.08 \\ 5.82$
N. J. Agr. Expt. Sta., Bull. 459, 16 (1927).	7.00	0.04
11. J. Agr. Expt. Sta., Dutt. 459, 10 (1921).		744
Naco Kalibor.		
See "Kalibor."		
Naphicide.		
(NITRATE AGENCIES CO., BAYONNE	NT)	
	Guaranteed.	Found.
Nicotine	2.00	2.23
N. J. Expt. Sta., Bull. 407, 11 (1924).		
Naptho.		

(MILLER PRODUCTS CO., PORTLAND, ORE.)

Naphthalene.....

Ore. Agr. Expt. Sta., Cir. 64, 15 (1925).

Guaranteed.

9.10

Found.

9.8

## MPOSITION OF COMMENCENTE INCESTIGES, ---

		-				
TAT	apt	ha	la	n	0	
7.4	a v v	пο	10	ш	v	ю

(MILLER PRODUCTS CO., PORTLAN	D, ORE.)	
	Guaranteed.	Found.
Naphthalene	85.00	89.00
Ore. Agr. Expt. Sta., Cir. 84, 15 (1927).		

## Navco Crude Oil Emulsion.

## Navco Miscible Oil.

See "Oil Emulsions, Mineral."

#### Navco Weed Destroyer.

(CHAS. C. NAVLET CO., INC., SAN	JOSE, CAL.)	
	Guaranteed.	Found.
Arsenious oxide	30.00	31.48
Cal. Dept. Agr., Spec. Pub. 75, 25 (1927).		

#### Navco Whale Oil Soap.

See "Soaps."

## Neutral Emulsion.

See "Oil Emulsions, Mineral."

## New Insect Destroyer.

(DEWANE BOGUE, MEDINA, N. Y.)

Found: The substance is largely kerosene oil. N. Y. Agr. Expt. Sta., Bull. 284, 301 (1904).

## New Jersey Dry-Mix-Sulfur-Lime.

(J. R. GILLAM & BRO., BURLINGTON, N. J.)

Guaranteed: Sulphur 62.00 per cent.

Found: Sulphur 55.58 per cent; coarser than 100 mesh, 5.70 per cent; passes 100 mesh, 94.30 per cent; 200 mesh, 35.00 per cent; 300 mesh, 1.80 per cent.

N. J. Agr. Expt. Sta., Bull. 459, 8 (1927).

## New Jersey Dry Mix Sulfur-Lime.

	(JERSEY ORCHARD SUPPLY CO., BUR	LINGTON, N. J.)	
		Guaranteed.	Found.
Sulphur		62.00	66.74
	Expt. Sta., Bull, 424, 10 (1925).		

## New Nicotine Contact Mixture.

See "Nicotine Dusts."

#### Niagara "All in One".

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

	Guaranteed.	Found.
Copper	4.00	6.30
Total arsenic, metal	1.80	1.82
Water-soluble arsenic, metal	- 0.50	
Nicotine	1.10	1.11
Sulphur	19.00	
Guaranteed lead arsenate 9.5 per cent: Bordea	ux mixture 19	9 per cent;

Guaranteed lead arsenate 9.5 per cent; Bordeaux mixture 19 per cent; inert matter 51.40 per cent.

Conn. Agr. Expt. Sta., Bull. 242, 157 (1922).

## Niagara All-in-One Dust No. 6.

See "No. 6 All-in-One Dust."

## Niagara A-1 Dust Mixture.

See "Nicotine Dusts."

Niagara Copodust.

See "Copodust."

Niagara Copotex.

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y.)

m	Guaranteed.	Found.
Total arsenic, metal	. 6.25	6.58
water-soluble arsenic, metal	0.50	0.27
Copper	. 6.80	6.62
N. J. Agr. Expt. Sta., Bull. 459, 13 (1927).		

## Niagara D. C. Dust No. 3.

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y.)

	Guaranteed.	Found.
Copper		8.40
Maine Agr. Expt. Sta., Official Inspections 122,	86 (1926).	

## Niagara D 18 Dust Mixture.

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y)

[발생] 경기 : [[[[[[] [[] [[] [[] [[] [[] [[] [[] [		
Total arsenic, metal Water-soluble arsenic, metal Copper	Guaranteed. 6.25 0.50 6.80	Found. 7.02 0.20 6.70
N. J. Agr. Expt. Sta., Bull. 459, 14 (1927).	0.80	0.70

# Niagara Dormant Dust.

See "Dormant Dust".

#### Niagara Dry Mix.

	(NIAGARA SPRAYER CO., MIDDLEP	ORT, N. Y.)	
Sulphur Cal. Dept.	Agr. Spec Pub 75 28 (1027)	Guaranteed 61.00	Found. 63.39

#### Niagara D 6 Dust

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

	,		
	Guaranteed.	Found.	
Copper	6.80	6.94	
Cal. Dept. Agr., Spec. Pub. 75, 43 (1927).			

### Niagara D 20 Dust.

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

	Guaranteed.	Found.
Copper	6.80	6.99
10tal alseinc, metal	171	5.10
Water-Soluble arsenic, metal	0 50	0.04
Guaranteed: Monohydrated copper sulphate arsenate 13.30 per cent.	19.00 per cent;	calcium
Cal. Dept. Agr., Spec. Pub. 75, 44 (1927).		

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

## Niagara D 25 Potato Dust.

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y.)

(MINOAKA SIKATEK CO., MIESELICA	Guaranteed.	Found.
Copper	8.60	8.90
Guaranteed: Monohydrated copper sulphate 24	4.00 per cent.	
Conn A or Expt. Sta., Bull, 258, 370 (1924).		

## Niagara Dust Mixture No. 3 with Nicotine.

See "Nicotine Dusts."

#### Niagara Dust Mixture with Sulphur and Nicotine.

See "Nicotine-Sulphur Dusts."

## Niagara 85-15 Dusting Mixture.

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y.)

Guaranteed: Total arsenic 2.92 per cent; water-soluble arsenic not more than 0.50 per cent; sulphur 83.00 per cent; lead arsenate 14.70 per cent; inert matter 2.30 per cent.

Found: Total arsenic 3.02 per cent. Conn. Agr. Expt. Sta., Bull. 242, 157 (1922).

## Niagara 80-10-10 Mixture.

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

Guaranteed: Total arsenic 1.95 per cent; water-soluble arsenic not more than 0.50 per cent; sulphur 78.00 per cent; lead arsenate 9.80 per cent; inert matter 12.20 per cent.

Found: Total arsenic 2.26 per cent.

Conn. Agr. Expt. Sta., Bull. 242, 157 (1922).

#### Niagara 80-20 Mixture.

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

	Guaranteed.	Found.
Total arsenic, metal	3.90	4.67
Water-soluble arsenic, metal	0.50	0.18
Sulphur	77.00	79.26
Cal Debt Agr Spec Pub 75 27 (1927)		

# Niagara Entodust, No. 1

See "Arsenate of Lead."

Niagara "50-50".

See "50-50".

Niagara Kolodust.

See "Sulphur."

#### Niagara Koloform.

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

Guaranteed: Sulphur 61.00 per cent.

Found: Sulphur 71.11 per cent; coarser than 100 mesh, 1.53 per cent; passes 100 mesh, 98.47 per cent; 200 mesh, 53.67 per cent; 300 mesh, 10.50 per cent.

N. J. Agr. Expt. Sta., Bull. 459, 8 (1927).

## Niagara Kolotax.

1	NIAGARA	SPRAYER	CO	MIDDLEPORT.	NT TT )	
	A CONTRACTOR OF STREET	OT TOTAL	CO.,	MIDDLEPORT.	NVI	

- Co., MIDDELI	, middle oki, N.Y.)		
Total arsenic, metal. Water-soluble arsenic, metal. Sulphur.	Guaranteed.	Found. 2.16 0.15 88.52	
N. J. Agr. Expt. Sta., Bull. 459, 13 (1927)	11.00	00.04	

# Niagara Mixture No. 150, 70-10-20.

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

	Guaranteed.	Found.
Total arsenic, metal.	1.95	1.90
Water-soluble arsenic, metal	0 50	0.20
oulphur	67.00	69.85
N. J. Agr. Expt. Sta., Bull. 459, 15 (1927).		00.00

## Niagara Mixture S 303.

# Niagara New Nicotine Contact Mixture D-1

# Niagara New Nicotine Contact Mixture D-11

## Niagara Nicotine Contact Mixture.

See "Nicotine Dusts."

## Niagara Nicotine Sulfur Dust No. 7.

See "Nicotine-Sulphur Dusts."

# Niagara 90-10 Dusting Mixture (Pomodust).1

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

Sulphur Total arsenic, metal Water-soluble arsenic, metal	1.95 0.50	Found. 87.79 1.84 0.24	Control of the contro	Found. 89.73 1.84 0.12	
Lead Arsenate	9.80 370 (1924)	and B	ull. 272, 146	 3 (1925)	

## Niagara No. 4 Nicotine-Lime Dust.

# Niagara No. 6 Dust Mixture, with Nicotine.

# Niagara No. 10 Dust Mixture, with Nicotine.

See "Nicotine Dusts."

# Niagara No. 3 Dust Mixture with Nicotine and Basic Lead Arsenate.

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

Nicotine	Guaranteed.	Found.
Nicotine	1.00	1.25
1 Otal alseme, metal	910	2.25
Water-soluble arsenic, metal.	0.50	0.06
Lead oxide		12.50
Cal. Dept. Agr., Spec. Pub. 75, 38 (1927).		

## Niagara P. D. B. Paradichlorobenzene.

See "Paradichlorbenzene."

# Niagara Potato Dust Mixture without Poison.

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

Guaranteed: Copper 6.50 per cent; monohydrated copper sulphate 19.50 per cent; inert matter 80.50 per cent.
Found: Copper 7.56 per cent.

Conn. Agr. Expt. Sta. Bull. 242, 157 (1922).

## Niagara 70-10-20 Mixture.

See "Niagara Mixture No. 150."

# Niagara Soluble Sulphur Compound.

(NIAGARA SPRAYER CO., MIDDLEPORT, N.Y.)

Guaranteed: Sodium polysulphide 40.00 per cent; sodium thiosulphate 18.00 per cent; free sulphur 3.00 per cent; inert matter 39.00 per cent. Found: Total sulphur 41.40 per cent, present mostly as thiosulphate and free sulphur. Not in original package and probably decomposed.

Conn. Agr. Expt. Sta., Bull. 258, 370 (1924).

## Niagara Soluble Sulphur Solution.

See "Soluble Sulphur Solution."

# Niagara Special Mixture M. 163.

	(NIAGARA SPRAYER CO., MIDDL	EPORT, N.Y.)	
		Guaranteed.	Found.
Copper		10.40	10.90
N T Agr	Expt. Sta., Bull, 459, 15 (1927).		

## Niagara Special Mixture No. 161.

See "Special Mixture No. 161."

## Niagara Special Mixture No. 274.

(NIAGARA SPRAYER CO., MIDDLEPO	ORT, N.Y.) Guaranteed.	Found.
Total arsenic, metal	0.98 0.50 82.00	1.27 0.20 82.78

## Niagara Special Mixture No. 221

(NIAGARA SPRAYER CO., MID	DLEPORT, N.Y.)	
Total arsenic, metal	Guaranteed 0.98 0.50 73.00	Found. 1.51 0.35 71.70

## Niagara Special Mixture 154 A.

# Niagara Special Mixture S 155 A.

See "Nicotine-Sulphur Dusts."

#### Niagara Special 68.

(NIAGARA SPRAYER CO., MIDDLE	PORT, N. Y.)	
(Mindakii bilane	Guaranteed.	Found.
Total arsenic, metal	1.94	2.31
Water-soluble arsenic, metal	2 02	$0.20 \\ 4.10$
Maine Agr. Expt. Sta., Official Inspections 1	14, 88 (1924).	

<sup>1</sup> Two Samples.

# Niagara Special 210 Dust Mixture.

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y.)

	Guaranteed.	Found.
Total arsenic, metal	7.40	2.68
water-soluble arsenic, metal	0.50	0.24
Copper	6.80	8.21

N. J. Agr. Expt. Sta., Bull. 459, 15 (1927)

## Niagara Sulphur Dust.

See "Sulphur Dust."

## Niagara Sulphur with 7 per cent Nicotine Solution.

# Niagara Sulphur with 10 per cent Nicotine Solution.

See "Nicotine-Sulphur Dusts"

Niagara Vitedust.

See "Vitedust."

## Nickel Carbonate.

(MANUFACTURER UNKNOWN)

Found: Nickel 44.64 per cent; nickel carbonate 90.28 per cent; trace of sulphate present.

Conn. Agr. Expt. Sta., Bull. 242, 160 (1922).

Nico Dust.

See "Nicotine Dusts."

## Nico-Dust for Walnut Worm and Aphis.

(WALNUT GROWERS' SPRAY MFG. CO., LOS ANGELES, CAL.)

Guaranteed: Nicotine 0.40 per cent; basic lead arsenate 10.00 per cent; total arsenic 1.50 per cent; water soluble arsenic trace, inert matter (lime) 89.60 per cent.

Found: Nicotine 0.62 per cent; lead oxide 11.95 per cent; arsenic oxide 2.89 per cent; total arsenic 2.19 per cent; water-soluble arsenic 0.08 per cent.

Cal. Dept. Agr., Spec. Pub. 51, 50 (1925).

## Nico Dust No. 75.

(NICO-DUST MFG. CO.)

	Guaranteed.	Found.
Total arsenic, metal		5.65
Water-soluble arsenic, metal.		0.18
Cal Data Ass. St. D. J. 77. 07. (2007)	22.00	15.05
Cal. Dept. Agr., Spec. Pub. 75, 27 (1927).		

## Nico Fume (Liquid).

See "Nicotine Sulphate Solutions."

#### "Nico-Fume" Paper.

(	ТОВАССО	BY-PRODUCTS	&	CHEM.	CORP.,	LOUISVILLE,	KY.)

	Guaranteed.	Found.
Nicotine	20.00	21.10
Ore. Agr. Expt. Sta. Cir. 84, 10 (1997)		-1.10

## Nico Fume Tobacco Powder.

See "Nicotine Dusts."

#### COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

#### Nico Garden Dust.

Nico Soan.

See "Nicotine Soaps."

Nico-Sulphur Dust No. 5.

Nico-Sulphur Dust No. 6.

Nico-Sulphur Dust No. 8.

See "Nicotine-Sulphur Dusts."

Nicota.

Nicoticide.

See "Nicotine Sulphate Solutions."

Nicotine Capsules.

(LANGLEY & MICHAELS).		
	Guaranteed.	Found.
Nicotine	13.00	13.48
C 1 D-14 Ann Shor D-1 75 25 (1097)		

Cal. Dept. Agr., Spec. Pub. 75, 35 (1927).

## Nicotine Contact Mixture.

See "Nicotine Dusts".

Nicotine Dusts.

See Table XVII.

Nicotine, Floral.

See "Nicotine Sulphate Solutions."

Nicotine-Lime Dust.

See "Nicotine Dusts."

Nicotine Miscible Oil.

(MICHEL & PELTON CO.)

Guaranteed: Water 10.00 per cent.

Found: Nicotine 1.00 per cent; soap 5.40 per cent; phenols 5.70 per cent; tar oil 21.10 per cent; mineral oil 61.60 per cent; water 4.00 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 39 (1927).

Nicotine, Rose.

See "Nicotine Sulphate Solution."

Nicotine Soaps.

See Table XVIII.

Nicotine Spray.

See "Nicotine Soaps."

## Nicotine-Sulfur-Lead Arsenate Dust.

(WALNUT GROWERS' SPRAY MFG. CO., LOS ANGELES, CAL.)

	Guaranteed.	Found.
Sulphur	60.00	44.19
- TCO time	Control of the contro	1.55
arsenate	10 00	0.27
Tisellic, metal	2 00	0.06
Inert matter	28.00	

Cal. Dept. Agr., Spec., Pub. 34, 49 (1923).

-	30
	2
-	1
-	,
- 5	-
-	300
- 1-	4
	)
100	4
U	1
-	4
	1
-3333	100
100	200
(	)
	7
OMPOSITION OF	33
. 100	
-	
-	,
1	4
- 3	
1	
	2
COMMERCIAL	1
-	1
-	4
	4
ho	
~ <	9
-	20.
(Z	1
	-
~	,
-	
-	4
-	-
12	400
-	4
1	3
1	4
TI	7
- 2	
U.	1
	L
-	3
0.00	2
	te i
-	1
-	4
100	
-	1
-	39
I	1
1000	53
U	1
4	
INSECTICIDES, ETC.	4
L.	i
i jes	1
200	4
(	2
	48

	Ni	cotine		Fine finer	ness, than	
Manufacturer, Distributor or Brand	Guaranteed	Found	Inert matter, Guaranteed	100 mesh	200 mesh	Publication
Anaheim Feed & Fuel Co., Anaheim, Cal. Double	76	%	%	%	%	
Bomberger Seed Co., Modesto, Cal. Double Nico-		2.71				Cal. Dept. Agr., Spec. Pub. 34, 44 (1922).
Bomberger Seed Co., Modesto, Cal. Nico Dust <sup>1</sup> California Associated Raisin Co., Fresno, Cal. Dou-		1.29				Cal. Dept. Agr., Spec. Pub. 34, 44 (1922). Cal. Dept. Agr., Spec. Pub. 34, 42 (1922).
California Associated Raisin Co., Fresno, Cal. Nico-		1.40	,			Cal. Dept. Agr., Spec. Pub. 34, 44 (1922).
California Pest Control Co. Calpest Garden Dust. California Spray Chemical Co., Watsonville, Cal. California Spray Chemical Co., Watsonville, Cal. California Spray Chemical Co., Watsonville, Cal.	3.00 2.00 2.00		95.00 98.00			Cal. Dept. Agr., Spec. Pub. 34, 42 (1922). Cal. Dept. Agr., Spec. Pub. 75, 36 (1927). Cal. Dept. Agr., Spec. Pub. 34, 42 (1922). Cal. Dept. Agr., Spec. Pub. 34, 43 (1922).
California Spray Chemical Co., Watsonville, Cal	3.50	3.60				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
California Spray Chemical Co. Watsonville Col	4.00	3.91				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
California Sprayer Co., Calispray Dust No. 12 California Sprayer Co. Calispray Insecticide Dust	3.60 1.70	3.45 1.70				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927). Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
No. 11	0.50	1.22		90.00	66.00	Cal. Dept. Agr., Spec. Pub. 66, 25 (1926).

California Sprayer Co. Calispray Insecticide Dust No. 15	2.40	2.32	97.80	85.50	72.50	Cal. Dept. Agr., Spec. Pub. 66, 25 (1926). Cal. Dept. Agr., Spec. Pub. 34, 43 (1922).
Dosch Chemical Co., Louisville, Ky. Dosch No. 10 Nico-Dust <sup>1</sup> Hall Tobacco Chemical Co., New York, N. Y. Fum-	3.80		96.20			Cal. Dept. Agr., Spec. Pub. 34, 42 (1922).
igator Heightstown Hardware Co., Heightown, N. J. Ax-	12.50	9.80				Ore. Agr. Expt. Sta., Cir. 64, 11 (1925).
$\mathrm{fixo^2}$	1.25	1.47	98.75		.,	Conn. Agr. Expt. Sta., Bull. 258,371 (1924).
Insecticide Supply Co., Los Angeles, Cal. Dustall No. 3	0.90	1.33				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
No. 6	2.00	2.30			,	Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
Insecticide Supply Co., Los Angeles, Cal. Dustall No. 8	2.75	3.48				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
No. 10	3.50	3.58				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
International Milling Co., Impco Extra Strong Insecticide Dust.  International Milling Co., Impco Insecticide Dust.	2.50 2.00	1.89 1.95		79.00 86.00	59.00 64.00	Cal. Dept. Agr., Spec. Pub. 58, 36 (1925). Cal. Dept. Agr., Spec. Pub. 58, 36 (1925).
International Milling Co., Impco Strong Insecticide Dust	1.20	1.06		86.00		Cal. Dept. Agr., Spec. Pub. 58, 36 (1925).
Manufacturer unknown	2.00	$\begin{bmatrix} 2.81 \\ 3.25 \\ 3.91 \end{bmatrix}$	• • • • •	• • • • • • • • • • • • • • • • • • • •	••••	Conn. Agr. Expt. Sta., Sample 9663. Conn. Agr. Expt. Sta., Sample 4170. Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
Niagara Sprayer Co., Middleport, N. Y. Niagara A-1 Dust Mixture		3.01				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
Niagara Sprayer Co., Middleport, N. Y. Niagara Dust Mixture No. 3 with Nicotine	1.00	1.13			• • • •	Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
Niagara Sprayer Co., Middleport, N. Y. Niagara Mixture S 303	1.00	1.35		93.00	82.00	Cal. Dept. Agr., Spec. Pub. 66, 25 (1926).

Lime carrier.
 Total ash 96.50 per cent; insoluble ash 1.28 per cent; calcium oxide 31.85 per cent; magnesium oxide 23.57 per cent.

	Nic	otine		Fine	eness, than	
Manufacturer, Distributor or Brand	Guaranteed	Found	Inert Matter, Guaranteed	100 mesh	200 mesh	Publication
Niagara Sprayer Co., Middleport, N. Y. Niagara	%	%	%	%	%	
New Nicotine Contact Mixture D-1	1.25	1.43	98.75			Conn. Agr. Expt. Sta., Bull. 258, 371(192
New Nicotine Contact Mixture D 11. Niagara Sprayer Co., Middleport, N. Y. Niagara	1.25	1.49				Cal. Dept. Agr., Spec. Pub. 58, 36 (1925)
Nicotine Contact Mixture	2.20	2.38			,	Conn. Agr. Expt. Sta., Bull. 242, 155(1922)
liagara Sprayer Co., Middleport, N. Y. Niagara No. 4 Nicotine-Lime Dust	1.90	1.95				Cal. Dept. Agr., Spec. Pub. 34, 43 (1922)
iagara Sprayer Co., Middleport, N. Y. Niagara No. 6 Dust Mixture with Nicotine	2.00	2.38		90.00	80.00	Cal. Dept. Agr., Spec. Pub. 58, 36 (1923)
iagara Sprayer Co., Middleport, N. Y. Niagara No. 10 Dust Mixture with Nicotine	3.25	3.89		89.00	72.00	Cal. Dept. Agr., Spec. Pub. 58, 36 (1928)
iagara Sprayer Co., Middleport, N. Y. Nicotine- Lime Dust <sup>1</sup> .	2.20	2.44	97.80			Cal. Dept. Agr., Spec. Pub. 34, 43 (1922)
iagara Sprayer Co., Middleport, N. Y. Nicotine- Lime Dust	1.10	1.23	98.90			Cal. Dept. Agr., Spec. Pub. 34, 43 (1922)
iagara Sprayer Co., Middleport, N. Y. Nicotine- Lime Dust	1.90	1.81	98.10			Cal. Dept. Agr., Spec. Pub. 34, 43 (1922)
iagara Sprayer Co., Middleport, N. Y. Nicotine- Lime Dust	0.70		99.30			Cal. Dept. Agr., Spec. Pub. 34, 43 (1922)
iagara Sprayer Co., Middleport, N. Y. No. 2 Niagara A-1 Dust	2.70	2.92				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927)

Niagara Sprayer Co., Middleport, N. Y. No. 5 <sup>1</sup> Nico-Dust Mfg. Co., Los Angeles, Cal. Nico-Dust	1.75	1.76	98.25			Cal. Dept. Agr., Spec. Pub. 51, 48 (1925).
No. 3	.095	0.99				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
No. 6	2.00	2.21				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
No. 8	2.75	3.84				Cal. Dept. Agr., Spec. Pub. 75, 36 (1927).
No. 10.  San Jose Spray Mfg. Co., San Jose, Cal. S. J.  Tobacco By-Products & Chem. Corp., Louisville, Ky.	3.50 1.75	4.62 1.82		99.00	86.00	Cal. Dept. Agr., Spec. Pub. 75, 36 (1927). Cal. Dept. Agr., Spec. Pub. 58, 36 (1925).
Black Leaf F 2  Tobacco By-Products & Chem. Corp., Louisville, Ky.	2.00	0.84				N. J. Agr. Expt. Sta., Bull. 441, 9 (1926).
Nico Fume Tobacco Powder.  L. D. Waller Seed Co., Guadalupe, Cal. Nico-Dust <sup>1</sup> Walnut Growers Association. Double Nico-Dust	12.50	12.71 3.20 2.41	94.00			Conn. Agr. Expt. Sta., Bull. 272, 147(1925). Cal. Dept. Agr., Spec. Pub. 34, 42 (1922). Cal. Dept. Agr., Spec. Pub. 34, 44 (1922).
Walnut Growers' Spray Mfg. Co., Los Angeles, Cal. Nico-Dust No. 2 <sup>1</sup>	0.75		99.25			Cal. Dept. Agr., Spec. Pub. 34, 44 (1922).  Cal. Dept. Agr., Spec. Pub. 51, 48 (1925).
Nico-Dust No. 5	1.50 0.56	1.60 0.45	98.50			Cal. Dept. Agr., Spec. Pub. 75, 36 (1927). Cal. Dept. Agr., Spec. Pub. 34, 42 (1922).

 $<sup>^1</sup>$  Lime carrier.  $^2$  Total ash 96.50 per cent; insoluble ash 1.28 per cent; calcium oxide 31.85 per cent; magnesium oxide 23.57 per cent.

	Nicotine		Soap			
Manufacturer or Distributor and Brand	Guaranteed	Found	Guaranteed	Found	Water Found	Publication
	%	%	%	%	%	
Alhambra Spray Co., Alhambra, Cal. Alhambra Nico-Soap		1.68		16.90		Cal. Dept. Agr., Spec. Pub. 51, 49 (1925).
An-Fo Mfg. Co., Los Angeles, Cal. An-Fo Nicotine Spray <sup>1</sup>	0.80	0.95	12.50	9.80	83.40	Cal. Dept. Agr., Spec. Pub. 75, 39 (1927).
An-Fo Mfg. Co., Los Angeles, Cal. An-Fo Nicotine Spray <sup>1</sup>	1.00 0.12		12.50 37.70		77.80 53.06	Cal. Dept. Agr., Spec. Pub. 75, 39 (1927). N. Y Agr. Expt. Sta., Bull. 384, 298(1914).
Juice <sup>3</sup>	1.00	2.11	37.704	38.054		Mich. Agr. Coll. Expt Sta., Spec. Bull. 74, 10 (1915).
The Charles H. Lilly Co., Seattle, Wash. Tobacco Soap	2.00	2.10	6.00	6.30		Ore. Agr. Expt. Sta., Cir. 84, 10 (1927).
Animal Soap	1	0.40				Cal. Dept. Agr., Spec. Pub. 34, 47 (1923).
Chas. C. Navlet Co., San Jose, Cal. Nico Whale Oil Soap		None		20.60		Cal. Dept. Agr., Spec. Pub. 34, 47 (1923).

Pine oil guaranteed 1.50 per cent.
 Fatty acids and resin, found 32.46 per cent; combined alkali, guaranteed 7.80 per cent; found 9.43 per cent; unsaponified, found

5.05 per cent.

<sup>3</sup> Potash guaranteed 7.80 per cent; found 8.04 per cent.

<sup>4</sup> Includes resin.

TABLE XIX.	Nicot	INE SULI	PHATE SO	DLUTION
	Nico	otine		
Manufacturer or Distributor and Brand	Guaranteed	Found	Inert matter, Guaranteed	Publication
Detroit Nicotine Co., Detroit, Mich. "To-Bak-	%	%	%	
ine" Liquid	45.00	46.02		Mich. Agr. Coll. Expt. Sta., Spec. Bull. 74, 9, (1915).
Germain Seed & Plant Co., Los Angeles, Cal. Floral Nicotine  Hall Tobacco Chemical Co., St. Louis, Mo. Hall's Kentucky Tobacco Products Co., Louisville, Ky.	10.00 40.00	11.14 40.61	90.00	Cal. Dept. Agr., Spec. Pub. 34, 52 (1922). Conn. Agr. Expt. Sta., Bull. 272, 147 (1925).
Black Leaf 40	40.00	41.95	60.00	Cal. Dept. Agr., Spec. Pub. 34, 52 (1922).
Nicotine Sulfate Nicotine Mfg. Co., St. Louis, Mo. Nikoteen Nicoteen Mfg. Co., St. Louis, Mo. 30% Nikoteen Nicotine Production Corporation. N. P. C.	40.00 30.00 30.00	$\begin{vmatrix} 40.30 \\ 31.56 \\ 30.77 \end{vmatrix}$	70.00	Ore. Agr. Expt. Sta., Cir. 84, 10 (1927). Cal. Dept. Agr., Spec. Pub. 34, 52 (1922). N. J. Agr. Expt. Sta., Bull. 441, 9 (1926).
Nicotine Sulphate	40.00 25.00 10.00	40.64 45.30 11.19		Cal. Dept. Agr., Spec. Pub. 75, 34 (1927). N. Y. Agr. Expt. Sta., Bull. 348, 94 (1912). N. Y. Agr. Expt. Sta., Bull. 384, 297 (1914).
Purity Chemical Products Co. Santa Rosa Ca. Nicota <sup>1</sup>	2.00	2.33 26.01		Cal. Dept. Agr., Spec. Pub. 75, 34 1927). U. S. D. A Bur. Chem., Bull. 68, 47 (1902).
tine Tobacco By-Products & Chemical Corp., Inc., Louisville, Ky. Black Leaf 40	10.00	9.92		N. Y. Agr. Expt Sta., Bull. 384, 297 (1914). Conn. Agr Expt. Sta., Bull. 242, 154 (1922) and Bull. 272, 147 (1925).
Tobacco By-Products & Chemical Corp., Inc., Louisville, Ky. 40% Nicotine Sulphate	40.00	40.42		N. J. Agr. Expt. Sta., Bull. 459, 7 (1927).
Tobacco By-Products & Chemical Corp., Inc., Louisville, Ky. Nico Fume (Liquid)	40.00	42.90		Conn. Agr. Expt. Sta. Bull. 272, 147, (1925).

 $<sup>^{1}</sup>$  Water guaranteed 85.00%; found 96.10%.

BULLETIN 300

TABLE XX. NICOTINE-SULPHUR DUSTS

	Nicotine		Free	Free Sulphur			Fineness, finer than			
Manufacturer or Distributor and Brand	Guaranteed	Found	Guaranteed	Found	Combined sulphur,	Inert matter, Guaranteed	100 mesh	200 mesh	Publication	
American Milling & Warehouse Co.	%	%	1 %	%	%	%	1 %	1 %		
American Milling & Warehouse Co. (defunct) Impco		0.29 0.96		49.98 39.39			78.00	52.00	Cal. Dept. Agr., Spec. Pub. 66, 24 (1926). Cal. Dept. Agr., Spec.	
California Associated Raisin Co., Fresno, Cal <sup>1</sup> .  The California Sprayer Co., Calispray		0.88		52.44	0.31				Pub. 34, 45 (1922). Cal. Dept. Agr., Spec. Pub. 34, 45 (1922).	
Dust No. 1	1.70	1.30	72.00	73.85					Cal. Dept., Agr., Spec. Pub. 75, 37 (1927).	
Dust No. 2	2.70	2.43	70.00	67.10					Cal. Dept. Agr., Spec. Pub. 75, 37 (1927).	
No. 3	3.30 1.60			64.60 66.55		3.40			Cal. Dept. Agr., Spec. Pub. 75, 37 (1927). Cal. Dept. Agr., Spec.	
Dosch Chemical Co., Louisville, Ky. Dosch	1.80	1.81	58.00	60.16		40.20			Pub. 34, 46 (1922). Cal. Dept. Agr., Spec.	
Dosch Chemical Co., Louisville, Ky. Dosch	2.00	4.03		54.58					Pub. 34, 46 (1922). Cal. Dept. Agr., Spec.	
Dosch Chemical Co., Louisville, Ky. Dosch	2.00	1.55	59.00	58.00		39.00			Pub. 34, 46 (1922). Cal. Dept. Agr., Spec.	
F. A. Frazier Co. Frazier's 5X <sup>3</sup>	1.50	1.52	40.00	35.70					Pub. 34, 46 (1922). Cal. Dept. Agr., Spec. Pub. 75, 37 (1927).	

L. O. Haupt, Hanford, Cal. 1		0.97		68.10					Cal. Dept. Agr., Spec. Pub. 34, 45 (1922).
Hood River Spray Co., Hood River, Ore. N. S. P. Dust		1.89 1.31		64.00 65.65					Ore. Agr. Expt. Sta., Cir. 84, 11 (1927). Cal. Dept. Agr., Spec. Pub. 34, 46 (1922).
Niagara Sprayer Co., Middleport, N. Y. Niagara Dust Mixture with Sulphur and Nicotine No. 5 Niagara Sprayer Co., Middleport, N. Y. Niagara No. 7	1.75 2.25		48.00 58.00		4.16	39.75			Cal. Dept. Agr., Spec. Pub. 75, 37 (1927). Cal. Dept. Agr., Spec. Pub. 51, 49 (1925).
Niagara Sprayer Co., Middleport, N. Y. Niagara Special Mixture 154A Niagara Sprayer Co., Middleport, N. Y.	1.75	2.32	58.00	62.02			95.00	73.00	Cal. Dept. Agr., Spec. Pub. 66, 24 (1926). Cal. Dept. Agr., Spec.
Niagara Special Mixture S155A Niagara Sprayer Co., Middleport, N. Y.	2.25	3.25	38.00	41.03			92.00	72.00	Pub. 66,24 (1926)
Niagara Sulphur with 7% Nicotine Solution 1	2.55	2.59	66.00	68.57	1.40	.,.,			Cal. Dept. Agr., Spec. Pub. 51, 49 (1925).
Niagara Sulphur with 10% Nicotine Solution <sup>1</sup>	3.75	3.27	63.00	64.13	2.14				Cal. Dept. Agr., Spec. Pub. 51, 49 (1925). Cal. Dept. Agr., Spec.
No. 10 <sup>1</sup>	3.25	4.22	58.00	56.29	5.74	39.75			Pub. 51, 49 (1925). Cal. Dept. Agr., Spec.
CalSan Jose Spray Mfg. Co., San Jose, Cal	3.50		50.00						Pub. 34, 45 (1922). Cal. Dept. Agr., Spec. Pub. 75, 37 (1927).
Standard Chemical Co., Oakland, Cal. Sulfur-Nicotine Compound <sup>4</sup> L. D. Waller Seed Co., Guadalupe, Cal	1.20 2.84		43.00	47.00 66.09		49.80			Cal. Dept. Agr., Spec. Pub. 34, 48 (1922). Cal. Dept. Agr., Spec. Pub. 34, 46 (1922).

Lime carrier.
 Calcium hydroxide guaranteed 24.00 per cent.
 Sulphur classed as inert matter on label.
 Soap guaranteed 6.00 per cent; found 4.60 per cent.

Free Sulphur.	Pound Pound Pound Pound Thert matter, Guaranteed 100 mesh 200 mesh Publication	% % % % %	50.00 Cal. Dept. Agr	5 (19 Agr., 4 (19	45.00 46.36   90.00   72.00   Pub. 58, 35 (1295).	40.00 32.78 90.50 60.50 Pub. 66, 24, (1926).	40.00 37.18 86.00 45.00 Pub. 66, 24 (1926).	48.10 49.99 Cal. Dept. Agr., Spec. Pub. 34, 45 (1922).	60.00 50.33 38.50 Cal. Dept. Agr., Spec. Pub. 34, 45 (1922).	50.00 50.26 44.00 Cal. Dept. Agr., Spec. Pub. 34, 45 (1922).	En on 14 on last. Agr., Spec
Nicotine	beetnaranto	% %	0.96 1.01	2.00 1.86 4	1.75 1.83 4	2.00 2.20 4	2.75 2.99 4	3.70 2.78 4	1.50 1.16 6	1.96 1.80 5	0 08 1 09 2
	Manufacturer or Distributor and Brand.	To be to be the second of the	Walnut Growers' Association <sup>1</sup>	Walnut Growers' Spray Mfg. Co., Los Angeles, Cal. Nico Garden Dust	55		Sulphur Dust No. 8	Spray Mfg Co. Los	Spray Mfg Co. 1 os	Mfg Co Los	Spiral title: Co., Los

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

Nicotine-Sulfur Paste. (STANDARD CHEMICAL CO., OAKLAN.		December
Nicotine	Guaranteed. 1.20 43.00 6.00 49.80	Found. 1.56 45.25 0.25 7.07
Nicotine Sulphate Dust.		
See "Nicotine Dusts."		
Nicotine Sulphate Solution. See Table XIX.		
Nicotine-Sulphur Dusts. See Table XX.		
Nico-Tone.		
(THE KIL-TONE CO., VINELAND, See also "Green Cross Nico-Tone."	N. J.) Guaranteed.	Found.
Nicotine	1.25	0.67
Nico-Tone B.		
(THE KIL-TONE CO., VINELAND, Nicotine	N. J. Guaranteed. 3.00	Found. 2.17
N. J. Agr. Expt. Sta., Bull. 407, 10 (1924).		
Nico Whale Oil Soap.		
See "Nicotine Soaps."		
Nictone.		
(PARKE, DAVIS & CO., DETROIT, Found: Nicotine 9.87 per cent.  N. Y. Agr. Expt. Sta., Bull. 384, 297 (1914).	місн.)	
Nikoteen. See "Nicotine Sulphate Solution."		
Nikoteen Aphis-Punk. (NICOTINE MFG. CO., ST. LOUIS,	мо.)	

(NICOTINE MFG. CO., ST. LOUIS, MO.)

Guaranteed: Nicotine 700 grains per package. Found: Nicotine 7.96 per cent; 239.5 grains per package.

N. Y. Agr. Expt. Sta., Bull. 348, 94 (1912).

## 90-10 Dusting Mixture.

See "Niagara 90-10 Dusting Mixture."

## No-Mor-Fly

(NO-MOR PRODUCTS CO., LOS ANGELES, CAL.)

Found: Oil 100.00 per cent; water trace, ash none to 0.003 per cent. Wintergreen and cassia present.

Cal. Dept. Agr. Spec. Pub. 51, 56 (1925).

## COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

## North State Insecticide.

(NORTH STATE INSECTICIDE CO., RICHMOND, VA.)

Found: Insoluble in hydrochloric acid, 24.91 per cent; cupric oxide 12.43 per cent; iron and aluminum oxides 13.98 per cent; calcium oxide 9.07 per cent; arsenious oxide 20.32 per cent; sulphur trioxide 12.95 per cent; acetic acid and water by difference, 6.34 per cent.

U. S. D. A., Bur. Chem. Bull. 68, 27 (1902).

#### Nox-A-Vermin.

(COAST CHEMICAL CO., LOS ANGELES, CAL.)

Phenols	22.40

Cal. Dept. Agr., Spec. Pub. 34, 36 (1923).

#### Noxem Squirrel and Gopher Poison.

See "Strychnine Preparations."

## Noxicide.

See "Phenol Soap Solutions."

#### Nox-Weed.

(C. W. MORRELL, LOS ANGELES, CAL.)

Found: Arsenious oxide 16.74 per cent; sodium hydroxide 26.80 per cent; sodium carbonate 42.00 per cent; sodium chloride small amount. chromic oxide 1.93 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 20 (1923).

## Nox Worm Dry Powdered.

(A. F. DINGLEY, SOUTH PORTLAND, ME.)

	Guaranteed.	Found.
Total arsenic, metal	16.00	19.82
Water-soluble arsenic, metal	5.00	0.25
Copper	4.00	5.30
Maine Agr. Expt. Sta., Official Inspections 122,		

#### N. S. P. Dust.

See "Nicotine-Sulphur Dusts."

## No. 1 Niagara Entodust.

See "Arsenate of Lead."

#### No. 3 Dust.

(LEGGETT & BRO., INC., NEW YOR	K, N. Y.)	
	Guaranteed.	Found.
Total arsenic, metal	16.00	16.30
Water-soluble arsenic, metal	0.50	0.04
Copper		4.85

N. J. Agr. Expt. Sta., Bull. 441, 12 (1926).

#### No. 3 Dust Mixture.

See "Nicotine Dusts."

## No. 4 Nicotine-Lime Dust.

#### No. 5 Nicotine Dust.

### No. 6 Dust Mixture with Nicotine.

## No. 10 Dust Mixture with Nicotine.

See "Nicotine Dusts."

#### No. 6 All-in-One Dust.

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y.)

	Guaranteed.	Found.
Nicotine	1.00	1.14
Total arsenic, metal	1.76	2.10
Water-soluble arsenic, metal	0.50	0.26
Lead oxide		6.45
Sulphur	75.00	76.16
Cal. Dept. Agr. Spec. Pub. 75, 38 (1927).		

## No. 6 Disinfectant.

See "Phenol Soap Solutions."

#### Nu Rex O.

(CALIFORNIA REX SPRAY CO., SACRAMENTO, CAL.)

	Guaranteed.	Found.
Arsenic oxide	5.55	6.62
Copper	12.70	11.50
Ove Agr Expt Sta Cir 84 13 (1927)		

#### Nyal Roach Powder.

(NYAL CO., SAN FRANCISCO, CAL.)

	Guaranteed.	Found.
Talc	55.00	46.32
Cal Debt Agr Spec Pub 75 63 (1927)		

#### 0

#### Oakland Ant Co.

(OAKLAND ANT EXTERMINATING CO., OAKLAND, CAL.)

Found: Arsenious oxide 0.29 per cent; arsenic, metal 0.22 per cent. Cal. Dept. Agr., Spec. Pub. 66, 19 (1926).

## Oil, Fish.

The U.S. Department of Agriculture recommends the following specifications for fish oil for spraying purposes:

The oil should be light pressed and unadulterated, and have the following specifications.

Specifications.	
Saponification value	190 to 193
Iodine value	$\dots$ 139 to 193
Specific gravity at 15° C	$\dots 0.927 \text{ to } 0.933$
Free fatty acid	Less than 5 per cent.
J. S. D A Debt Bull 1439 (1926)	

## Oil, Menhaden, Light Pressed

(MANUFACTURER UNKNOWN)

Found: Specific gravity, 15° C., 0.932; saponification number 187; iodine number 173; free fatty acids 3.46 per cent. Conn. Agr. Expt. Sta., Sample 421.

Oils, Mineral.

See Table XXI.

#### Oil Emulsions, Mineral.

See Table XXII.

Manufacturer or Distributor and Brand	Oil	Baumé gravity, degrees	Specific gravity, 20° C.	Saybolt Viscosity, 100° C., seconds	Unsulphonated	Evaporation, 60-65° C.1	Cold Test	. Publication
A - : : 1010 C - T - : 0.1 A	%				%			
Associated Oil Co., San Jose, Cal. Avon Brown Neutral	100.00	21.80			63.20			Cal. Dept. Agr., Spec. Pub. 75, 65 (1927)
Associated Oil Co., San Jose, Cal. Crude Oil <sup>2</sup>		18.40 19.50						Cal. Dept. Agr., Spec. Pub. 34, 39 (1922) Cal. Dept. Agr., Spec. Pub. 51, 57 (1925)
G. F. Doleshal Products Co., Pasadena, Cal. Sure-Shot <sup>4</sup>	99.805	39.60			75.90			Cal. Dept. Agr., Spec. Pub. 51, 57 (1925)
Hackney Chemical Co., Parlier, Cal. Di-Fli Home Spray	100.00	39.30						Cal. Dept. Agr., Spec. Pub. 58, 48 (1925).
Hockwald Chemical Co., San Francisco, Cal. Fli-Mo-Cide	100.006	44.10				. ,		Cal. Dept. Agr., Spec. Pub. 58, 48 (1925).
Lacko Specialty Co., Inc., San Francisco, Cal. Fly Croke	100.00	43.50						Cal. Dept. Agr., Spec. Pub. 58, 48 (1925)
Lacko Specialty Co., Inc., San Francisco, Cal. Lilacko Spray	99.80	43.10				,		Cal. Dept. Agr., Spec. Pub. 58, 48 (1925)
Cal. Chemco Insect and Germ Destroyer <sup>7</sup>	100.00	50.50	0.9201	222	$\begin{bmatrix} 52.10^8 \\ 56.20 \end{bmatrix}$	3.22	+7°F.	Cal Dept. Agr., Spec. Pub. 51, 56 (1925). Conn. Agr. Expt. Sta., Sample 6729.
Manufacturer unknown. No. 14 Mineral Oil				246	97.80	1.71		Conn. Agr. Expt. Sta., Sample 8213.
Oil				90	92.80	1.38		Conn. Agr. Expt. Sta., Sample 8214.

Manufacturer unknown. No. 16 Mineral Oil				331	96.40	1.68		Conn. Agr. Expt. Sta., Sample 8215.
Manufacturer unknown. No. 14 Spray Oil		1	0.9306	604	56.00	1.77	+27°F.	Conn. Agr. Expt. Sta., Sample 6727.
Oil			0.9280	473	56.40	1.69	+16°F.	
Mystic XX Spray <sup>9</sup> Standard Oil Co. of New York. Socony	100.00				77.40			Cal. Dept. Agr., Spec. Pub. 58, 43 (1925)
Motor Oil—heavy			$\begin{bmatrix} 0.9148 \\ 0.9073 \end{bmatrix}$	915 291			+18°F.	Conn. Agr. Expt. Sta., Sample 6725. Conn. Agr. Expt. Sta., Sample 6726.
Motor Oil—light Standard Oil Co. of New York, Socony Motor Oil—medium			0.9129	443	52.40			
Standard Oil Co. of New Jersey. Acto. Standard Oil Co. of New Jersey. Marcol			0.0120	173 96	96.00 96.00			Conn. Agr. Expt. Sta., Sample 9369. Conn. Agr. Expt. Sta., Sample 9371.
Standard Oil Co. of New Jersey. Wyrol Sun Oil Co., Philadelphia, Pa. XXX				140	96.00		Same Company and the	G 1 0 0 0 1 00F0
Pale Spindle OilSun Oil Co., Philadelphia, Pa. LIV			0.9257	169	55.20			
Newport Pale			0.9307	224			+16°F.	
New York Red			0.9360	449	53.20	1.77	+28°F.	Conn. Agr. Expt. Sta., Sample 6722.

¹See Scott, "Standard Methods of Chemical Analysis" 1st ed. p. 575.
²Light oils 4.00 per cent; medium oils 31.25 per cent; heavy oils 64.75 per cent.
³Distillation test: initial boiling-point 105° C.; distilling 105-200° C., 5.19 per cent; 200-250° C., 9.86 percent; 250-305° C., 15.30 per cent; residue 69.44 per cent. Trace of water. Asphalt.
⁴Guaranteed active ingredients 100.00 per cent.

\*Weter 0.2 per cent\*

<sup>\*</sup>Guaranteed active ingredients 100.00 per cent.

\*Water 0.2 per cent.

\*Ash 0.002 per cent.

\*Sample is a mixture of kerosene and gasoline with a little oil of wintergreen.

\*Fraction boiling above 150° C.

\*Precipitated chalk, guaranteed 0.34 per cent; none found. Water, trace.

Evaporation 60-65° C.1  Publication  Publication		2.22 +27°F. Conn. Agr. Expt. Sta., Sample 6721.	5.08   -27°F.   Conn. Agr. Expt. Sta., Sample 6714.	2.44 -18°F. Conn. Agr. Expt. Sta., Sample 6713.	1.69 +14°F. Conn. Agr. Expt. Sta., Sample 6710.	58   +43°F.   Conn. Agr. Expt. Sta., Sample 6711.	71 +85°F. Conn. Agr. Expt. 88 +21°F. Conn. Agr. Expt.	05 +18°F. Conn. Agr. Expt. Sta., Sample	93 +41°F. Conn. Agr.	+21°F.   Conn. Agr. Expt. Sta., Sample	Cal. Dept. Agr., Spec. Pub. 58, 48 (1925).
DetanondluanU	%	53.40 2.	58.00 5.	76.60 2.	77.80 1.	75.00 1.	54.20	54.	54.00 55.40	58.80 0.	
Saybolt Viscosity, 100° C., seconds		332	103	196	434	652	29110	46311	70413	213	i
Specific gravity, 100° C., seconds		0.9347	0.9254	0.9395	0.9450	0.9518	0.9705 907,	0.9342	0.9281 $0.9230$	0.9070	1:
Baumé gravity, degrees		:		i	:	:					42.20
liO	%							:			100.0014 42.20
Manufacturer or Distributor and Brand	Sun Oil Co., Philadelphia, Pa. LXII	Toledo Red	D <sub>2</sub> X			. :×	100	The Texas Co. Altair Oil	The Texas Co. No. 776 Oil	Tidewater Oil Co. Tycol 122	Van Winkle Coffee Co., San Francisco, Cal. Old Rip Van Winkle Fly Spray

<sup>1</sup>See Scott "Standard Methods of of Chemical Analysis" 1st ed. p. 575. <sup>10</sup>Guaranteed 300. <sup>10</sup>Guaranteed 500.

water, trace.

O. K. Ant Paste.

(WESTERN WHOLESALE DRUG CO.)

Guaranteed: Arsenic 8.39 per cent. Found: Arsenious oxide 11.33 per cent; arsenic, metal 8.58 per cent; invert sugar 21.45 per cent; sucrose 17.27 per cent; water 33.02 per cent. Cal. Dept. Agr., Spec. Pub. 58, 17 (1925).

O-Kay Gopher Poison.

Okay Poisoned Wheat.

O.K. Squirrel Poison.

See "Strychnine Preparations."

Old Rip Van Winkle Fly Spray.

See "Oils, Mineral."

Ongman's Dip and Disinfectant.

See "Phenol Soap Solutions".

Orange Oil Emulsion.

Orchard Dormant Soluble Oil.

Ortho Apricot Oil.

Ortho Crude Oil Emulsion.

See "Oil Emulsions, Mineral."

Ortho Double Nicotine Dust.

See "Nicotine Dusts."

Ortho Garden Dust.

(CALIFORNIA SPRAY CHEMICAL CO., WATSONVILLE, CAL.)

	Guaranteed.	Found.
Nicotine	2.00	2.11
Arsenic oxide	4.00	3.82
Copper		2.54
Cal Debt Agy Shee Pub 75 38 (1027)		

Cal. Dept. Agr., Spec. Pub. 75, 38 (1927).

Ortho Gopher Poison.

See "Strychnine Preparations."

Ortho Kleenup.

Orthol.

See "Oil Emulsions, Mineral."

Ortho Liquid Soap.

See "Soaps."

Ortho Miscible Oil.

Ortho Orange Oil Emulsion.

See "Oil Emulsions, Mineral."

Ortho Paradichlorobenzene.

See "Paradichlorbenzene."

		Oil			Wa		Soap	Constants of Separated Oil				
Manufacturer or Distributor or Brand	Guaranteed	Found	Phenols—Found	Soap—Found	Guaranteed	Found	Ash other than Ash—Found	Baumé Gravity, degrees	Saybolt Viscosity, 100°F.,	Unsulpho- nated	Publications	
	%	%	%	%	%	%	%			%		
J. Willis Adriance, Stockton, Cal.		94.00	5.90								Cal. Dept. Agr., Spec.	
Germo		60.40	10.20	4.90			1.31				Pub. 34, 39 (1923). Cal. Dept. Agr., Spec. Pub. 34, 36 (1923).	
American Soap Co., Oakland, Cal. American Jazz Spray <sup>2</sup>	70.00	66.00		4.05				28.00			Cal. Dept. Agr., Spec. Pub. 34, 37 (1923).	
Anaheim Spray Chemical Co., Anaheim, Cal. Blue Ribbon Orchard Spray		$79.50^{3}$		4.00	20.00	16.20		33.05	53	86.00	Cal. Dept. Agr., Spec. Pub. 75, 51 (1927).	
Anaheim Spray Chemical Co., Anaheim, Cal. E. M. F. Orchard Spray		68.15		6.20		24.70		37.10		66.75	Cal. Dept. Agr., Spec. Pub. 66, 32 (1926).	
Associated Oil Co., San Jose, Cal. Petrotine	83.00	86.30.				12.60		29.80		87.60	Cal. Dept. Agr., Spec. Pub. 75, 48 (1927).	
Balfour, Guthrie & Co., San Francisco, Cal. Orchard Dormant Soluble Oil.		81.95	3.684	5.28	13.00	8.00	0.81	20.80		52.00	Cal. Dept. Agr., Spec. Pub. 51, 37 (1925).	
Balfour, Guthrie & Co., San Francisco, Cal. Soluble Oil Spray <sup>5</sup> Balfour, Guthrie & Co., Los Angeles,	87.00	86.37	2.80	4.03			1.57				Cal. Dept. Agr., Spec. Pub. 34, 36 (1923).	
Cal. Universal Brand Neutral Emulsion	70.00	71.20				26.60		22.50		58.80	Cal. Dept. Agr., Spec. Pub. 75, 48 (1927).	
Balfour, Guthrie & Co., Los Angeles, Cal. Universal Brand Triona	70.00	75.40				23.30		31.95	58	86.80	Cal. Dept. Agr., Spec. Pub. 75, 48 (1927).	

Balfour, Guthrie & Co., Los Angeles,		86.603					1000				Cal. Dept. Agr., Spec.
Cal. Universal Brand Triumph		$ 86.60^3 $	4.40	3.90	$ 13.00^3 $	$  5.00^{3}$		28.40			Pub. 75, 53 (1927).
Balfour, Guthrie & Co., Los Angeles,											Cal. Dept. Agr., Spec.
Cal. Universal Brand Triumph	$ 87.00^3 $	88.403	4.80	3.50		$3.90^{3}$		29.10			Pub. 75, 53 (1927).
Balfour, Guthrie & Co., Los Angeles,											Cal. Dept. Agr., Spec.
Cal. Universal Dormant Soluble Oil.		85.703	1.80	7.10	13.003	$3.80^{3}$		19.90			Pub. 75, 53 (1927).
Balfour, Guthrie & Co., San Francisco,											Cal. Dept. Agr., Spec.
Cal. Universal Mealy Bug Spray Oil		75.40	7.80	9.40	13.00	7.00		23.20		44.20	Pub. 66, 32 (1926).
Balfour, Guthrie & Co., Lindsay, Cal.									DESCRIPTION OF THE PROPERTY OF		Cal. Dept. Agr., Spec.
Universal Medol Emulsion	70.00	66.40				30.20		28.20		52.70	Pub. 66, 30 (1926).
Bean Spray Pump Co., San Jose, Cal.											01010
Bean's Bug-Go, Crude	$ 75.00^{3}$	$77.05^3$		1.39		21.60		22.55			Pub. 66, 32 (1926).
David Burke & Co. Bejo Orchard Spray		89 40		2.00	30.00	6.80		40 10		76 90	Cal. Dept. Agr., Spec.
											Pub. 66, 32 (1926).
Bean's Bug-Go, Crude David Burke & Co. Bejo Orchard Spray  David Burke & Co. Pacific Orchard		77.40	5.80	4.90	25.00	11.40		39 10		70 60	Cal. Dept., Agr., Spec.
Spray			0,00					00.10			Pub. 66, 32 (1926).
California Pest Control Co. Calpest											Cal. Dent Agr Spec
Summer Spray	80 003	85 853				13 15		31 10			Pub. 75, 48 (1927).
California Rex Spray Co., Benicia, Cal.	00.00	00.00				10.10		01.10			Cal. Dept. Agr., Spec.
Rex Emulso	75 008	85.85 <sup>3</sup> 75.70 <sup>3</sup>		1 40		22 40		24 20		19.40	Pub. 66, 33 (1926).
California Rex Spray Co., Benicia, Cal.	1.0.00	10.10		1.10		10	1 10	21.20		13.10	Cal. Dept. Agr., Spec.
Rex Miscible Oil, Heavy		80.403	4 40	6.50	8 003	8 303		27.90			Pub. 75, 53 (1927).
California Rex Spray Co., Benicia, Cal.					The state of the s		100000000000000000000000000000000000000				
Rex Miscible Oil, Light		90 303	2 70	6 20	8 003	0 403		20 20			Cal. Dept. Agr., Spec.
		80.303	5.10	0.20	0.00	3.40		29.20			
California Rex Spray Co., Benicia, Cal.		74.703		1 606		22 058	100	30.15		62 00	Cal. Dept. Agr., Spec.
Vulture Oil	15.00	14.70	• • • •	1.00		44.95°		30.13		63.00	Pub. 75, 51 (1927).
								I constant		100	

<sup>&</sup>lt;sup>1</sup>Inert matter 24.60 per cent.

<sup>2</sup>Soap guaranteed 4.00 per cent; inert matter guaranteed 26.00 per cent; found 30.00 per cent; Baumé gravity of oil, guaranteed 30.00.

Rosin present.

<sup>3</sup>Per cent by volume.

<sup>4</sup>Phenols guaranteed 87.00 per cent. This emulsion contains rosin.

<sup>5</sup>Inert matter guaranteed 13.00 per cent; found 6.00 per cent. Rosin present.

<sup>6</sup>Guaranteed 1.00 per cent.

	Oi	1	ਧ		Wat	er	Soap		onstants parated		
Manufacturer or Distributor or Brand	Guaranteed	Found	Phenols—Found	Soap—Found	Guaranteed	Found	Ash other than Ash—Found	Baumé Gravity, degrees	Saybolt Vis- cosity,100°F., seconds	Unsulpho- nated	Publications
	%	%	%	%	%	1 %	%			%	
California Spray Chemical Co., Watsonville, Cal. Kleenup Oil		77.10				21.80	0.07	22.70		45.50	Cal. Dept. Agr., Spec. Pub. 66, 30 (1926).
California Spray Chemical Co., Watsonville, Cal. Ortho-Apricot Oil	75.00	74.00				26.00		29.10		59.80	Cal. Dept. Agr., Spec. Pub. 66, 30 (1926).
California Spray Chemical Co., Watsonville, Cal. Ortho-Crude Oil Emulsion	75.00 <sup>3</sup>	75.603		••••		22.70		22.10		38.80	Cal. Dept. Agr., Spec. Pub. 75, 49 (1927). Conn. Agr. Expt. Sta.,
California Spray Chemical Co., Watsonville, Cal. Ortho-Kleenup Grade A <sup>7</sup>							0.29		102	53.20	Sample 8861. Conn. Agr. Expt. Sta.,
California Spray Chemical Co., Watsonville, Cal. Ortho Kleenup Oil		112						,	102	55.60	Sample 8045.
California Spray Chemical Co., Watsonville, Cal. Ortho-Kleenup Oil, Grade B.	75.003	78.203				21.65		21.85		58.40	Cal. Dept. Agr., Spec. Pub. 75, 48 (1927). Cal. Dept. Agr., Spec.
California Spray Chemical Co., Watsonville, Cal. Orthol	75.003	77.503	none			$21.80^{3}$	0.13	40.60		94.50	Pub. 58, 42 (1925). Cal. Dept. Agr., Spec.
California Spray Chemical Co., Watsonville, Cal. Orthol Concentrate	80.003	79.603				18.70		30.70		86.80	Pub. 75, 49 (1927). Conn. Agr. Expt. Sta.,
California Spray Chemical Co., Watsonville, Cal. Orthol-K (Heavy)									111	80.80	Sample 9743. Cal. Dept. Agr., Spec.
California Spray Chemical Co., Watsonville, Cal. Ortho Miscible Oil	85.003	87.253	2.72	6.14		3.65		21.55		50.90	Pub. 66, 33 (1926).
	-					-		-		The state of the	

California Spray Chemical Co., Watsonville, Cal. Ortho Orange Oil Emulsion <sup>8</sup>					25 00	24 70	0.48	40.30		96.70	Cal. Dept. Agr., Spec.
California Spray Chemical Co., Wat- * sonville, Cal. Ortho Orange Oil Emul-					25.00	24.70	0.40	40.50			0.1 D
sion <sup>8</sup>	75.00	80.30			25.00	18.20	0.51	40.80		95.90	Cal. Dept. Agr., Spec. Pub. 51, 32 (1925).
sonville, Cal. Ortho Penetrating Oil	75 000	70 702	1 00			05 10		25 40			Cal. Dept. Agr., Spec.
Spray		$72.50^{3}$								53.80	Pub. 58, 44 (1925). Conn. Agr. Expt. Sta.,
sonville, Cal. Volck Concentrate California Spray Chemical Co., Wat-		83.55			E CONTRACTOR					• • • • •	Sample 2504. Conn. Agr. Expt. Sta.,
sonville, Cal. Volck <sup>9</sup>	83.00	79.19				1 30 THE CO.	The second second		10 10 10 10	94.40	Sample 8934. Conn. Agr. Expt. Sta.,
sonville, Cal. Volck Emulsion California Spray Chemical Co., Wat-									108	93.60	Sample 8047. Cal. Dept. Agr., Spec.
sonville, Cal. Volck Light Carco Spray Co., Taçoma, Wash. Oil		81.803						34.00			Pub. 75, 49 (1927). Ore. Agr. Expt. Sta.,
Emulsion		86.60									Cir. 84, 13 (1927). Conn. Agr. Expt. Sta.,
Keresol		80.0010	4.15	12.00							Bull. 258, 373 (1924). Conn. Agr. Expt. Sta.,
Soluble Spray Oil <sup>11</sup>		85.10									Bull. 272, 148 (1925).
son's Poultry Spray											Cal. Dept. Agr., Spec. Pub. 51, 45 (1925).
Zeno Miscible Oil Spray <sup>13</sup>		84.22	4.00	4.78			0.94				Cal. Dept. Agr., Spec. Pub. 34, 36 (1923).

 $<sup>^3</sup>$ Per cent by volume.  $^7$ Total nitrogen, 0.15 per cent; ammonia (NH<sub>3</sub>) 0.12 per cent; casein 0.19 per cent. This is an ammonium caseinate emulsion.  $^8$ Ash contains copper and calcium.  $^9$ Ammonia (NH<sub>3</sub>), found 0.06 per cent.

<sup>&</sup>quot;Specific gravity, 25° C., 0.9333; unsaponifiable matter 85.10 per cent. Rosin present. A light petroleum oil containing a sodium soap. 

<sup>12</sup>Guaranteed 15.00 per cent. Sample does not contain rosin.

<sup>13</sup>Inert matter guaranteed 16.00 per cent, found 6.00 per cent.

	(	Di1	þı		Wat	er	Soap		onstants parated		
Manufacturer or Distributor or Brand	Guaranteed	Found	Phenols—Found	Soap—Found	Guaranteed	Found	Ash other than Ash—Found	Baumé Gravity, degrees	Saybolt Viscosity, 100°F., seconds	Unsulpho- nated	Publications
	%	%	%	%	%	1 %	1 %			%	
Eastbay Chemical Co., Emeryville, Cal. Zeno Miscible Oil Spray <sup>14</sup>		83.00	4.20								Cal. Dept. Agr., Spec. Pub. 34, 36 (1923).
Herbicide Chemical Laboratories. Fumispray	54.57	53.99		$3.13^{15}$	45.29	42.20	0.01			65.30	Cal. Dept. Agr., Spec. Pub. 51, 35 (1925).
Herbicide Chemical Laboratories. Fumispray		55.76		3.54	40.00	39.00	0.08			1.72	Cal. Dept. Agr., Spec. Pub. 51, 35 (1925).
Hood River Spray Co., Hood River, Ore. Oil Emulsion	84.00	90.70									Ore. Agr. Expt., Sta. Cir. 84, 13 (1927).
Hood River Spray Co., Hood River, Ore. Oil Emulsion	75.00	73.80									Ore. Agr. Expt. Sta., Cir. 84, 13 (1927).
Leffingwell Rancho Co., Whittier, Cal. XXX Heavy Emulsion	85.003	87.803				12.50		38.10	69	98.40	Cal. Dept. Agr., Spec. Pub. 75, 49 (1927).
Leffingwell Rancho Co., Whittier, Cal. XXX Medium Emulsion	85.003	87.703				12.40		33.50	63	88.80	Cal. Dept. Agr., Spec. Pub. 75, 49 (1927).
Los Angeles Chemical C., Los Angeles, Cal. Mission Brand Insecto No. 1	90.003	$91.10^{3}$				8.42		26.40	75	70.00	Cal. Dept. Agr., Spec. Pub. 75, 49 (1927).
Manufacturer unknown. Sulco-V. B.			3.97								Conn. Agr. Expt. Sta., Bull. 242, 162 (1922).
McClure Chemical Laboratories. Calox		89.20	3.33	4.00	16.00	2.87		29.00		56.50	Cal. Dept. Agr., Spec. Pub. 58, 44 (1925).
McClure Chemical Laboratories. Renol (Miscible Oil)		84.60	3.50	8.50	6.00	2.40		38.60		69.60	Cal. Dept. Agr., Spec. Pub. 58, 44 (1925).

G. P. McNear Co., Petaluma, Cal. Four and One Spray		92.70	7.2016			0.10	0.01		•••	74.20	Pub. 51, 45 (1925).
G. P. McNear Co., Petaluma, Cal. McNear's Poultry House Spray Michel & Pelton Co., Mapco Miscible Oil		94.24 83.60	$\begin{bmatrix} 5.76^{17} \\ 5.20 \end{bmatrix}$	7:05	10.00	trace 2.50		21.05			Cal. Dept. Agr., Spec. Pub. 75, 56 (1927). Cal. Dept. Agr., Spec. Pub. 75, 53 (1927).
Chas. C. Navlet Co., Inc., San Francisco, Cal. Navco Crude Oil Emul-											Cal Dont Am Span
sion <sup>18</sup>		66.12			18 17 1 T. 32			1		31.70	Cal. Dept. Agr., Spec. Pub. 51, 33 (1925). Cal. Dept. Agr., Spec.
cisco, Cal. Navco Miscible Oil		85.403	5.20	6.30	$12.00^{3}$	$2.80^{3}$		22.30			
Peerless Spray Chemical Co., Covina, Cal. Peerless Spray Emulsion	85.003	85.273						27.77	85	86.40	
Perfecto Spray Mfg. Co., Los Angeles, Cal. Perfecto Spray Oil	83.003	85.733				13.98		31.33	45	74.30	
	86.00	88.70	present			11.17	1.22	2			
Rochester Rex Co., Rochester, N. Y. Emulso	80 116								157	61.00	Conn. Agr. Expt. Sta., Sample 8235.
R. R. Rogers Chemical Co., "Cee-Pee- Dee" Carbolized Petroleum Distillate		79.78	19.3220		2.00	0.80	0.06	24.90		47.50	
San Jose Spray Mfg. Co., San Jose, Cal. Cot Oil		79.903	0.10		30.00	18.50	0.21	30.60		62.20	Cal. Dept. Agr., Spec. Pub. 66, 31 (1926).
San Jose Spray Mfg. Co., San Jose, Cal. Crude Oil Emulsion		81.193	0.20			19.01	0.44	19.30			Cal. Dept. Agr., Spec. Pub. 66, 31 (1926).
Oracio or Birtholt			0.20					10.00			1 45. 50, 51 (1520).

<sup>&</sup>lt;sup>3</sup>Per cent by volume.

<sup>14</sup>Inert matter guaranteed 16.50 per cent; found 12.50 per cent.

<sup>15</sup>Guaranteed 0.14 per cent.

<sup>16</sup>Guaranteed 2.00 per cent.

<sup>17</sup>Guaranteed 4.80 per cent.

<sup>18</sup>Ether-insoluble matter (glue) 1.65 per cent. Rosin present.

<sup>19</sup>Guaranteed mineral oil 83.00 per cent; saponifiable oil 3.00 per cent; potassium oxide 1.00 per cent.

<sup>20</sup>Guaranteed 20.00 per cent.

	0	il	ıd		Wa	ater	Soap		onstants parated		
Manufacturer or Distributor or Brand	Guaranteed	Found	Phenols—Found	2	Guaranteed	Found	Ash other than Ash—Found	Baumé Gravity, degrees	Saybolt Viscosity, 100° F., seconds	Unsulpho- nated	Publications
Can Iona Conor Mfr. Co. Son Iona Cal	%	%	%	%	%	%	%			07,	Cal. Dept. Agr., Spec
San Jose Spray Mfg. Co., San Jose, Cal. Exelol		83.143			25.00	15.62		22.53		68.80	Pub. 75, 49 (1927).
San Jose Spray Mfg. Co., San Jose, Cal. San Jose Summer Oil San Jose Spray Mfg. Co., San Jose, Cal.	80.003	86.013				13.58		30.60	106	100.00	Cal. Dept. Agr., Spec Pub. 75, 49 (1927). Cal. Dept. Agr., Spec
Skalene		78.543	5.60	4.20	10.00	11.75		27.30		64.40	Pub. 66, 33 (1926).
San Jose Spray Mfg. Co., San Jose, Cal. Skalol		82.703	4.60	5.30	10.00	6.20		21.20			Cal. Dept. Agr., Spec Pub. 75, 53 (1927). Cal. Dept. Agr., Spec
Rosa, Cal. Sarpa Poultry Spray <sup>21</sup>		98.10	1.60			0.30	none			42.00	Pub. 51, 57 (1925).
Santa Rosa Poultry Association, Santa Rosa, Cal. Sarpa Poultry Spray Sherwin-Williams Co., Cleveland, Ohio.		76.11	21.88			2.00	0.01				Cal. Dept. Agr., Spec Pub. 51, 57 (1925). Cal. Dept. Agr., Spec
Citro-Mulsion <sup>22</sup>	69.00	72.90	0.56		31.00	23.20	0.05			67.10	Pub. 51, 33 (1925).
Sherwin-Williams Co., Cleveland, Ohio. Citro-Mulsion <sup>23</sup>	73.00	72.87	0.38			23.80	0.08			63.40	Cal. Dept. Agr., Spec Pub. 51, 33 (1925). Conn. Agr. Expt. Sta.
Free Mulsion		85.7024							162	59.40	Sample 7776.
Sherwin-Williams Co., Cleveland, Ohio. Spray Mulsion		83.60	2.90	8.10	5.00	4.00		24.70			Cal. Dept. Agr., Spec Pub. 75, 53 (1927).
Sherwin-Williams Co., Cleveland, Ohio. Summer Mulsion	75.00	74.90				24.40		36.50		95.20	Cal. Dept. Agr., Spec Pub. 75, 50 (1927).

						1	1			
Sherwin-Williams Co., Cleveland, Ohio. Winter Mulsion	75.34	0.23			23.80	0.11	27.20			Cal. Dept. Agr., Spec. Pub. 66, 31 (1926).
Standard Oil Co. of Indiana. Dendrol <sup>25</sup>	 	• • • •		••••				142	62.80	Conn. Agr. Expt. Sta., Sample 7777.
Standard Oil Co. of Indiana. L 21 Oil <sup>26</sup>	 					0.06		193	60.80	Conn. Agr. Expt. Sta., Sample 8862.
Sterling Spray Co. S. O. S. Sterling Oil Spray	92.48		1.90		4.55		36.65	37	76.10	Cal. Dept. Agr., Spec. Pub. 75, 51 (1927). Conn. Agr. Expt. Sta.,
Miscible Oil <sup>27</sup>	 79.00					2.85				Bull. 258, 373 (1924). Conn. Agr. Expt. Sta.,
Spray Oil	 							374	46.00	
Van Antwerp Drug Corp., Mobile, Ala. Schnarr's Insecticide <sup>28</sup>	 49.13				33.68	1.44				Bull. 258, 373 (1924).

<sup>&</sup>lt;sup>3</sup>Per cent by volume.

<sup>21</sup>Specific gravity, 0.89002.

<sup>22</sup>Ether-insoluble (gums) 2.05 per cent.

<sup>23</sup>Ether-insoluble (gums) 2.18 per cent.

<sup>24</sup>Cc per 100 gm.

<sup>25</sup>Alcohol 0.23 per cent.

<sup>26</sup>Total nitrogen 0.03 per cent, ammonia nitrogen 0.01 per cent.

<sup>27</sup>Unsaponifiable matter 79.00 per cent; saponified matter 16.00 per cent.

<sup>28</sup>Total fatty acids 8.25 per cent; free fatty acids 0.77 per cent.

## Ortho Penetrating Oil Spray.

See "Oil Emulsions, Mineral."

## Ortho Penetrating Poisoned Barley.

See "Strychnine Preparations."

#### Ortho Poison Brand.

(CALIFORNIA SPRAY CHEMICAL CO., WATSONVILLE, CAL.)

	Guaranteed.	Found.
Arsenious oxide	7.10	6.90
Arsenic, metal	1.75	5.22
Cal. Dept. Agr., Spec. Pub. 51, 18 (1925).		

#### Ortho Walnut Worm and Aphis Dust.

(CALIFORNIA SPRAY CHEMICAL CO., WATSONVILLE, CAL.)

	Guaranteed.	Found.
Nicotine	0.90	1.17
Total arsenious oxide	3.00	3.22
Total arsenic, metal	1 96	2.10
Water-soluble arsenic, metal		0.06
Lead oxide		10.00
Cal. Debt. Agr. Spec. Pub 75 38 (1927)		

## Ortho Walnut Worm Dust.

(CALIFORNIA SPRAY CHEMICAL CO., WATSONVILLE, CAL.)

	Guaranteed.	Found
Arsenic	. 1.96	1.72
Cal Debt Aar Shee Pub 75 96 (1097)		

#### Ortho Weed Killer.

(CALIFORNIA SPRAY CHEMICAL CO., WATSONVILLE, CAL.)

	Guaranteed.	Found.
Arsenious oxide	29.00	29.73
Cal. Dept. Agr., Spec. Pub. 75, 25 (1927).		

Ortho Whale Oil Soap.

See "Soaps."

#### Otaylite.

## (MANUFACTURER NOT STATED)

Found: Water 8.20 per cent; soap 14.38 per cent; free sulphur 8.13 per cent; sodium carbonate 22.50 per cent; silica 28.46 per cent; iron and aluminum oxides 1.40 per cent; calcium carbonate 13.05 per cent; magnesium carbonate 3.48 per cent. No arsenic or lead present.

Cal. Dept. Agr., Spec. Pub. 34, 59 (1923).

## The Owl Argentine Ant Poison.

(THE OWL DRUG CO., SACRAMENTO, CAL.)

	Guaranteed.	Found.
Arsenic	0.20	0.18
Cal. Dept. Agr., Spec. Pub. 75, 24 (1927).		

#### Owl Prussic Acid.

See "Hydrocyanic Acid".

#### COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

## The Owl Sheep Dip.

See "Phenol Soap Solutions."

## "Oxygenic" Powder.

(OXYGENIC INDUSTRIAL CO., LOS ANGELES, CAL.)

Found: Water 7.48 per cent; copper sulphate (CuSO<sub>4.5</sub>H<sub>2</sub>O) 28.45 per cent; calcium oxide 32.12 per cent; gypsum 11.29 per cent; ammonium chloride 4.59 per cent; potassium chloride 3.42 per cent; sodium chloride 0.50 per cent; magnesium chloride 2.85 per cent; magnesium sulphate 3.10 per cent; silica 2.40 per cent; iron and aluminum oxides 0.58 per cent; carbon dioxide 3.22 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 60 (1923).

## P.

## Pacific R & H Hydrocyanic Acid.

See "Hydrocyanic Acid."

## Panama Spray Compound.

(CENTER PHARMACY, OAKLAND, CAL.)

No statement of active or inert ingredients. Not analyzed. Cal. Dept. Agr., Spec. Pub. 66, 36 (1926).

#### Paracide.

See "Paradichlorbenzene."

#### Paradichlorbenzene.

See Table XXIII.

#### Paragrene.

(FRED L. LAVANBURG, NEW YORK, N. Y.)

Found: Cupric oxide 23.46 per cent; combined arsenious oxide 17.52 per cent; free arsenious oxide 23.08 per cent; acetic acid 6.72 per cent; gypsum 19.31 per cent, sodium sulphate 2.26 per cent; sodium chloride 0.25 per cent; ferric oxide 0.20 per cent; moisture 6.20 per cent. The substance is a mixture of Paris green, gypsum and white arsenic.

Univ. of Cal. Coll. of Agr., Expt. Sta., Bull. 151, 23 (1903).

## Paris Green.

See Table XXIV.

#### Parker's Magic Discovery.

See "Lime-Sulphur Solution."

#### Par'Oidium.

(F. C. BOUCHER & CO., ST. PAUL, MINN., IMPORTERS).

Found: Ash insoluble in hydrochloric acid 7.10 per cent; carbon dioxide 2.88 per cent; sulphur trioxide 17.12 per cent; ferric oxide 10.63 per cent; calcium oxide 15.58 per cent; magnesium oxide 0.73 per cent; free sulphur 19.12 per cent; tobacco and water (by difference) 26.84 per cent. Substance is a mixture of gypsum, sulphur, ferric oxide, small amounts of sand, and tobacco.

U. S. D. A., Bur, Chem., Bull. 68, 54 (1902).

# TABLE XXIII. PARADICHLORBENZENE

		dichlor- nzene	, ,	
Manufacturer or Distributor and Brand	Guaranteed	Found	Meiting Point	Publication
	%	%		
Braun-Knecht-Heimann, San Francisco, Cal California Spray Chemical Co., Watsonville, Cal.	99.00	98.80		Ore. Agr. Expt. Sta., Cir. 84, 14 (1927).
Ortho  Hooker Electrochemical Co., Paracide.  Mechling Bros. Mfg. Co. Camden, N. J. Mechling's P. T. R	99.00 100.00	98.50 99.70		Cal. Dept. Agr., Spec. Pub. 75, 65 (1927). Cal. Dept. Agr., Spec. Pub. 75, 66 (1927),
ling's P. T. B.  Niagara Alkali Co. Niagara P. D. B.  Southern California Disinfecting Co., Los Angeles  Cal. Cedar Blooks		99.17	$\begin{bmatrix} 56 \\ 53.2 \end{bmatrix}$	Conn. Agr. Expt Sta., Bull. 242, 160 (1922). Cal. Dept. Ag., Spec. Pub. 51, 56 (1925).
	100.00	93.60		Cal. Dept. Agr., Spec. Pub. 75, 66 (1927).

<sup>&</sup>lt;sup>1</sup> No ash,

## TABLE XXIV. PARIS GREEN

T	ABLE X	XIV. 1	PARIS C	GREEN		
		Arseniou	ıs Oxide			
	To	otal	Water-	Soluble		
Manufacturer or Distributor and Brand	Guaranteed	Found	Guaranteed	Found	Cupric Oxide Found	Publication
	%	%	%	%	%	
Acme White Lead & Color Works, Detroit, Mich Bowker Chemical Co., New York, N. Y Brunswig Drug Co. Pfeiffer's Chipman Chemical Engineering Co., Inc., Bound	50.00 50.00 50.00	55.23	3.50 3.50 3.50		24.60 29.55	Ore. Agr. Expt. Sta., Cir. 84, 9 (1927). N. J. Agr. Expt. Sta., Bull. 459, 5 (1927). Cal. Dept. Agr., Spec. Pub. 75, 23 (1927).
Brook, N. J. Chipman	50.00	55.17 55.48 56.99	3.50	1.15	31.28 30.42 30.07	N. J. Agr. Expt. Sta., Bull. 459, 5 (1927). Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928). Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).
The Glidden Co., Cleveland, Ohio. Glidden	50.00	55.32 55.04 55.68 55.58 54.53 54.33 56.72 54.34 53.15	3.50 3.50 3.50 3.50 3.50 3.50 3.50 3.50	3.26 3.07 2.43	31.60  31.68 30.19 30.01 29.80	Cal. Dept. Agr., Spec. Pub. 75, 23 (1927). N. J. Agr. Expt. Sta., Bull. 459, 5 (1927). Cal. Dept. Agr., Spec. Pub. 75, 23 (1927). Cal. Dept. Agr., Spec. Pub. 75, 23 (1927). N. J. Agr. Expt. Sta., Bull. 459, 5 (1927). N. J. Agr. Expt. Sta., Bull. 424, 7 (1925). Cal. Dept. Agr., Spec., Pub. 75, 23 (1927). N. J. Agr. Expt. Sta., Bull. 407, 7 (1924). N. J. Agr. Expt. Sta., Bull. 424, 7 (1925)
Sherwin-Williams Co., Cleveland, Ohio. Sherwin-Williams	50.00	55.83	3.50	2.74		Cal. Dept. Agr., Spec. Pub. 75, 23 (1927).
way's	50.00	55.42 57.20	3.50	1.10 3.02		Cal. Dept. Agr., Spec. Pub. 75, 23 (1927). Maine Agr. Expt Sta., Official Inspections 126; 84 (1927).
Walnut Growers' Spray Mfg. Co. Golden State Western Wholesale Drug Co. Devoe Western Wholesale Drug Co. Star	50.00	56.69	3.50 3.50 3.50	2.37 2.24 1.98		Cal. Dept. Agr., Spec. Pub. 75, 23 (1927). Cal. Dept. Agr., Spec. Pub. 75, 23 (1927). Cal. Dept. Agr., Spec. Pub. 75, 23 (1927).

#### P. B. K.

(JOHN LUCAS & CO., INC., PHILADELPHIA, PA.) (See also "Green Cross P. B. K.")

Total arsenic, metal Water-soluble arsenic, metal Copper	Guaranteed. 21.30 1.00 6.00	Found. 21.24 1.64 9.00
--	--------------------------------------	---------------------------------

# P. D. Q. Argentine Ant Solution.

See "Hirschev."

## P. D. Q. Insect Powder.

(WORCESTER COMPOUND CO., WORCESTER, MASS.)

Found: Moisture 2.62 per cent; free sulphur 15.59 per cent; ash insolround: Moisture 2.02 per cent; iree sulpnur 10.09 per cent; asn insoluble in hydrochloric acid 64.51 per cent; ash soluble in hydrochloric acid 15.23 per cent; coal-tar products (by difference) 2.05 per cent; calcium, magnesium, potassium, sodium, sulphate and phosphate present. A mixture of sulphur, coal-tar products and earth.

U. S. D. A., Bur. Chem., Bull. 68, 49 (1902).

## Peerless Spray Emulsion.

Penetrating Oil Spray.

See "Oil Emulsions, Mineral."

Perfecto Rosin Paste.

See "Soaps."

Activ

Perfecto Spray Oil.

See "Oil Emulsions, Mineral."

# Pest-Go.

(PEST-GO, INC., PORTLAND, ORE.)

re salts	Guaranteed.	Found. 68.80
Agr. Expt. Sta., Cir. 84, 15 (1927).		

Pestroy.

See "Bordeaux Mixture-Lead Arsenate."

## Peterman's Roach Food.

(WM. PETERMAN, NEW YORK, N. Y.)

Found: Borax 20.60 per cent; balance potato or pea meal and red coloring matter.

U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

Petrotine.

See "Oil Emulsions, Mineral."

Pheno-Dip.

See "Phenol Soap Solutions."

See Table XXV.

Phenol.

Phenolene.

See "Phenol Soap Solutions."

	Phenols	v.				Constants of	its of	
		2				Separated Oil	00 001	
	Guaranteed	Pound	Water	daA qso2-noV	sliO	Baumé Gravity, Degrees	-nonqlusnU bəts	Publication
	%	%	%	%	%		%	
De tox		24.60	0.50			:		Cal. Dept. Agr., Spec. Pub. 58, 46 (1925).
- Santo	:	33.60	1.00		65.40		:	Cal. Dept. Agr., Spec. Pub. 75, 56 (1927).
	15.00	14.70	0.40	:	82.80	23.50	22.50	Cal. Dept. Agr., Spec. Pub. 58, 46 (1925).
		22.00	09.0	0.13	0.13 75.66	0.80 39.03	39.03	Cal. Dept. Agr., Spec. Pub. 51, 44 (1925).
		14.92	trace	0.05	84.48	0.70	trace	Cal. Dept. Agr., Spec. Pub. 51, 44 (1925).
		37.70	$1.60^{2}$	trace	59.60	12.30	7.20	Cal. Dept. Agr., Spec. Pub. 51, 44 (1925).
	:	17.00	0.20	:	80.50	20.80	36.10	Cal. Dept. Agr., Spec. Pub. 58, 46 (1925)
R.R. Rogers Chemical Co., San Francisco,	40.00 42.00 14.00	42.00	14.00		44.00			Cal. Dept. Agr., Spec. Pub. 75, 56 (1927).
T117-74 11-78	25.00 26.92	26.92	1.60	0.03	70.98	1.60	3.60	Cal. Dept. Agr., Spec. Pub. 51, 44 (1925).
		12.60	trace		86.50	0.07 86.50 25.50 49.80	49.80	Cal. Dept. Agr., Spec. Pub. 51, 44 (1925).

Adulterated with mineral oil. Guaranteed none.

323

	Phe	enols	s	oap	W	ater			Oils		
Manufacturer or Distributor and Brand	Guaranteed	Found	Gnaranteed	Found	Guaranteed	Found	Non-Soap Ash	Total	Tar	Mineral	Publication
Acme White Lead & Color Works, Detroit, Mich. Acme Sheep & Cattle	%	%	%	%	%	%	%•	%	%	%	
An-Fo Mfg. Co., Los Angeles, Cal.		10.70		24.30	10.00	10.40	0.60		52.00		Cal. Dept. Agr., Spec. Pub. 75, 57 (1927).
An-Fo Disinfectant		16.10		21.81	10.00	9.60	2.09		49.25		Cal. Dept. Agr., Spec. Pub. 75, 57 (1927).
An-Fo Sheep Dip		17.30		20.90	10.00	8.20	1.05		50.40		Cal. Dept. Agr., Spec. Pub. 75, 57 (1927).
At-Ko Sheep DipBuker's Bird Store, San Francisco,		22.20		5.86	10.00	2.00	0.35		68.29		Cal. Dept. Agr., Spec. Pub. 75, 57 (1927).
Cal. Buker's Sheep Dip		25.10	• • • • •	14.40	10.00	6.40			22.70	29.40	Cal. Dept. Agr., Spec. Pub. 75, 57 (1927).
Cal. Creo Fenol Sheep Dip.: The G. E. Conkey Co., Cleveland,		14.00		16.00	10.00	8.00	0.50		59.70		Cal. Dept. Agr., Spec.  Pub. 75, 57 (1927).  Cal. Dept. Agr., Spec.
Ohio. Conkey's Noxicide <sup>1,2</sup> Economy Hog & Cattle Powder Co.		16.80		23.30	9.00	8.60	0.60	49.20			Pub. 51, 45 (1925). Cal. Dept. Agr., Spec.
Economy Germicide Dip		15.90		14.20	10.00	6.60	1.20		60.90		Pub. 75, 57 (1927). Cal. Dept. Agr., Spec.
Insecticide Sheep Dip		13.00	6.00	7.90		29.80	4.80		45.003		Pub. 75, 57 (1927). Cal. Dept. Agr., Spec.
and Cattle Wash	9.15	11.00	24.94	7.60		15.30	2.40		63.204		Pub. 75, 57 (1927).
W. McNess Krenol Dip'& Disinfectant		19.40		26.20	8.00	8.00	0.50		43.60		Cal. Dept. Agr., Spec. Pub. 75, 57 (1927).

Germain Seed & Plant Co., Los Angeles, Cal. Cann's Canco Germo Mfg. Co. Germo Carboline.		10.25 9.50		20.57 13.60	10.00 12.00	14.10 12.00	1.06 0.70		52.68 62.60		Cal. Dept. Agr., Spec. Pub. 75, 58 (1927). Cal. Dept. Agr., Spec. Pub. 75, 58 (1927).
Germo Mfg. Co. Germo Cresosote Dip Germo Mfg. Co. Germo Sheep Dip	••••	12.40 11.30		14.70 19.60	12.00 12.00	7.40 10.40	0.60		63.00 55.70		Cal. Dept. Agr., Spec.
Heinrich Chemical Co., Oakland, Cal. Ongman's Dip & Disinfectant Dr. Hess & Clark. Dr. Hess Dip and		12.50		10.30	8.00	5.80	1.70		25.80	44.00	Cal. Dept. Agr., Spec. Pub. 75, 58 (1927). Cal. Dept. Agr., Spec.
Disinfectant	14.00	14.93	20.00	19.07	9.50	8.87	1.25		$54.03^{5}$		Pub. 75, 58 (1927). Cal. Dept. Agr., Spec.
Hockwald Chemical Co. Hockwald's Sheep Dip		14.50		17.80	12.00	7.80	0.50		56.80		Pub. 75, 58 (1927).
Kirk, Geary & Co. Kirk Geary Dipsol Sheep Dip	16.00										lent Agr Spec
Sheep Dip		16.60		22.70	10.00	10.80			48.30		Pub. 75, 58 (1927).
Langley & Michaels Co., Germosol Special Disinfectant Dr. LeGear Medicine Co., St. Louis,						17.00					Cal. Dept. Agr., Spec. Pub. 75, 58 (1927).
Mo. Dr. LeGear's Dip and Disinfectant <sup>1,7</sup> .  Arthur R. Maas Chemical Co. Cresylol.		12.51		18.90	12.00	10.80	0.20	55.97			Cal. Dept. Agr., Spec, Pub. 51, 45 (1925). Cal. Dept. Agr., Spec. Pub. 75, 58 (1927).
lol Mefford Chemical Co. Sheep Dip	••••	14.76 3.50		15.60	15.00	10.60 9.20	0.43			57.95	Pub. 75, 58 (1927). Cal. Dept. Agr., Spec. Pub. 51, 45 (1925).

¹Contains rosin.
²Sp. Gr. of oil, Baumé, 3.10.
³Guaranteed 54.40 per cent.
⁴Guaranteed 36.60 per cent.
⁵Guaranteed 53.00 per cent.
⁶Guaranteed 65.00 per cent.
²Sp. Gr. of oil, Baumé, 2.95. Unsulphonated, 1.80 per cent.

	Pho	enols	S	oap	M	ater			Oils		
Manufacturer or Distributor and Brand	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	Non-Soap Ash	Total	Tar	Mineral .	Publication
Michel & Pelton Co., San Francisco,	%	%	%	%	%	%	%	%	%	. %	Cal. Dept. Agr., Spec
Michel & Pelton Co., San Francisco	50.00	52.43			21.00	21.18					Pub. 51, 43 (1925).
Cal. Liquor Cresolis Compound . Michel & Pelton Co., San Francisco,		47.75			15.00	16.00					Cal. Dept. Agr., Spec Pub. 51, 43 (1925).
Cal. Liquor Cresolis Compound Michel & Pelton Co., San Francisco,		53.03			21.00	19.85					Cal. Dept. Agr., Spec Pub. 51, 43 (1925).
Cal. Mapco Sheep Dip		18.70		8.60	10.00	2.20	0.30		28.30	40.00	Cal. Dept. Agr., Spec Pub. 75, 59 (1927).
Stock & Poultry Dip & Disinfectant <sup>1,8</sup> H. K. Mulford Co. Krelos		23.12 28.70		30.20 13.70	8.00 10.00	8.20 9.00	0.16 1.80	36.99	45.60		Cal. Dept. Agr., Spec. Pub. 51, 44 (1925). Cal. Dept. Agr., Spec.
Ness & Co., Darlington, England. Thymo-Cresol <sup>9</sup> No-Vermo Mfg. Co. Sol-o-cree <sup>1, 10</sup>		6.00		18.80	10.00	9.80	0.40	63.20			Pub. 75, 59 (1927). U. S. D. A., Bur. Chem. Bull. 68, 57 (1902). Cal. Dept. Agr., Spec.
The Owl Drug Co. The Owl Sheep Dip Pacific Chemical Co., Los Angeles,		20.00		14.20	10.00	8.00	1.50		55.70		Pub. 51, 45 (1925). Cal. Dept. Agr., Spec. Pub. 75, 59 (1927).
Parke, Davis & Co., Detroit, Mich		50.75				21.20					Cal. Dept. Agr., Spec. Pub. 51, 43 (1925).
Kreso Dip No. 1		22.55		20.80	8.00	8.60	2.65		44.60		Cal. Dept. Agr., Spec. Pub. 75, 59 (1927).

							de la			Designation.	
Polk Miller Products Corp. Sergeant's Disinfectant Pratt Food Co. Pratt's Dip and Disinfectant W. T. Rawleigh Co., Oakland, Cal.	9.50	22.10 9.20		23.00 15.50			0.70		$\begin{vmatrix} 42.60 \\ 64.90^{11} \end{vmatrix}$		Cal. Dept. Agr., Spec. Pub. 75, 59 (1927). Cal. Dept. Agr., Spec. Pub. 75, 59 (1927).
Rawleigh's Stock Dip and Disfectant		16.40	,,,,	17.90	8.00	8.40	1.20		54.30		Cal. Dept. Agr., Spec. Pub. 75, 59 (1927).
Cal. Robinson's Red Label Sheep Dip		20.12		9.41	10.00	2.00	0.07	13 10 10 10	65.40		Cal. Dept. Agr., Spec. Pub. 75, 59 (1927). Cal. Dept. Agr., Spec.
Francisco, Cal. Arcol	50.00	51.85	20.00	19.78	25.00	22.25	0.09		5.34		Pub. 75, 59 (1927).
R. R. Rogers Chemical Co., San Francisco, Cal. Ravol R. R. Rogers Chemical Co., San	50.00	43.50	20.00	21.50	27.00	24.01	0.63		8.40		Cal. Dept. Agr., Spec. Pub. 75, 59 (1927). Cal. Dept. Agr., Spec.
Francisco, Cal. Vermol	50.00	50.50	20.00	22.80	25.00	19.90	0.68		4.79		Pub. 75, 59 (1927).
Scott & Gilbert Co. Scott & Gilbert Sheep Dip The Shaw-Batcher Co. Capital		19.20		18.45			0.80		51.45		Cal. Dept. Agr., Spec. Pub. 75, 59 (1927). Cal. Dept. Agr., Spec.
Sheep Dip and Cattle Wash <sup>12</sup>											Pub. 75, 59 (1927).
The Sherwin-Williams Co., Cleveland, Ohio. Pheno Dip The Sherwin-Williams Co., Cleve-		26.50		23.90	8.00	8.20	0.30		38.80		Cal. Dept. Agr., Spec.  Pub. 75, 60 (1927).  Cal. Dept. Agr., Spec.
land, Ohio. Phenolene <sup>1</sup>		15.20		21.10	10.00	8.40	0.23	54.31			Pub. 51, 45 (1925).
The Sherwin-Williams Co., Cleveland, Ohio. S. W. Sheep Dip Spratt's Patent, Ltd., San Francisco,		10.40		22.10	10.00	10.00	0.10		55.30		Cal. Dept. Agr., Spec. Pub. 75, 60 (1927). Cal. Dept. Agr., Spec.
Cal. Spratt's Dip United Drug Co. No. 6 Disinfectant		$\begin{vmatrix} 23.70 \\ 29.40 \end{vmatrix}$		$\begin{vmatrix} 20.80 \\ 14.70 \end{vmatrix}$	10.00 10.00	8.00 7.80	2.90 2.20		43.80 45.40		Pub. 75, 60 (1927). Cal. Dept. Agr., Spec.
											Pub. 75, 60 (1927).

<sup>&</sup>lt;sup>1</sup>Contains rosin.

<sup>8</sup>Sp. Gr. of oil, Baumé, 3.70. Unsulphonated, trace.

<sup>9</sup>Substance is a creosote soap emulsion.

 $<sup>^{10}{\</sup>rm Sp.}$  Gr. of oil, Baumé, 22.40. Unsulphonated, 39.90 per cent.  $^{11}{\rm Guaranteed}$  55.00 per cent.  $^{12}{\rm Emulsion}$  broken. Not analyzed.

	Phe	Phenols	S S	Soap	M	Water			Oils		
Manufacturer or Distributor and Brand	Guaranteed	Found	Guaranteed	Found	Guaranteed	Found	deA qso2-noV	IstoT	TsT	Mineral	Publication
Geo Z Wait Co Dr G Z Wait's	%	%	%	%	%	%	%	%	%	- %	
Sheep Dip.	15.00	13.70		21.40		10.20		0.10	$52.20^{11}$		Cal. Dept. Agr., Spec. Pub. 75, 60 (1927).
Dip. Transfer Transfer Cal. Sucception of the	15.00	14.20		21.20		9.60		$51.10^{7}$			Agr., 6 (19
Liquor Cresolis Compound	40.00	45.50	:	34.1013 28.00		17.00	0.50		1.80	:	Cal. Dept. Agr., Spec. Pub. 75, 60 (1927).
		21.40		6.90	10.00	3.20	0.20		28.10		Cal. Dept. Agr., Spec. Pub. 75, 60 (1927).
Watkin's Germicide Dip and Dis- infectant	18.00	17.40	24.40	22.70		8.40	1.40		49.0014		Cal. Dept. Agr., Spec Pub. 75, 60 (1927).
Cal. Carco Natholeum Dip West Disinfecting Co. Karspray		12.50 47.00	: i	21.80 21.40	10.00	10.40 21.40	1.80	3.65	51.60		Cal. Dept. Agr., Spe. Pub. 75, 60 (1927). Cal. Dept. Agr., Spe.
West Disinfecting Co. Licresolis	:	55.63			15.00	13.20		i		:	Pub. 66, 33 (1926). Cal. Dept. Agr., Spec
Western Wholesale Drug Co., Los Angeles, Cal. Solution of Cresol.	50.00	48.30		:		19.20		į			Pub. 51, 43 (1925). Cal. Dept. Agr., Spec. Pub. 51, 43 (1925).
septic Sheep Dip	:	18.10			15.00	8.00				:	Cal. Dept. Agr., Spec Pub. 51, 44 (1925).

## TABLE XXVII. PHOSPHORUS PREPARATIONS

		ree phorus	
Manufacturer or Distributor and Brand	Guaranteed	Found	Publication
Allan-Pfeiffer Chemical Co., St. Louis, Mo. Allan's Lightning Roach Paste¹ American Druggists Syndicate. A.D. S. Rat and Roach Paste.  Barnard & Co., Boston, Mass. Roach and Water Bug Exterminator².  C. S. Brown & Co., Chicago, Ill. Fidelity Cockroach Paste³.  Buffalo Specialty Co. Rat Nip  B. E. Cole, Hollister, Cal. Squirrel Poison-Phosphorus⁴.  Common Sense Mfg. Co. Common Sense Rat Exterminator.  The DePree Co., San Francisco, Cal. Pied Piper Rat and Roach Paste.  C. W. Hill Chemical Co., Los Angeles, Cal. Rat Poison (Mission)⁵.  Hudelson & Damrell, Modesto, Cal. Gold Crown Poison.  Hudelson & Damrell, Modesto, Cal. Gold Crown Poison Barley⁵.  Jacob's Pharmacy, Atlanta, Ga. Tiger Paste².  H. K. Mulford Co. Mulford Phosphorus Paste.  Nyal Co., San Francisco, Cal. Sure Kill Rat and Roach Paste.  The Rat Biscuit Co. Rat Bis-Kit Paste.  The Rat-Mum Co. Rat-Mum.  Stearns' Electric Paste Co. Stearns' Electric Rat and Roach Paste.  United Drug Co. Elkay's Rat and Roach Paste.  United Drug Co. Elkay's Rat Snap².  Zeno Products, Inc., New York, N. Y. Zeno.	2.00 2.50 0.90 2.00 2.00 2.00 2.00 1.50 2.00 3-11/6 2.00 3-11/6 2.00	% 1.94 1.90 1.37 0.77 1.20 0.026 1.59 1.30 0.60 8.53 0.07 1.93 2.34 1.58 2.28 0.11 2.16 1.73 0.63 0.11	U. S. D. A. Bur. Chem., Bull 68, 45 (1902).  Cal. Dept. Agr., Spec. Pub. 75 47 (1927).  U. S. D. A. Bur. Chem., Bull 68, 45 (1902).  U. S. D. A. Bur. Chem., Bull 68, 45 (1902).  Cal. Dept. Agr., Spec. Pub. 75, 47 (1927).  Cal. Dept. Agr., Spec. Pub. 51, 52 (1925).  Cal. Dept. Agr., Spec. Pub. 58, 41 (1925).  Cal. Dept. Agr., Spec. Pub. 75, 47 (1927).  Cal. Dept. Agr., Spec. Pub. 75, 47 (1927).  Cal. Dept. Agr., Spec. Pub. 75, 47 (1927).  Cal. Dept. Agr., Spec. Pub. 66, 35 (1926).  U. S. D. A. Bur. Chem., Bull. 68, 45 (1902).  Cal. Dept. Agr., Spec. Pub. 75, 47 (1927).  Cal. Dept. Agr., Spec. Pub. 58, 41 (1925).  Cal. Dept. Agr., Spec. Pub. 58, 41 (1927).  Cal. Dept. Agr., Spec. Pub. 75, 47 (1927).  Cal. Dept. Agr., Spec. Pub. 58, 41 (1925).  Cal. Dept. Agr., Spec. Pub. 51, 52 (1925).  Cal. Dept. Agr., Spec. Pub. 51, 52 (1925).  Cal. Dept. Agr., Spec. Pub. 51, 52 (1925).

¹Contains corn starch and glucose.
²Contains wheat starch and molasses.
³Contains wheat starch, molasses and glucose.
⁴Manufacture discontinued.
⁵Guaranteed 97.00 per cent. inert matter (honey). Manufacture discontinued.
⁶No potassium cyanide found present.
³Guaranteed hydrocyanic acid 5/16 per cent; hydrochloric acid 5/16 per cent;
hydrofluoric acid 5/16 per cent; "vertigrese," blue stone and water 1 4/16 per cent;
lard, molasses, flour and corn meal 94 1/16 per cent. No hydrocyanic, hydrochloric or hydrofluoric acids or copper found.

Phenol Soap Solutions.

See Table XXVI.

Phosphorus Preparations.

See Table XXVII.

Pied Piper Rat and Roach Paste.

See "Phosphorus Preparations."

Pink Arsenoid (Lead Arsenite.)

(ADLER COLOR & CHEMICAL WORKS, NEW YORK, N. Y.)

Found: Lead oxide 49.58 per cent; combined arsenious oxide 40.02 per cent; free arsenious oxide 3.24 per cent; moisture 0.31 per cent; organic matter, lead sulphate, etc., 6.85 per cent. The substance is colored with a pink aniline residue.

Univ. of Cal. Coll. of Agr. Expt. Sta., Bull. 151, 26 (1903).

#### P. K. B.

(JOHN LUCAS & CO., INC., PHILADELPHIA, PA.)

	Guaranteed.	Found.
Total arsenic, metal	21.00	21.64
water-soluble arsenic, metal	1.00	0.87
Copper		15.50
N. J. Agr. Expt. Sta., Bull. 441, 13 (1926).		

#### Plant Oil.

(B. G. PRATT CO., NEW YORK, N. Y.)

Found: Largely vegetable oils.

N. Y. Agr. Expt. Sta,. Bull. 384, 302 (1914).

#### Plant Tonic and Insecticide.

(BYRNE MFG. CO., OAKLAND, CAL.)

Declared composition: Tobacco stems, lime, sulfur, borax, oil of sassa-

fras (synthetic), geranium oil and water.

Found: Nicotine 0.22 per cent; free sulphur 0.39 per cent; faint test for borax; very small amount of calcium oxide; sassafras and geranium oils present.

Cal. Dept. Agr., Spec. Pub. 34, 60 (1923).

#### Poison Bait Compound.

(CHIPMAN CHEMICAL ENGINEERING CO., INC., NEW YORK, N. Y.)

Guaranteed: Sodium arsenite 2.50 per cent; inert matter 97.50 per cent; total arsenic 1.60 per cent; water-soluble arsenic 1.60 per cent.

Found: Total arsenic (metal), 0.98 per cent.

Conn. Agr. Expt. Sta., Sample 1825.

#### Poison Barley.

See "Strychnine Preparations."

#### Poisoned Ant Syrup.

(SAN DIEGO HORTICULTURA)	COMMISSIONER,	SAN DIEGO,	CAL.)
		Guaranteed.	Found.
dium arsenite		0.15	0.23

Cal. Dept. Agr., Spec. Pub. 66, 19 (1926).

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

## Poisoned Barley.

#### Poisoned Grain.

## Poisoned Grain Squirrel Poison.

See "Strychnine Preparations".

#### Poison Green.

(BANNENMAN CHEMICAL CO., SYRACUSE, N. Y.)

Found: Substance is artifically colored calcium arsenite. Arsenious oxide, 38.4 per cent.

N. Y. Agr. Expt. Sta., Bull. 348, 98 (1912).

## Poison Spray Tabs-Bordeaux Arsenate.

(EARL THOMAS CULTURE CORP, NEW YORK, N.Y.)

	Guaranteed.	Found.
Total arsenic, metal	5.00	5.51
Water-soluble arsenic, metal	0.05	0.07
Copper		12.40
Maine Agr. Expt. Sta., Official Inspections 110,	56 (1923).	

#### . Sta., Official Inspections 110, 50 (1923)

#### Poison Wheat.

See "Strychnine Preparations."

#### Pomodust.

See "Niagara 90-10 Dusting Mixture."

#### Potassium Cyanide.

(MANUFACTURER NOT STATED).

Found: Moisture 0.22 per cent; potassium cyanide 93.80 per cent sodium cyanide 0.87 per cent; sodium chloride 1.15 per cent; potassium ferrocyanide 0.75 per cent; fine dust 3.22 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 25 (1923).

#### Potato Dust Mixture.

See "Nicotine Dusts."

## Potato Dust Poison.

(DELORO CHEMICAL CO., DELORO, ONT., CANADA).

Found: Total arsenic oxide 9.62 per cent; water-soluble arsenic oxide 0.10 per cent; cupric oxide 8.07 per cent.

Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).

#### Potato Scab Destroyer.

(AM. HORT. DISTRIBUTING CO., MARTINSBURG, W. VA.)

Found: Substance is an artifically colored solution of formaldehyde.

N. Y. Agr. Expt. Sta., Bull. 384, 301 (1914).

## Poultry House Spray.

(PURITY CHEMICAL PRODUCTS CO., SANTA ROSA, CAL.)

Guaranteed: Phenols 25.00 per cent; base, sediment and moisture 2.00 per cent.

Found: Phenols 25.00 per cent; mineral oils 53.90 per cent; water 0.40

Cal. Dept. Agr., Spec. Pub. 75, 56 (1927).

333

#### Pratt's Carboleine.

BULLETIN 300

See "Oil Emulsions, Mineral."

## Pratt's Dip & Disinfectant.

See "Phenol Soap Solutions."

## Pratt's Lice Killer.

(PRATT FOOD CO.)

(FRAIL FOOD CO.)	the last of the same of the sa	
	Guaranteed.	Found.
Nicotine	0.34	0.72
Naphthalene	5.60	5.88
Pyrethrum	4.00	
Cal. Dept. Agr. Spec. Pub. 75, 40 (1927).		

#### Premium Intestinal Cleaner.

(GERMAIN SEED & PLANT CO., LOS ANGELES, CAL.)

	Guaranteed.	Found.
Sodium bicarbonate	15.00	15.84
Cal. Dept. Agr., Spec. Pub. 75, 65 (1927).		

## Preparation B.

(THE INSECTICIDE CO., SAN FRANCISCO, CAL.)

Guaranteed: Inert matter, trace of butter color. Found: Mineral oils 10.00 per cent; yellow color present; no ash. Cal. Dept Agr., Spec. Pub. 51, 60 (1925).

## Preparation No. 3

(THE INSECTICIDE CO., SAN FRANCISCO, CAL.)

See "Sodium Fluoride."

#### Preparation No. 5.1

(THE INSECTICIDE CO.	., SAN FRANC	CISCO, CA	AL.)	
	Guaranteed.	Found.	Guaranteed.	Found.
Calcium arsenate		$\frac{3.48}{2.83}$	$3.25 \\ 2.50$	$\frac{3.80}{2.71}$
Cal. Dept. Agr., Spec. Pub. 75,	26 (1927).			

## Prussic Acid.

See "Hydrocyanic Acid."

#### P. S. C. Condensed Rosin Spray.

P. S. C. Fish Oil Soap.

See "Soap."

Puritol.

See "Albatross Puritol."

Purity.

(	PURITY	CHEMICAL	PRODUCTS	CO.)

	Guaranteed.		Found.
Sodium hypochlorite	4.00	,	4.24
Available chlorine			4.04
Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).			

## Pyrethro Fly Fluid.

(W. T. RAWLEIGH CO., FREEPORT, ILL.)

Not analyzed. Cal. Dept. Agr., Spec. Pub. 58, 48 (1925).

## Pyrethrum

See Table XXVIII.

#### Pyrox.

See "Bordeaux Mixture-Lead Arsenate."

#### Q.

#### Qua-Sul.1

(A. R. GRE	GORY, SAN	FRANCIS	so, CAL.)		
	ar. Found.	Guar.	Found.	Guar.	Found.
Sodium polysulphide Sodium thiosulphate	$\begin{array}{ccc} & 7.04 \\ & 3.04 \end{array}$		$\frac{8.00}{3.17}$	$\frac{4.45}{7.16}$	$7.24 \\ 3.60$
Active sulphur	19.00	6.00	20.50		19.00
Cal Dept. Agr. Spec. Pub.	66, 23 (192	6).			

## Quick Action Fly Killer.

(OUICK ACTION PRODUCTS CO., BELL,	CAL.)	
	Guaranteed.	Found.
Baumé gravity, 15.5°C	3.00	$   \begin{array}{r}     39.90 \\     96.74 \\     2.49   \end{array} $
Cal Debt Agr Spec Pub 51 55 (1925).		

## Ouick Action Lice Powder.

(QUICK ACTION PRODUCTS CO., LOS AN	GELES, CAL.)	
	Guaranteed.	Found.
Cresol oil (Phenols)	1.00	0.05
Hydrocarbons		
Naphthalene		5.92
Sodium fluoride	10.00	8.20
Inert matter	79.00	,,,,

#### Oykade.

(THE CHLORINE PRODUCTS CO., NEW YORK, N. Y.)

Found: Grams per 100 cc. total solids 1.26; available chlorine 0.40; total chlorine, 0.71; calcium oxide, 0.64 per cent. Substance is a mixture of calcium hypochlorite and calcium chloride in solution.

Conn. Agr. Expt. Sta., Bull. 258, 376 (1924).

Cal. Dept. Agr., Spec. Pub. 51, 59 (1925).

## R.

#### Rajah Argentine Ant Poison.

(THE OWL DRUG CO., SAN FRANCISCO, CAL.)

	Guaranteed.	Found.
Total arsenic, metal	$0.20 \\ 0.20$	$0.16 \\ 0.16$
Cal Debt Agr Spec Pub 34 18 (1923).		

#### Rat Bis-Kit Paste.

See "Phosphorus Preparations."

5

<sup>&</sup>lt;sup>1</sup> Two grades.

<sup>1</sup> Three Samples.

PYRETHRUM	
X XVIII.	
TABLE	

Manufacturer or Distributor and Brand Buhack Producing & Mfg. Co. Buhack Insect Powder	sect Bither Bither S. % 6. % Ash 6. % Ash 7	6. Stringet Ether 6. Stringet		Publication  Cal Dept. Agr., Spe. Pub. 75, 63 (1927).
	9.00	9.00 6.80	6.75	Cal. Dept. Ag ., Spec. Pub. 75 63 (1927).  6.75 U. S. D. A, Bur. Chem., Bull. 68 39 (1902).

D	-	+	T	x

(CHEMICAL SPECIALTIES CO., LOS ANG	ELES. CAL.)	
	Guaranteed.	Found.
Barium carbonate	20.00	26.21
Inert matter	80.00	
Cal Dept. Agr., Spec. Pub. 66, 34 (1926).		

Rat-Go.

See "Leinen."

Rat-Mum.

Rat-Nip.

Rat Poison (Mission).

See "Phosphorus Preparations".

Rat Scent.

See "Strychnine Preparations".

Rat Tex.

(CALIFORNIA REX SPRAY CO., WATSONVILLE, CAL.) Found. Guaranteed. 31.66 Arsenious oxide..... 30.00

Cal. Dept. Agr., Spec. Pub. 75, 26 (1927).

See "Phenol Soap Solutions."

Rawleigh's Insect Powder.

See "Pyrethrum."

Rawleigh's Louse Powder.

(W. T. RAWLEIGH CO., FREEPORT, ILL.)

Not analyzed.

Cal. Dept. Agr., Spec. Pub. 58, 48 (1925).

Rawleigh's Stock Dip & Disinfectant.

See "Phenol Soap Solutions."

Red Cross Ant Destroyer.

(RED CROSS CHEMICAL CO., CINCINNATI, OHIO.)

Found: Borax 11.07 per cent; calcium oxide 13.12 per cent; sulphur trioxide 18.56 per cent; gypsum and sassafras root present.

U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

Red Label Sheep Dip.

See "Phenol Soap Solutions."

Reliance Argentine Ant Poison.

(LANGLEY & MICHAELS CO., SAN FRANCISCO, CAL.)

Guaranteed: Arsenic, metal 0.11 per cent; sodium arsenite 0.20 per cent. Found: Arsenic, metal 0.13 per cent; sodium arsenite 0.22 per cent; invert sugar 7.88 per cent; cane sugar 40.29 per cent; dextrin, etc., 1.82 per cent; water 47.67 per cent.

Cal. Dept. Agr., Spec. Pub. 58, 16 (1925).

Renol.

See "Oil Emulsions, Mineral."

## Revenge Lice Destroyer.

(I. D. RUSSELL CO.)		
Naphthalene Sulphur Phenols Phenol and creosote.  Cal. Dept. Agr. Spec. Pub. 75, 41 (1927).	5.00	Found. 3.57 5.35 0.80 5.92

#### Rex Emulso.

See "Oil Emulsions, Mineral.":

## Rex Fly Tox.

(CALIFORNIA REX SPRAY CO., BENICIA, CAL.)

Guaranteed: Active ingredients 100.00 per cent.

Found: Specific gravity, 0.8147; oil 100.00 per cent; no water or ash; kerosene 92.50 per cent; safrole present.

Cal. Dept. Agr., Spec. Pub. 51, 56 (1925).

## Rex Miscible Oil.

## Rex Vulture Oil.

See "Oil Emulsions, Mineral."

## Rid-O-Germ.

(WESTERN CHEMICAL CO., INC.)

	Guaranteed.	Found.
Available chlorine		2.68
Water	87.28	87.45
Cal. Dept. Agr., Spec. Pub. 75, 61 (1927).		

## Rip Van Winkle Fly Spray.

See "Oils, Mineral."

# Roach and Croton Bug Exterminator.

(PERFECTION MFG. CO., JERSEY CITY, N. J.)

Found: Borax 47.61 per cent; balance pyrethrum, corn meal and pink coloring matter.

U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

## Roach and Water Bug Exterminator.

See "Phosphorus Preparations."

Roach Doom.

See "Murray."

#### Roachine.

(BROWN & ALLEN, ATLANTA, GA.)

Found: Borax 90.00 per cent; balance pyrethrum and blue coloring matter.

U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

#### Roach Liquid.

(THE PIED PIPER SERVICE, PROVIDENCE, R. I.)

Found: Substance is a mixture of kerosene and methyl salicylate.

Conn. Agr. Expt. Sta., Bull. 258, 376 (1924).

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC. Roach Powder.

(THE PIED PIPER SERVICE, PROVIDENCE, R. I.)

Found: Water 14.57 per cent; nitrogen 0.50 per cent; protein 3.13 per cent; starch 13.73 per cent; ash 40.60 per cent; sodium oxide 18.89 per cent; boric oxide 18.32 per cent; chloride 5.12 per cent; traces of silica, iron, sulphate and phosphate. The substance is a mixture of a cereal, borax and salt.

Conn. Agr. Expt. Sta., Bull. 258, 376 (1924).

## Roachsault.

(BARRETT CHEMICAL CO., NEW YORK, N. Y.)

Found: Sodium fluoride 85.00 per cent; water 2.53 per cent; aluminum and iron oxides 0.27 per cent; fine sand 9.60 per cent; sodium sulphate (by difference), 2.60 per cent.

U. S. D. A., Bur. Chem., Bull. 68, 53 (1902).

#### Rosin Spray.

See "Soaps."

#### Rough on Rats.

(E. S. WELLS ESTATE.)	Guaranteed.	Found.
Arsenious oxide Arsenic, metal Barium carbonate	56.00 20.00	98.19 $74.37$ $0.00$
Cal. Dept. Agr., Spec. Pub. 75, 27 (1927).		

# Royal Roach Powder.

(H. C. DUBRING & CO., CHICAGO, ILL.)

Found: Borax 30.94 per cent; arsenious oxide 0.73 per cent; arsenious oxide 4.69 per cent; some calcium and dye stuff. Substance is probably a mixture of borax, London Purple and pyrethrum.

U. S. D. A., Bur. Chem., Bull. 68, 30 (1902).

#### S.

#### Salairacine.

(J. D. MCGREGOR, STAMFORD, CONN.)

Two Samples:

#### Sample 17745.

Found: Moisture 0.90 per cent; acid-insoluble matter, 0.56 per cent; nitrate nitrogen none; lead oxide 24.21 per cent; arsenic oxide 11.73 per cent; iron and aluminum oxides 0.50 per cent; calcium oxide 25.64 per per cent; magnesium oxide 17.33 per cent; carbon dioxide (by difference) 19.13 per cent.

## Sample 19740.

Found: Nitrate nitrogen 1.95 per cent; arsenic oxide 10.54 per cent; calcium oxide 17.67 per cen .

Conn. Agr. Expt. Sta., Bull. 242, 161 (1922).

#### Sanders Blue Dust 5-11.

(NEW YORK INSECTICIDE CO., MEDI	NA, N. Y.)	
	Guaranteed.	Found.
Copper	9.00	11.10
Maine Agr. Expt. Sta., Official Inspections 122,	86 (1926).	

#### Sander's Dust.

(MANUFACTURER UNKNOWN).

Found: Copper 6.10 per cent; arsenic 2.93 per cent: Arsenic is present as calcium arsenate.

Conn. Agr. Expt. Sta., Bull. 258, 370 (1925).

## San Jose Summer Oil.

See "Oil Emulsions, Mineral."

## San-U-Zay Scale Oil.

(THE GARDINER-JOHNS CO., ROCHESTER, N. Y.)

Found: The substance is largely a mixture of mineral and animal oils. N. Y. Agr. Expt. Sta., Bull. 384, 301 (1914).

## Sapho "Fly X."

(THE KENNEDY MFG. CO., MONTREAL, CANADA.)

Found: Specific gravity, 19° C., 0.844; flash point 67° C.; fire point 77° C.; residue at 100° C., 0.15 per cent; methyl salicylate 2.08 per cent. Base is kerosene.

Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).

## Sapho Liquid.

(THE KENNEDY MFG. CO., MONTREAL, CANADA).

Found: Specific gravity,  $19^\circ$  C., 0.844; flash point  $67^\circ$  C.; fire point  $78^\circ$  C.; residue at  $100^\circ$  C., 0.81 per cent; phenol 1.68 per cent. Base is kerosene.

Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).

## Sarpa Poultry Spray.

See "Oil Emulsions, Mineral."

#### Scalecide.

(B. G. PRATT CO., NEW YORK, N. Y.)

Found: Substance is a mixture of mineral and vegetable oils and naphthalene.

N. Y. Agr. Expt. Sta., Bull. 384, 302 (1914).

## Scalybark Insecticide.

(DR. MESSIG, SANTA ANA, CAL.)

Found: Water 82.92 per cent; solids 17.08 per cent; lime-sulphur 3.00 per cent; sodium chloride 3.50 per cent; soap 3.00 per cent; free sulphur and vegetable matter, etc. 4.30 per cent; no lead; a minute quantity of arsenic. The substance is partly a yellow liquid and partly a black solid containing some small leaves; it has an ammoniacal odor.

Cal. Dept. Agr., Spec. Pub. 34, 57 (1923).

## Schnarr's Insecticide.

See "Oil Emulsions, Mineral."

## Schrader Argentine Ant Powder.

(SCHRADER CHEMICAL CO., SAN FRANCISCO, CAL.)

Guaranteed: Sodium fluoride 68.00 per cent; sodium acid fluoride 8.00 per cent; all closed dalmatian flowers 7.20 per cent.

Found: Total fluorine 38.33 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 63 (1927).

## COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

#### Schrader's Argentine Ant Syrup.

(SCHRADER CHEMICAL CO., SAN FRANCISCO, CAL.)

Guaranteed.

Found. 0.20

## Schrader's Red Ant Powder.

(SCHRADER INSECT POWDERS CO., SAN FRANCISCO, CAL.)

Guaranteed: Sodium fluoride 55.00 per cent; sodium acid fluoride 10.00 per cent; all closed dalmatian flowers 20.00 per cent; inert matter 15.00 per cent.

Found: Total fluorine 30.14 per cent; organic matter 21.13 per cent.

Cal. Dept. Agr., Spec. Pub. 66, 37 (1926).

## Scott & Gilbert Sheep Dip.

See "Phenol Soap Solutions."

#### Scrofularia.

(BENJ. HAMMOND, FISHKILL-ON-HUDSON, N. Y.)

Found: Moisture 4.91 per cent; ash 19.52 per cent; ether extract 7.35 per cent. Substance is a mixture of tobacco and pyrethrum colored with lead chromate.

U. S. D. A., Bur. Chem., Bull. 68, 49 (1902).

## Selenine.

(CHARLES DICKENS, OAKLAND, CAL.)

Found: Barium polysulphide 4.33 per cent; barium thiosulphate 0.39 per cent; selenium 0.23 per cent; tellurium trace.

Cal. Dept. Agr., Spec. Pub. 58, 50 (1925).

#### Semesan.

(E. I. DUPONT DE NEMOURS & CO., INC., WILMINGTON, DEL.).

Guaranteed: Hydroxymercurichlorphenol 35 per cent.; inert matter

65.00 per cent.

Found: Mercury 21.32 per cent; hydroxymercurichlorphenol 36.68 per cent; ash 66.30 per cent; calcium oxide 16.72 per cent; silica 0.68 per cent; iron and aluminum oxides 0.35 per cent; magnesium oxide 0.35 per cent; carbon dioxide 16.79 per cent. The mercury compound is para-hydroxymercuri-orthochlorphenol, probably present as the sodium salt. The inert matter is probably a mixture of calcium hydroxide and sodium carbonate.

Conn. Agr. Expt. Sta., Sa nple 8932.

#### Semesan Bel.

(E. I. DUPONT DE NEMOURS & CO., INC., WILMINGTON, DEL.). Guaranteed: Hydroxymercurichlorphenol 10.00 per cent. Found: Mercury 6.52 per cent.

Cal. Dept. Agr. Spec. Pub. 75, 62 (1927).

## Sergeant's Disinfectant.

See "Phenol Soap Solutions."

69.69

#### 70-10-20.

(J. R. GILLAM & BRO., BURLINGTO	ON, N. J.)	
Total arsenic, metal	Guaranteed. 1.80	Found. 1.80
Water-soluble arsenic, metal	68.00	0.30

N. J. Agr. Expt. Sta., Bull. 441, 11 (1926).

## 70-10-20 Mixture.

See "Niagara Mixture No. 150."

# 70-10-20 Sulfur Lead-Lime Dust.

(LUCAS-KIL-TONE CO., VINELAND, N. I.)

Guaranteed: Sulphur 70.00 per cent.

Found: Sulphur 66.48 per cent; total arsenic 1.93 per cent; watersoluble arsenic 0.26 per cent.

N. J. Agr. Expt. Sta., Bull. 459, 11 (1927).

## 70-20-10 Dusting Mixture.

(MECHLING BROS. CHEMICAL CO., CAMDEN, N. J.)

	Guaranteed.	Found.
Total arsenic, metal	2.00	2.65
water-soluble arsenic, metal	0.50	0.12
Sulphur	69.00	65.47
N I Am Embt Sta D. II 150 10 (1005)	00.00	00.47
N. J. Agr. Expt. Sta., Bull. 459, 12 (1927).		

#### 75-5-20

(J. R. GILLAM & BRO., BURLINGTON, N. I.)

Found: Total arsenic 1.28 per cent; water-soluble arsenic 0.26 per cent; sulphur 72,00 per cent.

N. J. Agr. Expt. Sta., Bull. 441, 11 (1926).

#### S & G Poultry Lice Powder.

See "Sodium fluoride."

#### Sheep Dip.

See "Phenol Soap Solutions."

## Simplex.

(H. G. SMITH CO., UTICA, N. Y.)

기계가 가는 아니다 그리지 않는데 되어 없는데 그리는 아들이 다	Guaranteed.	Found.
Sulphur	. 20.00	21.07
Alseme		2.61
Lead oxide	••••	9.75
Mich. Agr. Coll. Expt. Sta., Spec. Bull. 74, 11	(1915).	

## Sin-O-lor.

(HUGH KNIGHT, RIVERSIDE, CAL.)

Found: Substance is a purple liquid containing nearly 1 per cent potassium permanganate and 0.62 per cent sodium sulphate. Cal. Dept. Agr., Spec. Pub. 34, 58 (1923).

## S. J. Nicotine Dust.

See "Nicotine Dusts."

Skalene.

## COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

#### Skalol.

See "Oil Emulsions, Mineral."

#### Skinner's Arsenate Bordeaux.

See "Bordeaux Mixture-Lead Arsenate."

## Skinner's Special 30-70 Copper Lime Dust.

(SKINNER MACHINERY CO., DUNED	IN, FLA.)	
	Guaranteed.	Found.
Copper	10.00	12.11
N. J. Agr. Expt. Sta., Bull. 407, 16 (1924).		

## Slugall.

(GERMAIN	SEED	&	PLANT	CO.,	LOS	ANGELES,	CAL.)	)

	Guaranteed.	Found.
Nicotine	0.10	0.12
Total arsenious oxide	1.00	0.37
Water-soluble arsenious oxide	0.63	0.25
Naphthalene	5.00	1.10
Cal Debt Agr Spec Pub 75 38 (1927)		

## Slug Shot.

## (HAMMOND'S SLUG SHOT WORKS, BEACON, N. Y.)

(HAMMOND S SECO SHOT WORKS, BEN	10011, 11. 1.)	
	Guaranteed.	Found.
Free sulphur	6.00	
Total arsenic, metal	0.79	0.89
Water-soluble arsenic, metal	trace	0.12
Copper	0.91	1.01
Nicotine	trace	0.04
Crude carbolic acid	0.40	present
Copper sulphate and copper arsenate declared pres	sent.	
C 4 F H C B H 949 150 (1999)		

# Conn. Agr. Expt. Sta., Bull. 242, 159 (1922).

## Slug Slugger.

(MICHEL & PELTON CO.)		
	Guaranteed.	Found.
Calcium arsenate	3.00	4.76
Cal Debt Agr Shec Pub 75 27 (1927)		

#### Small's Snail Poison

	(SMALL'S SEED CO., RIVERSIDE,	CAL.)	
		Guaranteed.	Found.
lcium arsenate.		2.80	3.80

Cal. Dept. Agr., Spec. Pub. 75, 27 (1927).

Ca

#### Snailex.

#### (THE FERTILSPRAY CO. LOS ANGELES CAL.)

(THE PERTILSTRAT CO., EOS ANGELS	es, car.,	
	Guaranteed.	Found.
Ferrous sulphate	20.00	20.06
Ferric sulphate	20.00	32.39
Copper sulphate	0.45	1.25
Cal Dept Agr Spec Pub 75 45 (1927)		

#### Snail Foil.

(AN-FO MFG. CO.)		
	Guaranteed.	Found.
Free sulphur	3.00	2.88
Phenols	2.00	2.28
Sodium fluoride	1.00	1.36
Mineral oil	8.00	8.77
Cal Debt Agr Spec Pub 75 40 (1927)		

#### Snail Poison.

See "Small's Snail Poison."

## Snarol.

(THE ANTROL LABORATORIES, INC., LOS ANGELES, CAL.)

	Guaranteed.	Found.
Calcium arsenate	3.50	5.17
Total arsenic, metal		1.95
Inert matter	96.50	
Conn. Agr. Expt. Sta. Sample 8935		

Soap.

See Table XXIX.

Soap, Sulfo-Tobacco Plant & Animal.

See "Nicotine Soaps."

Soap, Sulpho-Tobacco.

See "Sulpho-Tobacco Soap."

Soap, Tobacco.

See "Nicotine Soaps."

## Soda-Sulfur Compound.

(W. C. COLLINS, SAN FRANCISCO, CAL.)

Found: Water-soluble matter, 68.44 per cent; sodium polysulphide 0.80 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 32 (1923).

Sodium Fluoride.

See Table XXX.

## Sodium Fluosilicate.

(VIRGINIA-CAROLINA CHEMICAL CORP., RICHMOND, VA.)

Found: Sodium fluosilicate 79.39 per cent; sodium fluoride 16.45 per cent; sodium carbonate 0.75 per cent; undetermined 3.41 per cent.

Conn. Agr. Expt. Sta., Sample 7505.

Solbar.

See "Barium Tetrasulphide."

Sol-o-cree.

See "Phenol Soap Solutions."

Soluble Arsenoid.

See "Watson."

Soluble Oil.

Soluble Oil Spray.

See "Oil Emulsions, Mineral."

# Soluble Sulphur Compound.

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

See "Nicotine Dusts."

## Soluble Sulphur Solution.1

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y.)

Guaranteed. Found. Guaranteed. Found.

#### Solution of Cresol.

See "Phenol Soap Solutions."

S. O. S. Sterling Oil Spray.

See "Oil Emulsions, Mineral."

## Sow-Bug Destroyer.

(CHAS. C. NAVLET CO., SAN FRANCISCO, CAL.).

Guaranteed: Copper aceto-arsenite 8.00 per cent. Found: Arsenious oxide 10.94 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 27 (1927).

Sow-Bug-Go.

See "Leinen."

## Sow-Bug-Killer.

(SAN JOSE SPRAY MFG. CO., SAN JOSE, CAL.)

Guaranteed: Copper aceto-arsenite 8.00 per cent. Found: Arsenious oxide 10.76 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 27 (1927).

#### Special Dry Mix.

(THE KIL-TONE CO., VINELAND, N. J.)

Found: Sulphur 54.64 per cent.

N. J. Agr. Expt. Sta., Bull. 441, 7 (1926).

#### Special Filled Capsules No. 173935.

(ELI LILLY & CO., INDIANAPOLIS, IND.)

Found: Nicotine 15.82 per cent.

Cal. Dept. Agr., Spec. Pub. 66, 26 (1926).

# Special Mixture 154A.

See "Nicotine-Sulphur Dusts."

## Special Mixture No. 161 Potato Dust.

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y.)

	Guaranteed.	Found.
Total arsenic, metal	7.40	8.65
Water-soluble arsenic, metal	0.50	0.22
Copper	6.00	7.31

N. J. Agr. Expt. Sta., Bull. 407, 14 (1914).

Special Mixture M 163.

Special Mixture No. 221.

Special Mixture No. 274.

<sup>1</sup> Two Samples.

		S	oap			,				
Manufact	r	otal				pun	ate		puno	
Manufacturer or Distributor and Brand	Guaranteed	Found	Rosin Soap	Fish Oil Soap	Water-Found	Total Alkali Found	Sodium Carbonate	Non-soap Ash	Inert Matter—Found	Publication
California Pest Control Co. Calpest	%	%	%	1 %	1 %	%	1 %	%	%	
California Spray Chemical Co Watson		24.86		7	69.38	2.53			1	Cal. Dept. Agr., Spec. Pub. 75, 54 (1927).
California Spray Chemical Co Watson		13.70			78.401			1.10		Cal. Dept. Agr., Spec. Pub 58, 46 (1925).
Roy E. Campbell. Alhambra Cal	30.00	30.80			61.60					Cal. Dept. Agr., Spec. Pub 75, 54 (1927).
General Chemical Co., New York, N. V.		20.80			35.25		42.40			Cal. Dept. Agr., Spec. Pub 34, 59 (1922).
Germain Seed & Plant Co. Los Angeles										Conn Agr. Expt. Sta., Bull 242, 160 (1922).
Germain Seed & Plant Co. Los Angeles		71.20	none	71.20					23.585	Cal. Dept. Agr., Spec. Pub 51, 40 (1925).
Johnson & Musser Seed Co. Los Angeles		34.70	6.10	28.60					63.90	Cal. Dept. Agr., Spec. Pub 51, 40 (1925).
Langley & Michaels Co. Hard Fish Oil	64.40	64.71 39.60		••••	$     \begin{array}{r}       34.30 \\       29.608     \end{array} $					U. S. D. A. Bur. Chem. Bull. 68, 35 (1902). Cal. Dept. Agr., Spec. Pub.
Manufacturer unknown. Lennox To- bacco <sup>9</sup>										75, 54 (1927). Cal. Dept. Agr., Spec. Pub 34, 47 (1922).

Michel & Pelton Co., San Francisco, Cal.										Cal. Dept. Agr., Spec. Pub.
Mapco Rosin Soap for Spray			43.30	18.30			7.70		29.1010	51, 40 (1925).
Michel & Pelton Co., San Francisco, Cal. Mapco Whale Oil		53.30			43.4011	0.53			,	Cal. Dept. Agr., Spec. Pub. 75, 54 (1927).
Miller Products Co., Portland, Ore., Whale Oil	60.00	57.90				••••				Ore. Agr. Expt. Sta., Cir. 84,
Mt. Hood Soap Co., Portland, Ore. Felbro Whale Oil	58.50	76.10								Ore. Agr. Expt. Sta., Cir. 84,
Chas. C. Navlet Co. Navco Whale Oil		25.90			69.40					0 1 5
Pacific Soap Co. P. S. C. Condensed Rosin Spray	55.00	53.36			33.70	$11.08^{12}$				Cal. Dept. Agr., Spec. Pub. 75, 54 (1927).
Pacific Soap Co. P. S. C. Condensed Rosin Spray	65.00	46.10			35.75	14.20				Cal. Dept. Agr., Spec. Pub. 75, 54 (1927).
Rosin Spray	75.00	52.00			31.7411					
	15.00	10.00		• • • •	16.67	4.37				Cal. Dept. Agr., Spec. Pub. 75, 55 (1927).
Perfecto Spray Mfg. Co., Los Angeles, Cal. Perfecto Rosin Paste		58.25			$35.10^{13}$			3.99		Cal. Dept. Agr., Spec. Pub. 58, 46 (1925).

<sup>&</sup>lt;sup>1</sup>Water guaranteed 67.00%.

<sup>2</sup>Trace of phosphate present.

<sup>3</sup>Two samples of this soap were examined only to see whether they were potassium or sodium soaps.

One was a sodium soap, the dispersion of the disp

<sup>4</sup>Glycerine 4.21 per cent.
Inert matter guaranteed 30.00%.
Fatty anhydrides 25.30 per cent; resin anhydrides 34.67 per cent; sodium oxide 2.92 per cent; potassium oxide 1.82 per cent.
Value guaranteed 35.60%.
No nicotine found.
Inert matter guaranteed 30.00%.
Water guaranteed 40.00%.
Alkali guaranteed 40.00%.
Water guaranteed 36.00%.
Water guaranteed 36.00%.

## Special 210 Dust Mixture.

See "Nicotine Dusts."

# Special Mixture S 155 A.

See "Nicotine-Sulphur Dusts."

## Spratt's Dip.

See "Phenol Soap Solutions."

## TABLE XXX. SODIUM FLUORIDE

		lium oride			
Manufacturer or Distributor and Brand	Guaranteed	Found	Publication		
Danaguia Dana Co	%	%	Cal. Dept. Agr., Spec. Pub. 51,		
Brunswig Drug Co		93.70	59 (1925). Cal. Dept. Agr., Spec. Pub. 75, 62 (1927).		
The Insecticide Co., San Francisco, Cal. Preparation No. 31		93.27	Cal. Dept. Agr., Spec. Pub. 75, 63 (1927).		
C. U. McClellan & Co. McClellan's	95.50	95.31	Cal. Dept. Agr., Spec. Pub. 75, 62 (1927).		
Mefford Chemical Co	90.00	97.56	Cal. Dept. Agr., Spec. Pub. 75, 62 (1927).		
Montgomery, Ward & Co., Portland, Ore	65.00	98.60	Ore. Agr. Expt. Sta., Cir. 84, 15 (1927).		
Scott & Gilbert Co. S&G Poultry Lice Powder	90.00	96.58	Cal. Dept. Agr., Spec. Pub. 75, 62 (1927).		

<sup>&</sup>lt;sup>1</sup> Ash found 5.40 per cent.; "inert fossil flour", guaranteed 7.60 per cent.

## Spratt's Lice Powder.

(SPRATT'S PATENT, LTD.)		
Ash (talc)	Guaranteed. 40.00	Found. 37.06
Cal. Dept. Agr., Spec. Pub. 75, 63 (1927).		

(KIRK, GEARY & CO.)		
Arsenious oxide	Guaranteed. 30.00	Found. 26.74

## Spray Mulsion.

See "Oil Emulsions, Mineral."

			Publication	100000000000000000000000000000000000000	Y. Agr. 384, 298 (	N. Y. Agr. Expt. Sta. 348, 95 (1912).	Cal. Dept. Agr., Spec 34, 59 (1922).	Ore. Agr. Expt. Sta., C 15 (1927).	Conn. Agr. Expt. Sta., ple 3778.	Cal. Dept. Agr., Spec 66, 36 (1926).
	bnuoT—retteM trenI			%						
	Non-soap Ash			%					0.95	1.10
	Sodium Carbonate			%					0.15	
	Total Alkali Found			%	7.2815	5.91			6.56	
	Water—Found			%	28.74	23.73	74.10		30.97	18.50
	Soap		Fish Oil Soap	%	:		i			
			Rosin Soap	%			:			
		Total	Found.	%	66.25	71.12	25.90	30.40	67.57	79.10
			Guaranteed.	%				32.10 30.40		
	Manufacturer or Distributor and Brand			i i	ierson, Albany, N. Y. Fir-Tree	A. Frice, Albany, N. Y. Stott's ree Oil <sup>16</sup>	se Soap Works, San Jose, Cal.	1	7illiams Co., Glastonbury, Conn.	s & Moore, Stockton, Cal. Wil-

cent; unsaponified matter (mostly phenol) 5.01 per cent. free caustic 0.18 per cent.

#### Squirrel Poison.

See "Strychnine Preparations."

# Squirrel Poison-Phosphorus.

See "Phosphorus Preparations."

#### Stallard's New Lice Exterminator.

(MANUFACTURER NOT STATED.)

Found: Moisture 0.16 per cent; free sulphur 73.51 per cent; ferric oxide 23.73 per cent; silica 1.53 per cent.

Cal. Dept. Agr., Spec. Pub. 51, 11 (1925).

#### Star Water.

(MANUFACTURER UNKNOWN.)

Found: Available chlorine 2.64 per cent. Conn. Agr. Expt. Sta., Bull. 258, 377 (1924).

#### Stearns' Electric Rat and Roach Paste.

See "Phosphorus Preparations."

# Steraklene.

(CONNECTICUT CHEMICAL & DISINFECTANT CO.	, NEW HAVEN,	CONN.)
	Guaranteed.	Found.
Sodium hypochlorite	8.00	6.13
Available chlorine		5.86
Inert matter	92.00	
Conn. Agr. Expt. Sta. Sample 9466		

#### Sterilac

(THE STERILAC CO., NORTH CHICAGO, ILL.)

Guaranteed: Active ingredient 94.00 per cent; inactive (mono-hydrated sodium carbonate), 6.00 per cent. "An alkaline chloramine mixture readily soluble in water."

Found: Chloramine T, 94.24 per cent. Conn. Agr. Expt. Sta., Sample 9658.

#### Sterling Argentine Ant Poison.

(WESTERN WHOLESALE DRUG CO., LOS ANGELES, CAL.)

	Guaranteed.	Found
Arsenic, metal	0.18	0.12
Cal. Dept. Agr., Spec. Pub. 58, 17 (1925).		

#### Sterling Oil Spray.

See "Oil Emulsions, Mineral."

#### Sterlingworth Cut Worm Killer.

					A CONTRACTOR OF THE PARTY OF TH
1	CTEDITATO	CITEMICAT	CO	CAMBRIDGE.	MACC

(SIERBING CHEMICHE CO., CHMEKIE	or, minor.	
	Guaranteed.	Found.
Total arsenious oxide	1.00	1.99
Water-soluble arsenic, metal	0.07	1.40
Conn. Agr. Expt. Sta., Bull. 242, 159 (1924).		

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC. Sterlingworth Weed Killer.

(STERLING CHEMICAL CO., CA	MBRIDGE, MASS.)	
	Guaranteed.	Found.
Arsenic oxide	30.00	30.10
Cal Dobt 1 av Shee Pub 31 90 (1093	<b>\</b>	

Stott's Fir-Tree Oil Soap.

See "Soap."

#### Strawberry Weevil Bait.

(M. R. FORSELL, SEATTLE, WASH.)

Found: Total arsenic oxide 1.54 per cent; water-soluble arsenic oxide 0.61 per cent; calcium oxide 0.42 per cent; magnesium oxide 1.42 per cent; moisture 7.24 per cent. Substance is dried apple waste plus magnesium arsenate.

Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).

### Strychnine Preparations.

See Table XXXI.

# Success Special Insecticide.

(VIOLONI INCLUDINI CO., LOS MIN	delles, cal.)	
	Guaranteed.	Found.
Sulphur	6.00	7.20
Sodium nitrate	23.00	22.44
Sodium carbonate	39.90	32.99
Soap	10.00	9.27
Water		27.41
Inert matter		
Cal. Dept. Agr., Spec. Pub. 58, 49 (1925).		

#### Success Tree Spray.

#### (H. D. BLANCHARD, PORTERVILLE, CAL.)

	Guaranteed.	Found.
Water		8.72
Sulphur	20.00	15.63
Sodium nitrate	20.00	19.62
Sodium carbonate	35.00	38.70
Soap	12.00	12.04
Insert matter	11.00	
C 1 D 4 C. D 1 F4 FF (100F)		

Cal. Dept. Agr., Spec. Pub. 51, 55 (1925).

#### Sulco-V. B.

See "Oil Emulsions, Mineral."

#### Sulfene.

### (F. A. FRAZIER CO., SAN FRANCISCO, CAL.)

	Guaranteed.	Found.
Sodium polysulphide	64.00	65.76
Sodium thiosulphate	22.00	25.60
Free sulphur	2.00	0.45
Sodium sulphate	4.00	
Water	8.00	

Cal. Dept. Agr., Spec. Pub. 66, 23 (1926).

## TABLE XXXI. STRYCHNINE PREPARATIONS

	Strych	nnine	
Manufacturer or Distributor and Brand	Guaranteed	Found	Publication
	1		
G. H. Benedict and Co. O. K. Squirrel	%	%	Cal. Dept. Agr., Spec. Pub
Poison <sup>1</sup>	0.40	0.17	75, 46 (1927). Cal. Dept. Agr., Spec. Pub
Brunswig Squirrel Annihilator		0.28	66, 28 (1926).
California Spray Chemical Co., Wat- sonville, Cal. Ortho Gopher Poison California Spray Chemical Co., Wat-	90.00	89.60	Cal. Dept. Agr., Spec. Pub 75, 46 (1927).
sonville, Cal. Ortho Penetrating Poisoned Barley	0.3125	0.443	Cal. Dept. Agr., Spec. Pub 66, 27 (1926).
sonville, Cal. Ortho Penetrating Poisoned Barley	0.3129	0.38	Cal. Dept. Agr., Spec. Pub 66, 27 (1926).
Coffin-Redington Co., Noxem Squirrel and Gopher Poison	0.30	0.29	Cal. Dept. Agr., Spec. Pub 75, 46 (1927).
B. E. Cole, Hollister, Cal. County Farm Bureau Mixture <sup>2</sup>		0.55	Cal. Dept. Agr., Spec. Pub 51, 51 (1925).
Diamond Milling Co., Livermore, Cal.		EXT. 0.000	Cal. Dept. Agr., Spec. Pub
Poisoned Barley	$\begin{bmatrix} 0.41 \\ 0.25 \end{bmatrix}$	$\begin{bmatrix} 0.33 \\ 0.26 \end{bmatrix}$	58, 38 (1925). Cal. Dept. Agr., Spec. Pub 75, 46 (1927).
Herbert F. Dugan. Strychnine Alka-		00.00	Cal. Dept. Agr., Spec. Pub
Dupins Chemical Co., Fort Dodge, Ia.		99.66	75, 46 (1927). Cal. Dept. Agr., Spec. Pub
Gopher-Get-'er <sup>3</sup>	0.25	$\begin{bmatrix} 0.310 \\ 0.37 \end{bmatrix}$	34, 53 (1923). Cal. Dept. Agr., Spec. Pub
Fort Dodge Chemical Co., Fort			75, 46 (1927). Cal. Dept. Agr., Spec. Pub
Dodge, Ia. Gopher Death <sup>4</sup> Germain Seed & Plant Co., Los An-		0.35	34, 53 (1923). Cal. Dept. Agr., Spec. Pub
geles, Cal. Go-For-Gopher <sup>5</sup>	0.31	0.05	51, 51 (1925).
Germain Seed & Plant Co., Los Angeles, Cal. Go-For-Gopher <sup>5</sup>	31.00	0.15	Cal. Dept. Agr., Spec. Pub 51, 51 (1925)
Kirk-Geary Co., Sacramento, Cal. Capital Squirrel Poison	0.24	0.23	Cal. Dept. Agr., Spec. Pub 58, 38 (1925).
Kirk-Geary Co., Sacramento, Cal. Dead Shot Squirrel & Gopher Killer	0.25	0.25	Cal. Dept. Agr., Spec. Pub 58, 38 (1925).
Kirk-Geary Co., Sacramento, Cal.		Lie Miles	Cal. Dept. Agr., Spec. Pub
Dead Shot Squirrel & Gopher Killer Langley & Michaels Co., San Fran-	0.24	0.21	58, 38 (1925).
cisco, Cal. Hall's Lightning Squirrel and Gopher Poison	0.3476	0.326	Cal. Dept. Agr., Spec. Pub 66, 28 (1926).

<sup>1</sup>Cyanide, guaranteed 0.10 per cent; found trace.

<sup>2</sup>Sample is poisoned wheat sweetened with saccharin.

<sup>3</sup>Sample is a poisoned mixture of raisins and figs.

<sup>4</sup>Tablets sweetened with sugar and having an odor of anice.

<sup>5</sup>Sample is poisoned raisins.

<sup>6</sup>Strychnine sulphate.

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC. TABLE XX XI. STRYCHNINE PREPARATIONS—Concluded.

	Stryc	hnine	
Manufacturer or Distributor and Brand	Guaranteed	Found	Publication
2 Nr. 1 1 C C B	%	%	
Langley & Michaels Co., San Francisco, Cal. Hall's Lightning Squirrel and Gopher Poison John F. Leinen Chemical Co., San	0.326	0.306	Cal. Dept. Agr., Spec. Pub. 66, 28 (1926).
Francisco, Cal. Leinen's Poisoned Barley	0.3125	0.449	Cal. Dept. Agr., Spec. Pub. 66, 28 (1926).
Francisco, Cal. Leinen's Poisoned Barley	0.40625	0.36	Cal. Dept. Agr., Spec. Pub. 66, 28 (1926).
Francisco, Cal. Morse's Gopher Poison	0.3125	0.27	Cal. Dept. Agr., Spec. Pub. 75, 46 (1927).
Francisco, Cal. O-Kay Gopher Poison	0.25	0.23	Cal. Dept. Agr., Spec. Pub. 75, 46 (1927).
Francisco, Cal. O-Kay Poisoned Wheat	0.25	0.24	Cal. Dept. Agr., Spec. Pub. 58, 39 (1925).
Long & Gretter, Monterey, Cal. Long and Gretter's Squirrel Poison	0.20	0.324	Cal. Dept. Agr., Spec. Pub. 66, 28 (1926).
Roy Specialty Co. El Roy Gopher Poison <sup>7</sup> Scott & Gilbert Co., San Francisco,		0.25	Cal. Dept. Agr., Spec. Pub. 51, 51 (1925). Cal. Dept. Agr., Spec. Pub.
Cal. Cespi Poisoned Barley Scott & Gilbert Co., San Francisco,	0.30	0.288	66, 28 (1926). Cal. Dept. Agr., Spec. Pub.
Cal. Cespi Poisoned Wheat Geo. Z. Wait Co. Wait's Squirrel and	0.30	0.33	Cal. 75, 46 (1927). Cal. Dept. Agr., Spec. Pub.
Western Wholesale Drug Co. Western	0.20	0.21	58, 40 (1925). Cal. Dept. Agr., Spec. Pub.
W. M. Willett, San Francisco, Cal.	0.27	0.26	66, 28 (1926).
Wakelee's Squirrel & Gopher Poison	0.25	0.61	Cal. Dept. Agr., Spec. Pub. 58, 40 (1925).
Williams & Moore. Moore's Prepared Squirrel Poison	0.32	0.33	Cal. Dept. Agr., Spec. Pub. 75, 46 (1927).
C. G. Woods Chemical Co. Gopher Scent. C. G. Woods Chemical Co. Rat Scent	$0.20 \\ 0.20$	0.30 0.29	Cal. Dept. Agr., Spec. Pub. 75, 46 (1927). Cal. Dept. Agr., Spec. Pub.
			75, 46 (1927).

<sup>6</sup>Strychnine sulphate <sup>7</sup>Sample is a mixture of poisoned raisins, wheat, etc. <sup>8</sup>Arsenic guaranteed 0.75 per cent; found 0.57 per cent.

#### Sulfo.

(MILLER PRODUCTS CO., PORTLAND, ORE.)

CONNECTICUT EXPERIMENT STATION

Guaranteed. Found. Sulphur..... 90.00 85.50 Ore. Agr. Expt. Sta., Cir. 84, 11 (1927).

#### Sulfocide.

(B. S. PRATT CO., NEW YORK, N. Y.)

Two samples:

## Sample 18457.

Guaranteed: 30 per cent sulphur as sodium polysulphide and thio-

Found: Total sulphur 29.25 per cent; thiosulphate sulphur 1.84 per cent; sulphate sulphur 0.10 per cent; sulphide sulphur 27.31 per cent. Conn. Agr. Expt. Sta., Bull. 242, 160 (1922).

Sample 2506.

Guaranteed: Sodium polysulphide 39-40 per cent; sodium thiosulphate

1-2 per cent; inert matter 58-60 per cent.

Found: Total sulphur 33.78 per cent; monosulphide sulphur 7.79 per cent; thiosulphate sulphur 2.56 per cent; sulphate sulphur 0.25 per cent; polysulphide sulphur 23.18 per cent; sodium polysulphide (calculated as the pentasulphide), 38.48 per cent.

Conn. Agr. Expt. Sta., Bull. 272, 146 (1926).

#### Sulfodust.

See "Sulphur."

Sulfo-Tobacco Plant and Animal Soap.

See "Nicotine Soaps."

Sulfur.

See "Sulphur."

Sulfur Dusting Mixture Formula No. 5.

See "Naco."

Sulfur-Lime.

See the following:

Dry Mix Sulfur-Lime.

80-20.

Kasulime.

Leinen's Mildew-Go.

Mechling's Dry Mix.

New Jersey Dry Mix Sulfur-Lime.

Niagara Dry Mix.

Sulfur-Nicotine Compound.

See "Nicotine-Sulphur Dusts."

# Sulpho-Arsenate Mixture.

(THE KIL-TONE CO., VINELAND,	N. J.)	
	Guaranteed.	Found.
Sulphur	68.00	68.57
Total arsenic, metal		2.47
Water-soluble arsenic, metal		0.19
N. J. Agr. Expt., Sta., Bull. 407, 13 (1924).		

COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

# Sulpho-Arsenate Powder.

(THE KIL-TONE CO., VINELAND,	N. J.)		
(See also "Green Cross")			
	Guaranteed.		Found.
Total arsenic, metal	9.48		9.69
Water-soluble arsenic, metal	0.66	V.S	0.19
Sulphur	48.50		49.73
N. J. Agr. Expt. Sta., Bull. 441, 12 (1926).			

#### Sulphur Carbolate of Lime.

See "Whitney."

# "Sulpho-Tobacco" Soap.

(R. W. HUNT, SAN JOSE, CAL.)

	Guaranteccu.	Touriu.
Nicotine	0.18	0.03
Sulphur	3.88	0.78
Cal. Debt. Agr., Spec. Pub. 34, 47 (1923).		

# Sulpho-Tone.

(LUCAS-KIL-TONE CO., VINELAND, N. J.)

Guaranteed: Sulphur 60.00 per cent.

Found: Sulphur 62.48 per cent; coarser than 100 mesh, 0.86 per cent; passes 100 mesh, 99.14 per cent; 200 mesh, 57.37 per cent; 300 mesh, 14.75 per cent.

N. J. Agr., Expt. Sta., Bull. 459, 8 (1927).

#### Sulphur.

See Table XXXII.

# Sulphur-Arsenate Dust 83-15. (IOHN BACON, GASPORT, N. Y.)

	Guaranteed.	Found.
Total arsenic, metal	2,80	2.44
Water-soluble arsenic, metal	0.15	0.14
Sulphur	83.00	84.75
Conn. Agr. Expt. Sta., Bull. 272, 147 (1925).		

#### Sulphur-Arsenate Dust 90-10. (TOTTAL DAGGE GAGDODE NEW )

(JOHN BACON, GASPORI. N.	Y.)	
	Guaranteed.	Found.
Total arsenic, metal	1.90	1.86
Water-soluble arsenic, metal	0.10	0.11
Sulphur	88.50	88.27

Conn. Agr. Expt. Sta., Bull. 272, 147 (1925).

	Sul	ohur				
Manufacturer or Distributor and Brand	Guaranteed	Found	Water	Ash	Type	Publication
Batelle & Rennick, New York, N. Y.	%	%	%	%		
99½% Pure Superfine Commercial Cabco Supply Stores. Cabco Flour Cabco Supply Stores. Cabco	90.00	$\begin{array}{c} 99.70 \\ 91.40^{2} \\ 93.50 \end{array}$			• • • •	N. J. Agr. Expt. Sta., Bull. 459, 8 (1927) Cal. Dept. Agr., Spec. Pub. 75, 28 (1927) Cal. Dept. Agr., Spec. Pub. 75, 28 (1927)
California Associated Buyers Co., Fresno-Selma, Cal. "Lilly"	100.00	99.86	0.12	0.02	Sublimed	Cal. Dept. Agr., Spec. Pub. 58, 23 (1925)
California Rex Spray Co. Rex Sulfur Paste	46.00	48.92				Cal. Dept. Agr., Spec. Pub. 34, 57 (1922
California Spray Chemical Co., Watsonville, Cal. Ortho Dusting	99.00	99.32	0.09	0.12	Milled	Cal. Dept. Agr., Spec. Pub., 58, 23 (1925
California Spray Chemical Co., Watsonville, Cal. Ortho-Milled <sup>3</sup>	85.00	87.29		,		Cal. Dept. Agr., Spec. Pub. 34, 26 (1922
California Spray Chemical Co., Watsonville, Cal. Ortho Soil		99.50	0.33	0.10	Milled	Cal. Dept. Agr., Spec. Pub. 58, 23 (1925
California Spray Chemical Co., Watsonville, Cal. Ortho	99.00	99.48	0.45	0.08	Milled	Cal. Dept. Agr., Spec. Pub. 58, 23 (1925
California Spray Chemical Co., Watsonville, Cal. Ortho Wettable		92.65 99.69	1.83 0.21	3.52 0.10		Cal. Dept. Agr., Spec. Pub. 58, 23 (1925 Cal. Dept. Agr., Spec. Pub. 34, 26 (1922
D. F. DeBernardi & Co. L. Chambon Fils	100.00	99.88	0.12	none	Sublimed	Cal. Dept. Agr., Spec. Pub. 58, 23 (1925
F. A. Frazier Co. Frazier's B-7 Dusting <sup>5</sup>	100.00	99.67	0.21	0.12	Milled	Cal. Dept. Agr., Spec. Pub. 51, 21 (1925)

			elementario este a maiori.		-	
F. A. Frazier Co. Frazier's Dusting Sulphur B-6	90.00	91.62	0.22	8.16	Milled	Cal. Dept. Agr., Spec. Pub. 58, 23 (1925).
General Chemical Co., San Francisco, Cal. Orchard Brand Atomic	45.00	46.10	52.15			Cal. Dept. Agr., Spec. Pub. 75, 28 (1927).
General Chemical Co., San Francisco, Cal. Orchard Brand Dritomic Granucci Bros., San Francisco, Cal.	90.00	91.20	0.80			Cal. Dept. Agr., Spec. Pub. 75, 28 (1927).
Columbus Brand		$97.15^{6}$			Milled	Cal. Dept. Agr., Spec. Pub. 58, 23 (1925).
Columbus Sublimed		99.40	0.33	0.28	Sublimed	Cal. Dept. Agr., Spec. Pub. 58, 23 (1925).
Columbus Ventilated	z.:	99.70	0.14	0.16	Milled	Cal. Dept. Agr., Spec. Pub. 58, 23 (1925).
N. J. Square Brand H.&H. Colloidal Holland & Holland XXXXX Brand	45.00	55.22				N. J. Agr. Expt. Sta., Bull. 459, 9 (1927).
Resublimed Flowers	100.00	99.50	0.30	0.20		Cal. Dept. Agr., Spec. Pub. 75, 28 (1927).
Soufre Sublimed Flowers	100.00	99.60	0.40			Cal. Dept. Agr., Spec. Pub. 75, 28 (1927).
Ground		98.00 36.30	0.23	0.61	Milled	Cal. Dept. Agr., Spec. Pub. 58, 23 (1925). Conn. Agr. Expt. Sta., Bull. 258, 371
Manufacturer unknown. Superfine <sup>7</sup>	2.330					(1924). Conn. Agr. Expt. Sta., Sample 6914.
Mechling Bros. Chemical Co., Cam-						
den, N. J. Sulfur Dust <sup>8</sup>		99.50			1	N. J. Agr. Expt. Sta., Bull. 459, 8 (1927).

<sup>&</sup>lt;sup>1</sup>Passes 100 mesh 97.2 per cent; 200 mesh 43.5 per cent; 300 mesh 14.5 per cent.

<sup>2</sup>Calcium hydroxide, guaranteed 10.00 per cent; found 7.14 per cent.

<sup>3</sup>Inert matter guaranteed 15.0 per cent.

<sup>4</sup>Passes 200 mesh 97.0 per cent.

<sup>5</sup>Passes 200 mesh 95.0 per cent.

<sup>6</sup>Copper sulphate 2.73 per cent.

<sup>7</sup>Passes 300 mesh 100 per cent.

<sup>8</sup>Passes 100 mesh 99.0 per cent; 200 mesh 24.6 per cent; 300 mesh 9.2 per cent.

J	
0	
32	
200	
out.	
200	
-	
222	
COMPOSITION OF	
~	
800	
1	
100	
COMMERCIAL	
_	
1892	
86	
-	
2	
-	
88	
=	
Ľ	
É	
j	
Ĵ	
Ĭ	
JE	
) IDE	
SE CIT	
SECI	
YDES.	
YDES.	
IDES.	
IDES. H	
IDES. E	
IDES. EX	
INSECTICIDES. ET	

	Sul	phur				
Manufacturer or Distributor and Brand	Guaranteed	Found	Water	Ash	Type	Publication
Meyer, Wilson & Co., San Francisco,	%	%	%	%		
Cal. Flowers of Sulphur K. R Murphy Oil Co., East Whittier, Cal. Chas. C. Navlet Co., Berkeley, Cal.	99.00	99.10 99.73	$0.90 \\ 0.27$	none		Cal. Dept. Agr., Spec. Pub. 75, 28 (1927). Cal. Dept. Agr., Spec. Pub. 34, 26 (1922).
Neutral Brand Dusting	99.00	98.20	0.30	1.10		Cal. Dept. Agr., Spec. Pub. 75, 28 (1927).
Chas. C. Navelt Co., Berkeley, Cal. Neutral Sulphur, Superfine Flour Niagara Sprayer Co., Middleport, N.	99.50	99.13	0.24	0.63	Milled	Cal. Dept. Agr., Spec. Pub. 58, 23 (1925).
Y. Niagara Dusting Sulphur. Sulfodust <sup>9</sup>	93.00	92.40				Conn. Agr. Expt. Sta., Bull. 258, 371 (1924).
Niagara Sprayer Co., Middleport, N. Y. Niagara Kolodust	90.00	90.10				Cal. Dept. Agr., Spec. Pub. 75, 28 (1927).
Y. Niagara Superfine Dusting	99.50	99.70	0.20	0.10		Cal. Dept. Agr., Spec. Pub. 75, 28 (1927).
Niagara Sprayer Co., Middleport, N. Y. Vesuvius Sublimed Flowers Dr. O'Toole, Sacramento, Cal	100.00	99.85 99.46	0.09 0.10	$0.06 \\ 0.44$		Cal. Dept. Agr., Spec. Pub. 66, 21 (1926). Cal. Dept. Agr., Spec. Pub. 34, 26 (1922).
Pacific Guano & Fertilizer Co., Berkeley, Cal. Cropmaker Flowers Pacific Guano & Fertilizer Co., Berkeley	100.00	99.50	0.40	0.10		Cal. Dept. Agr., Spec. Pub. 75, 28 (1927).
ley, Cal. Producer Flowers Pacific Guano & Fertilizer Co., Berke-	100.00	99.64	0.25	0.11		Cal. Dept. Agr., Spec. Pub. 75, 28 (1927).
ley, Cal. Sunkist Dusting	99.00	97.45	0.35	1.60		Cal. Dept. Agr., Spec. Pub. 75, 28 (1927).

	SET ELECTIVE					
Pascal, Dubedat & Co., P. D. C. Rooster Flowers	99.62	99.44 99.57 99.71	0.39 0.38 0.08	0.17 $0.05$ $0.21$	Sublimed	Cal. Dept. Agr., Spec. Pub. 66, 21 (1926). Cal. Dept. Agr., Spec. Pub. 58, 24 (1925). Cal. Dept. Agr., Spec. Pub. 66, 21 (1926).
cisco, Cal. Alfa Brand Dry Wettable Flowers	90.00	92.50			••••	Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
cisco, Cal. Anchor Sublimed Velvet Flowers	100.00	99.80	0.20			Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
100 % Pure	100.00	100.00			Sublimed	Cal. Dept. Agr., Spec. Pub. 58, 24 (1925).
San Francisco Sulphur Co., San Francisco, Cal. Diamond S Flour San Francisco Sulphur Co., San Fran-	100.00	99.65	0.23	0.12	Milled	Cal. Dept. Agr., Spec. Pub. 58, 24 (1925).
cisco, Cal. Eagle Sublimed Flowers	100.00	99.38	0.22	0.40		Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
San Francisco Sulphur Co., San Francisco, Cal. Electric	99.80	99.85	0.02	0.11		Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
cisco, Cal. Owl Superfine Flour	99.50	98.69	0.18	0.56		Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
San Francisco Sulphur Co., San Francisco, Cal. Red Flag Velvet Flowers San Francisco Sulphur Co., San Fran-	100.00	99.64	0.28	0.09	Sublimed	Cal. Dept. Agr., Spec. Pub. 58, 24 (1925).
cisco, Cal. Swan Brand Superfine Ventilated Dusting	98.50	98.13	0.30	1.57		Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
San Francisco Sulphur Co., San Francisco, Cal. Ventilated <sup>10</sup> San Francisco Sulphur Co., San Fran-		99.59	0.25	0.16		Cal. Dept. Agr., Spec. Pub. 34, 26 (1923.)
cisco, Cal. XL Sublimed	99.80	99.83	0.07	0.10	Milled	Cal. Dept. Agr., Spec. Pub. 66, 21 (1926).
The Sherwin-Williams Co., Cleveland, Ohio. Sherwin-Williams Dusting.	100.00	99.80	0.07	0.13	Milled	Cal. Dept. Agr., Spec. Pub. 58, 24 (1927).

 $<sup>^9</sup> Inert$  matter guaranteed 7.00 per cent.  $^{10} Passes\ 200$  mesh 97.00 per cent.

359

	Sul	Sulphur				
Manufacturer or Distributor and Brand	Guaranteed	Found	Water	hsA	Type	Publication
The Sherwin-Williams Co., Cleveland, Ohio Sherwin-Williams Illtra	%	%	%	%		
Brand Flowers. The Sherwin-Williams Co., Cleveland,	100.00 99.59	99.59	0.22			Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
Onio. Sherwin-Williams Zephyr Brand Flowers. Stauffer Chemical Co. Anchor Velvet	100.00	99.43	0.38	0.20		Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
Flowers. T. D. Urbahns Sacramento Cal	100.00 99.60	99.60	0.10	0.30		Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
Ortho Sprays—Milled <sup>11, 12</sup>	85.00	85.00 90.24	2.03			Cal. Dept. Agr., Spec. Pub. 34, 26 (1922).
	70.00 71.84	71.84				Cal. Dept. Agr., Spec. Pub. 34, 26 (1922).
Bear Brand Bleaching. Western Sulphur Co. Oakland Cal.	100.00 99.50	99.50	0.50			Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
Cream of Sulphur	90.00	90.13	0.51	7.83	Milled.	Cal. Dept. Agr., Spec. Pub. 58, 25 (1925).
Poppy Dusting Co. New York N V	98.00	80.66 00.86	0.15	0.78		Cal. Dept. Agr., Spec. Pub. 75, 29 (1927).
Sulphur Dust 100%-300 Mesh <sup>44</sup> .  Yolo County Purchasing & Selling						Conn. Agr. Expt. Sta., Sample 3620.
Corp., Woodland, Cal. Orthons	99.00	99.00 99.46 0.32	0.32	0.23		Cal. Dept. Agr., Spec. Pub. 34, 26 (1923).
				The state of the s		

thiosulphate 4.58 per cent; gypsum 1.63 per cent; casein, 0.46 per cent. r cent. aranteed: calcium trut matter guaranteed sses 200 mesh 100 per sses 200 mesh 83.16 r

# Sulphur Compound.

(BOGART CHEMICAL CO., NEW YORK, N. Y.)

(See also "Agricultural Sulphur Compound" and "Niagara Soluble Sulphur Compound").
Found: Soluble sulphur 9.35 per cent.

N. Y. Agr. Expt. Sta., Bull. 348, 93 (1912).

#### Sulphur Dust.

(NIAGARA DUST CO., KENTVILLE, N. S., CANADA.)

Found: Sulphur 78.60 per cent; total arsenic oxide 4.74 per cent; water-soluble arsenic oxide 0.06 per cent; lead oxide 9.75 per cent; colloidal clay (Bentonite) present.

Canada Dept. Agr., Div. Chem., Rept. Dominion Chemist (1928).

# Sulphur Smoke Dusting Mixture

See "Naco."

# Sulphur with 7 Percent Nicotine Solution.

# Sulphur with 10 per cent Nicotine Solution.

See "Nicotine-Sulphur Dusts."

#### Summer Mulsion.

#### Summer Oil.

See "Oil Emulsions, Mineral."

# "Sun" Argentine Ant Poison.

(SUN DRUG CO., LOS ANGELES, CAL.)

	Guaranteed.	Found.
Total arsenic, metal	0.20	0.30
Water-soluble arsenic, metal		0.30
Cal Dept Agr Spec Pub 34 18 (1923)		

#### Sun Miscible Oil.

#### Sunoco Spray Oil.

See "Oil Emulsions, Mineral."

#### Sure Death Rat Killer.

(PFEIFFER CHEMICAL CO.)

Found: Arsenious oxide 97.22 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 27 (1927).

#### Sure Death to Insects.

(H. C. DUSENBERRY, NEW YORK, N. Y.)

Found: Substance is gasoline possibly containing oil of citronella. U. S. D. A., Bur, Chem., Bull. 68, 56 (1902).

# Sure Destruction for Cockroaches and Ants.

(INSECT EXTERMINATOR MFG. CO., COUNCIL BLUFFS, IOWA). Found: Borax 99.50 per cent; balance pink coloring matter. U. S. D. A., Bur. Chem., Bull. 68, 43 (1902).

## Sure Kill Rat and Roach Paste.

See "Phosphorus Preparations".

## Sure-Noxem.

(	DEVOE	8	RAYNOLDS.	CHICACO	TLT.
,	(22,02	cc	TITT TTI OFFIDE	CHICACO.	ILLIL 0 /

	Guaranteed.	Found.
Arsenious oxide	2.00	2.10
Sulphur	6.00	16.60
Ore Agr Expt Sta Cir 64 11 (1025)		

#### Sure-Shot.

See "Oils, Mineral."

S. W. Sheep Dip.

See "Phenol Soap Solutions."

#### T

## Talbot's Bed Bug Powder.

(TALBOT MFG. CO.)

Guaranteed: Sodium acid fluoride 8.00 per cent. Found: Sodium fluoride 58.46 per cent; ash 4.67 per cent. Cal. Dept. Agr., Spec. Pub. 75, 63 (1927).

#### Talcimized Sodium Fluoride.

See "Sodium Fluoride."

#### Tar Acids and Petroleum Oils.

(R. R. ROGERS CHEMICAL CO.)

Guaranteed: Phenols 20.00 per cent.

Found: Phenols 21.60 per cent; oils 77.35 per cent; non-soap ash 0.02 per cent; moisture 0.40 per cent. Constants of separated oil: Baumé gravity 23.3; unsulphonated oil, 44.40 per cent.

Cal. Dept. Agr. Spec. Pub. 51, 44 (1925).

#### Thrip Juice.

See "Nicotine Soaps."

#### Thymo-Cresol.

See "Phenol Soap Solutions."

#### Tiger Paste.

See "Phosphorus Preparations."

#### Tizit Tree Spray.

## (TIZIT SPRAY MFG. CO., LOS ANGELES, CAL.

	Guaranteed.	Found.
Total arsenic, metal	4.00	3.41
Water-soluble arsenic, metal	0.20	2.25
Lead oxide	16.00	15.55
Sulphur	22.00	23.00
Sodium carbonate	16.00	17.08
Sodium oleate	25.00	24.98
Inert matter	17.00	

Cal. Dept. Agr., Spec. Pub. 51, 54 (1925).

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

#### Tobacco Dusts.

See Table XXXIII.

## Tobacco, Ground.

See "Tobacco Dusts."

# Tobacco Naphtholene Mixture.

(H. A. STOOTHOFF CO., MT. VERNON, N. Y.)

	Guaranteed.	Found.
Nicotine	0.50	1.02
N. Y. Agr. Expt. Sta., Bull. 384, 297 (1914)		

# Tobacco Soap.

See "Nicotine Soaps."

#### Tobacco Tea.

(INDIGO BLUING CO)

Found: Nicotine 0.28 per cent. Cal. Dept. Agr., Spec. Pub. 34, 52 (1923).

### "To-Bak-ine" Liquid.

See "Nicotine Sulphate Solutions."

# Todco, Argentine Ant Poison.

(OWL DRUG CO., SAN FRANCISCO, LOS ANGELES, CAL.) Guaranteed: Arsenic 0.20 per cent. Found: Arsenic, 0.15 per cent; cane sugar 59.24 per cent. Cal. Dept. Agr., Spec. Pub. 58, 17 (1925).

# Tonicide.

See "Blue Label."

#### Tree Paint.

(HOOD RIVER SPRAY CO., HOOD RIVER, ORE.)

	Guaranteed.	Found.
Copper	1.00	0.70
Ore. Agr. Expt. Sta., Cir. 84 15 (1927)		

#### Tree Spray.

See "Tizit" and "Victory."

#### Treevax.

(TREE VAX CHEMICAL CO., HICKSVILLE, OHIO).

Found: Sulphur 62.00 per cent; potassium nitrate 27.00 per cent; ferric oxide 11.00 per cent.

Mich. Agr. Coll. Expt. Sta., Spec. Bull. 74, 11 (1915).

#### Trelife.

# (THE FERTILSPRAY CO., LOS ANGELES, CAL.)

Form	Guaranteed.	Found.
Ferrous sulphate	20.00	20.98
		36.85
Copper sulphate. Cal. Dept. Agr., Spec. Pub. 75, 45 (1927).	0.45	1.27

Dust	
TOBACCO	
E XXXIII.	
TABLE	

Nicotine 120.	Manufacturer or Distributor and Brand  Ouaranteed  Pound  Wicotine-sulpha	% % %		panta Kosa, Cal		Corona Chemical Co., Newark, N. J 0.50 0.82	v. Dosch 1.90 0.	ster, Pa   1.00   0.	Impco	$0.50 \mid 0.50 \mid 0.50$	San Brancisco Cal 1 16 9 00	Cal 2.78 5.00	0.50 1.30	0.50   0.50	0 U.	, 1.	Montgomery, Ward & Co., Portland, Ore	1.00
ųs:	Inert Matter, Guaranteed Passing 200 me	% %								3.50	08.00		9.50		)0.96 			
	· Publication		1. Agr. Expt. Sta., Sample 471	Cal. Dept. Agr., Spec. Pub. 34, 50 (1923).	1. Agr. Expt. Sta., Bull. 242, 155	Agr., Spec. Pub.	Agr., Spec. Pub. 34, 50 (	I. Agr. Expt. Sta Bull.		Cal. Dept.	Conn. Agr. Expt	Dept. Agr., Spec. 1 up. 34, 50	Dept. Agr., Spec. Pub. 34, 50 (	N. J. Agr. Expt. Sta., Bull. 441,	Col Deat Act Sta., Bull. 272, 148	Cal. Dept. Agr., Spec. Pub. 34, 50 (1923).	ta., C	Cal. Dept. Agr., Spec. Pub. 75, 54 (1927).  Cal. Dept. Agr., Spec. Pub. 34, 50 (1923).

CONNECTICUT EXPERIMENT STATION

Total ash 39.60 per cent; acid-insoluble ash 15.30 per cent; total nitrogen 1.99 per cent.

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC. Trepep.

(PERFECTO SPRAY MFG. CO., LOS ANGELES, CAL.)

Guaranteed: Water 80.00 per cent. Found: Water 83.60 per cent; soap 3.69 per cent; nicotine 0.26 per cent; copper 0.23 per cent; protein 5.13 per cent; phenols 0.90 per cent; calcium oxide 4.70 per cent; sulphur 0.79 per cent; free ammonia 0.36 per cent. Cal. Dept. Agr., Spec. Pub. 66, 37 (1926).

#### Triona.

#### Triumph.

See "Oil Emulsions, Mineral."

#### Tuber Tonic.

(SHERWIN-WILLIAMS CO., CLEVELAND, OHIO.)

Found: Total arsenic 23.40 per cent; water-soluble arsenic 0.86 per cent; copper 19.70 per cent.

N. J. Agr. Expt. Sta., Bull. 459, 16 (1927).

#### Tux-E-Do Mixture.

(SCHOONMAKER & SON, CEDAR HILL, N. Y.)

Found: Total arsenious oxide 2.69 per cent; water-soluble arsenious oxide 1.77 per cent; cupric oxide 5.90 per cent.

N. Y. Agr. Expt. Sta., Bull. 348, 90 (1912).

#### Twenty Plus Disinfectant.

(MICHEL & PELTON CO., SAN FRANCISCO, CAL.)

Guaranteed. Found. 11.91 Available chlorine..... 5.67

Cal. Dept. Agr., Spec. Pub. 51, 53 (1925).

#### 210 Dust Mixture Special.

See "Nicotine Dusts."

Tydol 122.

See "Oils, Mineral."

# U.

## U. C. Worm Capsules.

(JONES DRUG STORE, PETALUMA, CAL..)

Found: Nicotine 15.59 per cent.

Cal. Dept. Agr., Spec. Pub. 66, 26 (1926).

#### Universal Brand Neutral Emulsion.

Universal Brand Triona.

Universal Brand Triumph.

Universal Dormant Soluble Oil.

Universal Mealy Bug Spray Oil.

Universal Medol Emulsion.

See "Oil Emulsions, Mineral."

#### University Capsules.

(THE UNIVERSITY CAPSULE CO., SAN FRANCISCO, CAL.)

[1] 문항 [1] 이 전 전 전 전 전 전 전 전 전 전 전 전 전 전 전 전 전 전	,,	
	Guaranteed.	Found.
Nicotine	13.00	14.20
Cal Debt Agr Sher Pub 66 26 (1026)		

# University Poultry Worm Capsules.

(NOVATO PHARMACY, NOVATO, CAL.)

Found: Nicotine 14.81 per cent. Cal. Dept. Agr., Spec. Pub. 66, 26 (1926).

# University Worm Capsules.

(CHICKEN PHARMACY, PETALUMA, CAL.)

27.	Guaranteed.	Found.
Nicotine	13.00	13.39
Cal. Dept. Agr., Spec. Pub. 75, 35 (1927).		

## University Worm Capsules No. 1.

(PETALUMA AVIAN PATHOLOGY LABORATORY, PETALUMA, CAL.

	Guaranteed.	Found.
Nicotine	15.00	14.83
Cal. Dept. Agr., Spec. Pub. 66, 26 (1926)		

## Urania Green.

(IMPORTED FROM GERMANY).

Found: Total arsenious oxide 57.05 per cent; water-soluble arsenious oxide 4.18 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 16 (1923).

## Uspulun.

(THE BAYER CO., INC., NEW YORK, N. Y.)

Guaranteed: Hydroxymercurichlorphenol sulphate 30.00 per cent. Found: Mercury 19.57 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 62 (1927).

#### V.

# Vaporizing and Fumigating Insecticide.

(EASTERN CHEMICAL CO., BOSTON, MASS.)

	Guaranteed.	Found.
Nicotine	37.00	30.87
Camphor	23.00	13.64
Alcohol	32.00	53.49
U. S. D. A., Bur. Chem., Bull. 68, 47 (1902).		

#### Veltha.

(WM. WOOD & SON, WOODGREEN, LONDON, ENG.).

Found: Sand 32.73 per cent; water 27.00 per cent; carbon 2.31 per cent; sulphur trioxide 19.90 per cent; ferrous oxide 17.90 per cent. Substance is partially dehydrated ferrous sulphate with 35.00 per cent sand and carbon.

U. S. D. A., Bur. Chem., Bull. 68, 48 (1902).

#### Vermol.

See "Phenol Soap Solutions."

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC.

#### Verm-O-Spray.

(VERM-O-SPRAY PRODUCTS CO., WEST HAVEN, CONN.)

Found: Specific gravity, 15.6° C., 0.8825; trace of ash; flash point 61.0°C.; fire point 64° C.; salicylate present. Substance is a mixture of kerosene and methyl salicylate.

Conn. Agr. Expt. Sta., Bull. 258, 376 (1924).

#### Victory Tree Spray.

(VICTORY TREE SPRAY CO., LOS ANGE	LES, CAL.)	
	Guaranteed.	Found.
Total arsenic oxide	4.50	5.74
Water-soluble arsenic, oxide	0.50	1.37
Lead oxide	15.00	14.83
Sulphur	25.00	24.85
Sodium oxide	10.00	13.02
Soap		3.80

Cal. Dept. Agr., Spec. Pub. 66, 18 (1926).

#### Vigorol.

(TREE CHEMICAL CO., SAN JUAN BAUTISTA, CAL.).

Guaranteed: Potassium cyanide 5.47 per cent.

Found: Potassium cyanide, 1.66 per cent; nitrate nitrogen 0.31 per cent; ammonia nitrogen 2.24 per cent; organic nitrogen 0.65 per cent; total nitrogen 3.20 per cent; phosphorous pentoxide 3.15 per cent. This is a preparation intended to be injected into the sap of trees. It was found to injure trees, and renewal of license to sell was refused in California.

Cal. Dept. Agr., Spec. Pub. 51, 12, 58 (1925).

#### Vitedust.

(NIAGARA SPRAYER CO., MIDDLEPORT, N. Y.)

	Guaranteed.	Found.
Copper sulphate	11.00	9.40
Lead arsenate	14.00	12.80
Ore. Agr. Expt. Sta., Cir. 64, 11 (1925).		

#### Volck.

#### Vulture Oil.

See "Oil Emulsions, Mineral."

#### W

#### Dr. G. Z. Wait's Sheep Dip.

See "Phenol Soap Solutions."

# Wait's Squirrel and Gopher Poison. Wakelee's Squirrel and Gopher Poison.

See "Strychnine Preparations."

#### Walnut Worm Dust.

See "Ortho Walnut Worm Dust."

#### War on Ants.

(CANNON CHEMICAL CO.)

Guaranteed: Sugar 20.00 per cent. Found: Sodium fluoride 45.27 per cent; organic matter 25.05 per cent; sugar 19.72 per cent.

Cal. Dept. Agr., Spec. Pub. 75, 63 (1927).

See "Phenol." Waters Bros. Crude Carbolic Acid.

Watkin's Germicide Dip and Disinfectant.

Watkin's Germicide Dip and Disinfectant See "Phenol Soap Solutions."

# Watkin's Louse Killer.

(J. R. WATKINS CO.)		
	Guaranteed.	Found.
Nicotine	0.10	0.23
Naphthalene	10.00	10.07
Cal Debt Age Shee Pub 75 11'(1027)		

#### Watson's Soluble Arsenoid.

(JOHN WATSON CO., INC., HOULTON, ME.)

Guaranteed: Sodium arsenate 80.00 per cent; arsenic, metal 32.20 per cent.

Found: Arsenic, metal 40.09 per cent.

Maine Agr. Expt. Sta., Official Inspections 118, 70 (1925).

#### Weedicator.

(WEEDICATOR WEED CO.		
	Guaranteed.	Found.
Arsenious oxide		7.84
Water	70.00	70.00
Cal. Dept. Agr., Spec. Pub. 75, 25 (1927).		

#### Weed Killer.

(CHIPMAN CHEMICAL ENGINEERING CO., BOU	ND BROOK, N.	J.)
	Guaranteed.	Found.
Salts	45.00	44.10
Ore. Agr. Expt. Sta., Cir. 84, 15 (1927).		

#### Weed Killer.

(HOOD RIVER SPRAY CO., HOOD RIVER, ORE.) Found: Arsenious oxide 30.00 per cent.

Ore. Agr. Expt. Sta., Cir. 84, 15 (1927).

#### Weed Killer.

(U. S. SMELTING, REFINING & MINING CO., SALT	LAKE CITY,	UTAH).
	Guaranteed.	Found.
Sodium arsenite	43.00	44.50
Ore. Agr. Expt. Sta., Cir. 84, 15 (1927).		

# Weed-Killer No. 1 Thin Liquor.

(F. W. WAITE, EL CENTRO, CAL.)

Found: Arsenious oxide 1.13 lb. per gal.; sodium chloride, 1.94 lb. per gallon.

Cal. Dept. Agr., Spec. Pub. 34, 20 (1923).

# Weed Killer, No. 2, Thick Liquor.

(F. W. WAITE, EL CENTRO, CAL.)

Found: Arsenious oxide 8.10 lb. per gal.; sodium chloride 0.14 lb. per gallon.

Cal. Dept. Agr., Spec. Pub. 34, 20 (1923).

# COMPOSITION OF COMMERCIAL INSECTICIDES, ETC. Weed Killer, Solid.

(F. W. WAITE, EL CENTRO, CAL.)

Found: Arsenious oxide 1.56 per cent; sodium hydroxide 57.60 per cent; sodium carbonate 38.16 per cent; sodium chloride 1.42 per cent.

Cal. Dept. Agr., Spec. Pub. 34, 20 (1923).

#### Weevil Bait.

See "Strawberry Weevil Bait."

#### Wescco Bleaching Water.

Found: Available chlorine 2.89 grams per 100 cc. Conn. Agr. Expt. Sta., Bull. 258, 377 (1924).

# Western Poisoned Barley.

See "Strychnine Preparations,"

Whale Oil Soan.

See "Soap."

# White Arsenoid (Barium Arsenite).

(ADLER COLD & CHEMICAL WORKS, NEW YORK, N. Y.)

Found: Barium carbonate 44.05 per cent; barium chloride 13.05 per cent; barium oxide 8.18 per cent; free arsenious oxide 27.64 per cent; lead carbonate 1.86 per cent; silica 0.20 per cent; water 4.00 per cent.

Univ. of Calif. Coll. of Agr. Expt. Sta., Bull. 151, 25 (1903).

#### Whitewash.

See "Wyandotte Detergent."

# Whitney's Sulphon Carbolate of Lime.

(WHITNEY CO., NATICK, MASS.)

Found: Sand, 1.14 per cent; carbon dioxide 8.60 per cent; sulphur trioxide 0.16 per cent; calcium oxide 57.42 per cent; magnesium oxide 7.37 per cent; phenol anhydride 1.68 per cent; water and pink dye by difference, 23.63 per cent.

U. S. D. A., Bur. Chem., Bull. 68, 51 (1902).

# Williams and Moore Spray Soap.

See "Soap."

### Winter Mulsion.

See "Oil Emulsions, Mineral."

#### W. & M. Antiseptic Sheep Dip.

See "Phenol Soap Solutions".

# Wyandotte Detergent (Whitewash).

(MANUFACTURER NOT STATED).

Found: Moisture 4.50 per cent; sand 69.22 per cent; sodium carbonate 10.60 per cent; calcium oxide 1.20 per cent; magnesium oxide 0.35 per cent; soap 10.40 per cent; nicotine none. This substance was sold as a wash to prevent sunburn of trees, but was found of no value for this purpose.

Cal. Dept. Agr., Spec. Pub. 34, 59 (1923).

Wyrol.

See "Oils, Mineral."

CONNECTICUT EXPERIMENT STATION

XL all.

(G. H. RICHARDS, LONDON, ENG.)

Found: Nicotine 3.41 per cent. N. Y. Agr. Expt. Sta., Bull. 348, 94 (1912).

XX Mystic XX Spray.

See "Oils, Mineral."

X X X Heavy Emulsion.

X X X Medium Emulsion.

See "Oil Emulsions, Mineral."

Yellow Label Hydroxide Paste.

See "Mechling."

Youell's Rat Snap.

See "Phosphorus Preparations."

Zeno.

See "Phosphorus Preparations."

Zenoleum Lice Killer.

(ZENNER DISINFECTANT CO., DETROIT, MICH.)

Found: Nicotine 0.51 per cent.

Mich. Agr. Coll. Expt. Sta., Spec. Bull. 74, 11 (1915).

Zeno Miscible Oil Spray.

See "Oil Emulsions, Mineral."

Zinc Arsenite.

See "Arsenite of Zinc."

Zinc Bordeaux.

(GENERAL CHEMICAL CO., NEW YORK, N. V.)

(OBTIDITIES CHEMICIES CO., ILE II TOL		
	Guaranteed.	Found.
Total arsenic, metal	10.16	10.22
Water-soluble arsenic, metal	1.00	0.25
Copper	15.70	16.07
Zinc	13.31	14.41
Conn Agr Frest Sta Bull 242 154 (1922)		

BULLETIN 301

FEBRUARY, 1929

# Connecticut Agricultural Experiment Station

New Haven, Connecticut

# **CONTROL STUDIES** ON THE PLUM CURCULIO IN

CONNECTICUT APPLE ORCHARDS

PHILIP GARMAN AND M. P. ZAPPE

The Bulletins of this Station are mailed free to citizens of Connecticut who apply for them, and to other applicants as far as the editions permit.

# CONNECTICUT AGRICULTURAL EXPERIMENT STATION

#### OFFICERS AND STAFF

# BOARD OF CONTROL

	DOARD OF CONTROL
His Excellency,	Governor John H. Trumbull, ex-officio President
George A. Hopson, S. Wm. L. Slate, Direct Joseph W. Alsop Elijah Rogers Edward C. Schneide	Secretary         Mt. Carmel tor and Treasurer           New Haven         Avon           Southington         Middletown
Trancis T. Emcom.	
	STAFF.
E. Administration.	H. JENKINS, PH.D., Director Emerius.  WM. L. SLATE, B.Sc., Director and Treasurer.  MISS L. M. BRAUTLECHT, Bookkeeper and Librarian.  G. E. GRAHAM, In charge of Buildings and Grounds.
Chemistry: Analytical Laboratory.	E. M. Balley, Ph.D., Chemist in Charge. C. E. SHEPARD OWEN L. NOLAN HARRY J. FISHER, A.B. W. T. MATHIS DAVID C. WALDEN, B.S. FRANK C. SHELDON, Laboratory Assistant. V. L. CHURCHILL, Sampling Agent. MRS. A. B. VOSBURGH, Secretary.
Biochemical Laboratory.	H. B. VICKERY, PH.D., Biochemist in Charge. GEORGE W. PUCHER, PH.D., Research Assistant. MISS HELEN C. CANNON, B.S., Dietitian.
Botany.	G. P. CLINTON, Sc.D., Botanist in Charge. E. M. STODDARD, B.S., Pomologist. MISS PLORENCE A. MCCORMICK, PH.D., Pathologist. HAROLD B. BENDER, B.S., Graduate Assistant. A. D. McDonnell, General Assistant. MRS. W. W. Kelsey, Secretary.
Entomology.	W. E. Britton, Phd., Entomologist in Charge: State Entomologist. B. H. Walden, B. Agr. M. P. Zappe, B.S. Philip Garman, Ph.D. Roger B. Friend, Ph.D. John T. Astworth, Deputy in Charge of Gipsy Moth Work. R. C. Botsford, Deputy in Charge of Mosquito Elimination. J. P. Johnson, B.S., Deputy in Charge of Asiatic and Japanese Beetle Quarantines. Mrs. Gladys Brooke, B.A., Secretary.
Forestry.	WALTER O. FILLEY, Forester in Charge. H. W. HICOCK, M.F., Assistant Forester. J. E. RILEY, JR., M.F., In Charge of Blister Rust Control. MISS PAULINE A. MERCHANT, Secretary.
Plant Breeding.	DONALD F. JONES, S.D., Geneticist in Charge. W. R. SINGLETON, S.M., Assistant Geneticist. H. R. MURRAY, B.S., Graduate Assistant. Mrs. R. A. HUNTER, Secretary.
Soils.	M. F. Morgan, M.S., Agronomist. H. G. M. Jacobson, M.S., Assistant Agronomist. Herbert A. Lunt, Ph.D., Research Assistant in Forest Soils. DWIGHT B. DOWNS, General Assistant.
Tobacco Sub-station at Windsor.	PAUL J. ANDERSON, Ph.D., Pathologist in Charge. T. R. SWANBACK, M.S., Scientific Assistant. MISS DOROTHY LENARD, Secretary.

- (1) Arsenate of lead¹ used at 1½ pounds per 50 gallons of spray mixture—applied four times, two applications being subsequent to the calyx period preferably in a pink, calyx, 7-day and 2-weeks or 17-day schedule. Continue until curculios are well under control or if the infestation is small in the beginning, then omit the 7-day application on all except outside rows around the orchard. Apply sprays after calyx at the rates indicated on page 409
- (2) Treat interplanted peaches, or peaches in orchards nearby, using lead arsenate as per spray calendar recommendations and cultivate thoroughly up to the trunks during July.
- (3) Take care of wild apples or unsprayed trees in immediate neighborhood if possible, either by removal of trees, spraying them at calyx period with lead arsenate and fish oil sticker, or by collection and destruction of early dropped fruits.
- (4) Destroy fence rows bordering orchard and remove stone walls if possible.
- . (5) For trees in the home garden, additional measures such as collection of dropped fruits, may be practised, keeping them picked up and destroyed from the middle of June on—and the destruction of beetles caught by jarring them from the trees onto a sheet.

Recommendations for Control of Curculios on Apple Trees in Connecticut

<sup>&</sup>lt;sup>1</sup>Pb H ASO4—Acid lead arsenate. Commercial arsenate of lead is commonly sold in Connecticut in this form.

#### CONTENTS

: 10 - 10 - 10 - 10 - 10 - 10 - 10 - 10	
Recommendations for Control of Curculios on Apple Trees in	age
Connecticut  Control Studies on the Plum Curculio in Connecticut Apple Orchards  Nature and Extent of the Damage	372
Orchards.	373
Injury to Apples	73
Life History. 3 Length of Life Periods, 1924–1925. 3 Length of Periods under Different Conditions, and Cause of the Variations.	73
Length of Periods, 1924–1925.	76
Variations	76
Life History in 1928. 3 Description 3	77
Abundance of the Court is the court of the C	80
Periods of Activity	81
Abundance of Beetles on the Trees	81
Egg-Laying Activity of Dig.	83
Emergence of the Second Proof Addition	86
Habits of the Curculio. 38 Hibernation and Spring Emergence 38	38
Hibernation and Spring Emergence 39 Food of the Beetle 39	91
Moisture Requirements 39	)2
Number of Feeding and Egg-Punctures.	
Total Number of Beetles Developing in Andrew Summer. 39	
Monogo to Emily 11 D 11 Property and Possible	14
Reaction of Beetles to Various Support 1 Aug. 39	
Materials	.=
Parasites 39	9
	- T
Predaceous Enemies 40	
Diseases	-
Popular 1 11 601 11 11 11 11 11 11 11 11 11 11 11 11 1	1
Conditions Surrounding Ten Representative Orchards 40	
Value of Exposing Drops to Direct Souli 1	
Conditions Surrounding Ten Representative Orchards 400 Preliminary Tests with Various Insecticides 400 Value of Exposing Drops to Direct Sunlight and Collecting Them from Beneath the Trees 400 Spraying Experiments for Curculio Control, 1924–1928 400 Spray Apparatus and Methods Used 400	7
Spray Apparatus and Mathada II 1924–1928. 40	
Spray Apparatus and Methods Used	
Materials Used 408 Experiment Station Orchard 410 Results in 1925 410	
Results in 1926 414	Į.
Results in 1927 417 Results in 1928 419	
Experiments at Shepard's Orchard M	
Results in 1925	
Results in 1027 425	
Results in 1927   425   Results in 1928   425   Results of Different Sprays for Curculio Control 1924–1928   429   Miscellaneous Results   429	
Results of Different Sprays for Curculio Control 1924–1928 429	
Cost of Materials 430	
Spray Burn430Arsenical Residue431General Conclusions431	
General Conclusions 431 Acknowledgments 433	
Acknowledgments	
The state of the s	

# Control Studies on the Plum Curculio<sup>1</sup> in Connecticut Apple Orchards

# PHILIP GARMAN AND M. P. ZAPPE

The plum curculio has been recognized as a major enemy of fruits in New England since Colonial days. Records of its occurrence in this country may be found in literature appearing between 1735 and 1750², and judging from the amount written since that time it has continued its depredations undiminished in severity, ranking today as one of the most important fruit pests. It is probably a native American insect, which became destructive on the introduction of fruit culture by the colonists.

The studies described in this paper were commenced in 1923, and therefore cover a period of six seasons. Though most of the observations were made at the Station farm at Mount Carmel, the orchard of Mr. C. E. Shepard of the West Woods section of Mount Carmel was placed at our disposal for experimental work. The systematic spraying tests recorded on pages 407-434 represent five seasons' work at the Station farm and four seasons at Mr. Shepard's orchard.

# NATURE AND EXTENT OF THE DAMAGE

As the name implies, plums are the favorite food of the curculio and damage to these fruits is always severe. Peaches are likewise attacked and although not seriously in Connecticut, much loss results in some regions. Apricots are about equally preferred to to plums while cherries, nectarines, quinces and pears are sometimes injured. Haws are also said to be infested. Perhaps the most serious damage by the curculio in Connecticut occurs on apples, not necessarily because of the amount of feeding or egglaying, but because of the importance of the apple crop in the State and the difficulty of controlling it under ordinary conditions.

# INJURY TO APPLES

Curculios puncture the fruit of the apple both to feed and to lay eggs. Egg punctures are made during the spring and early summer, whereas feeding punctures may be made at any time. Egg punctures are characteristic for the insect, consisting of small crescentic excavations (Plate IV, a) which may enlarge as the apple grows, often resulting in large arrow-shaped scars (Plate V, a)

<sup>&</sup>lt;sup>1</sup>Conotrachelus nenuphar Herbst. Order Coleoptera—Family Curculionidae.

<sup>&</sup>lt;sup>2</sup>See Quaintance and Jenne, 1912, p. 15.

a quarter of an inch or more across. The crescentic excavation usually prevents the destruction of the egg by growth of the fruit, the egg being placed in the flap at the center of the excavation. Feeding punctures consist of small circular holes often excavated around the edges by the feeding of the beetle, the fall punctures

being largely of this type (Plate VI, a, b).

The number of punctures per apple varies with the density of the infestation. On trees sprayed with lead arsenate according to the usual schedule, the number varies from one to five, though occasionally more are encountered. On unsprayed fruit, the number frequently reaches 20 or more per single fruit. Where one or two punctures are present, the value of the fruit is not seriously affected except that it should be removed1 from the highest grade (Fancy). Where 10 to 20 punctures occur, the value is naturally much less. A few external punctures do not, however, impair the keeping qualities and there is no danger of wormy fruit even where more than this occur, since no larvae develop in fruit remaining on the tree. Expanded curculio marks from punctures made early in the season frequently cover a considerable area and may hurt the sale if put on the market in competition with perfect fruit. The greatest damage, however, results from early feeding punctures which deform the fruit and from larvae which partly develop but are later crushed by the growth of the apple (Plate V, b). Such fruit is much reduced in value.

As to the amount of the injury occasioned, it may be said that a large per cent of the fruit is often affected on unsprayed trees, varying with the abundance of the beetles and the size of the crop. Our experience indicates that they may injure as much as 90 per cent of the entire crop, rendering a large part of it unsalable or fit only for cider. On large trees (12 years or more), a medium crop usually shows greater injury than either a very light or heavy one. Ordinarily 50 to 60 per cent of the fruit is damaged in infested orchards where no spraying is done. The following table shows the degree of injury which is found on untreated trees in sprayed orchards, the percentages representing curculio marked fruit:

		TABLE	1					
	Orchard	1921	1922		1924 Cent	1925	1926	1927
Milford	A	57	67	60	48		23-60	
Expt. Sta., Mt. Carmel					64	61	50	55
Shepard's, Mt. Carmel.	C			201		79	26	40

The above figures would not be especially significant unless a comparison were given with fruit from average commercial orchards. Counts were therefore made in a few orchards in order

to obtain figures bearing on the amount of damage to fruit as handled by the average Connecticut orchardist.

TABLE 2—RESULT OF SCORES IN COMMERCIAL ORCHARDS—1928

TABLE	Location		Number		
Variety	of Orchard	Total Apples	Marked by Curculio	Per Cent Marked	Treatment
		749	128	17.2	Dormant and pink
Baldwin	Cheshire	749	120	11.2	sprays dusted 7, 14, 30 days after.
Baldwin	Cheshire	661	36	5.4	Same as above plus prepink spray.
Baldwin	Wallingford	1,333	40	3.3	Dormant oil, pink, calyx, dust 10 days later, July 30.
Wagener	Wallingford	749	49	6.5	
Baldwin	Wallingford	1,179	166	14.2	
McIntosh	Cheshire	535	68	12.7	Pink spray, calyx dust (90-10) 7, 14, 30 days later.
Baldwin	Cheshire	749	128	17.4	Same as above.
Baldwin	Cheshire	659	36	5.5	Same as above plus prepink spray.
Baldwin	Cheshire	1,376	228	16.5	Dormant, pink, de- layed calyx sprays; dusted afterwards.
Baldwin	Cheshire	554	123	22.2	
Baldwin	Branford	1,421	42	2.95	Dormant, pink calyx and 4 others last on August 1; all sprays.
Baldwin	Branford	945	102	10.7	Dormant, pink, calyx and 3 others; all sprays.

## LIFE HISTORY

The life history of the plum curculio is very well known and has been worked out carefully by many investigators. Therefore, it will be unnecessary to extend this phase of the report unduly, but it will be desirable to give a general outline of the life cycle in Connecticut, developing such points as affect control measures.

Eggs are laid in the crescent-shaped punctures already mentioned. On apples, these punctures begin to be noticed shortly after the petals fall when the apples have reached one-fourth to one-half inch in size. After this, there is a period of three to five weeks (Fig. 28) in the field when egg-laying is continued but the peak of oviposition occurs about June 15, dropping off rapidly and being similar to the abundance curve (Fig. 26, b). In insectary cages, egg-laying has been observed to continue until the second week in July or even the last of July while newly made egg-punctures may be observed on apples in the field as late as July 25.

<sup>&</sup>lt;sup>1</sup>Public Acts, State of Conn. 1919, Chapter 295, Section 2.

# LENGTH OF LIFE PERIODS 1924-1925

The eggs hatched in an average of seven days according to our observations, varying from five to 16 days. The egg and larval period in the fruit averaged for this locality 22.9 days, varying from 17 to 39 days; while the average time from the entrance of the larvae into the soil to the formation of the adult was 21.2 days varying from 10 to 31 days. Our total period then from deposition of eggs to emergence of adults averaged 53.4, and varied from 31-67 days. The adult was observed to spend some time in the soil after transforming and there seemed to be considerable lack of uniformity in the maturity reached before they came from the soil. This probably accounts for the rather large variation encountered in the total length of the life cycle. There is, however, considerable difference in the length of the life periods in different years and the summer emergence of the adult has varied accordingly. The following records give a general picture of the periods in 1924-1925.

Table 3		
	Average	Davs Range
Eggs in fruit	7.0	5-16
Larvae in fruit	15.9	12-23
Larvae in soil	11.6	6-16
Pupae in soil	11.0	8-25
Adult in soil	9.8	
Total	55.4	31-80

# Length of Periods under Different Conditions and Cause of the Variations

It seems probable that the total life period given in the preceding table is shorter than actually occurs in the field in some seasons and the following figures bear on this point. Larvae emerging from dropped fruit collected under experimental trees were placed in field cages in 1924-1925, and the time to the emergence of the beetles observed. It will be seen that the average time spent in the ground is 42.2 days which if added to the periods spent in the fruit as egg and larva: namely, 22.9 days, would equal 65.1 days. This, it would seem, more nearly approaches conditions in the orchard. We know, for instance, that the greatest number of eggs laid about June 15 (1923-1924) in Connecticut and the total period of 65 days would bring the period of adult emergence (second brood) to the maximum for August 19. In 1924 and 1925, the curve (see chart on page 389) came to the maximum on September 1, although since that time it has fallen more nearly on the theoretical date, August 15-20 in 1927, and 10-15 in 1928. Another fact

which supports the late summer emergence idea is the percentage , of fall feeding punctures found on the fruit.

PLUM CURCULIO ON APPLE

TABLE	: 4
Number Observed	Average Time Spent in Soil, Days
38	37.8
$\frac{5}{1}$	$\frac{36.0}{32.0}$
$\begin{array}{c} 28 \\ 22 \end{array}$	42.6
13	50.1 $34.3$
$\frac{1}{12}$	49.0
<u>—12</u>	$\frac{52.0}{}$
Total120 Av. of all obs Av. of Avera	ervations42.2 days ges41.7 days

Thus, from the variety Dutchess (unsprayed) examined on August 26, 1927, .09 per cent were marked with this type of feeding punctures; whereas, Russets picked and scored during October ran as high as 2.0 per cent; Greenings scored September 11 averaged 1.4 per cent while Baldwins scored October 27 but picked about one week earlier averaged 6.5 per cent on some unsprayed trees. The greatest amount of fall feeding is thus seen to occur after the first of September and since the adults will begin to feed within a few days after emergence, they could not well have emerged much earlier than this date. On the other hand, during 1928 when there was a very early emergence of adults, few or no feeding punctures were found on early fruit which would tend to contradict the last statement.

# LIFE HISTORY IN 1928

Results of rearing curculios in 1928 indicated that there was a much shorter total life period than occurred in 1924 and 1925 and as a direct result an earlier emergence of adults during August, than was experienced during these years. One hundred individuals reared from eggs during 1928 gave an average cycle of 47.9 days and 28 days from emergence from the fruit until appearance above the soil as adults. Larvae placed in the soil on the same dates presumably under identical conditions (placed together in small jars) varied as much as three weeks in their period of emergence. In all cases, a normal emergence curve was produced, the peak of the lots obtained early in the season being approximately 27 days, and those obtained later, about 30 days. The following table shows the results obtained for 1928 in rearing curculios from egg to adult and from larva to adult, the latter being obtained from dropped fruits collected in the orchard at Mt. Carmel.

TABLE 5—LENGTH OF TIME SPENT IN SOIL—1928

Date			1020	
Entered	Date	Number		Range
Soil	Emerged	Observed	Average	Days .
June 28	July 25–27	4	28	27-30
July 1	July 25-Aug. 6	107	27	24-36
July 7	July 29-Aug. 21	43	27	22-45
July 8	July 31-Aug. 15	44	28	23-28
July 8	Aug. 4-Aug. 21	119	36	27-44
July 8	July 29-Aug. 10	32	26	21-31
July 9	Aug. 1-Aug. 11	$3\overline{2}$	27	22-32
July 10	July 21-Aug. 11	95	$\overline{26}$	21-31
July 10	Aug. 1-Aug. 24	125	$\frac{1}{27}$	21-45
July 11	Aug. 6-Aug. 21	65	31	$\frac{27}{27-41}$
July 12	Aug. 4-Aug. 11	91	27	23-30
July 15	Aug. 7-Aug. 19	51	28	23-35
July 16	Aug. 1-Aug. 19	114	32	26-39
	Aug. 11-Aug. 23	64		
July 18	Aug. 10-Aug. 21		27	23-34
July 23	Aug. 19-Aug. 26	7	29	27-35
July 24	Aug. 17-Aug. 26	4	29	24-33
July 26	Aug. 19-Aug. 27	10	28	24-36
July 27	Aug. 23-Sept. 3	11	30	27-36
July 29	Aug. 23-Sept. 5	16	30	25-38
July 30	Aug. 21-Sept. 3	6	29	22-33
July 31	Aug. 30-Aug. 31	4	31	30-31
Aug. 3	Aug. 26-Sept. 3	4	26	23-29
	July 21-Sept. 5	1,048	28	21-45

TABLE 6—LENGTH OF LIFE CYCLE FROM EGG TO ADULT—1928

No.	Date Eggs Laid	Date Adult Emerged	Number Beetles	Period Days
1	May 30	July 26	1	57
2	June 12–13	July 26	$\frac{1}{2}$	43
$\frac{2}{3}$	June 12–13	July 28	2	45
4	June 10–11	Aug. 4	ĩ	54
5	June 15–16	Aug. 4	i	49
6	Tune 24	Aug. 4 Aug. 1	1.	38
6 7	June 24	Aug. 3	i	40
8.	June 24–5	Aug. 6	1	42
9	June 24–5	Aug. 7	î	43
10	June 21	Aug. 8	î	47
11	June 26	Aug. 9	ī	44
12	June 26	Aug. 15	3	50
13	June 26–7	Aug. 9	7	43
14	June 29	Aug. 15	2	47
15	June 29	Aug. 17	3	49
16	June 29	Aug. 19	2 3	52
17	June 27	Aug. 14	3	48
18	June 27	Aug. 15	1	49
19	June 30	Aug. 15	2	47
20	June 30	Aug. 19	1	51
21	July 1	Aug. 11	1	41
22	July 1	Aug. 15	1	45
23	July 11	Aug. 24	1	55
24	June 28	Aug. 10	3	43 -
25	June 28	Aug. 11	1	44
26	June 28	Aug. 13	1	46
27	June 28	Aug. 30	1	63
28	July 5	Aug. 26	2	52

TABLE 6-LENGTH OF LIFE CYCLE FROM EGG TO ADULT-1928-Concluded

PLUM CURCULIO ON APPLE

IBLE O	DENGIN	OF BILL CLOBE	ricom 200 ro.	110001	2020	0011011114
	No.	Date Eggs Laid	Date Adult Emerged	Number Beetles	Period Days	
				1	55	
	29	July 5	Aug. 29	1	46	
	30	July 2 July 3	Aug. 17			
	31	July 3	Aug. 13	1	41	
	32	July 3	Aug. 19	5	47	
	33	July 6	Aug. 15	5 2 2 1	40	
	34	July 6	Aug. 19	2	44	
	35	July 4	Aug. 6	1	31	
	36	July 4	Aug. 14	1 1	39	
	37	July 4	Aug. 17	1	42	
	38	July 4	Aug. 19	4	47	
	39	July 7	Aug. 31	1	55	
	40 .	July 11	Aug. 30	1	50	
	41	July 11	Aug. 31	1 1 1 1	51	
	42	July 11	Sept. 5	1	56	
	43	July 8	Aug. 19	ī	42	
	44	July 18	Aug. 22	ī	45	
	45	July 18	Aug. 25	î	48	
	46	July 8	Aug. 26	$\frac{1}{4}$	49	
	47	July 9	Aug. 24	î	46	
	48	July 9	Aug. 26	3	48	
	49	July 12	Aug. 30	1	49	
	50	July 12	Aug. 50		55	
	51	July 12	Sept. 5	1		
		July 10	Aug. 30	2 1 3 3 1	51	
	52	July 10	Aug. 31	3	52	
	53	July 14	Aug. 31	3	48	
	54	July 14	Sept. 9	1	57	
	55	July 22	Sept. 14	2	54	
	56	July 22	Sept. 17	1	57	
	57	July 19	Sept. 17	1	60	
	58	July 22-24	Sept. 17	2	56	
	59	July 22-24	Sept. 25	1	64	
	Total	average and ra	nge	100	17 1	21_64

Total, average and range..........100 47.1 31-64

A chart of the various activities of the curculio during the summer is shown in Figure 23. The first phase showing beetles coming from hibernation often begins in April but is not shown on the chart for lack of space. The peaks and relative abundance at different periods are shown together with the dates when the activity began, when it came to the maximum and finally ceased.

As a rule, the curculio lives longer in the adult stage than in any other. Adults of the second generation emerging from the soil during August and September hibernate and appear during May or June of the following year. Mating occurs in August or September but more often in Spring after coming from hibernation. Beetles sometimes live in captivity until October of the second year, making their total life 13 or 14 months. In most instances, however, beetles hibernating successfully, die before the middle of July and the total period for this stage may be said to vary from a few months (for those that fail to survive the winter) to nine to 13 months. The following chart (Fig. 23) indicates the general course of the life history.

381

BULLETIN 301

### DESCRIPTION

No extended scientific description will be attempted here, since the various stages have been frequently described and are shown in Plates I and II. The larva is sometimes confused with that of the Oriental fruit moth, and it may be said that the main gross

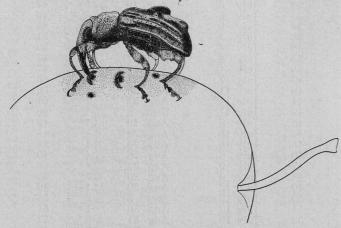


Fig. 21. The curculio beetle, its egg and feeding punctures. Enlarged 5 times.

points of difference lie in the curvature of the body and the color. The curculio almost invariably rests in a curved position while the fruit moth is straighter and more active and when mature is

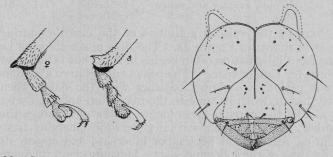


FIG. 22. Structural details of the curculio. Left, hind tibiae and tarsi of adult female and male. Right, head of larva, front view showing punctures and setae.

pink in color. The curculio larva is always white or slightly yellow and possesses no abdominal legs.

There has been some doubt in the authors' minds regarding the number of larval instars, but we have been able to check the statement of Quaintance and Jenne (1912) on p. 56 and find that there are four instars as stated there. The average width of the head capsule of 10-14 individuals of each instar are as follows: first, .28 mm.; second, .45 mm.; third, .69 mm.; and fourth, .96 mm. The larva is shown in Plate I, b and c, the head capsule in Fig. 22.

The pupa, shown in Plate I, d, is without protection except that it transforms in an earthen cell.

Adults, Plate II, a to d, are small brownish gray snout beetles with elevations of different heights on the wing covers, the beetle itself varying in length from tip of snout to end of abdomen from 5 to 7 mm.

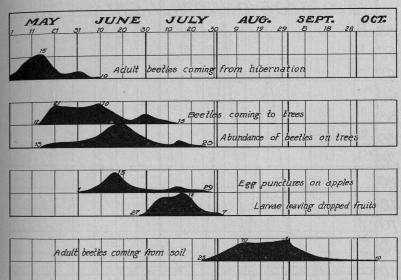


Fig. 23. Chart showing various activities of the curculio. Earliest emergence of hibernating beetles during our experiments occurred April 24. Adult beetles enter hibernation in August and September shortly after emerging from the soil.

When at rest, the adult folds its legs close to the body and if disturbed falls to the ground where it resembles closely a piece of bark or small cinder. The sexes are distinguished by the shape of the hind tibia as shown in Fig. 22.

Abundance of the Curculio in Its Various Stages and at Different Periods of Activity

# SPRING EMERGENCE OF THE ADULT

Emergence from hibernation begins on the approach of warm weather in the spring. The appearance of beetles depends upon temperature, and apparently upon rainfall in certain seasons. It has been noted by different authorities and observed during these investigations, that there may be an almost complete absence of beetles in the orchard until a given period dependent upon the temperature or other climatic conditions, when they come suddenly in considerable numbers. During the present investigation, records have been kept of emergence of the adults confined in field cages, from which data it appears that they begin to come from hibernation as a rule when the blossom buds of most varieties

BULLETIN 301

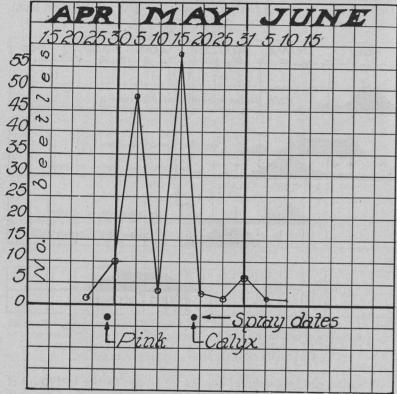


Fig. 24. Emergence of curculio beetles from hibernation in field cages 1925. Varying emergence is probably due to variations in temperature.

are turning pink. The earliest record of emergence thus far obtained is on April 24, 1925, and adults continued to come from the soil until June 12. The best records were obtained in 1925 and 1927 when three distinct waves of emergence were noted probably due to varying temperatures. In 1925, '26, '27 and '28, however, they appeared well on schedule, beginning to appear in numbers when the apple trees had reached the pink blossom bud stage. The data are shown graphically in Figures 23 and 24.

Temperature is, of course, an important factor affecting the time of emergence in spring, and especially the appearance of the beetles on the trees. According to various records (Quaintance and Jenne, 1912, pp. 118-119) the beetles begin usually to be found on the trees after a few days when the thermometer registers 60° F. (mean daily) or above. If the season is dry, the amount of rainfall decidedly affects their emergence from the soil. This is especially true of our observations in field cages where in 1926 the numbers emerging were decidedly increased after heavy rains. In 1927 this condition was not as marked but the following table will show that the greatest emergence occurred after the heaviest rainfall of that period.

Table 7—Emergence of Hibernating Beetles as Affected by Rainfall—1927

Date	Rainfall in Inches	Beetles Emerging in Cages	Temperature, Deg. F.
May 2	tr.		
3	.05	$\dot{2}$	52-62
4 5 7	$^{.2}$ tr.	i	50-62
10	.2	3	56-62
11	1.05	12	57-71
13			
14	. 55		
15		2	53-64
17	.1		
20	tr.	2	50-64

## ABUNDANCE OF BEETLES ON THE TREES

The beetles, however, do not appear on the trees until some time after the first emergence in cages, and the maximum abundance on apple trees is not reached until at least three weeks after the calyx or petal fall spray would normally be applied. On plums, however, their appearance on the trees is somewhat earlier corresponding with the blooming period of some varieties, and the maximum abundance and egg-laying activities do not correspond altogether with the maximum abundance on apples according to our figures. The following charts (Figures 26 and 27) have been prepared to show this phase of the beetles' activities. On apples they begin to appear sometime between the pink and calyx periods, but do not become abundant until later. From the calyx period on, they increase rapidly until about three weeks afterwards when they decrease and gradually disappear, a few stragglers remaining until the middle or even the latter part of July.

# EGG LAYING ACTIVITY AT DIFFERENT PERIODS

Extensive observations on the egg laying activities of the curculio in the field were made in 1928, results of fruit examinations being kept during the egg laying period. On apple trees it appears that the danger period in 1928 lay between June 1 and June 22 (calyx spray applied May 28 in our orchard) during which period sprays should be repeated at least twice for trees that are heavily infested. After June 22, there was a long-drawn-out period continuing until August when a few egg scars were made, which suggests the advisability of infrequent applications after the greatest feeding period is past in order to prevent the small amount of damage which occurs at the later time. Ordinarily, however, this has not been found necessary from the control standpoint, but sprays in July recommended for maggot control should play an important part in preventing damage from the late injury described. Figure 28 shows the injury to apples and other fruits resulting from egg punctures, and Plate IV shows typical egg scars.

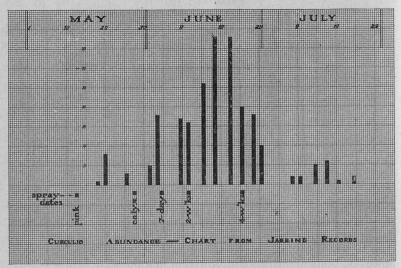


Fig. 25. Curculio abundance on fruit trees constructed from 1926 jarring records, together with spray dates for that year.

It has been thought that some differences might occur in different parts of the State in relation to the earliest egg laying activities. Thus, in 1928, according to the observations of Professor Manter, the calyx spray for apples was begun May 29 at Storrs and on the same date at Mt. Carmel. According to the bioclimatic law¹, there should be a variation of seven days between New Haven and Storrs, Conn., or 11 days between New Haven and Salisbury, while between Mount Carmel and Greenwich there should be no important difference. Such differences must vary from season to season, however, but evidently egg laying at Mount Carmel and Storrs commenced almost simultaneously in 1928.

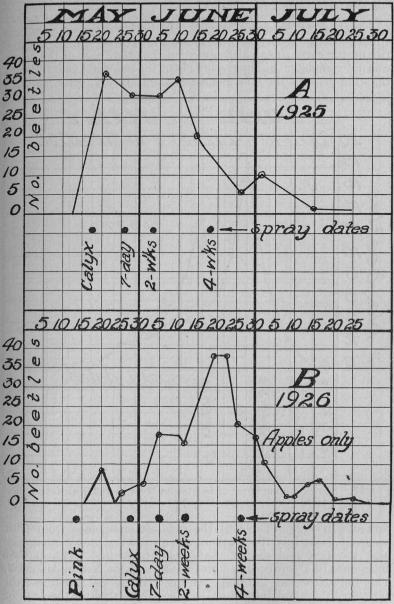


Fig. 26. Chart for comparison of (A) beetles coming to trees from hibernation as determined by removing beetles jarred at frequent intervals and (B) abundance on trees determined by jarring and releasing the beetles after counting.

<sup>&</sup>lt;sup>1</sup>Hopkins, A. D. The Bioclimatic law and its application to research and practice in Entomology, Jour. Wash. Acad. Sci., 1921; 11: p. 141. According to this law there should be a difference in periodic activity of four days for each one degree of latitude, 5 degrees of longitude and 400 feet of altitude.

# LARVAE LEAVING DROPPED FRUIT

Larvae of the curculio begin to leave the drops in the latter part of June and continue until August. In 1924, the peak was apparently reached about July 15. There was, however, a very sharp rise on the early side, which indicated that that part of the curve was not complete. Similar records kept the following year indicate that the peak of the emergence fell about July 1, some 15 days

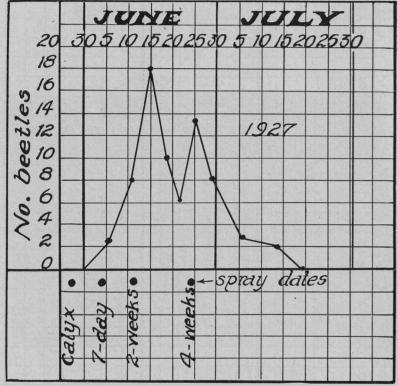


Fig. 27. Showing abundance of beetles on trees in 1927, as determined by jarring and releasing the beetles after counting.

earlier. In 1928, (Table 9) the greatest emergence from fruit occurred about July 10, but there was also a very heavy emergence near the first and during the days intervening. The earliest emergence this year occurred on June 27 and the last on August 7. This indicates that the main part of the brood leave the fruit between July 1 and July 15 or less than a month after the peak of egg laying on apples (Fig. 29). There is some variation from season to season in this phase of the curculios' activity though not so much as might be expected.

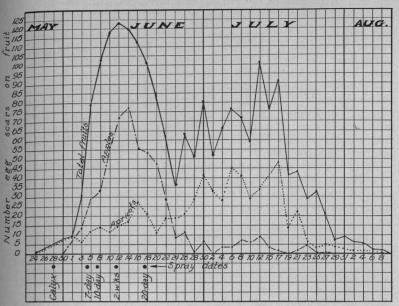


Fig. 28. Abundance of egg scars on fruit of apples, plums and apricots, shown under "total fruits" and apples and apricots separately. The curve for plums was essentially the same as that for apricots.

Curves obtained from examination of marked branches every other day.

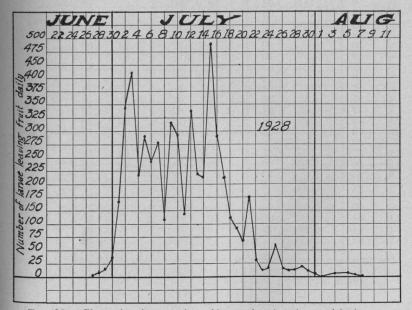


Fig. 29. Chart showing number of larvae leaving dropped fruit daily in 1928.

# EMERGENCE OF THE SECOND BROOD ADULTS

Adults begin to emerge in insectary cages towards the latter part of July, but few have been obtained in field cages before the first of August except in 1928. Thus in 1924 (Fig. 30), the peak came near the first of September; whereas, during 1926, it came between August 15 and 20, and in 1928, about August 10. The average peak of the adult emergence of the second brood lies about the middle of August, the variation being probably due to weather conditions. In order to predict the exact time of adult peak emergence it would be necessary to obtain temperature data similar to that obtained for the codling moth by Glenn, Shelford and Headlee.

Table 8 shows temperature and rainfall data obtained from the New Haven Weather Bureau for May to August, 1928, the temperature records from this station averaging two or three degrees higher than normally occur in the orchard.

TABLE 8—RAINFALL AND TEMPERATURE IN NEW HAVEN—1928

	M	av	T <sub>1</sub> .	ine				1020
	Mean Temp.	Rainfall Inches	Mean Temp.	Rainfall Inches	Mean	lly Rainfall	Mean	igust Rainfall
1	58	.04	66	inches	Temp.	Inches	Temp.	Inches
2	58	.01	72	94	72	Trace	72	. 02
3	61		62	.34	74	.12	80	
4	60		62	19	77		82	Trace
5	58		58	.42	78	.88	85	
6	60	Trace	62	1.04	76	1.13	82	.08
7	55	.03	66	. 58	68	1.29	67	.79
8	51		66	Trace	71		64	.21
9	50	iii	64		78		71	. 05
10	56		61	.49	80		78	
11	60	Trace	60		74	. 05	76	. 46
12	52	Trace	64	71.11	75	Trace	75	.24
13	50		66		78	.01	68	Trace
14	54		70	11:	72	.24	69	
15	54		68	.45	73	.84	76	
16	60		64		74	11.1.	80	
17	62		64		76		78	
18	57	.66	68	10	78		74	.15
19	57	.06	64	.18	82		76	.21
20	56	.28	63	1.51	82		75	
21	60	. 20	63		72	.04	70	
22	58		58	Trace	68		72	Trace
23	55	.06	57	. 20	68	.11	64	.32
24	56	.50	66	Trace	76	.04	64	. 56
25	56	THE PARTY OF ANY	72	.19	• 74	2.40	69	
26	59	.36	73		78		71	.02
27	58	.07	75		73		72	.29
28	60	.02	74	.04	74	.02	72	. 07
29	64		69	Trace	78	. 69	76	. 04
30	64		72	. 50	69		82	
31	63			.15	66		82	Trace
	00	11.1	1:-		68		68	Trace
Mon	thly				E P. C.			
	n 57.4		65.6		74.3		72 0	
Tota	al precip.	2.26		6.09	11.0	7.86	73.9	9 51
						1.00		3.51

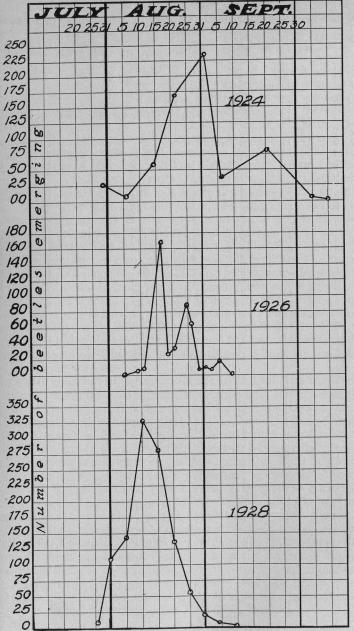


Fig. 30. Chart showing adult emergence from the soil during 1924, 1926 and 1928. Curve for 1928 represents number of emerging adults summed at the end of 5-day intervals.

CONNECTICUT EXPERIMENT STATION

Date	6622 Mixed Plums, Apples and Peach	2779 Peach	1423 Early Apple	3028 Medium Apple	707 Late Drops From	Totals	Apples Only
	Drops	Drops	Drops	Drops	Apple		0.1127
May 27	1					1	
May 28	6	• •			10.7.3	16	
May 29	14	•				14	
May 30						36	
May 31	151-14					00	
July 1	170		* * * * * *		Transfer Maria	170	
July 2	349					349	
July 3	417		6			423	6
July 4	214		3	HE!		217	3
July 5	170		14	9	F - 1 - 1	186	16
July 6	227		16	$\frac{2}{2}$	W111	245	18
July 7	$\frac{221}{224}$	14	30	8		276	38
	73	11	23	4	F	111	27
5	156	64	56	43		319	99
July 9	135	58	49	49		291	98
July 10	$\frac{155}{67}$	28	11	39		145	50
July 11	146	84	0	100		330	100
July 12		34	73	36		$\frac{330}{223}$	100
July 13	80 84	53	22	59	4	$\begin{array}{c} 223 \\ 212 \end{array}$	81
July 14		89	$\frac{22}{27}$			487	268
July 15	130 88	90		$\frac{241}{119}$	1	311	133
July 16			14		S. 1	$\frac{311}{212}$	132
July 17	41	39	6	126		119	68
July 18	20	31	1	67			54
July 19	14	26	0	54		94	
July 20	12	34	1	13	8	68	22
July 21	11	29	2	70	9	121	81
July 22	3	9	1	7	6	26	14
July 23	0	7	0	4	6	$\begin{array}{c} 17 \\ 20 \end{array}$	10 14
July 24	0	6	0	3	11		
July 25	3	35	1	10	12	61	23
July 26	2	5	4	3	5	19	12
July 27	1	6		3	3 2	13 14	6
July 28	4	6		$\begin{array}{c}2\\2\\2\\1\end{array}$	12		4
July 29	$\frac{1}{0}$	4		2	13	$\frac{20}{12}$	15
July 30	0	1		2	9		11 3
July 31	2		1	1	1	5	3
Aug. 1	0		4.00				
Aug. 2	0		V	4			
Aug. 3	1				3	4	3
Aug. 4	0				1000		
Aug. 5	0				1 .:		• • •
Aug. 6	0			1	2	3	3
Aug. 7	1		3 10.00	1	1	3	2
Totallary	ae 2,903	763	361	1,071	91	5,189	1,523

## HABITS OF THE CURCULIO

PLUM CURCULIO ON APPLE

## HIBERNATION AND SPRING EMERGENCE

It is fairly well established that the adult beetles winter in woods, hedge rows or stone walls near the orchard or even in the orchard itself. During the winter, the beetle is very difficult to find and an extended search in peach orchards, apple orchards or woods adjoining may fail absolutely to disclose the hibernating quarters. Probably the most successful attempt to discover the winter quarters is reported by Quaintance and Jenne (1912 p. 130) "At Youngstown, N. Y., in 1905 Mr. Johnson made frequent searches in the fall during October, and on the 14th of that month nine beetles were discovered in a slight depression under an apple tree. They were well covered with closely matted well decayed leaves.....Nine more beetles were found in a similar situation in an apple orchard on the 16th.....On November 7, six more specimens were taken beneath partly rotted leaves close to the soil. The beetles were wet and dull colored from their surroundings..... In the spring of 1905 Mr. Johnson made extended searches for beetles in fence rows, in peach, plum, apple and quince orchards in old stumps in adjoining woods, cracks in fences, under piles of wood and so forth..... None, however, was discovered... The following year, 1906, Mr. Johnson found on April 24, 10 beetles covered with leaves and decayed fruit on the surface of a young apple orchard in sod. At this time the blossom buds were just beginning to open."

In spite, however, of the usual failure to find hibernating beetles in or near orchards, it is almost invariably true that beetles first appear in greater numbers on rows adjoining fence rows or woods, and there can be little doubt that they seek shelter in such situations. A search in Connecticut in 1923 and 1924 failed to reveal the whereabouts of the beetle in the orchard, but a single specimen was found in a woods near one of the orchards. This beetle was found under leaves and trash near the surface of the soil. A second beetle was found during September, 1928 in a dried curled leaf on the edge of a nearby woods. In field cages it has been observed that the beetles leave the apples under which they crawl to feed before hibernation and apparently hide under leaves or trash in the immediate vicinity. In spring before emergence, a search in these cages will reveal beetles on the surface of the soil. Shortly afterwards they may appear in considerable numbers on the netting over the cage. Furthermore, curculios bumped from the trees in spring are frequently covered with clay, indicating

that they passed the winter in contact with the soil.

393

# FOOD OF THE BEETLES

Unlike some other insects the adults of the curculio require food. They will feed upon the leaves of peach and apple, and even upon the petals of the flowers (Plate III, b): Confined in cages with fruit blossoms, they often eat through the calyx cup, evidently in search of sweets. In confinement they have been observed to feed also on sugars of various kinds and seem very fond of honey. Much feeding is of course done on young fruits, of all species which the curculio infests. As to varieties of apples preferred either for food or oviposition there seems to be little choice since feeding and egg punctures have been observed on practically all varieties grown in this state. If there is any choice of variety on the part of the beetle, it is the Dutchess variety which seems to be severely injured in most Connecticut localities. The following table shows that the curculio will develop in many varieties. We have attempted to feed them on other fruits besides those in which the larvae develop but without much success except in the case of

TABL	- 16
I A BL	H. 1

Variety	No. of Apples	No of Adults Obtained
Baldwin	100	
Fall Pippin		$\frac{35}{2}$
Gravenstein	60	20
TT 11	100	90
Hurlburt	100	39
King	100	29
McIntosh	100	18
Russet	70	9
Stark	70	14

oranges, on the peel of which they will feed to some extent. They will also feed on such mixtures as casein and honey, with just enough water added so that it can be rolled into a ball, and they have been observed to feed on a sponge containing sugar with enough citric acid to make it quite sour. Saccharin was not observed to be attractive to the beetles for food.

# MOISTURE REQUIREMENTS

Adult curculios require considerable moisture during their lives. Not only does moisture or humidity influence their distribution in the United States, but a certain amount seems to be needed for proper functioning of the life activities. Trees with thick, heavy and abundant foliage, providing abundant dampness in the interior are often heavily infested. In cages, they may be frequently seen taking water from any convenient source or they may be trapped in bottles containing only moist blotting paper placed alongside of equally large bottles containing a natural food supply. Such an experiment was performed, using small glass bottles provided

with wire funnel traps. Apricots were used as the attracting food. The following results were obtained:

		TABLE	11	
Date	No. Beet les Used	Beetles in Moisture Bottle	Beetles in Food Bottle	Notes
May 31-June 1	7	5	2	
Tune 1-2	7	5	2	bottles reversed.
June 1–2	6	1	5	pair of beetles put in food bottle before
May 28-9	6	0	6	starting. check—nothing in moisture bottle.

It is not certain when the beetles take moisture in the field, but it is evident that they are much more active on warm, damp cloudy days than on dry clear ones. One must naturally conclude that moisture or water supply plays a very important role in the biology of the curculio.

Another meteorological factor that affects the curculio is high winds. It is well known that a slight jar is sufficient to make the insect feign death and drop to the ground. It is but natural, therefore, to find them much less abundant on trees, after or during a high wind when the branches are moving about.

# Number of Feeding and Egg Punctures

As already noted, the beetles will feed readily upon the fruit. The number of feeding and egg punctures varies in confinement but the following figures will indicate the number usually produced. Quaintance and Jenne give records of individuals making as many as 616 egg punctures although the average for all localities is much lower—101 for Myrtle, Georgia, and 31 for Siloam Springs, Arkansas. The average feeding punctures for these localities is much higher than the number of egg punctures—287 for Arkansas and 161 for Georgia. These figures, however, represent the combined feeding of one pair and indicate that the average feeding punctures per beetle is usually not over 150. Our figures vary from 35-105 egg punctures per female with an average of 79. Feeding punctures per female, however, averaged less than the total number of egg punctures, 46 in number although some individuals fed much more than this.

# Egg Laying and Feeding of Adults Emerging During Summer

As already mentioned, no eggs are laid in this locality by beeties emerging during the summer. The greatest damage to the fruit lies in the fall feeding punctures (Plates VI and VII, a) which,

however, have never appeared to be especially abundant in orchards where our experiments were conducted. The following table shows the relative abundance of this type of feeding punctures on sprayed fruit:

TABLE 12-AMOUNT OF FALL FEEDING OF THE ADULT CURCULIO

Date Scored	Total Fruits	Number with Fall Feeding Punctures	Per Cent with Fall Feeding Punctures	Treatment
August and Sept., 1928	21,362	44	$\begin{array}{c} 0.2 \\ 0.3 \end{array}$	Sprayed
August and Sept., 1928	16,181	51		Sprayed
Sept. and Oct., 1928	32,186	112	$0.3 \\ 0.8$	Sprayed
Sept. and Oct., 1928	18,075	151		Check—no
				treatment

The actual percentage of fall feeding punctures on sprayed fruit thus seems to be small and may be disregarded as far as preventive measures are concerned. There is often a considerably greater number on dropped mature fruits than on fruit picked from the trees which is thought to be due to beetles seeking hibernating quarters beneath the trees. The greatest amount of damage due to fall feeding of curculios is recorded in our data for 1926, when it averaged 6.5 per cent on unsprayed Baldwins.

# Total Number of Beetles Developing in Apples and Possible MENACE TO FRUIT OF THE FOLLOWING SEASON

In 1924, all dropped fruits were collected from a sprayed orchard containing two rows of unsprayed trees. All picked fruits were counted in scoring so that the relative number developing in the orchard is apparent. By making certain deductions for mortality, and allowing a conservative number of punctures per beetle (below the average in this case) the percentage of the crop which could be injured during the following year may be roughly estimated. While such figures can never be made to represent the actual condition in the field due to many undeterminable and variable factors, they do show that curculios developing in an apple orchard may offer a considerable menace to the crop during the succeeding year. The estimate is a very conservative one and it seems probable that the amount of damage is often considerably greater.

Total number of fruits in 1925	
Total larvae from drops in 1924	2,601
Total adults allowing 85% mortality (50%	
larval mortality, 70% for adults)	390
Total punctures allowing 100 per beetle	39,000
Total apples which could be injured al-	
lowing 1.5 punctures per apple	26,000
Per cent of total crop which could be	
injured	18.0

# PLUM CURCULIO ON APPLE REACTION OF ADULT BEETLES TO INSECTICIDES

In confinement the curculio is more or less easily poisoned with various arsenicals. In the field, however, a tree sprayed with the usual strength of lead arsenate will contain no curculios within 24 hours after spraying; whereas, on unsprayed trees in the immediate vicinity they may be very numerous. Experiments along this line were carried out in 1924 with always the same results; namely, the rapid disappearance of beetles after the poison was applied. This was determined by jarring the trees within 24 hours and placing sheets under the trees to catch fallen beetles. None were taken on the sheets. In cages the most rapid kill that we have yet obtained was 100 per cent in four days. which would indicate that some of the beetles at least should be obtained by jarring in 24 hours. No doubt a fatal dose is obtained shortly after the material is applied and the beetles go elsewhere to die. A report published in Bulletin 32 of the Georgia State Board of of Entomology is interesting in this connection. The authors here confined 372 beetles in an enclosed tree sprayed twice with 3-50 lead arsenate. The following day no beetles could be found feeding on the tree but they all died in 10 days. It seems probable that conditions obtaining here are very similar to field conditions although it does not necessarily prove that arsenate of lead as generally applied (where the dosage is smaller) is a repellant. Probably the substance is repellant in action only after the killing dose is obtained. Our tests, however, indicate a rapid disappearance from trees sprayed with 11/2 pounds per 50 gallons.

Various other insecticides have been employed in cage tests, including basic lead arsenate, sodium fluosilicate and ferrous arsenate (scorodite) but none have equaled the lead arsenate in killing power. Sodium fluosilicate mixed with four parts of lime showed considerable value but did not quite equal acid lead arsenate. Basic lead arsenate was only partially effective, a fact made further evident by field tests in 1928. To increase the killing power of certain poisons of low toxicity, various compounds such as lead and zinc stearate were added but without success. These are mentioned on page 406.

# REACTION OF BEETLES TO VARIOUS SUPPOSED ATTRACTIVE AND REPELLENT SUBSTANCES

Curculios are very sensitive to odors. It has been shown by Power and Chestnut<sup>1</sup> that the odorous constituents of apples consist of such compounds as acetaldehyde, amyl esters of formic acid and caproic acid and malic, caproic and capryllic acid. A subsequent research<sup>2</sup> indicated the presence of geraniol. Much

<sup>&</sup>lt;sup>1</sup>Journ. Amer. Chem. Soc. 42: No. 7: pp. 1509-1526: 1920. <sup>2</sup>Journ. Amer. Chem. Soc. 44: p. 1498: 1922.

time has been given to a consideration of these compounds from the standpoint of attractives and the only ones which could be detected to have much influence on the beetle are acetaldehyde and malic acid. Both the pure acetaldehyde and the acetaldehyde producing acetaldehyde-sodium bisulphite have been used with similar results. However, when used in the field the beetles have not been successfully trapped by any substance. Probably the scarcity of beetles is one factor influencing results, but another important factor is the high volatility, the odor disappearing within a short time after being placed in the open. This is also true of the repellents tried. Laboratory tests were conducted by placing small specimen bottles (capacity 23 cc.) in one end of a box three

Table 13—Results of Tests with Various Attractive Materials Using Long Box Plus Trap Bottles at End; Compared in Each Case with Natural Food

Substance Geraniol	Kind of Fruit Compared apricots <sup>1</sup>	Number in Fruit Bottle	Number in Test Bottle	Number Beetles Used 10 <sup>2</sup>
Granulated sugar and				
and water	apricots	8	0	10
Malic acid plus sugar	apples and	7	2	9
Time and Prince and Pr	apricots	6	1	7
Cal. malate Acet. sod. bi-	apples and	6	2	8
sulphite	peaches	1	6	7 8
surplince	peaches	6	2	8
Acetal., water, cal. malate, sugar	apricots	3	2	5
Acetaicai. maiate (dry).	apricots	(1	(2	TELL PROPERTY.
Moist blotting paper	apricots	$2 \left\{ \begin{array}{c} 1 \\ 1 \end{array} \right\}$	$5\left\{rac{2}{3} ight.$	7
Moist blotting paper	apricots plus	SIST MANAGE	8-50 Mining	
Wolst blotting paper	geraniol	2	5	7
Moist blotting paper	apricots plus			Ustrate a Feb
	pair beetles	5	1	6
Empty bottle	apricots	6	0	6

feet long by four inches wide by three inches high with openings at both ends. At one end was connected a short upright tube in which was placed a small electric light. This provided enough heat so that there was a gentle current of air through the box, from one end to the other. Each bottle was provided with a wire cone so that beetles could enter but would remain in the bottle. The tests were begun by placing the beetles in the far end of the box and allowing them to remain over night, the number found in each bottle being recorded the following day. The top of the box was provided with heavy red celluloid made for photographic work. It will be seen here that although the light factor is excluded by use of the red screen and the maze construction of the box, there is

still a disturbing element in the moisture which accompanies the attractive substances. It is also important as noted by others to use beetles at a time when naturally attracted to their native food plants, for if older beetles are supplied especially after most of the feeding and egg-laying are over, conflicting results may be expected. The Y tubes such as have been devised by McIndoo¹ have also been used with air currents from a suction pump. The unevenness of the air current through the tubes precluded consistent results in the few tests that were made with this type of apparatus.

In view of the fact that moisture apparently played such an important part in the above experiments another series was devised comparing equally large squares of blotting paper, one soaked in pure water, the other soaked in the solution of odorous substances to be tested. The paper did not take up such insoluble substances as lime, but some, of course, was deposited on them.

TABLE 14—COMPARISON OF EQUAL SIZE SQUARES OF BLOTTING PAPER
CONTAINING WATER WITH VARIOUS SOLUTIONS
OF ODORIFEROUS MATERIALS

Substances Compared	Number in Test Bottle	Number in Moisture Bottle	Number Beetles Used
Calcium malate and acetaldehyde  Iso-amyl n-capryllate acetaldehyde.	8	1	9
malic acid and lime	4	2	8
Same	5	2	8

From these results it may be seen that the substances tried had considerable greater attractiveness, yet when we refer to the preceding table it is evident that they do not have nearly as great attractiveness as the natural foods.

The action of repellents was tested in a similar manner except that air was blown through the cage with considerable force with an electric fan. The various substances were placed in the base of the funnel and the fumes blown into the cage with the air. In some of the tests, beetles were first collected on a peach shoot before introduction of the test odor, with the result that they were promptly driven from their natural food to the far end of the box. About 32 substances were tried in this manner with the following results:

### TABLE 15

Repellents and Activators
Capryl alcohol
Iso amyl-n-caproate
Iso amyl alcohol
Acetaldehyde
Carbolic acid

Repellents
Benzaldehyde
Capryl alcohol
Allyl isothiocyanate
Iso-amyl-n-caproate
Xylene

Substances Without Apparent Action Amyl alcohol (Iso) Amyl Formate Amyl n-capryllate N.-butyl alcohol Anise oil

<sup>&</sup>lt;sup>1</sup>Small green fruits.

<sup>2</sup>Two beetles failed to enter trap bottles.

<sup>&</sup>lt;sup>1</sup>Journal of Economic Entomology 19: 549-571: 1926.

#### TABLE 15—Concluded

Repellents and Activators Gasoline Benzaldehyde Xylene Allyl iso-thiocyanate Allyl alcohol Repellents Allyl alcohol

Substances Without Apparent Action Calcium cyanide Carbon disulphide Chloroform Cider Clove oil Dichloroethane Dinitrotoluene Ether Ethyl alcohol Lime sulphur Lysol Nicotine sulphate Octvl alcohol Paradichlorobenzene Propyl alcohol Vinegar

It is worthy of note that such substances as calcium cyanide have little effect, while capryl alcohol and benzaldehyde are strongly repellent. The almost immediate activation of the beetles on the introduction of strong capryl alcohol odor is surprising, the insects running about the cage as if mad.

Application of these apparently repellent substances in the field, however, resulted only in the rapid disappearance of the odor through volatilization, and they consequently had little effect in keeping the beetles from the trees. It is very difficult to select the various repellents in order of their potency and the foregoing list is not intended to convey the impression that one is necessarily more active than another. They have been divided into three lots for convenience. Some of the activator substances were apparently not repellent in the tests conducted and they have therefore been omitted from the repellent list.

There remains the discussion of the protective action of such materials as lime or other non-poisonous material. If we protect a portion of an apple, for instance, with a thick coat of lime, the beetles will feed on the unprotected side. Under natural conditions it is almost impossible to cover completely all sides of an apple unless extremely careful work is done and it is as nearly impossible to maintain this covering over any period of time. It is furthermore difficult to maintain a coat of any thickness so their ultimate value in this regard is doubtful. Lime is sometimes desirable in the spray mixtures, however, for other reasons.

# REACTION OF BEETLES TO LIGHT

Confined in cages, the adult curculio beetles are decidedly positive to light, but Quaintance and Jenne have shown that the activities of the female are about equally distributed during the day and the night. The adult beetles have been observed frequently in the field during the day at rest with legs folded, although when disturbed by jarring they may become active and take flight.

## HABITS OF THE LARVAE

On emergence from the egg, the curculio larva burrows into the center of the fruit where it continues to feed until a cavity of considerable size is excavated. After the apple drops and the larva becomes full grown, it enters the soil, penetrating to a depth of two inches or less. They rarely go more than three inches below the surface and 90 to 100 per cent will be found within the first two inches. Frequent disturbance of the soil by any means destroys the larvae and the value of cultivation in curculio control is therefore apparent. Undersized larvae often desert fruit and burrow in the ground, but it seems probable that these are stunted mature larvae, forced to abandon the fruit before full size is reached.

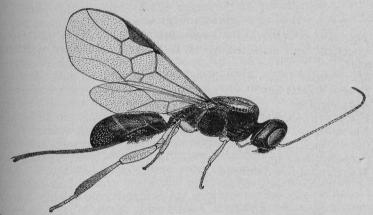


Fig. 31. Adult parasite *Triaspis curculionis* Fitch enlarged about 12 times. Left legs, wing and antenna removed.

# **PARASITES**

Parasites of the plum curculio have not been numerous at any time during this investigation. The most abundant, *Triaspis curculionis* Fitch, (Fig. 31) was observed in 1928 and it parasitized about 18 to 33 per cent of the larvae leaving the fruit July 1. The following table shows the relative numbers of this parasite obtained from a large number of larvae, the average parasitism being 5.0 per cent. The species emerged July 15 to August 21.

The only other parasite observed was the egg parasite Anaphoidea conotracheli Girault, found during the summer of 1928. This parasite is reported to kill as much as 85 per cent of the eggs of the plum curculio. Quaintance and Jenne mention nine other

TABLE 16—PERCENTAGE OF PARASITISM BY Triaspis curculionis FITCH

			011.011101110 11
Date Larvae Entered Soil	Number Curculio Larvae	Number Parasites Emerging	Per Cent Curculio Larvae Parasitized
June 28	6	2	33.3
July 1	170	31	18.2
July 2	22	7	31.8
July 12	50	1 .	2.0
July 7	96	1	$\overline{1.0}$
July 12	32	0	0.0
July 13	128	0	0.0
July 12	. 91	1	1.0
July 8	118	1	. 9
July 16	125	5	4.0
July 17	114	ĺ	.8
	953	48	5.0

parasites, a number of predaceous insects and other enemies. It will suffice here to list these additional enemies together with a few notes on their Connecticut status.

# HYMENOPTERA

Thersilochus (Porizon) conotracheli Rilev. Reported from Connecticut and collected by H. L. Viereck on flowers of Ribes species.

Microbracon (Bracon) mellitor Say.

Bracon dorsata Say.

Anisocyrta sp.

Pimpla (Epieurus) sp.

Eurytoma sp.

Cataloccus sp.

Cerambycobius sp.

Microbracon lixi Ashmead.

## DIPTERA

Miophasia aenea Wiedemann. Generally distributed over North and South America and probably occurring in Connecticut. Cholomyia inaequipes Bigot.

# PREDACEOUS ENEMIES

Ants, Dorymyrmex pyramicus Roger. We do not have this species in Connecticut although other species no doubt play an important part in destruction of curculio larvae.

Lacewings Chrysopa oculata Sav and other species present in Connecticut orchards.

Thrips. Reported to destroy eggs of the curculio. Observed frequently in or near egg scars and they had apparently destroyed the eggs present in them.

Carabid or ground beetles. Frequently present in Connecticut orchards.

Chauliognathus pennsylvanicus DeGeer. A small brown beetle frequently present. The larvae of this species is said to attack curculio larvae.

PLUM CURCULIO ON APPLE

Birds of various species are known to feed on curculios.

Moles destroyed a great many larvae in ground cages in 1924. It was found necessary to protect the bottom of these cages against their entrance.

# DISEASES

Larvae have been observed frequently to die of disease where the soil becomes very damp, and are frequently covered with a white fungus, probably Isaria or Sporotrichum species. The adult beetles have also been found diseased under similar conditions one of the causes being the green muscardine fungus Isaria anisopliae (Metch.) commonly attacking grubs in the soil. (See Pl. II, b.)

# CONTROL

Before introduction of the arsenicals as insecticides about 60 years ago, horticulturists used various means for combatting curculios. Cultivation, jarring the trees to capture the beetles, allowing live stock to run in the orchard in order to destroy the larvae in drop fruits are a few of the more successful means of control. Besides these there have been forty or more recommendations and suggestions varying from such means as placing chips under the trees to hanging dead mice therein in order to attract the adult beetles which were thought to lay eggs on such material.

The development of arsenicals as insecticides resulted in a material change in control procedures. Arsenate of lead, developed about 1893, brought a further increase in this means of control. Successful sprays were soon developed for control of the curculio and in late years this method has prevailed almost entirely, though accompanied by orchard sanitation, perhaps the most successful of the older means of control. The development of spray controls during the last 25 years and the trend of present day recommendations is well illustrated in the following summary of literature:

# RECOMMENDATIONS OF OTHER INVESTIGATORS

1905. Crandall, C. S., Illinois Agricultural Experiment Station, Bulletin No. 98. Extensive account of apple and plum curculios with detailed field experiments for control. Spraying operations considered unsuccessful; cultivation recommended.

1906. Crandall, C. S., Ibid., Bulletin 106, pp. 219-231.

1921. Fernald, H. T., In Applied Entomology, pp. 137-139. Control (p. 138). "No one method nor even all the methods of control taken together will give entire freedom from this pest. A combination of the treatments, however, will accomplish considerable in this line." Recommends removal of rubbish and hibernating quarters; (2) pruning trees to allow similarly the cotage (2) allowing forms and however. allow sunlight to enter; (3) allowing fowls and hogs to run under trees or thorough shallow cultivation from time larvae begin to leave fruit until

403

six weeks later; (4) spraying with arsenate of lead for apples—treatment commonly given for codling moth though similar later applications may also be necessary if the insects are abundant." (5) jarring the trees and collecting the beetles when only a few trees are involved.

1906. Forbes, S. A., Illinois Agricultural Experiment Station, Bulletin 108. Used arsenicals in field tests against the curculio on apples with an average increase of 63 per cent sound fruit over untreated trees.

1914. Headlee, T. J., Report of the Department of Entomology, New

Jersey Agricultural Experiment Station for 1913, p. 654. States that where curculio is present foliage must be kept covered with arsenical from time the "creature begins to feed until it disappears." Over-winter-for about a month.

1918. Headlee, T. J., Ibid., Report for 1917, pp. 437-438. "Attack seems to have come between blossom-fall and the ten-days-after-blossom-fall spraying." In orchards standing near woodlands or plantings interspersed with old stumps or carpeted with grass, the damage was particularly severe." Recommends clean culture during forepart of season followed by cover crop not producing a dense sod, removal of stumps and cleaning fence rows. Recommends spraying to preserve the coating intact for the first month after blossoms fall.

1919. Headlee, T. J., Ibid., Report for 1918, pp. 212-213. Recommends 7-day spray after calyx. Records two successful cases of control in orchards where curculios were abundant and where crop unsprayed was "ruined by the curculio". Recommends as a schedule for curculio control; "(1) before buds swell; (2) as blossom buds first show color; (3) directly after petals fall; (4) seven days later; (5) 17 days after blossoms fall; (6) June 20 to 30 for all fall and winter varieties. Better results obtained with lime-sulphur and arsenate of lead than with arsenate of lead alone; both have repellent action."

1921. Headlee, T. J., Ibid., Report for 1920, p. 449. Table showing comparison of dust and spray on apples; unsatisfactory control of curculios indicated with dust. Seven to nine per cent injured by curculio.

1923. Headlee, T. J., Ibid., Report for 1922, p. 373. Table showing comparison of dust and spray on apples including curculio records. Dusts 1924. It is a specific of the control curculio as well as sprays.

1924. Headlee, T. J., Ibid., Report for 1923, pp. 274-278, Tables 6, 7 and 8. Results of curculio spraying.

1925. Herrick, Glenn W., In Manual of Injurious Insects, pp. 156-157. "Control measures.—All fence rows, hedge rows and stone walls should be removed from about an orchard. Sunlight on the fallen fruit is fatal to the larvae within, hence judicious pruning to let in the light is useful. Cultivation during late July and early August will destroy pupae in the gallons of water just after the petals fall and again ten days later will be effective if the hibernating places have been destroyed."

effective if the hibernating places have been destroyed."

1918. Pickett, B. S., Watkins, O. S., Ruth, W. A., and Gunderson, A. J., Illinois Agricultural Experiment Station, Bulletin 206. Contains much valuable information on orchard sprays for the curculio and their results are ably discussed in tables and text. Page 492, under General Summary states: "Codling moth and curculio, as a rule were well controlled by applications of arsenate of lead"; page 493, controls obtained from 60 to 94 per cent" (in the most successful experiments in 1913) and the least effectively sprayed plats from 32-79 per cent. In 1914 the per cent and the least effectively sprayed plats from 45 to 87 per cent. In no case did spraying with arsenate of lead fail to exercise a decidedly least of the state of the st

1912. Quaintance, A. L., and Jenne, E. L., U. S. Department of Agriculture, Bulletin No. 103. The most extensive single publication on the plum curculio and its control yet published. Contains summaries of all previous work, data on life history and parasites and accurate studies

of control measures. Page 200 under conclusions states: "with a small amount of fruit and abundance of curculios the most thorough spraying will not serve to bring through a satisfactory amount of sound fruit"—"with a large crop of fruit and abundance of insects, results will likewise be disappointing". Recommends four sprays for apples using dilute fungicide and lead arsenate: (1) as cluster buds are out; (2) as petals fall; (3) three or four weeks after petals fall; (4) nine or ten weeks after petals fall. Secured controls amounting from 19 to 77 per cent increase in sound fruit from sprays tested. The best figures show 91.07 per cent sound fruit as a maximum obtained by the method advocated.

1922. Quaintance, A. L., and Siegler, E. H., U. S. Department of Agriculture, Farmers' Bulletin 1270, pp. 7-10. "Most practical means of control are spraying with arsenate of lead, and cleaning up of trash from the orchards and vicinity as well as thorough cultivation during the summer—the prompt collection and destruction of infested fallen fruit will also aid in reducing this pest." The first spray application to poison the beetles should be applied in pink cluster bud stage, and the second as soon as the petals have dropped, using arsenate of lead at the rate of one pound of powder or two pounds of paste to 50 gallons of water or fungicide. Supplemental treatments are desirable in orchards where the curculio is more than ordinarily destructive.

1914. Slingerland, M. V., and Crosby, C. R., Manual of Fruit Insects, pp. 243-251. Recommends for apples: two sprays as for codling moth just after petals fall and three weeks later—"but where the infestation is severe additional applications will be found necessary." Thoroughness of spraying is essential. Reliance should not be placed on any one method of attack. Clean cultivation, proper pruning, thorough cultivation at proper time are necessary.

1922. Snapp, O. I., Turner, William F., Roberts J. W., U. S. Department of Agriculture, Circular 216. Describes methods used for controlling curculio in the Georgia fruit belt on peaches. Recommends destruction of early drops or disking to destroy pupae and proper orchard sanitation. Iarring the trees also mentioned.

1924. Snapp, O. I., and Alden, C. H., U. S. Department of Agriculture, Bulletin 1205. Dusting and spraying peach trees after harvest for control of the plum curculio. General summary, p. 17 states, "Post-harvest treatments are not advisable except in cases where the curculio infestation has been severe during the peach season". Two applications of 10 per cent lead arsenate and 90 per cent hydrated lime dust are recommended for these treatments.

1928. Sanders, P. D., Trans. Peninsula Hort. Society, pp. 18-23 (Abstr. in Review of Applied Entomology Vol. X. VI: p. 451: 1928.) "Recommendations for control include destruction of overwintering adults by burning woodlands and hedge rows around orchards. Application to peaches of lead arsenate 1-50 or 5 per cent lead arsenate dust, once when petals have fallen and again as shucks are pushing off and cultivation under spread of the tree during second and third weeks after dropping of windfalls.

Besides the literature just quoted, there is considerable published data on the effect of dusts in curculio control with some difference of opinion regarding the merits of the method compared with sprays. Thus Quaintance (1921) p. 224 says "In the case of the plum curculio on apples, dusting compares favorably with spraying where the insect is not especially abundant"...."Under conditions of curculio abundance....dusting is not an effective control and spraying may not furnish the protection desired". The Indiana State report on Horticultural investigations (1919) states that "dusting controlled curculio and codling moth as well as spraying, whereas Cullinan and Baker, Bulletin 283 (1924) of the

Indiana Station state that dusting was inferior to sprays for curculio control in three out of five years work. The work of Stoddard and Zappe at Milford conducted during the years 1920-1924 showed consistent results1 in favor of spraying although the percentage gain from this method was never great. In addition there are the New Jersey reports comparing dust and spray indicating an advantage for sprays, and in general, opinions favor

sprays as better controls.

It is important before discussing any control program to know what conditions surround orchards where most damage is done. A number of typical fruit farms near New Haven were therefore studied. Nearly all Connecticut apple orchards are either cultivated in part or allowed to remain in sod, the orchards in one or the other being about equally divided. Few or no orchardists use clean cultivation. This is due in part to the nature of the land used for this purpose which is often hilly or rolling, and the danger of washing is considerable. There is also much waste or uncultivated land in many localities so that it is almost impossible to locate an orchard without placing it near a wood or within a few rods of numbers of wild apples. Such conditions are primarily responsible for severe infestations commonly found, and it must be recognized that under such conditions heavy infestations of curculio are the rule rather than the exception. Plate VII, b shows a typical wild apple tree.

# CONDITIONS SURROUNDING TEN REPRESENTATIVE ORCHARDS

Orchard No. 1. Wallingford. South side of one section near woods reported to be worst; inspection revealed much damage: no wild apples, however, in the vicinity although not far away was an unsprayed orchard; section protected on west by woods. In addition to the woods, a number of peach trees probably supplying many beetles were interplanted with the apples. In another section of the same orchard near a woods on the east side little damage could be seen. This, however, was some distance from the peach trees and the apple orchard mentioned.

Orchard No. 2. Durham. Orchard well in the open though .with a fence row on the south. On the north side from a quarter to one-half a mile distant were peach orchards of considerable size.

Orchard No. 3. Middlefield. Large extensive plantings, some of the apples interplanted with peaches. Worst infestations seem to be in these interplanted orchards.

Orchard No. 4. Wallingford. Interplanted orchard, worst conditions in section protected on west by woods, other parts of interplanted orchard said by owner to be less severely injured.

Orchard No. 5. Wallingford. Large plantings some near peach orchards apparently free or with few curculios. No fence rows or woods to protect the orchards although peaches are planted near some of them. The peaches are cleanly cultivated and dusted yearly with arsenate-sulphur-lime dust. Strip cultivations practiced in apple orchards. Plums in bearing near orchard.

Orchard No. 6. Milford. Damage worse on west side adjoining woodland. Orchard in sod or strip cultivation. Fence rows or stone walls around entire orchard, but the worst damage is next the woods as mentioned. Wild apples not far distant.

Orchard No. 7. Branford. Worst damage on outside rows next the woods. Wild apple trees not far distant, but no peaches. Orchard under strip cultivation.

Orchard No. 8. Mount Carmel. Strip cultivation practiced. Worst infestation on west side near fence row, although the latter is low. Young peach orchard also on west side. Other apple trees nearby, but sprayed.

Orchard No. 9. Mount Carmel. Protected on west by fence row and on southwest by wood lot. Wild apples numerous near the orchard. Peach orchards and plum trees to east.

Orchard No. 10. Mount Carmel. About one-fourth mile from No. 9. Not protected on west by fence row, but with bearing peach orchard on south and a few plums on north and west. Damage in this orchard, however, has never been severe for some unknown reason. Orchard in sod. Possibly the plums are sufficiently attractive to draw the beetles away from the apples. Plenty of hibernating quarters nearby.

About the only conclusions which can be drawn from this survey is that the worst damage is done in orchards protected by woods or fence rows on one side (which strange to say seems to be the west side in many of the orchards examined), while those not so protected are not as severely damaged. The presence of peaches in the vicinity seems to have some influence especially if accompanied by woods or fence rows and the presence of wild apples probably has something to do with total number of curculios in an orchard although the survey does not show anything striking in this regard. There is little doubt that beetles develop on these wild fruits since the fruits are abundantly marked and the adults have been jarred from the trees on several occasions. The occurrence of plums or apricots does not seem to influence the severity of infestation as much as peaches.

# PRELIMINARY TESTS WITH VARIOUS INSECTICIDES

We have already shown that curculios leave the trees shortly after applications of arsenic and it has been well demonstrated that they may be easily killed in confinement when fed on leaves sprayed with the usual dosages. Laboratory tests indicate that the Connecticut curculio is no different in these respects from others. Tables 17-19. It will be noted, however, that death is not immediate and sometimes does not occur until 10 to 12 days after introduction of poisoned food.

<sup>&</sup>lt;sup>1</sup>Conn. Agr. Expt. Station, Bull. 265; p. 293; 1924.

## TABLE 17-FIVE-DAY TESTS WITH ARSENICALS. FOLIAGE TESTS

N <sub>11</sub>	mber of	Date			umber Follow:		770 07	
	eetles	Begun	1	2	3 4	ing Da	Dead	Notes
Acid lead arsenate		Tune						
1.2 gm. per 250 cc.	11	25	2	8	10	11	100	
Calcium arsenate								Beetles
1.2 gm. per 250 cc.	8	25	0	5	5	8	100	excluded
Ortho-zinc arsenite				*				from
	11	25	0	3	4	11	100	water
Acid lead arsenate			1					supply.
1,2 gm. per 250 cc.								
molasses 10 cc	10	25	1	4	8	9	90	
Ball of casein molas-								
ses and lead arse-								
nate	8	25	2	5	8	8	100	
Check	10	26	0	0	0	0	00	

These tests were made in jars in the insectary, the twigs being sprayed with the solutions indicated and allowed to dry before placing in the jars with curculios.

#### TABLE 18-TWELVE-DAY TESTS WITH VARIOUS ARSENICALS

Number Substance Used Beetles Date	1	Dead in the Following Days 2 3 4 5 6 7 8 9 10 11 12											
Acid lead arsenate Aug.	20 C	-	3	-	3	U		0	ð	10	11	12	Dead
1.2 gm. per 250 cc. 10 20		1	3	0	3	0	5	0	0	7	0	10	100
Ball of casein with 50 gm. casein, 1 gm. lead arsenate, 10 gm. honey; water													
to make 125 gm 9 20 (	) (	0	2	0	0	0	5	0	0	9	0	0	100
Check—no treatment 7 20 (	) (	0	0	0	0	0	0	0	0	0	0	0	00

# Table 19—Laboratory Tests with Poison Dusts to Kill the Plum Curculio (20 beetles used in each jar)

Material	Date					De	ad in	the	fol1	owin	g da	vs					
	Begun		2	3	4	5	6	7	8	9	10	11	12	13	14	15	
Coomedite 9 ame	Aug.																
Scorodite 2 gms. Lead stearate 1	0																
gms		1	0	1	0	0	1	0	9	0	0	2	0	2	0	1	
Scorodite 10 gms.	. 20	-	U	1	U	U	1	U	4	U	U	J	U	J	0	+	
Zinc stearate	5																
gms	. 23	0	0	0	0	0	2	0	4	0	0	4	0	4	0	4	
Copper fluoride	2																
gms. lime 20 gm Lead stearate pure Zinc stearate pure	ns.23	1	0	1	0	0	2	0	2	0	0	2	0	3	0	3	
Lead stearate pure	e. 23	0	0	0	0	0	2	0	4	0	0	5	0	5	0	6	
Zinc stearate pure	. 23	1	0	1	0	0	1	0	0	0	0	1	0	1	0	1	2 lost
Magnesium fluorid	le																
20 gms. lead stearate 4 gms.	99	1	0	-1	0	0	0	0	1	0	0	9	0	9	0	1	0.1
Lead peroxide 1 lb		1	U	1	U	0	U	0	1	U	0	0	U	0	U	4	3 lost
As <sub>2</sub> 0 <sub>5</sub> ½ oz		0	0	2	0	0	2	0	2	0	0	2	0	3	0	3	1 lost
Lead arsenate	. 20	v	0		U	U	-	U	-	0	0	-	U	· ·	U	0	1 1050
(basic) pure	. 23	0	0	2 2	0	0	6	0	8	0	0	9	0	13	0	14	4 sick
Check	. 23	2	0	2	0	0	2	0	2	0	0	2	0	3	0	14 4	
Sodium fluosilicat																	
1 lb. Lime 4 lbs		0	0	6	0	10	0	0	0	11	0					0	2 escaped
Check		0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
Arsenate sulfur du		0	0	19	0	0	15	0	0	10	0	0	1 -	0	0	0	D
10% lead arsena Check		0	0	0	0		0	0		0	0	0	15	0	0	0	2 escaped
<u> </u>	. 10	U	U	U	U	U	U	U	U	U	U	U	1	U	U	U	

Note: Insecticides dusted on fruit and placed in jars containing curculio beetles.

# VALUE OF EXPOSING DROPS TO DIRECT SUNLIGHT AND COLLECTING THEM FROM BENEATH THE TREES

Several ground cages were supplied in 1928 with dropped apples and plums containing egg punctures of the curculio and presumably infested by the larvae. Some of these cages were exposed to direct sunlight; others were protected by cotton sheeting; still others by thick building paper excluding all direct sunlight. The following results were obtained.

Table 20	Number Beetles Emerging	Number Beetles per 100 Fruits
Protected with cotton sheeting(1) 200 apples	75	100 Truits
(2) 75 apples	25	36
Exposed to direct sunlight(1) 200 apples	25	
(2) 75 apples	6	11
Protected with building paper (1) 75 plums	11	14
Exposed to direct sunlight(1) 75 plums	9	12

There is undoubtedly some advantage from exposure of dropped fruits to direct sunlight but control by this means is far from complete in Connecticut. Raking drops into the open is a comparatively simple operation in completely cultivated orchards. but a difficult one in orchards kept in sod. The time required for collecting drops by hand from a 17-year old tree is at least 30 minutes for one man (tree in sod or partly cultivated) and probably one hour for large trees. To obtain the greatest benefit, this operation requires repetition at least twice during the summer and the cost may be figured at 50 cents to \$1.00 per tree for \$4.00 a day labor. At best, the operation is expensive, although it is fully recognized that the use of such labor often depends upon the value of the crop. At present the feeling among the growers in this section is that the plan is not practical. There is no reason, however, why such a method cannot be employed to advantage where a few trees are involved, though it should be recognized that complete control will not be obtained where there are other infested fruits in the vicinity: neither will the result of this work be evident until the following year.

# SPRAYING EXPERIMENTS FOR CURCULIO CONTROL—1924-1928

In view of the rather wide variation in recommendations for spray control of the curculio on apple and the failure in several instances to obtain satisfactory results in Connecticut with the schedules in use, a program of control experiments was devised and carried out with a view to finding more satisfactory measures than were available at the beginning. It became apparent at once that the 7-day spray advocated by the New Jersey Station

4.8

offered one solution although a difficult one from the standpoint of many growers with large orchards and limited equipment. If some means could be devised whereby this additional spray might be avoided it would be of considerable advantage. It became desirable to find out, therefore, which sprays, if any, could be omitted and it seemed also worth while to distribute the sprays more evenly throughout the life activities of the adult which as shown by the diagram, Figure 25, do not correspond fully with the pink-calyx-7-day-2-weeks schedule. However, the data obtained in 1928 on the egg-punctures of the beetle seem to indicate that a pink, calyx, 2 weeks, or 17-day schedule, piles up the poison on the trees before the peak of this activity is reached, and the amount of damage occasioned by late feeders, being relatively small, can probably be disregarded in commercial orchards of Connecticut. It also seemed desirable to try addition of such substances as molasses as an attractive food for the beetle or any new arsenical substitute which might come into use, because of the danger of foliage injury from the usual mixtures and the growing belief that the arsenical residue on sprayed fruit at harvest is too large. A comparison of dusts and sprays used in a pink, calvx, 7-day and two weeks schedule was tried out in 1928. Nicotine dust was also tried in 1924 and shown to be worthless for control.

# SPRAY APPARATUS AND METHODS USED

All trees were sprayed at first with 12-foot rods, provided with angle disc nozzles. Sprays were applied in 1924 and 1925 with an Arlington XL sprayer furnishing 150-200 pounds pressure and in 1926, 1927 and 1928 with a Bean sprayer furnishing about 250 pounds. Spiral nozzles of the type shown in Plate VIII, a and b, were used in 1926, 1927 and 1928 and were very effective in furnishing a thorough even coating on fruit and foliage. In 1928 an extension of about four feet was used on one of the rods allowing the operator to reach the tops of the trees more easily. With age, the spiral nozzles became worn through the center and allowed a solid stream to issue at the end. One of our nozzles, however, was used for three years before developing this defect, but it would no doubt last a much shorter time in large orchards where kept in continual use. When operating to best advantage, we were able with this apparatus, using two lines of hose, to apply about 1,000 gallons in the course of an eight-hour day.

The method of spraying consisted of passing completely around each tree, coating as thoroughly as possible all parts thereof. The amount applied to each tree varied with the size of the tree, but from calculations made at Shepard's orchard and that of the Experiment farm at Mount Carmel, the amount applied was approximately one gallon at 7-day and two week periods for

every 24 cubic yards. This will mean approximately one gallon or more for every 10 feet of circumference measured around the outer branches of such varieties as Baldwin, McIntosh and Greening, where these varieties are normal in shape and 15-20 years old. The exact amounts as figured would be as follows for the three varieties:

Table 21

Circumference, Baldwin Gallons Spray Greening McIntosh Feet

100 16.6 14.1 17.0 80 9.3 6.6 10.4

4.7

60

The volume increases rapidly with circumference, one large Greening tree 125 feet around containing about 760 cu. yards. At the rate mentioned this tree should require about 30 gallons at the 7-day period.

. . . .

Trees will naturally vary in the number of branches according to the method and amount of pruning and consequently the amount of foliage, but it is believed that the trees experimented upon in two different orchards represent fairly typical conditions for the average Connecticut orchard. The amount applied at the calyx period was about two-thirds the amount of the 7-day because of the smaller amount of foliage, while the pink spray required only one-third of the amount of the 7-day.

The main reason for using the circumference instead of the height in calculating volume lies in the fact that volume seems to be more closely correlated with circumference in our orchards when figured on the formula

$$_{\nu}$$
 (cu. yds.) =  $\frac{C}{4\pi x^{27}} \left( \frac{0}{2} - .144C \right) = \frac{C}{339.1} \left( \frac{0}{2} - .144C \right)$ 

The three varieties mentioned assume a fairly typical shape in middle-aged trees and in older trees if there is plenty of room around each tree. There are cases, however, where the shape of the tree is irregular, due to manner of growth or from being crowded. In the latter cases probably the best method of obtaining the amount of spray would be to rely on careful measurements as suggested by Smith (l. c.) but it is very doubtful whether the orchardist would consider such an operation as practical or even advisable. A certain amount of judgment should always be exercised by the operator, with the aim in view of covering the leaf and fruit surface as completely as possible. A rough estimate of the amounts needed on regular shaped trees can be obtained by

<sup>&</sup>lt;sup>1</sup>See Smith, Jour. Ec. Ent. 1927. Formula used  $v(\text{cu.ft.}) = \frac{C_2}{4\pi} \binom{9}{2} - .144\text{C}$ .

ATION BULLETIN 301

circumference measurements, as a basis for control operations. It should also be stated that the experimental trees were headed low, the outer branches being quite near to or touching the ground.

Dusting was done with a power Niagara duster, covering both sides of the tree as completely as possible at each application.

In the course of these experiments a total of more that 700,000 individual apples were examined, representing 1,400 barrels of fruit.

#### MATERIALS USED

Materials used throughout these experiments were standard materials sold on the market and consisted of lead arsenate, dry lime sulphur, casein lime and 40 per cent nicotine sulphate. Nicotine sulphate was omitted wherever possible and the experiments at the Station farm were conducted wholly without this material. At Shepard's Orchard it was used in the early sprays (pink and calyx) in 1925 and 1928, but not in 1926 and 1927.

The casein lime was a standard brand consisting of two parts of lime and one part casein analyzed by the Department of Chemistry and reported in 1925, Bulletin 272, p. 149, under analytical number

2475.

Two brands of lead arsenate were used which have also been analyzed by the Department of Chemistry and reported in Bulletin 272, p. 145, under analytical numbers 2474 and 2464. Both products contained approximately 30 per cent arsenic oxide and .18 per cent water soluble arsenic.

The fish oil used in 1926 and 1927 was purchased on the local market as light pressed menhaden oil, but differed considerably from that obtained from Mr. Ashworth in 1928, in being much thicker and heavier. The analysis of our 1928 material follows:

Specific gravity	.9324
Saponification number	187.4
Iodine number	173.3
Free fatty acids as oleic acid	3.46%

The two dusts used were three per cent nicotine dust, and 90-10 sulphur arsenate dust consisting of 90 per cent finely ground sulphur and 10 per cent lead arsenate.

# EXPERIMENT STATION ORCHARD

Spraying experiments were begun in 1924, using the Experiment Station orchard at Mount Carmel. This orchard consists of a number of varieties arranged in rows running the length of the orchard or part way through. There are 96 trees in all, arranged in six rows. As far as could be determined, the varieties appeared to show about equal infestations, but with a systematic decrease

in amount of injured fruit from the outside rows towards the center, and being greater in the lower parts of the orchard to the west. Work during this summer was more or less preliminary and consisted of small tests on 7-14 trees each, taking in entire rows or parts thereof, and scoring all fruit from the orchard. Owing to the variation in degree of infestation in the orchard it. was recognized that these results could not be fully relied upon. although giving a general indication of the amount of control. Consequently, the tests were continued in 1925 in different locations which procedure gave a considerably better idea of their value. In 1924, it will be seen that there were fully twice as many punctures on the early drop fruits as on the picked fruit and windfalls harvested later, except on the unsprayed trees where the number was about equal. Tree shaving all sprays had slightly more unmarked fruit than any others. The plot having no 7-day spray. however, averaged 85.5 per cent clean as compared with 86.6 per cent for the complete schedule. None of the remaining treatments averaged as high as these.

It should be mentioned here that several factors influence the abundance of curculios on the trees and have been taken into

consideration in evaluating results. They are:

(1) Location of the tree in the orchard—outside trees or trees

near fence rows are always more heavily infested.

(2) Number of apples per tree. Very few apples on a large sprayed tree are not usually injured. A medium or light crop, one barrel or less, on a tree normally producing six or seven, is usually injured severely while a maximum crop, well sprayed, usually shows a relatively small percentage of punctures.

(3) Size or volume of the tree itself has influenced results.

TABLE 22—RESULTS OF SPRAYING EXPERIMENTS FOR CONTROL OF PLUM CURCULIO ON APPLE—1924.—Detailed Record

Tree Nos.	Total No. Apples	No. Good Apples	No. Marked Apples	Total No. Punctures	Per Cent Marked Apples	No. Punctures per Apple	Kind of Fruit Scored	Treatment
A 1-3, 5-12, 14-16	664 480 669 3,409 5,222	$ \begin{array}{r} 165 \\ 96 \\ 410 \\ 2,051 \\ \hline 2,722 \end{array} $	499 384 259 1,358 2,500	$ \begin{array}{r} 1,354 \\ 1,124 \\ 654 \\ 3,238 \\ \hline 6,370 \end{array} $	75.0 80.0 38.7 39.8  47.8	$ \begin{array}{c} 2.04 \\ 2.34 \\ 0.98 \\ 0.95 \\ \hline 1.22 \end{array} $	Drops Drops Windfalls Picked Total	Molasses plus lead arsenate Calvx and 2 weeks
A 4 and A 13	275 89 154 265 783	$ \begin{array}{r} 36 \\ 2 \\ 24 \\ 60 \\ \hline 122 \end{array} $	239 87 130 205 ——————————————————————————————————	622 280 432 583 	86.9 97.7 84.4 77.3 84.4	2.26 3.15 2.80 2.20 	Drops Drops Windfalls Picked	None

Table 22—Results of Spraying Experiments for Control of Plum Curculio on Apple—1924—Detailed Record

			OI, IIIIII	1924	-Detail	ed Kec	ora	
Tree Nos.	Total No. Apples	No. Good Apples	No. Marked Apples	Total No. Punctures	Per Cent Marked Apples	No. Punctures per Apple	Kind of Fruit Scored	Treatment
B 1-3, 5-12, 14-16	621 515 482 4,428	406 305 404 3,756	215 210 78 672	450 467 114 1,199	34.6 40.7 16.2 15.2	$     \begin{array}{r}       0.72 \\       0.91 \\       0.24 \\       0.27     \end{array} $	Drops Drops Windfalls Picked	Calyx, 7-day and 2 weeks; No pink
	6,046	4,871	1,175	2,230	19.4	0.37	Total	
B 4 and B 13	322 93 292 315	. 89 10 210 220	233 83 82 95	515 273 166 208	72.3 $89.2$ $28.1$ $30.2$	$\begin{array}{c} 1.60 \\ 2.93 \\ 0.57 \\ 0.66 \end{array}$	Drops Drops Windfalls Picked	None
	1,022	529	493	1,162	48.2	1.13	Total	
C 1- 3, 5- 8	2,386 1,429 764 3,301	1,471 870 616 2,739	915 559 148 562	2,017 1,044 232 952	38.3 39.2 19.3 17.0	$\begin{array}{c} 0.84 \\ 0.73 \\ 0.30 \\ 0.29 \end{array}$	Drops Drops Windfalls Picked	Pink, 7-day and 2 weeks; No calyx
	7,880	5,696	2,184	4,245	27.7	0.54	Total	
C 4	34 133 363	7 178 135	27 55 128	$75 \\ 84 \\ 263$	$79.4 \\ 41.3 \\ 35.2$	$2.20 \\ 0.63 \\ 0.72$	Drops Windfalls Picked	None
	530	320	210	422	39.6	0.79	Total	
C 9–12, 14–16	2,477 1,638 1,336 4,302	786 307 339 1,270	1,691 1,331 997 3,032	4,135 4,531 3,306 9,546	68.2 81.2 74.6 70.5	1.67 2.77 2.47 2.22	Drops Drops Windfalls Picked	Commercial nicotine dust
	9,753	2,702	7,051	21,518	72.29	2.20	Total	
C 13	1,766 576 521 2,258	728 105 265 805	1,038 471 256 1,453	1,910 945 488 3,114	58.7 81.7 49.13 64.35	1.08 1.64 .93 1.38	Drops Drops Windfalls Picked	None
	5,121	1,903	3,218	6,457	62.83	1.26	Total	
D 1-3, 5-8	803 1,030 990 4,430	669 893 853 3,781	134 137 137 643	251 230 210 672	16.69 13.30 13.84 14.51	0.31 0.22 0.21 0.15	Drops Drops Windfalls Picked	Pink, calyx and 2 weeks; No 7-day
	7,253	6,202	1,051	1,363	14.49	0.19	Total	
D 4	317 228 294 1,129	107 63 99 326	210 165 195 803	458 375 448 2,360	66.24 72.37 66.33 71.12	1.44 1.64 1.52 2.09	Drops Drops Windfalls Picked	None
	1,968	595	1,373	3,641	69.77	1.85	Total	

Table 22—Results of Spraying Experiments for Control of Plum Curculio on Apple—1924—Detailed Record

Tree Nos.	Total No. Apples	No. Good Apples	No. Marked Apples	Total No. Punctures	Per Cent Marked Apples	No. Punctures per Apple	Kind of Fruit Scored	Treatment
D 9-12, 14-16	359 221 843 2,160	87 63 286 407	272 158 557 1,753	262 627 1,957 3,688	75.76 71.5 66.07 81.16	.72 2.83 2.32 1.70	Drops Drops Windfalls Picked	Commercial nicotine dust
	3,583	843	2,740	6,534	76.47	1.87	Total	
D 13	26 66 145 176	9 11 51 37	17 55 94 139	36 217 207 377	65.38 83.33 64.83 78.98	1.38 3.28 1.42 2.14	Drops Drops Windfalls Picked	None
	413	108	305	837	73.85	2.02	Total	
E 1-3, 5-12, 14-16	207 238 328 2,921	120 147 279 2,607	87 191 49 314	205 256 93 673	42.1 38.27 14.94 10.3	$0.99 \\ 1.07 \\ 0.28 \\ 0.2$	Drops Drops Windfalls Picked	Pink, calyx and 7-day; No 2 weeks
	3,694	3,053	541	1,227	14.66	0.33	Total	
E 4 and E 13	97 106 106 1,168	12 34 47 430	85 72 59 738	50 266 154 2,129	87.5 67.94 55.6 63.0	0.5 2.5 1.4 1.8	Drops Drops Windfalls Picked	None
	1,477	523	954	2,599	64.58	1.76	Total	
F 1-3, 5-12, 14-16	6,320 2,915 4,561 10,487	5,096 2,302 4,220 9,433	1,224 613 341 1,054	2,266 1,514 703 1,728	$19.37 \\ 21.03 \\ 7.45 \\ 10.05$	$0.35 \\ 0.52 \\ 0.15 \\ 0.16$	Drops Drops Windfalls Picked	Pink, calyx, 7-day and 2 weeks
	24,283	21,051	3,232	6,211	13.31	.25	Total	
F 4 and F 13	1,070 319 194 211	315 39 91 69	655 280 103 142	1,411 798 194 350	61.2 87.8 52.5 67.3	1.32 2.50 1.0 1.66	Drops Drops Windfalls Picked	None
	1,794	514	1,180	2,753	65.7	1.53	Total	
A, B, C D, E & F, 4 & 13	3,907 1,610 1,706 5,885	1,404 342 787 2,182	2,504 1,268 919 3,703	5.077 3,238 2,089 9,384	64.1 78.8 53.9 62.9	1.29 2.01 1.22 1.59	Drops Drops Windfalls Picked	Summary of all check trees
0	13,108	4,714	8,394	19,788	64.04	1.51	Total	
Special	290	148	142	321	48.96	1.10	Drops	Two sprays of lead arsenate
	831	563	268	519	32.3	.625	Picked	with baits.

# TABLE 23—Showing Comparison of Different Spray Treatments Experiment Station Farm—1924

BULLETIN 301

Treatment No Pink Spray Calyx, 7-day and 2 weeks	Per Cent Unmarked Fruit 62.5 83.9	Kind of Fruit Drops Picked	Total Per Cent of Unmarked Fruit
No Calyx Spray Pink, 7-day and 2 weeks	61.6	Drops Picked	72.22
No 7-day Spray Pink, calyx and 2 weeks	$\frac{85.2}{85.6}$	Drops Picked	85.5
No 2 weeks Spray Pink, calyx and 7-day	60.0 88.8	Drops Picked	82.67
All Sprays Pink, calyx, 7-day and 2 weeks	$\frac{80.1}{90.7}$	Drops Picked	86.69
Check—no spray	$\frac{31.6}{39.1}$	Drops Picked	35.96

# EXPERIMENT STATION FARM, 1924

#### Notes:

Trees planted 1911.

Percentage based on counts varying from 3,583 to 24,283 for each treatment.

Peach orchard alongside removed in winter of 1923-24; early drops collected and larvae trapped in cages 1924.

#### Spray formula used

Lead arsenate	3 pounds
Lime sulphur (liquid)	3 gallons
Casein lime	1 pound
Nicotine sulphate	1 pint
Water1	00 gallons

## Dates of Spray Applications

Pink														. ]	May	13	
Calyx.										×.					June	2	
7-day.									5 3						Tune	10	
2-weeks	3								3 4	163		3/1	0		Tune	16	

#### RESULTS IN 1925

The work was continued in 1925, shifting blocks, and also introducing several new treatments. First a schedule comprising pink, four days after calyx, five days later and two weeks after the five-day spray. (2) use of coated lead arsenate (home made). (3) double the usual strength of lead arsenate and (4) fish oil sticker in calyx spray without lime sulphur. This year the spray containing fish oil plus lead arsenate applied at the calyx period

without lime sulphur was slightly better than any other treatment, although the difference between it and the complete schedule was only 0.2 per cent. The "no calyx" treatment stood high in clean fruit indicating that beetles did not get started in the orchard this year until some time after this period. The trees from which the 7-day spray was omitted were low in percentage of clean fruit. In these tests, the complete schedule again stood practically as high as any other, being only .16 per cent less than the highest score.

TABLE 24—RESULTS OF SPRAYING EXPERIMENTS FOR CONTROL OF PLUM CURCULIO ON APPLE—1925

## EXPERIMENT STATION FARM: DETAILED RECORD

Tree Nos.	Kind of Fruit	Total No. of Apples	No. Marked by Curculio	Per Cent Marked Fruit	Total No. of Punctures	Average No. Punctures Per Apple	Treatment Received
A 1-3 5-8	Drops Picked	1,656 6,509	808 1,156	48.79 17.76	1,366 1,814	.82 .28	Pink: 6 days after calyx 8 days after 6-day;
	Tota1	8,165	1,964	24.05	3,180	.39	2 wks. after 8-day; No lime-sulphur at 6- day
A 4	Drops Picked	56 621	45 488	80.35 78.58	76 1,069	1.35 1.72	Check—no treatment
	Tota1	677	533	78.72	1,145	1.69	
A 9-12 14-16	Drops Picked	2,086 6,879	1,506 1,940	72.20 28.20	2,817 2,834	1.35	Pink; 6 days after caly with coated lead arsenate: 2-weeks
	Tota1	8,965	3,446	38.43	5,651	.63	ate, 2-weeks
A 13	Drops Picked	267 416	264 289	98.87 69.47	650 671	2.43 1.61	Check—no treatment
	Total	683	553	80.96	1,321	1.93	
B 1-3	Drops Picked	7,456	395 636	58.34 8.53	745 1,080	1.10	Pink; heavy dose lead arsenate at calyx and 2-weeks
	Total	8,133	1,031	12.67	1,825	.22	Z-weeks
B 4	Drops Picked	2,104 3,579	1,562 1,514	74.23 42.30	2,950 2,616	1.40	Check—no treatment
	Total	5,683	3,076	54.12	5,566	.98	
B 9-12 14-16	Drops Picked	1,887 1,933	83 149	4.40 7.71	122 233	.06	Pink; calyx with fish oil sticker, no lime-sulphur; 2 weeks.
	Total .	3,820	232	6.07	355	.09	surphur, 2 weeks.
B 13	Drops Picked	309 479	178 273	57.60 56.99	433 704	1.40	Check—no treatment
		788	451	57.23	1,137	1.44	
C 1-3 5-12	Drops Picked	5,596 17,708	1,052 1,358	18.80 7.67	1,591 2,228	.28	No 2-weeks; Pink; calyx; 7-day
14-16	Total	23,304	2,410	10.34	3,819	.16	
C 4 and 13	Drops Picked	71 699	365 365	57.74 52.21	97 784	1.36 1.12	Check-no treatment
	Total	770	406	52.72	881	1.14	
D 1-3 5-12	Drops Picked	6,545 13,318	518 720	7.91 5.40	886 1,362	.10	All sprays Pink; calyx; 7-day and 2-weeks
14-16	Tota1	19,863	1,238	6.23	2,248	.11	2- WCCKS

Table 24—Results of Spraying Experiments for Control of Plum Curculio on Apple—1925—Concluded

BULLETIN 301

# EXPERIMENT STATION FARM: DETAILED RECORD—Concluded

Tree Nos.	Kind of Fruit	Total No. of Apples	No. Marked by Curculio	Per Cent Marked Fruit	Total No. of Punctures	Average No. Punctures Per Apple	Treatment Received
D 4 and 13	Drops Picked	627 1,818	365 1,369	58.21 75.30	816 4,559	$\frac{1.30}{2.50}$	Check—no treatment
	Total	2,445	1,734	70.92	5,375	2.19	
E 1-3 5-12	Drops Picked	3,556 18,030	201 1,734	5.65 9.61	300 2,819		No calyx Pink; 7-day and 2-wks
14-16	Total	21,586	1,935	8.96	3,119	.14	
E 4 and 13	Drops Picked	1,015 2,672	425 2,039	41.87 76.30	1,112 6,139	1.09 2.29	Check—no treatment
	Total	3,687	2,464	66.82	7,251	1.96	
F 1-3 5-12	Drops Picked	9,620 22,592	1,629 3,268	16.93 14.46	2,929 6,075		No 7-day Pink, calyx and 2-wks
14-16	Tota1	32,212	4,897	15.20	9,004	.27	
F 4 and .13	Drops Picked	2,709 1,635	1,130 1,332	41.71 81.46	2,337 3,640	.86	Check—no treatment
	Tota1	4,344	2,462	56.67	5,977	1.37	
A, B, C, D, E, F,	Drops Picked	7,158 11,919	4,010 7,669	56.02 64.34	8,471 20,182	1.18 1.69	Summary of all checks
4 & 13	Tota1	19,077	11,679	61.22	28,653	1.50	
TT 1 1		1	144 405				

Total number apples scored 144,435

Note—"Drops" include early drops collected until about the middle of July.
"Picked" includes fruit taken from tree and also windfalls at time of harvesting crop.

TABLE 25—Showing Comparison of Different Spray Treatments

#### EXPERIMENT STATION FARM-1925

	Treatment	Per Cent. Unmarked Fruit	Kind of Fruit	Total Per Cent of Unmarked Fruit
1	No calyx Pink, 7-day and 2 weeks	$\frac{94.35}{90.39}$	Drops Picked	91.04
2	No calyx: Pink; 6 days after calyx; 8 days later; 2 weeks after 8 day	$\frac{61.21}{82.24}$	Drops Picked	75.95
3	No 7-day Pink, calyx and 2 weeks	$\frac{83.07}{85.54}$	Drops Picked	84.80
4	No 7-day Pink, calyx with fish oil; 2 weeks	$\frac{95.60}{93.30}$	Drops Picked	93.93
5	No 7-day: Pink—heavy dose lead arsenate at calyx; also at 2 weeks	$\frac{41.66}{91.47}$	Drops Picked	88.33

TABLE 25—Showing Comparison of Different Spray Treatments
—Concluded

# EXPERIMENT STATION FARM-1925-Concluded

		Per Cent Unmarked Fruit	Kind of Fruit	Total Per Cent of Unmarked Fruit		
	No 2 weeks	81.20	Drops	90.66		
6 I	Pink, calyx and 7-day	92.33	Picked	89.66		
7	All sprays	92.08	Drops	93.77		
	Pink, calyx, 7-day and 2 weeks	94.60	Picked	93.77		
	CI 1	43.98	Drops	38.78		
7	Check-no spray	35.66	Picked	98.78		

#### Notes:

Percentages based on counts varying from 3,820 to 32,212 apples for each plot.

Early drops collected 1925.

Amount of spray used 8-14 gallons per tree. All apples scored by individual examinaion.

Orchard cultivated until 1925 when middle section was left in sod.

#### Spray Formulas Used

Lead arsenate	3	pounds	
Nicotine sulphate		pint	
Casein lime	1	pound	
Lime sulphur (dry)	6	pounds	
Water	100	gallons	

Fish oil (light pressed menhaden)

	1 quart
Lead arsenate	3 pounds
	(No. 4 only at calyx)

Water......100 gallons

Lead arsenate....... 6 pounds
(No. 5 only at calyx and
2 weeks)

Water.....100 gallons

## Dates of Spray Applications

Pink,—April 29. Calyx,—May 19. 6-day,—May 25. 7-day,—May 26. 2-weeks,—June 1 and 2. 4-weeks,—June 16.

#### RESULTS IN 1926.

Coated lead arsenate was tried with a view to eliminating some of the sprays but without success. It should be said, however, that the material used was not the same as the product manufactured later and probably failed for this reason. Colloidal arsenate of lead was used as a substitute for acid lead arsenate with good results. During 1926, the plots treated with lead arsenate and fish oil at the calyx period fell below the complete schedule by 1.3 per cent. The total score of all trees receiving the treatment, however, fell somewhat below this due to the

high score of one outside tree. This tree seemed to be so much out of line with the results obtained from the rest that it seems justifiable to omit it from the final figures:

BULLETIN 301

TABLE 26—RESULTS OF FISH OIL—LEAD ARSENATE TEST

Tree No.	Total number apples	Number marked by curculio	Size of tree cubic yards	Per cent marked fruit
B1	2,840	705	330	24.8
B2	3,707	231	318	6.2
B3	1,976	.78	286	3.9
B5.	4,607	227	345	4.9
B6	2,550	90	400	3.5
B7	1,502	52	343	3.5
B8	735	18	278	2.4
Average:	marked fruit o	mitting Bl.		4.6

TABLE 27—RESULTS OF SPRAYING EXPERIMENTS FOR CONTROL OF PLUM CURCULIO ON APPLE—1927

#### EXPERIMENT STATION FARM: DETAILED RECORD

			'a Dilli	OII I HICH	· DEIA	ILED RECORD
Tree Nos.	Total No. of Apples	No. Marked by Curculio	Per Cent Marked Fruit	Total No. of Punctures	Average Punctur Per App	es Treatment
A 1-3 5-8	6,670	1,333	19.9	2,474	.37	No 7-day; pink, calyx with regular lead arsenate (no lime sulphur), 2-weeks
A 9-12 14-16	10,988	1,807	16.4	3,797	.34	No 7-day; pink, calyx with coated lead arsenate (no
A 4 and	1,552	1,207	77.1	4,104	2.64	lime sulphur), 2-weeks Check—no treatment
B 1-3 5-8	18,267	1,751	9.6	2,872	.16	No 7-day; pink; calyx with fish oil and lead arsenate
C 1-3 5-8	23,169	1,672	7.2	3,213	.14	fish oil and lead arsenate,
B4 and C4	3,221	871	27.0	1,649	.51	no lime sulphur Check—no treatment
B 9-12 14-16	4,816	522	10.8	634	.13	No calyx; pink; 4 days after calyx; 5 days later;
C 9-12 14-16	3,093	232	7.5	363	.12	12 days after 5-day (complete mixture)
B 13 C 13	3,181	1,667	52.4	3,575	1.12	Check—no treatment
D 1-3 5-8	5,965	87	1.46	153	.025	Complete schedule; pink (ferrous ars.); 7-day, calyx Colloidal lead ars.; 2-weeks
D 4 D 9-12 14-16	1,820 4,061	702 136	38.86 3.34	1,333 286	.73 .07	Check—no treatment All sprays; pink; calyx; 7-day; 2-weeks; complete mixture
D 13 E 1-3 4-12 14-16	398 2,889	113 143	28.4 4.9	264 389	.66	Check—no treatment No pink; calyx; 7-day and 2-weeks; complete mixture
E 4, 13 F 9-12	853	421	49.3	1,235	1.45	Check—no treatment
14-16 F 13 Checks	10,664 66 11,091	578 59 5,040	5.4 89.4 45.44	1,159 342 12,502	.10 5.18 1.12	No 7-day; pink; calyx and 2-weeks; complete mixture Check—no treatment

Notes—Drop fruits were not collected in this orchard in 1926. Spray dates: Pink, May 4. Calyx, May 28

Calyx, May 28
7-day, June 4
2-weeks, June 11
4-weeks, June 24

Formula: Complete mixture and fish oil same as in 1925.
Colloidal and ferrous arsenates at 3 pounds per 100 gallons.
Coated lead arsenate at 8 lbs. paste per 100 gallons.

#### RESULTS IN 1927

During this year the orchard was divided into three blocks of 24-30 trees each, the plots being sprayed with (1) pink, calvx. 7-day and 2-weeks schedule, (2) pink, calyx, with fish oil and lead arsenate, and 2-weeks sprays, (3) pink, calyx, 2-weeks and 4-weeks sprays. Casein lime was added to plot (2) at the calvx period causing much of the spray to run off the foliage and was thought to be responsible for the poor showing of this plot. The results again showed that the complete schedule (1) was better and on the end row, some 8-15 per cent better than other schedules. The total percentage, however, showed little difference (2-5 per cent), but the considerably improved control of this schedule since 1924 began to be apparent, indicating that the most consistently good results are to be expected from it. A diagram of the orchard with percentages of marked fruit on the count trees is shown below and indicates the variation in percentage of marked fruit in various locations. This year outlying wild apples were sprayed within a fourth of a mile of the orchard.

PLAN AND RESULTS OF EXPERIMENTS AT STATION FARM, MOUNT CARMEL
IN 1927

	IN 1927														
						N<-		<b>'</b>							
F X		3 11	4	5 X	6 34	7 X	8 5	9	10 X	<u>11</u>	12 X	13 4	14	15 X	16 X
EX	X	5	7	X	74	X	3	5	X	19	X	3	0	X	X
D X	X	13	3	X	73	X	2	2	X	44	X	1	6	X	X
C X	X	_		X	_	X		2	X	15	X	_	3	X	X
вх	X	_	8	X	_	X	5	2	X	21	X	0	1	X	X
A 19	15	17	23	<u>.                                    </u>	94	20	15	13	4	24	7	5	5	5	9
Treatments: pink, calyx, 2 weeks, 4 weeks.			check	pink, calyx with fish oil, 2 weeks.			check		k, ca week		7-da	y,			

Notes:

All percentages referred to nearest whole number; represent fruit marked by curculios.

Trees left blank bore no fruit or so little that the count was worthless.

All picked fruit from count trees scored.

Outlying wild apple trees sprayed with lead arsenate at calyx period.

Drop fruits not collected in 1927.

Row A next a fence row.

Total

8,216

3,297

Spray dates: Pink-May 6

Calyx—May 26-7 7 day—June 4 2 weeks—June 9 4 weeks—June 23

Formula:

Lead arsenate3 pounds.Lime sulphur (dry)6 pounds.Casein lime1 pound.Water100 gallons

Table 28—Results of Spraying Experiments for Control of Plum Curculio on Apple—1927

#### EXPERIMENT STATION FARM: DETAILED RECORD

Tree Nos.	Total No. of Apples	Number Marked by Curculio			Average Punctu Per Ap	res Treatment
A 1 A 2 A 3 A 4 A 5 B 3 B 4 D 4 E 3 E 4 F 3, 4	962 1,166 586 260 7 9 1,424 1,71 1,014 2,098 1,261 4,512	182 176 99 60 0 112 22 36 102 86 486	18.9 15.1 16.9 23.0 0.00 0.0 7.8 12.9 3.5 4.8 6.8 10.8	686 296 208 139 0 0 169 55 69 163 198 808	.71 .25 .35 .534 .0 .0 .12 .32 .068 .077 .15	Pink, calyx, 2-weeks, 4-weeks
Total	13,470	1,361	10.1	2,791	.20	
A 7 A 8 A 9 A 10 B 8 B 9 C 9 D D 9 E 8 E 9 F 8	1,586 2,366 260 178 19 450 626 667 1,452 2,792 2,496 4,543	321 357 33 7 1 10 16 12 35 93 118 238	20.2 15.1 12.7 3.9 5.2 2.2 2.5 1.8 2.4 3.3 4.7 5.2	667 601 44 10 2 16 29 16 52 180 205 382	.42 .25 .16 .056 .105 .035 .046 .024 .035 .064 .082 .084	Pink, calyx (with fish oil), 2-weeks
Tota1	17,435	1,241	7.11	2,204	.126	
A 12 A 13 A 14 A 15 A 16 B 14 B 13 C 14 D 13 D 14 E 13 E 14 F 13	2,139 218 466 2,004 3,855 167 50 5,890 626 2,089 396 3,185	156 1 222 97 359 2 0 168 74 118 14 0 139	7.3 .45 4.7 4.8 9.3 1.2 0 2.8 1.2 5.6 3.5 0 4.3	288 1 36 114 547 5 0 248 23 197 36 0 159	.13 .04 .077 .056 .14 .029 .0 .042 .036 .094 .091	Pink, calyx, 7-day, 2-weeks
Total	21,124	1,150	5.44	1,654	.078	
A 6 A 11 B 11 C 11 D 6 D 11 E 6 E 11 F 6	292 152 179 1,491 1,056 809 916 1,461 1,860	274 36 37 221 772 358 682 276 641	93.8 23.7 20.6 14.8 73.1 44.2 74.4 18.9 34.4	1,259 69 51 343 2,031 781 1,869 412 1,200	4.30 .45 .28 .23 1.92 .96 2.04 .28 .64	Check—no treatment

40.13 8,015

#### RESULTS IN 1928

The same plots were used in 1928 as in 1927, but with one different treatment; namely, a pink, calyx, 10-day, and 20-day schedule. There appeared to be little difference in any of the plots, indicating that they were about equal in curculio control. The pink, calyx, 7-day and two-weeks spray and the pink, calyx, 10-day and 20-day sprays, gave more uniform control. A comparison of several trees of the same variety and the same relative location in the orchard on the other hand, indicates a slight but not significant advantage for the schedule containing fish oil over that omitting it, Table 29. Wild apples in the vicinity of the orchard were sprayed again.

			Т. г.	LE 29		
			IAB	LE 49		
Variety	Tree No.	Total Apples	Height in Feet	Size cu. yds.	Per cent marked	Treatment
Baldwin	A2	2,179	20	277	13.3	Pink, calyx, 7-day, 2 weeks.
"	A13	2,811	19	274	5.1	pink, calyx with fish oil. 2 weeks.
u	A4	2,294	20	318	8.3	pink, calyx, 7-day 2 weeks.
"	A12	2,125	21	309	7.1	pink, calyx with fish oil, 2 weeks.
Greening	D4	1,318	20	310	7.7	pink, calyx, 7 day, 2 weeks.
u	D13	1,419	19	310	2.1	pink, calyx with fish oil, 2 weeks.
u	D3	1,417	18	233	4.3	pink, calyx 7-day, 2 weeks.
a	D14	1,325	18	312	5.0	pink, calyx with fish oil, 2 weeks.

The total percentage of marked fruit is almost identical on the outside row in the different plots and further indicates that the amount of control was almost the same in all three while mathematical computations indicate no significant advantage of any schedule. This year was a bad year for russeted fruit and scab. No McIntosh trees received the fish oil-lead arsenate spray in 1928, but it is reasonable to expect decreased control of scab when such schedules are used. On trees not scabbing badly results were good, the foliage being better and with less spray burn, due no doubt to decreased amount of spray. Advantages of this schedule lie in its greater economy since one application is omitted, and its greater safety—probable disadvantages in lessened fungous control.

A comparison was also made in 1928 of 90-10 sulphur arsenate dust and spray, both dust and spray being applied on the same schedules—pink, calyx, 7-day and 2-weeks. Results indicate a slightly better control of curculio by spraying although the differ-

ence is not great. Thus the per cent of unmarked fruit in the sprayed plot for this year totaled 96.6 per cent while the unmarked fruit on the dusted plot totaled 92.7 per cent. This corresponds in general with results obtained in the Milford experiments of Zappe and Stoddard.

BULLETIN 301

PLAN AND RESULTS OF EXPERIMENTS TO CONTROL CURCULIOS AT EXPERIMENT STATION FARM IN 1928

						180.00										
						*******	N<									
								V	     							
	1	2	3	4	5	6	6	8	8	10	11	12	13	14	15	16
F	X	X	2.8	1.7	45	X	X	4.3	3.2	X	X	28	25	2.2	X	X
E	X	X	3.1	4.9	X	20	X.	3.5	02	X	9.5	3.1	1.4	1.0	X	X
D	X	X	4.3	7.7	X	33	X	6.0	2.7	X	27	X	2.1	5.0	X	X
C	X	X	3.2	2.2	X	11	X	1.4	3.0	X	22	X	3.5		X	X
В	X	X	4.6	5.2	X	9	X	2.3	3.8	X	12	X	4.8	7.4	X	X
A	6.4	13.3	11.7	8.3	9.2	30	8.0 1	0.7	7.1	10.2	33	7.1	5.1	6.3	7.0	_
Treatments: pink, calyx, 7 day, 2 weeks.					check	pink, 20-da	calyx y.	x, 10	day,	check			caly:			

Notes:

Outlying wild apple trees sprayed with lead arsenate and fish oil sticker at calyx period. No drop fruits collected.

Spray dates: pink—May 8

Calyx—May 28 7-day—June 4 10-day—June 7 2-weeks—June 11 20 day—June 18

Formula—

TABLE 30—RESULTS OF SPRAYING EXPERIMENTS FOR CONTROL OF PLUM CURCULIO ON APPLES

#### EXPERIMENT STATION FARM-1928

		LAIL	ittime					
	Apples	Number Marked by Curculio	Fruit Pi	unctures	Average N Punctures Per Apple	Trea	tment	1
A123AA453BBCCDD434BBCCDDEFF	235 2,179 844 2,294 3,971 3,083 2,308 3,175 2,811 1,417 1,318 1,001 896 3,914 2,740	15 290 99 191 366 143 120 103 61 61 101 31 44 109 47	6.4 13.3 11.7 8.3 9.2 4.6 5.2 2.2 4.3 7.7 3.1 4.9 2.8	36 450 136 288 521 186 177 132 73 69 177 52 66 111	.15 .21 .16 .12 .13 .06 .08 .04 .02 .05 .13 .05 .07 .03	Pink, calyx, 7	-day, 2-	weeks
Totals	32,186	1,781 5	$.53 \pm 1.95$	2,531	.079			
A7 A8 A9 B8 B9 CC9 D8 D9 F8 F9	4,075 2,747 3,952 1,859 1,699 2,360 944 708 1,171 1,498 676 234 3,676	327 293 281 189 40 91 13 21 70 41 24 10 120	8.0 10.7 7.1 10.2 2.3 3.8 1.4 3.0 6.0 2.7 3.5 4.3 3.2	448 341 356 275 96 138 21 28 112 51 24 3 137	.11 .12 .09 .15 .06 .06 .02 .04 .09 .03 .03 .01	Pink, calyx, 1	0-day, 2	20-day
Totals	25,599	1,520 5	.93±2.29	2,030	.079			
A 12 A 13 A 14 A 15 B 13 B 14 C 13 D 13 D 14 E 12 E 12 F 13 F 14	2,125 2,811 3,613 2,233 2,218 1,055 3,204 1,419 1,325 3,185 1,258 695 479 1,400	153 144 228 379 106 78 112 30 67 98 18 7 120 31	7.1 5.1 6.3 17.0 4.8 7.4 3.5 2.1 5.0 3.1 1.4 1.0 25.0 2.2	237 209 295 617 146 128 138 40 100 133 12 6 167 47	.11 .07 .08 .27 .06 .12 .04 .03 .07 .04 .009 .009	Pink, calyx 2-weeks	with	fish oil,
Totals	27,020	1,571 5	.81±1.87	2,275	.084			
A 6 B 6 C 6 D 6 E 6 F 5 A 11 B 11 D 11 E 11 F 12	1,003 1,447 1,690 1,580 840 352 2,450 2,291 1,126 995 1,149 3,152	302 130 183 515 168 158 818 285 242 255 109 873	30.1 9.0 10.8 32.6 20.0 44.9 33.4 12.4 21.5 25.6 9.5 27.7	602 170 222 864 294 279 1,326 429 393 495 124 1,235	.60 .12 .13 .55 .35 .79 .54 .19 .35 .50	Check—no tr	eatment	1010 0011 10011 10011
Totals	18,075	4,038 22	.26±2.28	6,433	.356			

<sup>1</sup> Does not include fall feeding punctures.

TABLE 31—EXPERIMENTS TO CONTROL THE PLUM CURCULIO ON APPLE: SPRAY VERSUS DUST

Tree	Total Apples	Number Marked b Cur <b>c</b> ulio	Per Cent y Marked Fruit	Total Number of Punctures	Average Number Puncture Per Appl	s Treatment
N 3 N 4 N 6 N 7 N 8	380 432 4,042 2,236 2,034 1,118	119 60 121 9 30 0	31.3 13.9 3.0 .4 1.4	179 87 156 8 45 0	.47 .20 .4 .003 .02	Spray: pink, calyx, 7-day, 2-weeks
Totals	10,242	339 3	.3±3.3	475	.046	
R 5 R 6 R 7 R 8 R 9 S 7 S 8 S 9	1,983 896 1,993 891 2,469 513 1,242 3,994	181 57 90 70 127 6 40 259	9.1° 6.4 4.5 7.8 5.1 1.2 3.2 6.5	272 82 91 101 143 6 62 323	.14 .09 .04 .11 .06 .01 .05 .08	Dust: pink, calyx, 7-day, 2-weeks
Totals	13,981	830	$5.9 \pm 1.6$	1,070	.076	
N 2 O 1	1,068 202	525, 96	49.2 47.5	1,302 216	.122	Check—no treatment
Totals	1,270	621	48.89	1,518	.119	

## Experiments at Shepard's Orchard, Mount Carmel; Results in 1925

This orchard consists of 52 trees of three varieties arranged in six rows. It was divided into two plots with check row through the center. The orchard is provided with abundant sources of infestation, with sprayed and unsprayed apple trees on three sides and a bearing peach orchard on the fourth. In 1925, it received (1) a pink, calyx, 7-day and 2-weeks schedule and (2) a pink, calyx, 2-weeks and five weeks spray. Results favored the condensed schedule (1) by nearly seven per cent.

TABLE 32—RESULTS OF SPRAYING EXPERIMENTS TO CONTROL CURCULIOS: SHEPARD'S ORCHARD—1925

Total Apples	Total Injured	Per Cent Injured	Total Punctures	Average No. Punctures Per Apple	Kind of Fruit Scored	Treatment
421 11,071	113 2,295	26.84 20.73	266 4,718	.63 .43	Drops Picked	7-day April 29, May 19 May 27, June 2
11,492	2,408	20.95	4,984	.43	Total	
421 12,352	113 3,389	26.84 27.43	266 7,616	.63	Drops Picked	5-weeks April 29, May 19 June 2, June 22
12,773	3,502	27.41	7,882	.62	Total	
625 3,040	534 2,369	85.44 77.92	1,574 8,121	2.52 2.67	Drops Picked	Check-no treat-
3,665	2,903	79.20	9,695	2.64	Total	

Table 33—Comparison of Sprays for Curculio Control: Shepard's Orchard—1925

Treatment	Per Cent Unmarked Fruit	Kind of Fruit	Total Per Cent of Unmarked Fruit	
7-day	73.26	Drops	<b>50.05</b>	
7-day April 29, May 19 May 27, June 2	79.27	Picked	79.05	
5-weeks	73.16	Drops	<b>70.70</b>	
April 29, May 19, June 2, June 22	${72.57}$	Picked	72.59	
Check	14.56	Drops		
No Treatment	23.08	Picked	20.80	

Notes:

Trees planted in 1911.

Amount of spray used 5-7 gallons per tree. (Trees bearing most fruit smaller than trees at Experiment Station farm.)

Percentages based on counts varying from 3,665 to 12,773 apples for

This orchard is very heavily infested; has been in sod, but was sprayed regularly by owner in previous years. Cultivated in part during 1925.

#### Spray Formula Used.

Lead arsenate 3	pounds
	pint
Casein lime 1	pound
	pounds
Water100	gallons

#### RESULTS IN 1926

The two plots were reversed during 1926 and were the same except that schedule Number 2 received a 4-weeks after calyx application instead of a 5-weeks. This year the condensed schedule averaged .5 per cent less than the expanded pink-calyx-2-weeks-4-weeks schedule.

#### RESULTS IN 1927

Thinking that our 1926 results might have been influenced unduly by some unseen factor the same sprays were repeated on the same plots with the result that the expanded schedule averaged .1 per cent better. The amount of fruit on the two plots was slightly larger both years on the plot receiving the condensed schedule and if influenced by this factor alone the count should have been much better. The systematic increase in injury occurs on the orchard fringe but is greater towards the west side and should influence both plots alike. This infestation is apparently

Table 34—Results of Spraying Experiments for Control of Plum Curculio: Shepard's Orchard—1926

BULLETIN 301

Row	Kind of Fruit	Total No. of Apples	Number Marked by Curculio	Marked		Average No Punctures Per Apple	Treatment
C 1-4	Drops Picked	2,243 9,325	100 339	4.4 3.6	147 540	.065	Pink, calyx, 7-day and 2-weeks
D 1-4	Drops Picked	5,321 10,752	373 306	7.01 2.8	626 504	.117	
E 1-4	Drops Picked	3,110 5,681	151 91	4.8 1.6	286 196	.0919	
F 1-4 .	Drops Picked	2,403 8,553	85 196	3.5 2.26	121 295	.050 .034	
G 1-4	Drops Picked	583 3,792	26 75	4.4 1.97	37 105	.063	
H 1-4	Drops Picked	925 2,223	95 93	10.2 4.2	155 303	.167	
Tota1	Drops Picked	14,585 40,326	830 1,100	5.69 2.73	1,372 1,943	.094	
Total all	fruits	54,911	1,930	3.51	3,315	.060	
C 6-7	Drops Picked	1,261 4,546	84 200	6.6 4.4	122 292	.09	Pink, calyx, 2-weeks and 4-weeks
D 6-8	Drops Picked	2,193 6,731	96 86	4.4 1.2	263 115	.12	
E 6-9	Drops Picked	1,383 6,266	190 148	13.9 2.3	360 254	.26	
F 6-9	Drops Picked	310 1,624	9 33	2.9 2.0	14 66	.04	
G 6-8	Drops Picked	262 2,719	13 129	4.9 4.7	18 231	.07	
H 6-10	Drops Picked	1,127 2,239	158 183	14.0 8.2	240 406	.21 .18	
Total	Drops Picked	6,536 24,125	550 779	8.41 3.22	1,017 1,364	.15	
Total all	fruits	30,661	1,329	4.334	2,381	077	
Total CDEFG	Drops 5Picked	3,177 8,036	1,398 2,144	44.0 26.7	2,781 4,444	.87. .55	Check—no treat ment
Total all	fruits	11,213	3 542	31.58	7 225	.644	

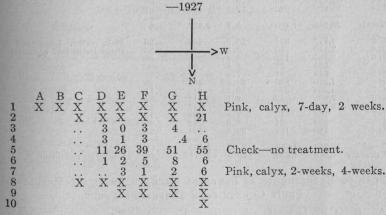
somewhat worse, however, on the northwest corner which falls within the plot receiving the condensed schedule. The only other factor entering the question is the size of the trees. These were measured and it was found that those receiving the condensed schedule are larger. From these facts it becomes apparent that although the count is almost identical, that plot receiving the condensed schedule is in reality better because it controlled equally well with a greater handicap.

TABLE 35—RESULTS OF SPRAYING EXPERIMENTS FOR CONTROL OF PLUM CURCULIO ON APPLE; SHEPARD'S ORCHARD—1927

Tree	Total No. of Apples		Per Cent Marked Fruit F	Total No. of unctures	Average No Punctures Per Apple	Treatment
Nos.	703	23	3.27	36	.051	Pink, calyx, 7-day, 2-weeks
E3	19	0 38	0.0 3.0	0 56	.0	
E3 F3 G3	1,236 836	38	4.5	66	.079	
G3 H2	746	160	21.45	361	.484	
D4	223	7	3.1	9	.040	
E4	95	1	$\frac{1.05}{2.8}$	2 59	.021	
F 4	1,645 868	46	.46	11	.012	
G 4	782	49	6.26	207	.264	
H4						
Totals	7,153	366	5.1	807	.112	
D 6	210	3	1.4	3	.014	Pink, calyx, 2-weeks, 4-weeks
E6	121	2	1.65	2	.016	
F 6	964	55 99	5.7	117 94	.121	
G 6	157 1,704	100	6.3 5.86	161	.598	
H 6 E 7	346	9	2.6	9	.026	
F7	1,441	16	1.1	24	.016	
G7	1,924	42	2.2	64	.033	
H7	1,308	81	6.2	106	.081	
Totals	8,175	407	4.98	580	.071	
D 5, E 5	1,566	628	40.1	1,073	.68	Check—no treatment
F 5, G 5,	11.0					

Note-Drop fruits were not collected in this orchard in 1927.

PLAN AND RESULTS OF EXPERIMENTAL WORK IN SHEPARD'S ORCHARD



Notes

Figures represent percentages of curculio marked fruit. All percentages are referred to the nearest whole number.

All picked fruit from count trees scored.

Trees left blank bore no fruit in 1927.

Spray dates—pink May 6-7

calyx—May 27 7-day—June 4 2-weeks—June 9

#### RESULTS IN 1928

This year the complete condensed schedule (pink, calyx, 7-day and 2-weeks) was retained on the same plot and compared with the same schedule on the remaining plot, varying only in the substitution of basic lead arsenate for the acid form commonly employed. The basic arsenate apparently reduced the amount of russet on Gravensteins by seven per cent, but did not so affect the McIntosh. On the other hand, canker worms did much more damage in this plot and the total percentage of fruit unmarked by curculio was considerably less—90.9 per cent compared with 94.3 per cent for the acid lead arsenate.

TABLE 36—EXPERIMENTS IN CURCULIO CONTROL; SHEPARD'S

			ORC	HARD	1940	
Tree Nos.	Total No. of Apples	Number Marked by Curculio		Total No. of Punctures	Average No Punctures Per Apple	Treatment
C33 E33 FG32 HCD44 EF44 G4	2,749 3,148 4,211 1,795 1,182 1,203 2,154 2,528 2,789 1,978 1,421	211 37 118 20 87 254 216 53 93 88 89	7.67 9.8 2.8 1.1 7.36 21.1 10.0 2.1 3.3 4.4 6.3	286 37 141 24 133 470 328 62 131 129 127	.10 .01 .03 .01 .11 .39 .15 .02 .05	Pink, calvx, 7-day, 2 weeks (acid lead arsenate)
H 4	800	220	27.5	409	.09	
Totals	25,958	1,480	5.7	2,277	.8	
C 6 D 6 E 6 F 6 G 6 H 6 C 7 E 7 F 7 F 7	1,741 2,140 2,792 565 649 1,210 1,894 3,983 2,617 164 825 780	351 60 178 27 51 289 385 82 109 13 31 183	20.2 2.8 6.4 4.8 7.9 23.88 20.32 2.1 4.2 7.9 3.8 23.5	488 75 270 35 73 434 212 111 147 18 42 294	.28 .03 .09 .06 .01 .36 .11 .03 .05 .11	Pink, calyx, 7-day and 2-weeks (basic lead arsenate)
Totals C <sub>5</sub> D <sub>5</sub> E <sub>5</sub> F <sub>5</sub>	19,360 4,848	1,759 1,435	9.18 29.6	2,199 2,211	.11 .45	Check, no treatment

PLAN OF EXPERIMENTS AT SHEPARDS ORCHARD, 1928; AND RESULTS OF FRUIT COUNTS FROM DIFFERENT TREES

Spray dates—Pink—May 10 Calyx—May 24 7-day—May 31 2-weeks—June 7

Formula same as used in Experiment Station Orchard, except for addition of nicotine sulphate at pink and calyx periods.

PLUM CURCULIO ON APPLE

# Assembled Results of Different Sprays for Curculio Control—1924–1928

The following table gives a general summary of the work with sprays conducted between 1924 and 1928 with percentages of fruit unmarked by curculios obtained in the different experiments. Treatments 1-8 were conducted in the orchard of the Experiment Station at Mount Carmel, 9-11 in the orchard of Mr. C. E. Shepard near Mount Carmel. The assembled results show the importance of the 7-day treatment, and the close similarity of the schedule containing fish oil sticker. The unusual uniformity of results obtained in 1928 is probably accidental, but shows that the various schedules employed are about equal in effectiveness; and we believe all of them have considerable merit for control of curculios under Connecticut conditions.

Table 37—Results of Different Sprays for Curculio Control 1924–1928—Picked Fruit Only

	Treatment	1924	Per C	Cent Clean 1	Fruit 1927	1928	
1	No pink	1924	1920	1920	1921	1920	
		83.9		95.04			
2	No calyx						
	Pink, 7-day, 2-weeks	82.5	90.4				
3	No 7-day						
1		85.6	85.54	94.6			
4	No 7-day						
	Pink, calyx, 2-weeks, 4-				89.9		
5	weeks	••••			09.9		
	Pink, calyx and 7-day	88 8	92.3				
6	Pink, calyx, 10-day, 20-day					94.07	
7	All sprays						
	Pink, calyx, 7-day, 2-weeks	90.7	94.6	96.65	94.6	94.4	
8	No 7-day						
	Fish oil plus lead arsenate						
	at calyx; pink, calyx						
			93.30	95.3	92.9	94.19	
9	5-weeks spray						
	Pink, calyx, 2-weeks, 5-						
1			72.5				
10	4-weeks spray						
	Pink, calyx, 2-weeks, 4-						
11	weeks			96.78	95.02		
11	All sprays		<b>FO</b> 0	0= 00	0.4.0	01.1	
	Pink, calyx, 7-day, 2-week	S	79.2	97.26	94.9	94.4	

### MISCELLANEOUS RESULTS

During the experiments conducted at the Experiment Station Farm, it was noted that a gradual reduction in curculio injury took place over that five-year period when the tests were conducted. The following figures are the average of marked fruit obtained from the check trees:

TABLE 38

Date	Per Cent Marked by Curculio	Not	es
1924	64	Drops colle	cted
1925	61	-"	"
1926	45	"	"
1927	40	Wild apples	sprayed
1928	22	« 1 ·	"

It seems that continued sprays of the kind employed, together with supplementary controls such as removal of drop fruits (1924-6) and spraying of outlying trees (1927-8) has considerable effect in reducing the total infestation. Also that several years of intensive work results in a steady decline in the total number of curculios in the orchard. It seems reasonable that after a point is reached, similar to 1928, in our spray experiments, that some of the sprays could be safely omitted without seriously influencing the amount of unmarked fruit. The orchardist should then be concerned chiefly with the outside trees of his orchard or the rows nearest the protecting woods if such woods are present.

#### COST OF MATERIALS

The question will naturally arise as to whether the gain from one extra spray, in decreased amount of curculio damage will pay the cost of the treatment.

Figuring on a cost of materials plus labor employed, and on the basis of our 1928 results it would require at least a five per cent gain in unmarked fruit to pay the cost of the treatment where fruit sells for 12 dollars a barrel, the general market quotation for the highest grade in 1928 (Nov. 25 quotation \$2-\$12). Where the value of the crop drops to \$6 a barrel or \$2, a much higher per cent of gain from curculio must be obtained if the cost of the operation is successfully met from this gain. The greatest differences between a three and a four spray schedule, except on outside rows, has not been over 10 per cent in any year, so that the increased value of the crop does not meet the cost of the application in uninjured fruit except for apples bringing the highest price.

The gain in value of clean fruit may be further offset by the practise of including apples marked by curculios (one to three external marks per fruit) in the higher grades (Grade A)—so that

although marked by such punctures the fruit still has considerable value on the market. We have allowed for this condition and assumed for convenience that this fruit will bring 75 per cent of the value of the highest grades although it may and often does bring a much higher per cent.

PLUM CURCULIO ON APPLE

Regarding the schedule including fish oil sticker it may be stated that although more economical than the other schedules, because of (1) omission of lime sulphur at calyx period, (2) omission of the 7-day application altogether, it cannot be said to control scab except on varieties not especially susceptible. It has, however, been used successfully on Baldwins, Spys, Gravensteins, Wealthys, Suttons, Hurlburts, Greenings and Starks. It may also be added that the European red mite did not become abundant on any of the varieties mentioned during the years when this schedule was used.

#### SPRAY BURN

Spray burn being of considerable popular interest at this time, the basic form of lead arsenate was used as noted above in Shepard's orchard. Little or no difference could be seen in the foliage of Gravensteins, Astrachans, or McIntosh, although Gravensteins showed seven per cent less russeted fruit than was present in the plot where acid lead arsenate was used. It is not altogether certain, however, that this difference was due to the spray employed. In our Mount Carmel orchard, the plot where fish oil and lead arsenate were used at the calyx period showed much better foliage than either of the other two spray tests. There was considerable russeted fruit throughout the orchard, most of which was probably due to factors not associated with the spray program.

### ARSENICAL RESIDUE

Analyses of apples sprayed with various combinations were made in 1928, from plots receiving the four-spray schedule, and the three-spray schedule including fish oil sticker at calyx. The latter block contained one Astrachan tree which was picked about August 10. Analysis by Mr. Fisher of the Department of Chemistry, showed .004 grains As<sub>2</sub>O<sub>3</sub> per pound of fruit taken from the side of this tree and .006 grains per pound from the top. The tolerance for export fruit is .01 grain arsenic trioxide per pound which is much above that found by Mr. Fisher. Rainfall during July and August averages about four inches per year for each of these months in this locality, but in 1928 the precipitation in July was more than seven inches. Analyses of fruit picked later in the season and also from trees receiving an arsenical spray the tenth of July in addition to the usual schedule (pink, calyx, 7-day and 2-weeks) showed similarly small quantities of arsenical residue, all of them much below the export tolerance. From this it seems that

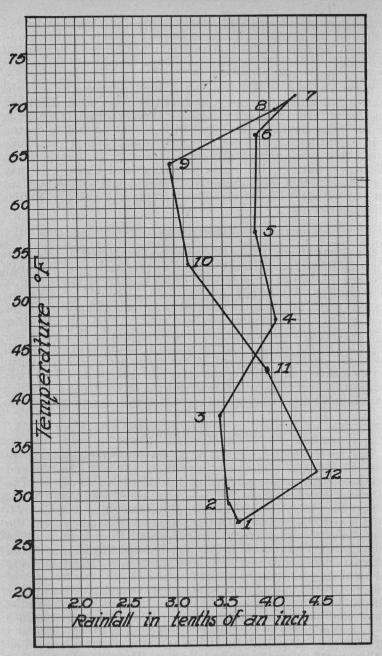


Fig. 32. Climatograph of New Haven district, compiled from U. S. weather bureau records covering the period between 1920 and 1927. The numbers 1, 2, 3, etc., represent the months January, February, March, etc.

there is little danger of harmful residues in Connecticut, especially on those apples picked after September 1, when normally eight or more inches of rain will have fallen. This is also illustrated in Fig. 32 which is a climatograph of the period between 1920 and 1927.

# GENERAL CONCLUSIONS

(1) The plum curculio is responsible for much damage to apples in Connecticut.

(2) It is single-brooded and all attempts to produce eggs from

beetles emerging during the summer, failed.

(3) The beetles emerge from hibernation near the first of May but do not appear in numbers on the trees till after the blossoms

(4) Beetles come to the trees from outside sources in greatest

numbers for a period of about 20 days in Connecticut.

(5) The peak of abundance and egglaving is reached near the middle of Tune, frequently the 15th, sometimes nearer the 20th.

(6) Beetles developing in dropped apples may offer a serious

menace to the succeeding crop.

(7) Sprays applied according to successful schedules outlined in this paper pile up the poison until the June peak of abundance is reached after which there is considerably less danger.

(8) Beetles continue on the trees (apples) in small numbers

until the last of Tulv.

(9) All attracting and repelling substances thus far have failed to exert any influence on the curculio in the field. Capryl alcohol is a powerful repellent when confined in cages.

(10) Laboratory tests indicate that the beetles are fond of sweets and may be poisoned in captivity by mixtures containing

(11) Effective poisons used in cage tests include, (1) acid lead arsenate, (2) basic lead arsenate, (3) sulphur arsenate dust 90-10, (4) calcium arsenate (5) ortho-zinc arsenite, and (6) sodium fluosilicate lime dust, 1-4.

(12) Parasites are not abundant in Connecticut, the most numerous being apparently Triaspis curculionis, which, however, did not kill over 33 per cent of the larvae obtained at any one time.

(13) Spray tests in three different orchards indicate:

(1) That a fair degree of control may be secured with three sprays in Connecticut, but better control with four using a pink,

calyx, 7-day and 2-weeks schedule as outlined.

(2) That occasionally a pink, calyx, 2-weeks, and 4-weeks schedule will give good control, while the omission of the 7-day and using fish oil and lead arsenate at the calvx period without lime sulphur closely approximates the results obtained with the four spray schedule but is on the whole less uniform. Such a schedule should be valuable on trees that burn easily and are not subject to scab infection, besides being more economical than a 4-spray schedule.

(3) That the most consistently good results have been obtained with the 4-spray schedule consisting of pink, 7-day and 2-weeks applications. This resulted in an average of 94.2 per cent unmarked fruit for all tests.

(4) There has been a gradual reduction in injured fruit since 1924, in the orchard where dropped fruits were collected and wild apples in the vicinity sprayed.

(5) Basic lead arsenate compared with acid lead arsenate and used at the same rate per gallon gave less control of curculio than the acid form. It also allowed too much damage from canker worms to be practical at the rate used.

(6) Sprays and dusts used in one field test in 1928 on a full schedule (pink, calyx, 7-day, and 2-weeks) showed the spray slightly superior in control but that dust has merit. This conforms in general with the Milford experiments of Stoddard and

(7) Spray residues remaining on the fruit at harvest have been small enough in all cases to conform with the export tolerance.

#### ACKNOWLEDGMENTS

The authors gratefully acknowledge all help received during the course of these investigations. The faithful help of those men, including Messrs. T. F. Cronin, A. E. Warren, J. L. Rogers, G. T. Thompson and J. F. Townsend, who did much of the laborious and tiresome portions of the work, is hereby acknowledged. We are especially indebted to Mr. E. M. Stoddard, and Mr. C. E. Shepard for use of orchards in which the work was done; to Mr. B. H. Walden who prepared the plates, and to Dr. W. E. Britton, who provided the necessary incentive and encouragement as well as important criticism of the manuscript as finally prepared.

#### REFERENCES TO LITERATURE.

- BLATCHLEY, W. S., and LENG, C. W. 1916. Rhyncophora or weevils of North America, pp. 469-470. Keys to species of Conotrachelus and description of *C. nenuphar* adults.
- Brooks, F. E., The cambium curculio Conotrachelus anaglypticus Say. 1924. In Journal of Agricultural Research, XXVIII; 377-386; 3 pls. Comparison of larvae of C. anaglypticus and C. nenuphar with figures of both.
- CHASE, W. W., Principal insects and diseases of the apple in Georgia 1913. Georgia State Board of Entomology. Bull. 38: pp. 15-20.
- CRANDALL, C. S. 1905. The curculio and the apple. Illinois Agr. Exp. Station, Bull. 98. 1906. Spraying apples. Illinois Agr. Exp. Sta., Bull. 106.
- 5. CULLINAN, F. P., and BAKER, C. E. 1924. Lime sulphur versus sulphur dust for apple spraying. Indiana Sta., Bull. 283. Dusting proves inferior to spray for curculio control in three out of five years.
- FERNALD, H. T. 1921. Applied Entomology, pp. 137-139. FORBES, S. A. 1906. Spraying apples for the plum curculio. Illinois Agr. Exp. Station, Bull. 108.
- GARMAN, PHILIP. 1927. In Proc. Conn. Pomological Society, pp.
- GARMAN, PHILIP. 1927. The problem of curculio control in Connecticut apple orchards. In Journ. Econ. Entomology, 20:
- 10. Gossard, H. A. 1920. Monthly bulletin Ohio Agr. Exp. Station, 5, No. 5, pp. 147-153. Comparison of dusts and sprays.
- 11. GOODWIN, W. H. 1917. The plum curculio. In Monthly Bulletin
- Ohio Agr. Exp. Station, No. 4, pp. 113-116, Fig. 4.

  12. Headlee, T. J. 1913. Report of the Department of Entomology of New Jersey Agr. Exp. Station, p. 654.

  1918. Ibid. Report for 1917, pp. 437-438.

  - 1919. Ibid. Report for 1918, pp. 212-213.
  - 1921. Ibid. Report for 1920, p. 449.
  - 1923. Ibid. Report for 1922, p. 373.
  - 1924. Ibid. Report for 1923, pp. 276-278.
  - The 7-day spray was developed by Dr. Headlee during the period covered by these reports.
- 13. HERRICK, GLENN W. 1925. Manual of Injurious Insects, pp. 156-157.
- HOOD, CLIFFORD E. 1926. Fish oil, an efficient adhesive in arsenate of lead sprays. U. S. Dept. Agr., Bull. 1439.
  HOPKINS, A. D. 1921. The bioclimatic law and its application
- to research and practise in entomology. In Journ. Washington Academy of Sciences, 11; p. 141.
- 16. Leiby, R. W., and Gill, J. B. 1923. The plum curculio on peaches in North Carolina, its history and control. North Carolina Department of Agriculture.
- 17. LYNCH, W. D., McDonnel, C. C., HAYWARD, J. K., QUAINTANCE, A. L., and WAITE, M. B. 1922. Poisonous metals sprayed on fruits and vegetables. U. S. Dept. Agriculture, Bull. 1027, p.
- 18. McIndoo, N. E. 1926. An insect Olfactometer. In Journ. Econ. Entomology, 19: pp. 545-571.
- 19. Morse, W. J., and Folsom, D. W. 1925. Maine Agr. Experiment Station, Bull. 325. Orchard experiments including curculio work and much other information.

437

20. O'KANE, W. C., HADLEY, C. H., and OSGOOD, W. A. 1917. Arsenical residues after spraying. N. H. Agr. Exp. Station, Bull. 183. Extensive poison experiments to determine effect of arsenical

CONNECTICUT EXPERIMENT STATION

21. PARROTT, P. J., STEWART, F. C., and GLASGOW, H. 1922. Spraying and dusting experiments with apples in 1922. N. Y. Agr.

Exp. Station, Circular 63.

22. PICKETT, B. S., WATKINS, O. S., RUTH, W. A., and GUNDERSON, A. J. 1918. Illinois Agr. Experiment Station, Bull. 206. Field experiments in spraying apple orchards in 1913 and 1914.

23. POWER, F. B., and CHESTNUT, V. K. 1920. Odorous constituents of apples. In Journ. American Chemical Society, 42; pp. 1509-

1921. Odorous constituents of peaches. In Journ. American Chemical Society, 43: pp. 1725-1739.

1922a. Odorous constituents of apples II, evidence of presence of geraniol. In Journ. American Chemical Society, 44: pp.

1922b. Confirmation of the occurrence of Linalyl esters in peaches. In Journ. American Chemical Society, 44; pp. 2966-2967.

1919. Public acts of State of Connecticut, 1919. An act concerning the grading of apples, Chapter 295.

1919. (Report of Horticultural Investigations.) Indiana Station Report, 1919: p. 47. "Dusting controlled codling moth and curculio as well as spraying".

1920. (Report on Horticultural Investigations.) Indiana Station

Report, 1920: p. 26.

26. QUAINTANCE, A. L., and JENNE, E. L. 1912. The plum curculio. U. S. Department of Agriculture, Bull. 103. The most extensive single publication on this insect, including life history, habits and control; complete bibliography to 1912.

27. Rumsey, W. E. 1909-1910. Spraying for the codling moth in 1910. West Virginia Agr. Exp. Sta., Report, pp. 29-37. Contains information on the curculio as well as the codling moth.

28. SANDERS, P. D. 1928. In Transactions of the Peninsular Horticultural Society, pp. 18-23.

29. SLINGERLAND and CROSBY. 1914. Manual of Fruit Insects. Pp. 243-251.

30. SMITH, R. H. 1928. An investigation of spray coverages and arsenical residue in relation to the control of the codling moth. In Journ. Econ. Entomology, 21: pp. 571-588.

31. SNAPP, O. I., et al. 1922. Controlling the curculio, brown rot and scab in the peach belt of Georgia. U. S. Dept. Agriculture,

32. SNAPP, O. I., and ALDEN, C. H. 1924. Dusting and spraying peach trees after harvest for control of the plum curculio. U.S. Dept. Agr., Bull. 1205; 1924.

33. SNAPP, O. I., et al. 1927. Experiments in the control of the plum curculio, brown rot and scab attacking the peach in Georgia. U. S. Dept. Agr., Bull. 1482.

34. SNAPP, O. I. 1923. Recent developments in plum curculio investigations in Georgia. In Journ. Econ. Entomology, 16: pp. 275-

34a. SNAPP, O. I., and SWINGLE, H. S. 1929. Preliminary report on attrahents for peach insects. In Journ. Econ. Entomology, 22: 98-101. Reports slight attractive powers for gallic acid, methyl formate and several other substances.

35. STODDARD, E. M., and ZAPPE, M. P. 1926. Further reports on spraying and dusting of apples. Conn. Agr. Exp. Station,

Bull. 285: pp. 228-234.

TALBERT, T. J. 1924. Spraying Missouri Fruits. Missouri Agr. Exp. Station, Bull 216: p. 6.

TURNEY, A. G. 1918. Orchard spraying experiments. Annual Report Dept. Agr. New Brunswick, pp. 98-101. "Thus far no satisfactory sprays have been developed for the apple maggot, green apple bug and plum curculio".

QUAINTANCE, A. L., and SCOTT, E. W. 1912. The one spray

method in control of the codling moth and plum curculio. U.S. Dept. Agriculture, Bureau of Entomology, Bull. 115, pt. 2:

pp. 87-112.

39. QUAINTANCE, A. L. 1921. Dusting v. spraying of apples. In Journ. Econ. Entomology, 14: pp. 220-225. Dusting compared with spraying for control of the curculio.

QUAINTANCE, A. L., and SIEGLER, E. H. 1922. The more important apple insects. U.S. Department of Agriculture, Farmers'

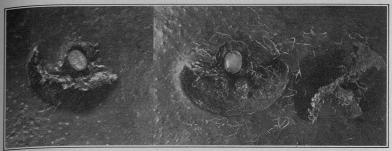
Bulletin 1270.

41. ZAPPE, M. P., and STODDARD, E. M. 1922. Results of dusting versus spraying in Connecticut apple and peach orchards. Conn. Agr. Exp. Station, Bull. 245, 1923.

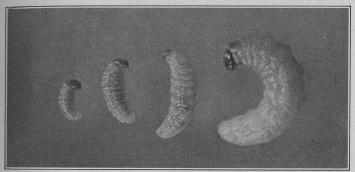
1923. Experiments in dusting versus spraying in Connecticut apple orchards. Conn. Agr. Exp. Station, Bull. 256: pp.

1924. Same, in Conn. Agr. Exp. Station, Bull. 265: pp. 286-294. Same, in Conn. Agr. Exp. Station, Bull. 275; pp. 272-278.

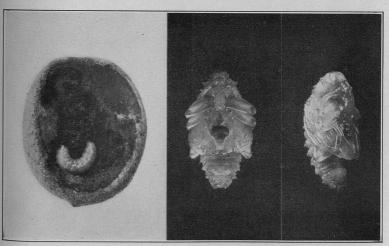
1927. Further reports on spraying and dusting of apples. Conn. Agr. Exp. Station, Bull. 294: pp. 256-261.



a. The plum curculio egg and crescent shaped egg scars, enlarged nine times.

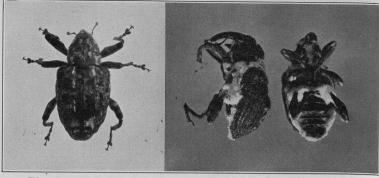


b. The four larval instars of the curculio, about three times natural size.

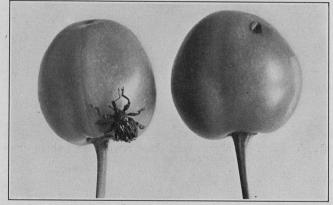


c. Full grown larva in peach, natural size.

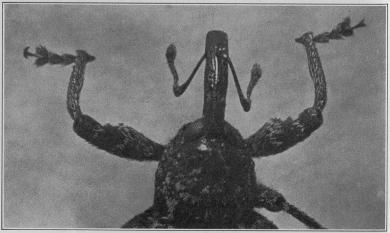
d. The pupa, enlarged about five times.



- a. The adult beetle curculio, enlarged six times.
- b. Beetles attacked by a fungus, *Isaria* sp.



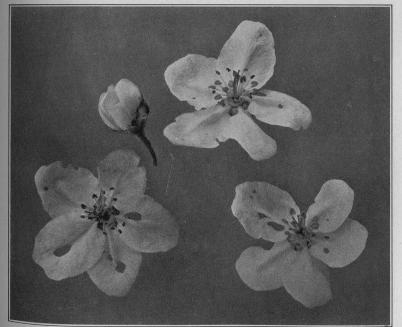
c. The adult beetle and its work on cherries, twice enlarged.



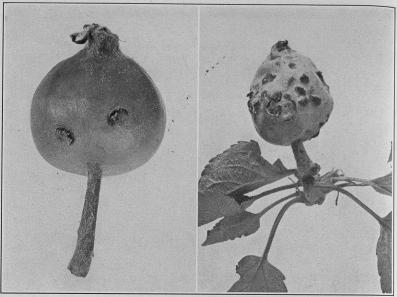
d. The head and beak of the adult curculio, enlarged about 20 times.



a. Jarring beetles from cherry trees in full bloom. Few or no beetles were found at this period.

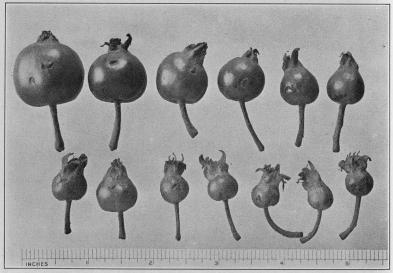


b. Feeding of adult beetles on apple blossoms.

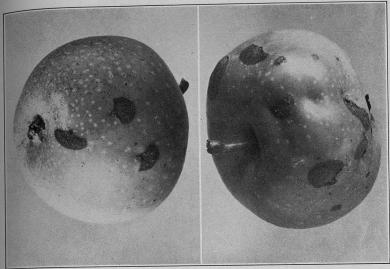


a. Egg scar on apple, enlarged twice.

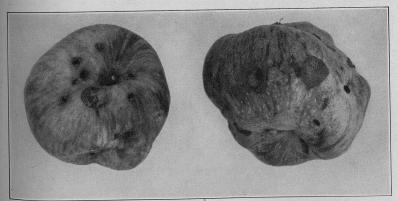
b. Result of numerous feeding punctures.



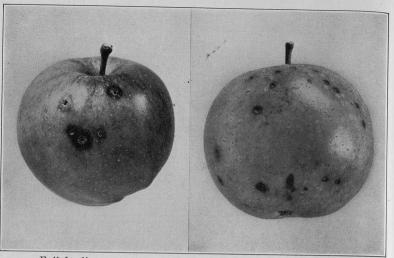
c. Size of dropped apples in which curculio larvae frequently develop. Slightly reduced.



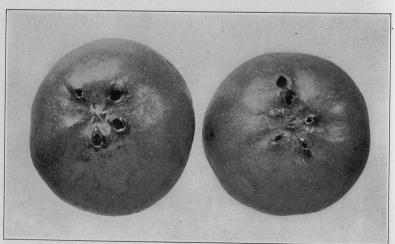
a. Expanded egg scars due to growth of the apple. The eggs or larvae were probably crushed by the growth of the fruit.



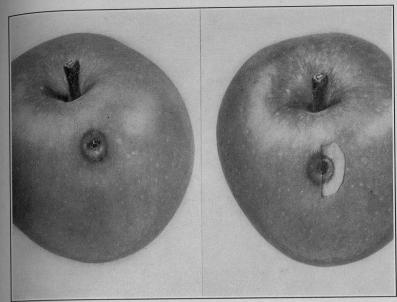
b. Deformed fruit resulting from numerous curculio egg scars and feeding punctures.



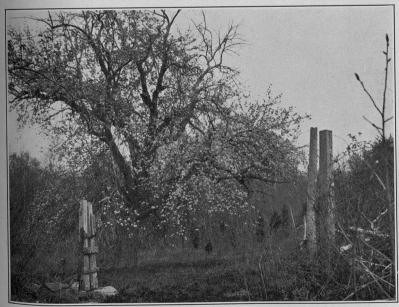
a. Fall feeding punctures of beetles emerging during the summer.



b. A common type of fall feeding puncture.

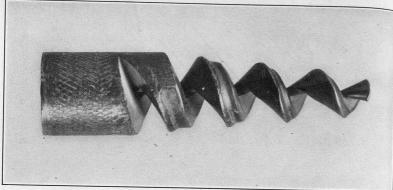


a. Fall feeding puncture cut away to show how puncture is excavated around the edges.



b. Wild apple tree growing under ideal conditions to promote development of curculios in large numbers. A decided menace if near a commercial orchard.





a. Type of spray nozzle used in some of our experiments. There is a graduated hole through the center of the nozzle which furnishes liquid to the spiral surface.



b. Delivery of spray from the nozzle shown in a. Note the solid coneshaped whirling spray which covers the tree with great rapidity.

Absence of projections on the end of the spray rod to catch on branches is a decided advantage. It can be thrust into the center of a tree covering parts not ordinarily reached with the usual spray outfit.

# Connecticut Agricultural Experiment Station

Nem Haven, Connecticut

# THE WILLOW SCAB FUNGUS

Fusicladium saliciperdum

G. P. CLINTON AND FLORENCE A. McCormick

The Bulletins of this Station are mailed free to citizens of Connecticut who apply for them, and to other applicants as far as the editions permit.

# CONNECTICUT AGRICULTURAL EXPERIMENT STATION

#### OFFICERS AND STAFF

#### BOARD OF CONTROL

[2] - [2] -
His Excellency, Governor John H. Trumbull, ex-officio President
George A Honson Secretary
TILLO LO DIAGO, D'HELLOT WHILL I TENSITY OF
Edward C. Schneider. Southington Francis F. Lincoln. Middletown Cheshire
STAFF.

Administration.	E. H. JENKINS, Ph.D., Director Emeritus.  WM. L. SLATE, B.SC., Director and Treasurer.  MISS L. M. BRAUTLECHT, Bookkeeper and Librarian. G. E. GRAHAM, In charge of Buildings and Grounds.
Chemistry: Analytical Laboratory.	E. M. BAILEY, PH.D., Chemist in Charge. C. E. SHEPARD OWEN L. NOLAN HARRY J. FISHER, A.B. W. T. MATHIS DAVID C. WALDEN, B.S. FRANK C. SHELDON, Laboratory Assistant. V. L. CHURCHILL, Sampling Agent. MFS. A. B. VOSBURGH, Secretary.

Biochemical Laboratory.	H. B. VICKERY, Ph.D., Biochemist in Charge. GEORGE W. PUCHER, Ph.D., Research Assistant. MISS HELEN C. CANNON, B.S., Dietitian.
	C. Chillon, D.D., Dielitan.

Botany.	G. P. CLINTON, Sc.D., Bolanist in Charge. E. M. STODDARD, B.S., Pomologist.
	MISS FLORENCE A. McCormick, Ph.D., Pathologist.
	HAROLD B. BENDER, B.S., Graduate Assistant
	A. D. MCDONNELL, General Assistant.
	MRS. W. W. KELSEY Secretary

Entomology.	W. E. BRITTON, Ph.D., Entomologist in Charge: State Entomologist
	D. II. WALDEN, B.AGR.
	PHILIP GARMAN, PH.D.
	ROGER B. FRIEND, PH.D.
	JOHN T. ASHWORTH. Deputy in Charge of Gipsy Moth Work
	R. C. BOTSFORD, Deputy in Charge of Mosquito Flimingtion
	J. P. Johnson, B.S., Deputy in Charge of Asiatic and Japanese Beetle Quarantines.

Mrs. Gladys Brooke, B.A., Secretary.	
WALTER O. FILLEY, Forester in Charge.	
H. W. HICOCK, M.F., Assistant Forester.	

J. E. RILEY, JR., MISS PAULINE A	M.F. In Char	ge of Blister	Rust Control.	

D1004 D 1!	
Plant Breeding.	Donald F. Jones, S.D., Geneticist in Charge.
	W. R. SINGLETON, S.M., Assistant Geneticist.
	W. R. BINGLETON, S.W., Assistant Geneticist.
	H. R. MURRAY, B.S., Graduate Assistant.
	MRS. R A HUNTER Secretary

M. F. Morgan, M.S., Agronomist. H. G. M. Jacobson, M.S., Assistant Agronomist.
HERBERT A. LUNT, M.S., Research Assistant in Forest Soils. DWIGHT B. DOWNS, General Assistant.

Tobacco Sub-station at Windsor.	PAUL J. ANDERSON, Ph.D., Pathologist in Charge. T. R. SWANBACK, M.S., Scientific Assistant.
	MISS DOROTHY LENARD, Secretary.

#### Press of The Wilson H. Lee Co., New Haven, Conn.

Forestry.

Soils.

# The Willow Scab Fungus

Fusicladium saliciperdum\*

#### G. P. CLINTON and FLORENCE A. McCormick

#### Introduction

The scab fungus of willows has been known in Europe for many years. In its parasitic stage it was found in North America for the first time in 1927 by the senior author (20). Years before, however, it had been reported, in what is claimed to be its saprophytic stage, from Greenland and Ellesmere Island by Rostrup (48). It is a relative of the common apple and pear scabs and has been called willow scab in Europe but its action on its most susceptible hosts has been so different from that of these scabs that the name willow leaf rot is more descriptive of its injury. It is the imperfect or conidial stage Fusicladium saliciperdum that, as a parasite, causes the injury to the willows while the asco or perfect stage occurs on the old dead leaves as a saprophyte only.

We have no fixed opinion whether this disease is native or has been introduced. The fact that it is very closely related to the poplar scab, which is native and has been known for years but which in 1928 assumed unusual vigor, might indicate it is also a native disease that, because of very favorable conditions, has suddenly sprung into unusual prominence from which it may subside, as it does in Europe.

On the other hand, while found here on wild willows, it seems to be confined, so far as seen by us, to the general vicinity where serious outbreaks have occurred on willows planted as shade trees. This might indicate its introduction on foreign stock planted in some of these infected regions, as has been claimed by some.

The willow disease must not be confused with the injury of willow-leaf beetles whose larvæ cause serious damage to the leaves by eating out holes, often skeletonizing them completely.

<sup>\*</sup>We are indebted for aid in obtaining data of various kinds for this paper to Messrs. E. M. Stoddard, A. D. McDonnell, A. A. Dunlap and Mrs. Lillian D. Kelsey, of the Botanical Department. Specimens of diseased willows, collected both within and outside the state, have been sent by a number of interested people. Permission was granted by Luther M. Keith of the State Highway Department to spray street trees and by Mr. and Mrs. Henry F. Parmelee, to spray trees on their estate; both at Norfolk, Conn. Alfred Rehder, of the Arnold Arboretum, has verified many of our willow determinations. Dr. C. W. Dodge, of the Farlow Cryptogamic Herbarium of Harvard, has helped us with the literature and exsiccati references. C. O. Erlanson, of the University of Michigan, sent us specimens of willow leaves recently collected in Greenland, upon which we found Venturia chlorospora in its asco stage. Written December, 1928.

Injury of this nature has also been very conspicuous in certain sections of New England in the past season. Seen from the distance it is difficult to tell which one is causing the trouble.

Cesati (17) discoverer of the saprophytic stage, Allescher and Tubeuf (7) first describers of the parasitic stage, Karsten (30, 31). Rostrup (47-52) and Aderhold (2-3) were some of the earlier European botanists who made notes on this fungus. Aderhold of Germany, was the first to definitely associate the two stages as now understood. In an article (2, pp. 80-3) published in 1897 on the scabs of birch, pear, poplar, apple, willow and ash he described both stages of the scabs on these hosts, placing the imperfect stage of each under the form genus Fusicladium and the perfect stage under the genus Venturia. The names applied to the two stages of the willow scab were designated as Fusicladium ramulosum, now known as F. saliciperdum, and Venturia chlorospora.

Recent investigations have been carried on by two European botanists concerning the parasitic nature of this fungus and its action on the willow. Dr. Marie Schwarz (60), of Holland in 1922, published a paper dealing with this and other parasitic fungi found on a species of weeping willow, Salix alba var. vitellina pendula, in the parks at Utrecht in 1920. Mrs. N. L. Alcock (4.5). of Edinburgh, Scotland, has more recently published two short papers dealing with its action on willow rods, Salix alba var. vitellina. Like Dr. Schwarz, she found a variety of other fungi associated with the death of the leaves and twigs some of which. besides the scab, were apparently parasitic. However, there has been some doubt even after these investigations as to how important the scab was in such outbreaks. We have attempted in our investigations to clear up some of the doubtful points in its life history and to determine methods of control by spraying.

# PART I. AMERICAN INVESTIGATIONS

#### DISTRIBUTION

The willow scab in its conidial stage was first definitely recognized in Connecticut, at Norfolk, in late June, 1927. Mrs. Parmelee who wrote to the Experiment Station of injury to her willows had, however, noticed the trouble the year before, but the cause had not been determined. Our visit to Norfolk was made some time after the leaves had been killed (Plate X, a) and many had fallen, so that at first we were not sure of the cause. thinking possibly a late frost, which was said to have occurred, might have been responsible. Identification was made more difficult by the fact that many of the leaves showed no fruiting stage of this fungus while other fungi had become more or less prominent on the dead tissues. The responsible fungus was soon located,

compared and found identical with European specimens of the willow scab. Now that its characteristic appearances are known it is easily recognized by us if present on the dead leaves or twigs.

Soon after its location at Norfolk, information was received from various sources of its presence in restricted localities in eastern New York, western Massachusetts and in near by locations in Connecticut, on both willow trees and shrubs. On a visit during August of 1927, the senior author also found the disease causing great damage to large willow trees at Weymouth and Digby in Nova Scotia (Plate XI, a, c,) and to a less extent at Greenwich in New Brunswick, Canada, and although the auto trip also included all the New England states except Vermont, it was not observed further in any of these.

To date, in Connecticut, the disease has been surely located in sixteen towns in the northwestern part of the state. These towns are Bridgewater, Canaan, Cornwall, Goshen, Kent, Litchfield, Norfolk, North Canaan, Salisbury, Sharon, Southbury, Torrington, Warren, Waterbury, Winchester, Woodbury. However, in three or four other towns removed from that region, occasional dead twigs of certain species of willows have been found that superficially resembled those seen on the willows in the infected district but, as the fruiting stage of the fungus was not found on any of them, we have left their identification as extremely doubtful. It is quite possible that in these cases the injury, resembling fireblight, was caused by bacteria, since they were sometimes found in the dead tissues. Bacterial diseases of this nature have been described elsewhere but as yet we have made no special study of such a trouble in Connecticut.

A camping trip, in late June, 1928, made by the senior author with students at Keene, New Hampshire, disclosed the fungus abundant on some highway trees in the town of Dublin. It has also been reported to us as occurring prominently in the Bridgewater valley in Vermont, although we have received no specimens from that state. In August it was found by the junior author in New York state on the Hudson-Hillsdale highway. Specimens have also been sent to us from the Lebanon mountain region in New York, found on the road from Pittsfield to Albany.

In July of this year a short auto trip from Canaan, Conn., north into the Berkshires through Sheffield, Great Barrington, Southbridge, Lenox and Pittsfield, Massachusetts, revealed the disease as common along the highway on the trees and shrub willows at these places. This is a region where large willows are very common and form a conspicuous part of the landscape. Specimens from this part of Massachusetts have also been sent to us for determination. The disease was seen by Mr. McDonnell of the botanical department in the vicinity of Easthampton and Northampton, but we have no record of its presence further eastward in Massachusetts, though it has been looked for as far east as Boston and from there north to the Maine line.

Early in the summer of 1928, Maine specimens were sent us from Salisbury Cove, on Mt. Desert Island, and from Brooklin, both in Hancock county. Later, in August, the senior author made an auto trip along Route 1 of the National Highway and saw diseased trees common from the state line at Calais all the way south to Bath, but few below the latter city. These infected trees were at or near the following cities: Calais, Pembroke, Dennysville, East Machias, Machias, Whitneyville, Harrington (Plate XII, c), Cherryville, Ellsworth, Salisbury Cove (Plate XII, d), Bar Harbor, Blue Hill, Brooklin, Bucksport, Lincolnville, Camden, Wiscasset, Wells, Kittery Point. The disease was reported by other observers from Belfast, Waldoboro and Gouldsboro.

Previous to the Maine trip the senior author also visited Nova. Scotia (including Cape Breton Island), Prince Edward Island and New Brunswick and saw the disease on the susceptible species Salix alba var. vitellina practically everywhere it occurred over the 1200 miles traveled by auto. This species is a common street and shade tree in this region. The towns, or vicinities near them. where the disease was seen were as follows: In Nova Scotia proper at Yarmouth, Argyle, Shelburne, Liverpool, Bridgewater, Lunenburg, Mahone Bay, Hubbards, Black Point, St. Margarets Bay. Halifax, Waverly, Oakfield, Enfield, Stubenacadia, Stewiacke (Plate XI, b), Brookfield, Truro, New Glasgow, James River. Antigonish, Heatherton, Afton, Monastery; in Cape Breton Island at Margaree Forks, Margaree Harbor (Plate XI. d). Cheticamp. Badeek, Sydney, Big Pond, Big Pond Center, Cleveland: in Prince Edward Island at Charlottetown, North River, New Haven (Plate XII, a); in New Brunswick, at Cape Tormentine, Sackville, Moncton, (Plate XII, b), Salisbury, Sussex, Hampton, St. George, St. Andrews and at various places between those last mentioned. We also understand that the disease has been found in Quebec though we have seen no specimens from there.

#### Hosts

## Fusicladium Stage:

More attention has been paid by us to determine the species of willows attacked in Connecticut than elsewhere, so most of the species reported here are from this state. Those on which we have found the fungus are as follows:

- 1. Salix alba: Conn., Me.
- 2. S. alba var vitellina: Conn., Mass., Me., N. H. (this may be S. alba x fragilis=S. rubens, according to Rehder), N. Y., Vt.; N. Scotia proper, C. Breton Isl., P. Edw. Isl., N. Brunsw.
- 3. S. cordata: Conn., Mass.; P. Edw. Isl.
- 4. S. discolor: Conn.

- 5. S. lucida: Conn.
- 6. S. nigra: Conn., Mass., N. Y.
- 7. S. pentandra: Conn., Me.; N. Scotia proper.
- 8. S. sericea: Conn.
- 9. Salix sps. undet.: Conn., Me.; N. Scotia.

# Venturia Stage:

From literature we have learned of the North American distribution of the asco stage as follows:

- 1. Salix arctica: Ellesmere Isl. (Rostrup), Greenland (Lind).
- 2. S. arctica var. Brownei: Greenland (Lind).
- 3. S. glauca: Greenland (Rostrup).
- 4. S. grænlandica: Greenland (Rostrup).
- 5. S. herbacea: Greenland (Rostrup).
- 6. S. reticulata: Greenland (Lind).

Besides the above we have recently received specimens of dead willow leaves collected June 1, 1928, at Englishman's Harbor, Disko Island, Greenland, by C. O. Erlanson of the University of Michigan. The leaves were collected for *Rhytisma salicinum* (Pers.) Fr., but also had other minute ascomycetes on them. On one of the leaves of these we found the asco stage of *Venturia chlorospora*, (spores more nearly var. *canescens* as they varied from 5.5-8u x 16-21u, chiefly 16-18u, and the asci from 14.5-16.5u x 48-54u, bristles on perithecia infrequent, perithecia commonly 90-100u in diameter), on this additional following host:

## 7. S. chloroclados x glauca: Greenland (Erlanson).

# Susceptibility and Resistance

There seems to be at least some difference in the susceptibility of different species of willows to this fungus (Plate IX, a-b, free). There has also been quite a difference in its attack on susceptible individual trees and shrubs of the same kind even when growing close together. This latter difference we explain on the supposition that the disease had previously become well established on the badly infected plants, through overwintering on the branches, but not as yet on the less infected individuals and that the spores are washed or carried over the former much more readily than to adjacent plants. In time, however, if conditions prove favorable, the latter should become badly infected. For example, along the highway from Norfolk to Canaan there are numerous large willows, some apparently of the same variety as those in the village vet the disease gradually decreases outside of Norfolk to become inconspicuous at Canaan several miles distant. Some of these trees, however, may be crosses with Salix alba (Plate IX, c). At Calais, Maine, there is a row of Salix alba var.

449

vitellina of about a dozen trees reaching from the water to the highway. Those next the road were badly diseased in August, 1928, while those near the water (Plate IX, d) showed very little injury.

On the other hand Salix alba (Plate IX, a, shows catalpa and willow), although reported infected in Europe, seems to be resistant here, since we have rarely found it on that species even where the disease exists nearby on its variety vitellina which is the most susceptible of all the willows yet observed.

On two of the estates at Norfolk are several trees determined by us as the Bay-leaf willow, Salix pentandra, but although these are very near badly diseased trees no disease was found on them in 1927 and only occasional infected leaves in 1928. This species, however, seems to be one of those infected in Maine, although not as seriously as var. vitellina. Salix nigra is another large shade tree that has been rather badly injured though the fungus does not seem to fruit so abundantly on it. Salix cordata (Plate X, c) is apparently the most susceptible of the native willow shrubs. So far we have not seen the disease on the weeping willow, Salix babylonica, although it occurs occasionally in the infected regions and is reported from Europe as a host.

#### INJURY TO TREES

The worst injury to large trees, mostly Salix alba var. vitellina, was seen at Norfolk, Conn. (Plate X, b), Hancock county, Maine, and general in Nova Scotia. In the last region, according to a farmer interviewed, the disease, as in Connecticut, appeared conspicuously in 1926 but much more so in 1927. When first seen by the senior author in August of the latter year (Plate XI, a, c), some of the large trees had no, or very few, leaves on them and could easily be mistaken for dead trees. In 1928 in different parts of Nova Scotia hundreds of dead trees were seen and many more so badly injured that another season's attack was likely to finish them (Plate XI, b, d). The situation in Maine, in Hancock county along the coast (Plate XII, c, d), was not much better.

At Norfolk, by the end of June, 1927, the very large trees of Salix alba var. vitellina, which are conspicuous shade trees in the village, were partly to largely defoliated although later some new leaves were put out. In 1928, with an unusually wet spring and summer, by the first of August there was not a single untreated tree of this variety (Plate X, b) that was not completely or very nearly completely defoliated and very few if any new leaves were put out. It looks as if many of these trees were doomed, as they have gone through at least three seasons of more or less complete defoliation, and early this season most of the smaller twigs and many of the large branches were dead. Already several fairly young trees have been killed or so severely injured that only the

main trunks are alive. These trees were so weakened by the general attack on the leaves and young twigs very early in the several seasons that starvation and winter injury killed the large branches which were not directly attacked.

Bad as healthy willows are in littering the ground with dead twigs, after passing through last winter, the litter of twigs on the ground under the infected trees was unusually conspicuous, although only a comparatively few had yet broken off. We have seen no fungous disease of trees where the injury has been so sudden and severe as from this fungus, though the chestnut blight and the white pine blister rust in the long run cause more serious financial loss and eventually just as serious injury to the trees.

The fungus carries over the winter on the young twigs infected the previous year. In the spring the Fusicladium stage appears on these and the spores are washed down on the very young leaves in the opening buds, so that their death may occur before they have reached any size, much as occurs with the leaves of the sycamore from the anthracnose fungus. Some young leaves, however, escape infection only to succumb later. If the moist favorable weather continues, nearly full grown, or even full grown, leaves may suddenly rot on the trees and adhere there for some time, presenting a very mournful sight. They then dry up and gradually fall off leaving the trees more or less completely defoliated. Bad defoliation two or three years in succession seems to be fatal since after the first year, little adventitious foliage is put out and gradual starvation results.

If the tissues of the leaves are fairly mature when first infected, the infection may stop after killing spots of varying size in the otherwise healthy green tissues (Plate XIII, b). Often where the large, but still young, leaves are attacked the rot spreads down the midrib to the base killing the tissues as it advances. Quite frequently in these cases it reaches through the petiole into the tissues of the young twigs and causes a more or less conspicuous canker there (Plate XIII, a). If girdling occurs the twig, with its attached leaves above, soon dies. These dead twigs and leaves assume a reddish-brown or blackish color according to the species of willow. The fruiting stage may or may not be found on the dead twigs and leaves, apparently developing much more on some willow species than on others and, of course, not developing on tissues that have been indirectly killed by the action of the fungus.

#### THE FUNGUS

The spring and summer of 1928 were unusually favorable not only for this scab but for all related scabs. Apple scab obtained an early start and its injury was much greater than in ordinary years; Pear scab, too, was more prominent than usual. Although known in New England for some time, we had not previously

listed the Poplar scab, Fusicladium radiosum (Lib.) Lind (F. ramulosum, F. Tremulae) from Connecticut. This year it has been found common not only in this state but in New Hampshire, Vermont and Massachusetts.

The chief aid in identifying the presence of the willow scab is the characteristic appearance of its fruiting stage. This develops usually on the under side of the leaves but is rather infrequent on the cankered areas of the young twigs. It may appear on these latter next year in early spring and produce the first infection of the leaves. It shows on the leaves as small dense olive-brown pustules which more or less cover the surface, but which particularly follow the downward course of the larger veins and especially the midrib (Plate XV, a).

The spores are olive to reddish-brown in color. They are truncate at the base, generally rounded at the apex, and have one, rarely two, or very rarely three septa. They vary from 12u to 25u long and from 6u to 10u wide. When two-celled, the basal cell is usually considerably longer and is somewhat broader (Plate XV, d). They are borne singly at the tips of the conidiophores. Apparently only one spore is formed on each conidiophore.

In cultures there is formed a dense velvety olive-brown growth on which the spores are usually abundantly developed in two to three weeks (Plate XV, b). The spores are somewhat larger than in nature, varying from  $9-12u \times 18-36u$ , and largely disappear through germination in old cultures. With age the mycelium is of similar color to the spores, moderately septate and branched and varies from 2.5u to 7.5u, chiefly 4-6u, in diameter. Not infrequently rounded, larger cells (9-15u) occur in the mycelium but are not thickened like chlamydospores. In old cultures imperfect perithecia appear somewhat sparingly. We have not yet been able by crossing with other species of Fusicladium or by stimulation to bring them to maturity. They appear similar to the immature perithecia we have found on sections of leaves, mentioned later.

The conidiophores on the leaves are closly compacted together to form the fruiting pustule. They arise from a more or less extended sclerotial mass of rounded cells (Plate XV, c) from which they are not always very clearly marked off. In general their erect habit, their brownish color, about the same as that of the spores, and their separation from each other above serve to distinguish them. They have one to several septa and at their base they gradually merge into the cells of the sclerotial mass. As a rule they are as long or longer than the spores and slightly narrower. Occasionally sections through the old leaves show more elongated, fewer and laxly clustered conidiophores that have borne spores both at and near their tips. We believe that these are always the conidiophores of the saprophytic Cladosporium.

Although the writers have made attempts to locate the Venturia stage of this fungus as described by Aderhold on the old leaves, have searched the twigs and leaves during the growing season and examined them especially at the time in the spring when infections from this stage would naturally take place, we have as yet failed to find it. Occasionally, however, in the sections made of infected leaves taken from the trees, we have found immature perithecia closely associated with the Fusicladium that may be the asco stage. Apparently, no one has yet made cultures from the Venturia stage and by this means, or by inoculations. proved its relationship to the Fusicladium. We have little doubt that the Venturia fungus eventually will be found here. We do doubt, however, if it is the asco stage, that it is as important in primary infection (unless occurring on the twigs on which it has not vet been reported) as the Fusicladium stage produced in early spring on the twigs from the overwintered cankers. This condition has also been found to be true with certain hosts of the pear and apple scabs in Connecticut and in England.

#### ASSOCIATED FUNGI

As stated earlier in the paper, certain of the workers with the Fusicladium stage of this fungus have found other fungi associated with it, some of which they have considered parasites and so apparently responsible for part of the trouble in the outbreak. Dr. Schwarz (60), for example, mentions three parasitic fungi, one of which, *Phoma intricans*, is described as new. The second is an ascomycete belonging to the genus Physalospora, determined by her as *P. Salicis* (Fckl.) Rab. She also notes that two other species of Physalospora, *P. gregaria* Sacc. and *P. Miyabeana* Fuk., have been mentioned as parasites of the willow. Lastly she gives *Discella carbonacea* (Fr.) Berk. & Br. as the third parasite.

Mrs. Alcock (5) in her article mentions the following fungi occurring on willow rods as parasites of greater or less intensity: *Physalospora gregaria* Sacc., *Cryptomyces maximus* (Fr.) Rhem, *Scleroderris fuliginosa* (Pers.) Karst. and *Myxosporium scutella* 

tum (Otth) Petrak.

Very recently Nattrass (42, 43), of England, has published articles on the parasitism of *Physalospora Miyabeana* Fuk., in relation to *Fusicladium saliciperdum*, as a cause of the disease of willows. Contrary to other investigators he is inclined to believe that the willow scab, at least in England, plays a secondary part to the Physalospora. He makes the following statement at the conclusion of his second article (43): "From these preliminary observations and experiments it appears that on the basket willow in Somerset *F. saliciperdum* cannot be regarded other than as a follower of *P. Miyabeana* and that it is in no way responsible for the symptoms of the disease on the leaves."